

2M4  
NOVEMBER 1973

MDC G4901

(NASA-CR-120251) G-189A ANALYTICAL  
SIMULATION OF THE INTEGRATED WASTE  
MANAGEMENT-WATER SYSTEM USING  
(McDonnell-Douglas Astronautics Co.)  
256 p HC \$16.00

N74-28629

CSSL 06I

G3/05

Unclas  
15905

267  
**G-189A ANALYTICAL SIMULATION OF THE  
INTEGRATED WASTE MANAGEMENT-WATER SYSTEM  
USING RADIOISOTOPES FOR THERMAL ENERGY**



MCDONNELL DOUGLAS ASTRONAUTICS COMPANY

MCDONNELL DOUGLAS

CORPORATION



**G-189A ANALYTICAL SIMULATION OF THE  
INTEGRATED WASTE MANAGEMENT-WATER SYSTEM  
USING RADIOISOTOPES FOR THERMAL ENERGY**

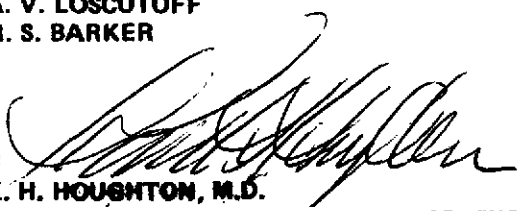
**NOVEMBER 1973**

**MDC G4901**

**PREPARED BY:**

**J. V. COGGI  
A. V. LOSCUTOFF  
R. S. BARKER**

**APPROVED BY:**

  
**K. H. HOUGHTON, M.D.**  
**CHIEF, BIOTECHNOLOGY & SPACE SCIENCES ENGINEER  
BIOTECHNOLOGY AND SPACE SCIENCES DEPARTMENT  
ENGINEERING DIVISION**

**PREPARED FOR THE NATIONAL AERONAUTICS AND SPACE ADMINISTRATION,  
MARSHALL SPACE FLIGHT CENTER, UNDER CONTRACT NAS 8-28982**

**MCDONNELL DOUGLAS ASTRONAUTICS COMPANY-WEST**

**5301 Bolsa Avenue, Huntington Beach, CA 92647**

## FOREWORD

The G-189A analytical simulation of the Integrated Waste Management-Water System Using Radioisotopes for Thermal Energy (RITE) was developed by the Biotechnology and Space Sciences Department, Engineering Division, of McDonnell Douglas Astronautics Company (MDAC), Huntington Beach, California, under Contract NAS8-28982. The simulation was prepared for the NASA Marshall Space Flight Center (MSFC) under the technical direction of Mr. S. E. Clonts of the Propulsion and Thermodynamics Division, Science and Engineering Directorate.

The study was supervised by K. H. Houghton, M.D., Chief Biotechnology and Space Sciences, and R. S. Barker, Chief, Environmental Engineering ECLS Section, MDAC. J. V. Coggi was the Study Manager for McDonnell Douglas. The study development and report preparation were done under the direction of J. V. Coggi with major assistance from: A. V. Loscutoff and R. S. Barker.

The authors wish to acknowledge the cooperation and support given to this program by R. W. Murray and J. D. Schelkopf of the General Electric Valley Forge Spacecraft Center.

## TABLE OF CONTENTS

<u>Section</u>	<u>Page</u>
1	INTRODUCTION AND SUMMARY - - - - - 1.1
2	TECHNICAL DISCUSSION - - - - - 2.1
3	COMPUTER MODEL DESCRIPTION - - - - - 3.1
3.1	Computer Model Organization - - - - - 3.1
3.2	Computer Model Documentation - - - - - 3.3
3.2.1	Low Temperature Heating Loop - - - - - 3.6
3.2.2	Flush Water Loop - - - - - 3.10
3.2.3	Air Loop - - - - - 3.15
3.2.4	Solid Waste Loop - - - - - 3.15
3.2.5	Vapor Loop - - - - - 3.21
3.2.6	Potable Water Storage Loop - - - - - 3.35
3.2.7	Cooling Loop - - - - - 3.40
3.2.8	High Temperature Heat Source - - - - - 3.40
4	CORRELATION OF EXPERIMENTAL AND ANALYTICAL DATA - - - - - 4.1
4.1	Pyrolysis Units - - - - - 4.2
4.2	Air Sterilizer - - - - - 4.2
4.3	Heat Pipe - - - - - 4.8
4.4	Heat Block - - - - - 4.10
4.5	Low Temperature Isotope Heater, Evaporator, and Condenser - - - - - 4.10
5	CHARACTERISTICS OF THE RITE SYSTEM - - - - - 5.1
5.1	RITE System Performance Characteristics - - - - - 5.1
5.1.1	Low Temperature Heating Loop - - - - - 5.1
5.1.2	Flush Water Loop - - - - - 5.8
5.1.3	Air Loop - - - - - 5.8
5.1.4	Vapor Loop - - - - - 5.11
5.1.5	Cooling Loop - - - - - 5.21
5.1.6	Potable Water Loop - - - - - 5.23

TABLE OF CONTENTS  
(Continued)

<u>Section</u>	<u>Page</u>
5.2 Transient Behavior - - - - -	5.23
5.2.1 Micturition - - - - -	5.23
5.2.2 Defecation - - - - -	5.28
5.2.3 Incineration - - - - -	5.28
5.3 RITE System Operational Envelope - - - - -	5.31
6 RITE SIMULATION DESIGN APPLICATIONS - - - - -	6.1
6.1 Condenser Coolant Inlet Temperature - - - - -	6.2
6.2 Component Design - - - - -	6.6
6.2.1 Evaporator - - - - -	6.6
6.2.2 Condenser - - - - -	6.10
6.3 Design Study Results - - - - -	6.10
7. FAILURE MODE ANALYSIS - - - - -	7.1
7.1 Cooling Loop Failure - - - - -	7.2
7.2 Incinerator Vent Valve Failure - - - - -	7.2
7.3 Low Temperature Heating Loop Failure Analysis - - -	7.5
8 REFERENCES - - - - -	8.1
Appendix A SUBROUTINE DOCUMENTATION - - - - -	A.1
A.1 EVAP	
A.2 CNDNSR	
A.3 INCIN	
A.4 HTPIP	
A.5 TANKG	
Appendix B INPUT DATA CARDS DESCRIPTION - - - - -	B.1
Appendix C OUTPUT DATA - - - - -	C.1
Appendix D GPOLY SUBROUTINES - - - - -	D.1

## LIST OF FIGURES

<u>FIGURE NO.</u>		<u>PAGE</u>
<u>Section 2</u>		
2.1	Integrated Waste Management-Water System Using Radioisotopes for Thermal Energy	2.2
2.2	RITE System Operation	2.4
2.3	G-189A Program Logic	2.9
<u>Section 3</u>		
3.1	G189 Schematic Representation	3.4
3.2	G189A Schematic of the RITE System	3.5
3.3	Low Temperature Heating Loop G189A Schematic	3.7
3.4	Isotope Heater Nodal Network	3.11
3.5	Flush Water Loop (Flow Code 4)	3.12
3.6	TANKG Mass and Energy G-189 Output Representation	3.14
3.7	G189 Component Definition Diagram for the Air Loop (Flow Code 3)	3.16
3.8	Air Sterilizer Schematic Defining Thermal Nodes	3.19
3.9	G189A Schematic of the Solid Waste Loop	3.20
3.10	G189 Component and Flow Schematic for the Vapor Loop	3.24
3.11	EVAP Subroutine Flow Chart	3.28
3.12	Mass Balance in the Evaporator	3.29
3.13	CNDNSR Subroutine Flow Chart	3.33
3.14	Condenser Subroutine G-189 Output Representation	3.34
3.15	G189A Schematic of the Condensate Collection Loop	3.36
3.16	G189A Schematic of the Potable Water Loop	3.37
3.17	G189A Schematic of the Cooling Loop	3.41
3.18	G189A Schematic of the High Temperature Heat Source Loop	3.43
3.19	Heat Block Model	3.45
3.20	Heat Block Nodal Network	3.46

## LIST OF FIGURES (continued)

<u>FIGURE NO.</u>		<u>PAGE</u>
<u>Section 4</u>		
4.1	Pyrolysis Unit Temperature Profile	4.3
4.2	Time-Temperature Data G-189A Model/Experimental for the Air Sterilizer	4.4
4.3	Time-Temperature Curve G-189A Model/Experimental	4.5
4.4	Air Sterilizer Thermal Model	4.6
4.5	Heat Rejection Rate vs. Source Temperature	4.9
4.6	Heat Block Heating Profile Correlation	4.11
4.7	Heat Block Nodal Network	4.12
<u>Section 5</u>		
5.1	Mass and Heat Balance for Low Temperature Heating Loop	5.2
5.2	Mass and Heat Balance for the Flush Water Loop	5.9
5.3	Air Loop Mass and Heat Balance During Micturition	5.10
5.4	Mass and Heat Balance for the Vapor Loop	5.12
5.5	RITE System Chemical Reactions Mass Balance	5.14
5.6	Urinal Input Flow Cycle	5.16
5.7	Blender Input Cycle (one/6 hr)	5.17
5.8	Trash Input Cycle (one/6 hrs)	5.18
5.9	Mass and Heat Balance for the Cooling Loop	5.22
5.10	Typical Potable Water Storage Tank	5.24
5.11	Evaporator Evaporation Rate During Micturition	5.25
5.12	Evaporator Temperature Response to Micturition	5.26
5.13	Heat Block Temperature Response to Micturition	5.27
5.14	Evaporator Temperature Response to Defecation	5.29
5.15	Flow Rates in Evaporator During Defecation	5.29
5.16	Temperature Response of the Air Sterilizer Due to Defecation Air	5.30
5.17	Incinerator Temperature Cycle	5.32
5.18	RITE System Operational Envelope (Condenser Coolant Temperature = 35°F)	5.33
5.19	Evaporator and Condenser Performance Evaluation (At GE Operating Point of 181°F LTHL Temperature)	5.35

# LIST OF FIGURES (continued)

<u>FIGURE NO.</u>		<u>PAGE</u>
<u>Section 5</u>		
5.20	RITE System Water Purification Efficiency	5.36
5.21	Heat Input to Evaporator	5.36
5.22	Evaporator Outlet Temperature as a Function of Wall Temperature	5.37
5.23	Evaporator Pressure as a Function of Wall Temperature	5.37
<u>Section 6</u>		
6.1	RITE System Operational Envelope Condenser Coolant Temperature 35°F	6.3
6.2	RITE System Operational Envelope With Condenser Coolant Temperature Equal to 40°F	6.3
6.3	RITE Operational Envelope with Radiator Coolant Temperature Equal to 50°F	6.4
6.4	Water Recovery Rate as a Function of Condenser Coolant Temperature	6.4
6.5	RITE System Operational Envelope With Evaporator UA = 92.6 Btu/hr-°F	6.7
6.6	RITE System Operational Envelope With Evaporator UA = 75 Btu/hr-°F	6.7
6.7	RITE System Operational Envelope With Evaporator UA = 150 Btu/hr-°F	6.8
6.8	RITE System Operational Envelope With Condenser UA = 75 Btu/hr-°F	6.8
6.9	RITE System Operational Envelope With Condenser UA = 117 Btu/hr-°F	6.11
6.10	RITE System Operational Envelope With Condenser UA = 150 Btu/hr-°F	6.11
<u>Section 7</u>		
7.1	Temperature Response to Loss of Cooling Accident	7.3
7.2	Condenser and Evaporator Flow as a Result of Loss of Cooling Accident	7.4
7.3	Temperature Response to the Incinerator Oxygen Cycle Failure	7.6



LIST OF FIGURES (continued)

<u>FIGURE NO.</u>		<u>PAGE</u>
<u>Section 7</u>		
7.4	Temperature Response to Loss of Flow and Loss of Coolant Failures	7.8
7.5	Evaporator Bypass Failure - Isotope Coolant Outlet Temperatures with Various Emergency Heat Exchanger UA Values	7.10
7.6	Effects of Evaporator Bypass and Emergency Heat Exchanger Coolant Supply Failures on Isotope and Isotope Coolant Temperatures (Shield Tank Drained)	7.11

## LIST OF TABLES

<u>TABLE NO.</u>		<u>PAGE</u>
<u>Section 2</u>		
2.1	Summary of RITE System Design Point Average Performance	2.3
2.2	G-189A Flow Code Options	2.10
2.3	RITE Simulation Study Organization	2.12
<u>Section 3</u>		
3.1	GPOLY Logic for the Low Temperature Heating Loop	3.8
3.2	GPOLY Logic for the Water Loop	3.13
3.3	GPOLY Logic for the Air Loop	3.17
3.4	GPOLY Logic for Solid Waste Loop	3.22
3.5	GPOLY Control Logic for Computing/Implementing Stable Time Increment (Vapor Loop)	3.25
3.6	GPOLY Logic to Attain Mass Balance in Vapor Loop	3.30
3.7	GPOLY Control Logic for Potable Water Storage Loop	3.38
3.8	GPOLY Logic for the Cooling Loop	3.42
3.9	GPOLY Logic for the High Temperature Heat Source	3.47
<u>Section 4</u>		
4.1	Correlations from Complete System Operations	4.14
<u>Section 5</u>		
5.1	RITE System Characteristics	5.3
5.2	Pyrolysis Vent Gas Analysis	5.13
<u>Section 6</u>		
6.1	Key System Parameters for Radiator Coolant Temperatures of 35, 40, 50°F	6.5
6.2	System Output for Evaporator UA's of 75, 92.6 and 150 Btu/hr-°F	6.9
6.3	Key System Parameters for Condenser UA's of 75, 117 and 150 Btu/hr-°F	6.12

## Section 1.0

### INTRODUCTION AND SUMMARY

An analytical simulation of the RITE-Integrated Waste Management and Water Recovery System using Radioisotopes for Thermal Energy has been prepared for the NASA-Manned Space Flight Center (MSFC). The RITE system (Reference 1 and 2) is the most advanced concept water-waste management system currently under development and has undergone extended duration testing in 1973. It has the capability of disposing of nearly all spacecraft wastes including feces and trash and of recovering water from usual waste water sources: urine, condensate, wash water, etc. All of the process heat normally used in the system is produced from low penalty radioisotope heat sources.

The analytical simulation has been developed with the G189A computer program (References 3 and 4). The objective of the simulation was to obtain an analytical simulation which can be used to a.) evaluate the current RITE system steady state and transient performance during normal operating conditions and also during off normal operating conditions including failure modes and b.) evaluate the effects of variations in component design parameters and vehicle interface parameters on system performance.

The programmed simulation was implemented so that data for the following general categories of system steady state and transient operating conditions were obtained:

- A. Steady state performance for design point and off design operating conditions. The simulation includes steady state mass and energy balances which enabled the following quantities to be determined:
  - 1. Water recovery efficiency; water output and loss rates
  - 2. Required gaseous expendables:  $O_2$  and  $N_2$
  - 3. Heating and cooling requirements
  - 4. Power requirements

- B. Transient performance for design point and off design operating conditions. Time line data for nominal and some off nominal events for the following activities were imposed on the simulation: micturation events, trash processing, defecations, and wash water processing. Some of the more significant transient output data obtained for these cases included:
1. Heat block cyclic temperature variations.
  2. Evaporator controller performance
  3. Incinerator cyclic temperature excursions
  4. Condenser and evaporator flow, pressure and temperature cyclic variations.
- C. Transient performance for failure mode conditions. Failure of the low temperature heating loop, the cooling loop, and the incinerator vent valving in different modes were investigated. Emergency equipment heat transfer capabilities were evaluated. The significant transient data obtained here were the transient fluid and radioisotope capsule temperatures.
- D. Steady state performance with variations in component design parameters and vehicle interface parameters. Key parameters which affect RITE performance are condenser vapor pressure and evaporator temperature level. These have been investigated by varying the evaporator heating fluid supply temperature and the flow area of the orifice in the steam outlet circuit for the condenser. Other parameters which were varied in this investigation include the condenser cooling fluid supply temperature level and the overall conductances (UA) of the evaporator and condenser.

This report describes the subject analytical simulation effort in the following sections. Section 2, Technical Discussion, contains descriptions of the RITE system and its operation, the G189A computer program emphasizing features required in the simulation of the RITE system, the study approach, summarized simulation results, and applications for the prepared RITE simulation. Section 3, the Computer Model Description, discusses the organization of the analytical simulation model of the RITE system and describes the modeling logic for the subsystems and individual components in the subsystems. Section 4, the Computer Model Correlation section describes the correlation of G.E. component test data and G189A program modeling data for the pyrolysis units, the air sterilizer, the heat pipe, the heat block, the low temperature isotope heater, the evaporator, and the condenser. Section 5, the Characteristics of the RITE System, describes the steady state or 24-hour average performance data for the various subsystems, some nominal transient conditions, and the operational envelope for the system. RITE Simulation Design Applications, Section 6, discusses the effects on system performance due to variations in component data from the design point values for the following: low temperature heating loop inlet temperature to evaporator, condenser vapor pressure, coolant inlet temperature to the condenser, and overall conductances, UA, from the heat transport fluid to the process liquid for the condenser and the evaporator. Section 7, Failure Mode Analysis, describes system transient performance subsequent to various failures in the low temperature heating loop, the cooling loop, and the incinerator vent valving.

## Section 2

### TECHNICAL DISCUSSION

#### 2.1 RITE System Descriptions and Operations

The RITE system is the most advanced concept water-waste management system under development by NASA and the AEC. It has the capability of recovering water from all spacecraft waste materials including feces and can also shred and process trash. It goes one step further than other systems by automatically pumping the brine-sludge residue from the water recovery unit to an incinerator that reduces the solid waste to an innocuous ash. All of the process heat used in the system is produced from radioisotope thermal energy (RITE) sources. A schematic of the system is shown in Figure 2.1. System specified performance data are given in Table 2.1.

Figure 2.2 shows input material rates for the system and nominal time-line data for material processing by various components in the system. The feces are conveyed into the blender via an air flow of approximately 40 CFM. Here the feces are mixed with the warm wash water and the resulting slurry is pumped along with the transport air to the evaporator.

The urinal also uses an air flow to convey the urine into a liquid gas separator. The separator centrifugally separates the phases with the air going directly to the suction blower and the urine metered into the evaporator. For urine addition, it is not necessary to repressurize the evaporator as is the case for commode and shredder operation. This permits uninterrupted evaporator operation for the frequent micturations. After urinal use, it is automatically flushed.

Trash is manually placed into the shredder as required for later processing when the commode is used. The plastic and paper trash and garbage are shredded and mixed with water and pumped into the evaporator. Coincident operation of the trash shredder and commode permits fewer interruptions of the evaporation process.

Wash water from the galley, shower and clothes washer are also periodically added to the evaporator as required to keep the evaporator water level above minimum.

**Figure 2.1 Integrated Waste Management-Water System Using Radioisotopes for Thermal Energy**

Table 2.1

SUMMARY OF RITE SYSTEM DESIGN POINT AVERAGE PERFORMANCE

Water Recovery Rate	2.3 lb/hr
Water Input Rate	2.38 lb/hr
Water Loss to Incinerator	.05 lb/hr
Water Loss to Condenser Vent	.03 lb/hr
System Conversion Efficiency	9.68%
Total Heat Input Rate	6550 Btu/hr
Heat from LTHL	5280 Btu/hr
Heat from High Temperature Isotope	1270 Btu/hr
Heat Loss to Condenser Cooling Loop	2456 Btu/hr
Heat Loss to Cabin Atmosphere	3768 Btu/hr
Heat Loss to Vent Gases and Water	326 Btu/hr
Electrical Energy Requirements	30 watts-hrs/day
Evaporator Wall/Vapor Temperature	115/105°F
Expendable Oxygen Usage Rate	.043 lb/hr
Expendable Nitrogen Usage Rate	.003 lb/hr



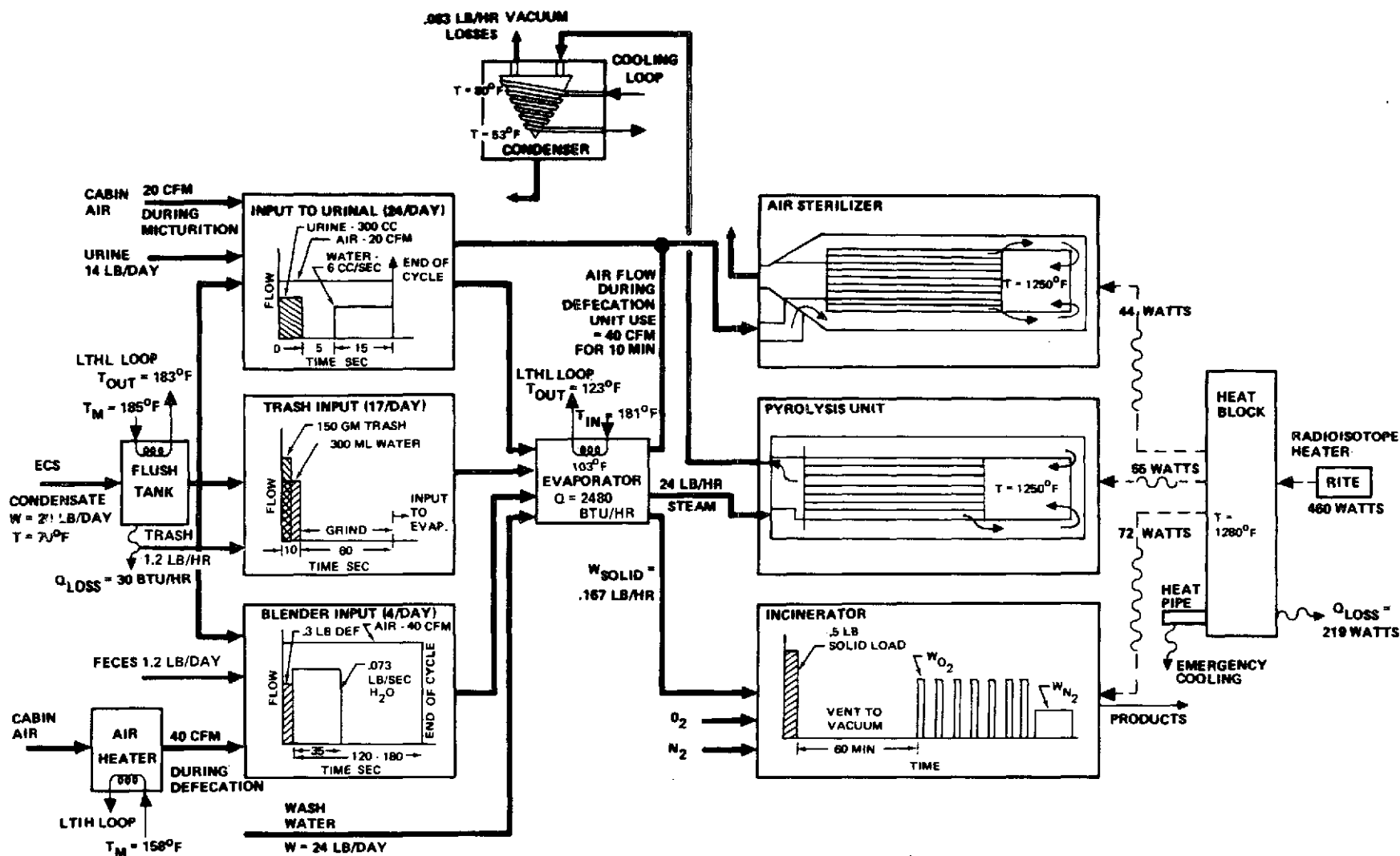


Figure 2.2 Rite System Operation

The evaporator receives all the wastes and provides solid/liquid separation and initial water purification by distillating the wastes at a reduced temperature and pressure of 105°F and 1.1 psia. The reduced temperature boiling minimizes volatilization of impurities. The evaporator has two phase separators. A low speed impeller (100 RPM) creates an artificial gravity field for nuclear boiling in zero gravity. The high speed impeller (2000 RPM) acts as a secondary phase separator to assure that liquid is not carried over with the steam or air flow.

The transport air flow used to convey the wastes into the collection devices and then into the evaporator, is created by a suction blower which then forces the air thru the air sterilizer before it is returned to the cabin. The air sterilizer provides sufficient retention time for the air at high temperatures to assure that all microorganisms are killed. A recuperative heat exchanger in the sterilizer uses most of the thermal energy in the exiting flow to heat the incoming air.

Three catalytic oxidizers are connected in parallel to process the steam from the evaporator. The three units are nested around the single heat source. Each oxidizer contains ruthenium catalyst on alumina pellets. The incoming steam at 120°F is preheated by the exiting steam and is finally heated to 1200°F in the catalyst zone. The impurities in the water vapor are oxidized and are vented from the condenser when the steam is condensed. The steam exits the catalytic oxidizers at approximately 300°F.

The steam passes from the pyrolysis units to the condenser. The steam is condensed between 75°F and 85°F and between .5 psia and .7 psia in the condenser. This pressure is less than evaporator pressure and forces the steam flow through the system. Gases in the steam flow (that are not eliminated in the pyrolysis units) are vented to space vacuum from the condenser.

The low speed impeller in the evaporator also circulates the waste liquid thru the solids pump. The solids pump consists of a filter, solids compression piston and auger. The solids separated during the distillation process are circulated thru the filter with the larger particles being retained as the liquid returns to the evaporator. Periodically, the solids collected in the filter are compressed to remove excess liquid. When the filter is filled with solids, the piston forces the solids into the auger where they are pumped into the incinerator. The incinerator consists of a motor driven shuttle in a sealed cylinder. The solids are pumped by the auger into the shuttle positioned at the cool end of the cylinder. After being filled, the shuttle is moved into the high temperature zone where the solids are initially dried and then thermally decomposed. Finally, oxygen is added and the remaining solids are incinerated and the gases vented. Incineration follows thermal decomposition to minimize the amount of oxygen required to dispose of the solids. After incineration, the shuttle is returned to the cool end and the remaining ashes are blown by use of a nitrogen purge into a filter for collection. The ash is approximately 1% of the initial solid waste input weight.

The purified water is pumped out of the condenser periodically. It passes through conductivity and pH sensors to a set of potable water storage tanks. The tanks are filled in rotation, and tested for chemical and microbiological purity prior to use.

The low-temperature RITE heater provides heat for the evaporator and the water storage tanks. The tanks are heated (160°F) to prevent microbial growth. The flush water tank is heated to 100°F and disinfectant is automatically added to this water to maintain sterility.

The system is automatically controlled to function in a fail-safe manner. Alarms warn of any failures that have occurred. If any component in the system ceases to function, components that supply effluent to it are automatically shut down.

Overheat of the RITE heaters is prevented by discharging surplus heat to compartment air. The high-temperature RITE heater discharges surplus heat through a heat pipe to the compartment air. The low temperature RITE heater is protected from overheat by a separate liquid loop and a heat exchanger which is interfaced with the low-temperature fluid loop. Excessive heater temperature (190°F) activates this loop.

## 2.2 G189A Program Descriptions

The G189A Generalized Environmental/Thermal Control and Life Support Systems (ETC/LSS) Computer Program (References 3 and 4) was selected for use in modelling the RITE system. It provides an extremely versatile and useful analytical tool for support of ETC/LS system development work beginning with preliminary design and proceeding through hardware design, system integration and test, and actual flight. A G189A simulation is useful in determining total system steady state and transient performance, establishing test conditions, and performing failure mode analyses.

The RITE system is simulated with the program by describing the equipment in terms of individual "components" which are connected by gaseous or liquid flow streams. Each particular type of component is simulated by an individual component subroutine contained within the G189A program. A component corresponds to all or part of a physical part; such as a tee, heat exchanger, etc.; or to a complete subsystem or process; such as incineration, pyrolysis, etc. The individual components of the simulation are arbitrarily assigned successive integer numbers which are used to define the component flow stream interconnections and to specify the computational sequence of the individual components. The interconnection and sequence data are specified by the user and may be easily changed. The computational sequence generally follows a path corresponding to the system flow stream paths.

The G189A component subroutines perform heat transfer, chemical reaction, and mass and energy balances for both steady state and transient operating conditions. System pressure drop versus fan or pump pressure rise balances can also be performed if desired. Examples of energy balances are: (1) the

summation of heat rejection from the heat block balanced with the total heat generated by the high temperature isotope and (2) the summation of condenser heat loads balanced with the heat rejected to the cooling loop. Examples of mass balances include the balancing of water vapor and  $\text{CO}_2$  generation rates from the evaporator against oxygen supply rates, pyrolysis rates, and condenser removal rates.

The G189A program is essentially comprised of seven sections: (1) a main subroutine or Master Control Block (MCB); (2) an Input Editor; (3) two user coded subroutines, GPOLY1 and GPOLY2, for implementing additional computational logic; (4) a set of ETC/LSS component subroutines; (5) a pressure drop analysis subroutine; (6) a set of utility subroutines; and (7) a SD-4060 plotting package.

Figure 2.3 presents a block diagram of the G189A program logic. The program operations and their sequence are determined by the Master Control Block (MCB), also known as subroutine ECLST. The operations performed by the MCB are indicated in the center block of Figure 2.3.

All component data and table data are dynamically loaded, at execution time, into a large single array which is predimensioned in the main program. The G189A program allocates storage within this single array as the simulation data are input and at completion the number of unused storage locations are printed. If the simulation data overflows the array execution is halted and a note to that effect is printed. This feature allows computer core storage requirements to be adjusted to each simulation and may result in cost savings and decreased turn-around time when operating on multi-processing or time sharing computer systems.

Another capability provided by the G189A program is the use of a flow code option. Table 2.2 specifies the type of flow data computed and stored within the program for each of the five basic flow code options (0-4). Addressing provisions for 100 flow code options (0-99) are provided within

2.9

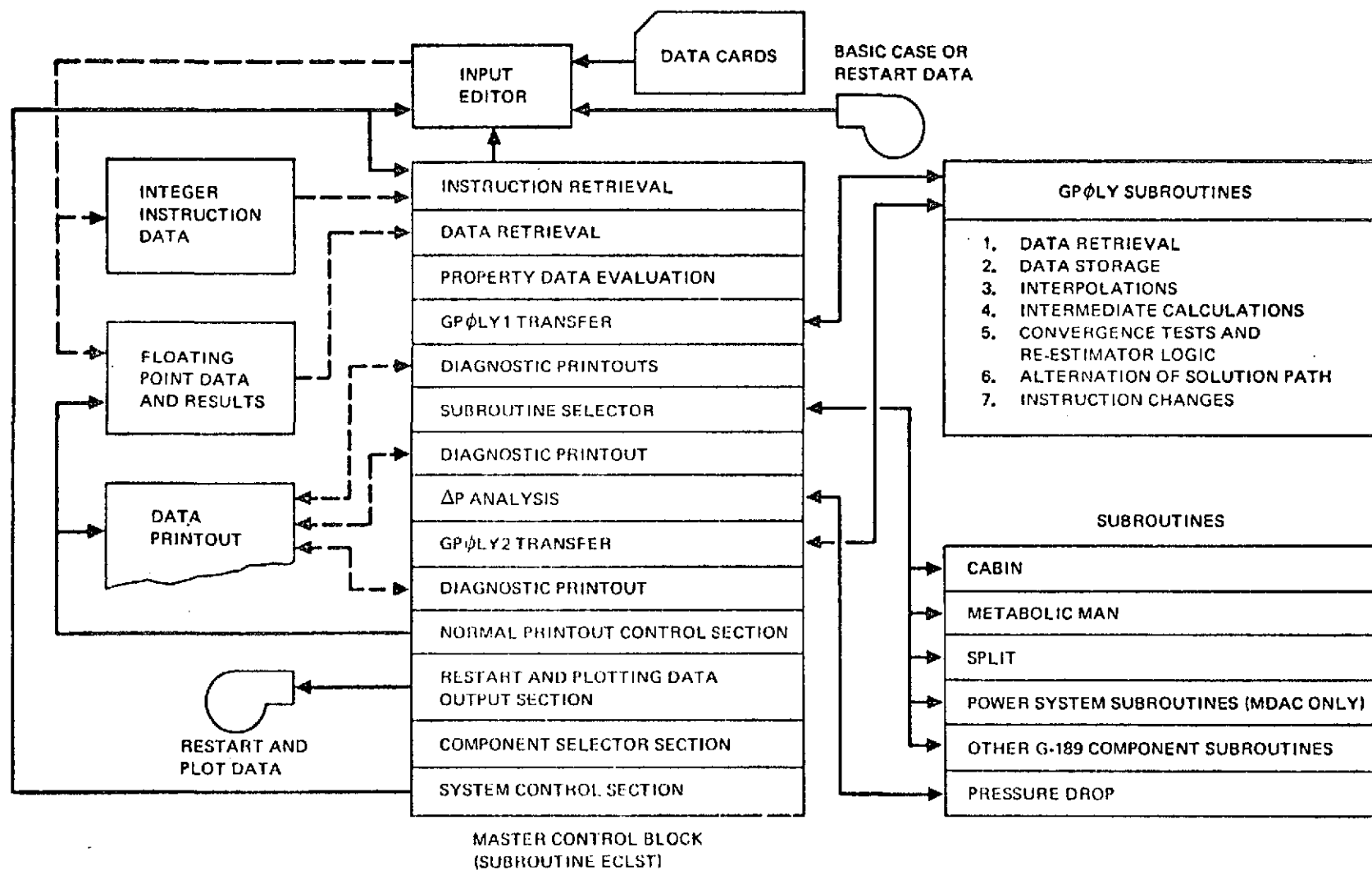


FIGURE 2.3 G-189A PROGRAM LOGIC

Table 2.2  
G189A FLOW CODE OPTIONS

	Flow: Flow Code =	Liquid 0	Gaseous 1 2 3	Liquid 4
1. Total Flow		↑	↑	↑
2. Temperature		↑	↑	↑
3. Upstream Duct Outlet Pressure		↑	↑	↑
4. Component Outlet Pressure		↑	↑	↑
5. Noncondensables Flow		↑	↑	↑
6. Condensable Vapor Flow		↑	↑	↑
7. Condensable Entrained Liquid Flow		↑	↑	↑
8. Noncondensables Specific Heat		↑	↑	↑
9. Noncondensables Molecular Weight		↑	↑	↑
10. Oxygen Flow		↑	↑	↑
11. Diluent Flow		↑	↑	↑
12. Carbon Dioxide Flow		↑	↑	↑
13. Trace Contaminants Flow		↑	↑	↑
14. Special Flow No. 1		↑	↑	↑
15. Special Flow No. 2		↑	↑	↑
16. Special Flow No. 3		↑	↑	↑
17. Special Flow No. 4		↑	↑	↑
18. Special Flow No. 5		↑	↑	↑
19. Special Flow No. 6		↑	↑	↑

the program. The basic liquid flow code option of zero requires the minimum amount of data storage locations (4) and contains total flow, temperature, and pressure data. A liquid flow code option of four reserves nine storage locations and allows the total flow to be subdivided into five constituent flows (Special Flow Nos. 2-6). These special flow constituents may be selected from up to 99 different species whose thermophysical property data are defined by the user. Three gaseous flow code options (1-3) are available.

The basic gaseous flow code option of one requires nine data storage locations and subdivides the total flow into noncondensables, condensable vapor, and condensable liquid flows. The gaseous flow codes of two and three further subdivide the noncondensables flow into fixed constituents and special flow constituents, these options are useful in tracking particular constituent flows within the advanced recovery systems such as carbon dioxide, hydrogen, methane, etc.

The G189A thermodynamic and mass balance equations are based on real gas mixtures and the thermodynamic properties (specific heat, molecular weight, viscosity, and thermal conductivity) of the total flow stream are computed from the input constituent property data and the calculated constituent flows. All component and table data are input using a standardized set of data card formats which provide ample room for comments. This feature allows the user to document all input data and, if desired, all component output data for an ETC/LSS simulation thereby providing a self-documenting set of input data which are always in agreement with the simulation.

### 2.3 Study Approach

The study approach is outlined in Table 2.3. As indicated the system was considered to be comprised of individual groups or "loops" of components which interface with each other and with boundary conditions. Schematics which are often called Component Definition Diagrams were prepared for these loops. These diagrams indicate assigned component numbers, assigned component subroutines, flow connections, and boundary



TABLE 2.3

## RITE SIMULATION STUDY ORGANIZATION

TASK REQUIREMENT	ORGANIZATION	PROGRAM NEW SUBROUTINES	CORRELATION	SUBSYSTEM SIMULATION	SYSTEM SIMULATION	APPLICATIONS
COMPONENT WORK EFFORT	G-189A SCHEMATICS MASS & HEAT BALANCE	EVAPORATOR CONDENSER INCINERATOR HEAT PIPE	PYROLYSIS AIR STERILIZER HEAT PIPE EVAPORATOR CONDENSER INCINERATOR HEAT BLOCK	AIR LOOP VAPOR LOOP SOLID WASTE LOOP, COOLING LOOP, FLUSH WATER LOOP LTHL, POTABLE WATER LOOP	SYSTEM	STEADY STATE TRANSIENT
TO NASA COR	G-189 SCHEMATICS HEAT IN MASS BALANCE	SUBROUTINE DOCUMENTATION	CORRELATION DESCRIPTION, DATA DECKS CONDENSER/ EVAPORATION OPERATION AT 2 PRESSURES	COMPUTER RESULTS DATA DECKS	COMPUTER RESULTS DATA DECK	ANALYTICAL RESULTS COMPARISON WITH 180 DAY RUN DATA
DATA REQUIREMENTS (G.E.)	SYSTEM DESCRIPTION			CORRECTED MASS @ HEAT BALANCE, CONTROL LOGIC	BASELINE 180 DAY RUN OPERATION DESCRIPTION	180 DAY RUN DATA

conditions. This procedure, along with developing system familiarization, resulted in the determination of the need for four new component subroutines and the need for modifications to one existing subroutine. The new subroutines are those for simulating the RITE evaporator (subroutine EVAP), condenser (subroutine CNDNSR), incinerator (subroutine INCIN) and heat pipe (subroutine HTPIP). The modified subroutine is used in simulating the water tanks (subroutine TANKG). The use of heating fluid coils on the RITE water tank walls necessitated that the modelling capability for this feature be added to subroutine TANKG.

As noted in Table 2.3 the individual loops are as follows: air loop, vapor loop, solid waste loop, cooling loop, low temperature heating loop, flush water loop and potable water loop. Once the loop of components were specified the next task was to obtain the parameter data for the individual components in the loops and to prepare the corresponding input data for the G189A program. The procedure for obtaining these data was as follows.

Plotted data showing the results from steady state and transient tests for some individual system components were provided by General Electric. The G189A program models for these components were correlated with the test data by adjusting such quantities as heat transfer coefficients and thermal masses so that computed temperatures closely agreed with the corresponding test data. In cases where component test data for use in determining component parameters were not available or values for the parameters themselves could not be provided by General Electric, two additional means for arriving at values for these parameters were undertaken. The first was to perform engineering calculations aimed at obtaining reasonable estimates for the parameters. The other means was to obtain computer runs for predicting performance of individual loops. Comparing test data for loop performance, such as supply and return temperatures at a given flow rate, with the corresponding analytical data often indicated the accuracy of parameter data for components in the loop. Simulations for the loops noted in Table 2.3 were run on the computer. Modifications in component data were made until satisfactory loop performance was achieved. When the loop simulations were considered

To be satisfactory they were combined into the total system simulation. The total system simulation has been used to a) evaluate the current RITE system steady state and transient performance during normal operating conditions and during off-normal operating conditions including failure modes and b) evaluate the effects of variations in component design parameters and vehicle interface parameters on system performance.

#### 2.4 Summary of Computed Results

Computed data have been obtained for steady state and transient operational conditions. Steady state simulation cases have been run in order to determine design point and off design operating conditions. Steady state or 24-hour average design point operating conditions are summarized in Table 2.1. Off design point operating conditions have been used in preparing the performance maps for Sections 5 and 6. Section 6 contains plotted data obtained for investigations of the effects of variations in boundary conditions and component heat transfer characteristics on system performance.

One of the more significant results of these investigations is that for a LTHL supply temperature of 180°F to the evaporator the optimum water recovery rate decreases from 2.3 lb/hr to 2.13 lb/hr as the condenser coolant inlet temperature is increased from 35°F to 50°F. The condenser pressure level for optimum water recovery varies from 0.63 psia to 0.8 psia for the cases. The importance of condenser pressure level is explained and demonstrated in Section 5.3. Briefly, for condenser pressure levels lower than the optimum, condensation rates and therefore recovery efficiency are reduced due to the small temperature difference between the condenser wall temperature and the saturation temperature of the vapor. For condenser pressure levels higher than the optimum, evaporation rates are reduced with correspondingly reduced recovery rates due to the higher back pressure which restricts output flow.

In investigating changes from the design point values in evaporator and condenser overall conductances (UA) between heat transport fluid and process liquid the following improvements in system performance were determined: increasing the evaporator UA from 92.6 Btu/hr °F to 150 Btu/hr °F increased

the water recovery rate from 2.3 lb/hr to 2.42 lb/hr while increasing the condenser UA from 117 Btu/hr °F to 150 Btu/hr °F increased the water recovery rate from 2.3 lb/hr to only 2.32 lb/hr.

Transient cases have been run in determining the dynamic performance of the system with nominal timelines for micturition, defecation, and incineration activities. The plotted transient results are given in Section 5. The above step function type time line activities cause damped temperature responses in system components such as the evaporator, heat block, and incinerator.

Transient failure mode data are presented in Section 7 for failures in the cooling loop, in the incinerator vent valving, and in the low temperature heating loop.

Transient evaporator and condenser temperatures for a failed cooling loop show the condenser wall temperature rising from 75°F and stabilizing at about 100°F with a corresponding slight increase in evaporator temperature. During these conditions the condenser recovery rate decreases to zero and the evaporator vapor output decreases to a low value due to the increased back pressure from the condenser.

The incinerator vent failure results in an increase in incinerator temperature from about 1150°F to about 1460°F due to the prolonged oxidation of solids within the evaporator. The isotope capsule temperature is only slightly perturbed during this condition due to the large thermal mass of the heat block and the components mounted in the heat block.

The failure of the LTHL were postulated to occur with loss of coolant flow, loss of coolant from the loop due to leakage, failure of the evaporator bypass valve or draining of liquid from the evaporator, and loss of water from the low temperature isotope shield tank with various types of simultaneous failures in the LTHL. Loss of coolant flow results in an increase in coolant temperature from 150°F to about 300°F with corresponding slight changes in isotope and shield tank temperatures. Loss of coolant from the

loop, however, results in an increase in isotope capsule temperature from 900°F to over 1000°F in about 23 minutes with the temperature still rising. This is due to the fact that with the loss of the conductive path through the fluid in the isotope capsule cooling jacket heat transfer between the isotope capsule and the tank is now mainly by radiation and is assumed to be much less than when the fluid is present in the isotope capsule cooling jacket.

Loss of heat removal from the LTHL due to either failure of the evaporator bypass valve in the full bypass position or due to draining the evaporator liquid is protected by the emergency heat exchanger loop. It was found that an emergency heat exchanger  $UA > 50.0 \text{ Btu/hr } ^\circ\text{F}$  is required to adequately control the coolant temperature to below 190°F.

With loss of water from the isotope shield tank and with the evaporator bypass valve failed in the full bypass condition, the plotted results show that the emergency heat exchanger loop maintains the coolant outlet temperature at 210°F and the isotope capsule at about 995°F. Of course with these conditions along with a loss of the emergency heat exchanger loop, the coolant and isotope capsule temperatures rather quickly rise to excessive levels.

With the exception of the LTHL failures, failures in the system would only interrupt RITE operation until repair is accomplished and, at least in the short run, would not endanger the health of the crew. Failures in the LTHL could result in the loss of isotope cooling which could lead to loss of capsule integrity and the escape of radioactive particles. Therefore, a secondary, passive, emergency cooling mode should be an integral part of the isotope design.

## 2.5 Applications for Prepared RITE Simulation

The prepared analytical simulation of the RITE system can be used to accomplish several different objectives. These objectives are of course related to the future research and development activities which the RITE system and derivative systems will experience. A partial list of possible future RITE activities along with corresponding applications for the prepared simulation is as follows:

### A. Improvements in Present System's Performance

Further G189A simulation investigations of component design changes could lead to additional insight concerning component modifications directed toward improving system performance. For example, better thermal insulation characteristics for components in the low temperature heating loop would reduce the present large heat loss to the environment from this loop.

### B. Changes in System Boundary Conditions Commensurate with Future Spacecraft Characteristics.

Investigations of system performance with revised liquid and solid waste material processing quantities and time lines, available coolant flow and temperature level, cabin atmosphere temperature, pressure, and composition, etc., appropriate to future spacecraft characteristics will determine the suitability of the RITE system to these environmental and operational characteristics.

### C. Scaling of RITE System Equipment

Investigations of system performance with scaled component characteristics as a result of changes in boundary conditions and/or accommodated crew size will determine optimal component characteristics. Examples of these are: required overall UA for heat transport to fluid or environment, volume, pumping power, weight, heating and cooling requirements and expendables required. This type of investigation could be directed toward establishing an experimental RITE configuration for a Sortie Lab. experiment.

#### D. Integration with Other Systems or Subsystems

Revising the present simulation to provide modeling for components from integrated systems or subsystems as a result of revised boundary conditions or other reasons would lead to a system simulation capable of supporting the design and development of the integrated system. For example, imposing a large wash water requirement on the system due to inclusion of a shower in the system might lead to integrating a multifiltration assembly with the PRITE system.

### Section 3

#### COMPUTER MODEL DESCRIPTION

The RITE system computer model is comprised of eight main subsystems. The subsystems and their function are defined below:

1. Low Temperature Heating Loop - Supplies heating fluid (water) to the evaporator, potable water storage tanks, flush water tank, hot water heater, and the air heater.
2. Flush Water Loop - Collects Environmental Control System (ECS) condensate input. Provides flush water for the urinal, blender and the trash shredder.
3. Air Loop - Provides drying air flow for the urinal. Sterilizes exhaust air in the air purifier.
4. Solid Waste Loop - Provides solid, liquid and gaseous inputs to the evaporator. Incinerates solids and then collects resultant ash in the ash collector.
5. Vapor Loop - Evaporates water vapor from the evaporator solid-liquid slurry. The vapor is purified in three catalytic burners and is then condensed in the condenser.
6. Cooling Loop - Provides cooling flow to the condenser and the water cooler.
7. Potable Water Loop - Collects and stores water from the condenser. Distributes hot and cold water as required by users.
8. High Temperature Heat Source - Provides high temperature heat to the incinerator pyrolysis units and the air sterilizer.

#### 3.1 Computer Model Organization

The G-189A RITE simulation model was divided into the corresponding eight subsystem loops. Each loop was developed independently prior to being integrated into a total system. The component numbers for each loop were assigned as shown below.



	<u>Subsystem</u>	<u>Component Numbers</u>
1	Cooling Loop	1 - 19
2	Air Loop	20 - 39
3	Flush Water Loop	40 - 59
4	Potable Water Storage	60 - 99
5	Low Temperature Heating Loop	100 - 179
6	Heat Block	180 - 199
7	Solid Waste Loop	200 - 219
8	Water Pyrolysis	220 - 239

The calculation begins at component 100, the radioisotope heater of the low temperature heating loop. In the first part of this loop the heating fluid is distributed to all user loads. The calculation then transfers to the flush water loop where flush water is provided for the urinal, shredder and blender as required. The calculation then transfers to the air loop. Urine, flush water and air enter this loop in accordance to the micturation schedule noted in Section 5. After liquids are separated and directed to the evaporator, the remaining air is blown by the fan to the air sterilizer and then back to the cabin.

Calculation control then shifts to the solid waste loop. In the solid waste loop the air, solids, and flush water requirements for the blender and trash shredder are collected and directed to the evaporator. The input schedules are defined by table data and are shown in Section 5. In the evaporator water vapor is generated and the solids are directed to the incinerator.

The calculation then shifts to the vapor loop. In this loop the gases given off by the evaporator are oxidized in the pyrolysis beds and the water vapor is condensed in the condenser.

The order of solution then continues with the supply portion of the cooling loop. The coolant loop provides coolant for the condenser and the water cooler. The next operation involves the potable water storage loop. In

this loop the water is collected from the condenser, stored, and then heated or cooled and dispensed upon demand. The calculation then returns to the cooling loop where the coolant is collected and returned to the source.

A heat balance is then performed on the heat block and the heat pipe. Finally the calculation returns to the low temperature heating loop. Here the fluid is collected from the heated equipment and returned to the isotope heater, component 100. The system is then ready to perform another pass.

The logic used to control the operation of each of the above loops is discussed in detail in the subsequent sections. The G-189A schematics of the loop are key elements in the description of the computer model. Therefore, a review is given first of the information contained in these diagrams.

### 3.2 Computer Model Documentation

The description of the computer model for each of the subsystems includes a G-189A block diagram schematic of the system. An example of the applications of a G-189A schematic is shown in Figure 3.1. Each block in the diagram corresponds to a major component of the subsystem. The block contains the computer component number, the component name and the name of the subroutine used to simulate that component. The circled numbers define the system flow splits and tees. The solid arrows connecting the component define the source of primary and secondary flow. Primary and secondary flow are noted by a P and S respectively. Also noted in the Figure is the flow code definition and a list of the constituents comprising that flow code. The solid-dash lines connecting the components define the component order of solution.

The subsequent writeup defines the operation of the loop and technical data for each component. Control for each component is defined by the GPOLY1 and GPOLY2 subroutines. The GPOLY1 and GPOLY2 logic for each component is defined side by side with the corresponding listing. Generally, these component descriptions do not include the detailed numerical data. Numerical values of the input data may be found in Appendix B (Input Data Card Listing).

The system G-189A schematic is shown in Figure 3.2. The system was divided into eight subsystems and the description of the computer model for each of these subsystems follows:

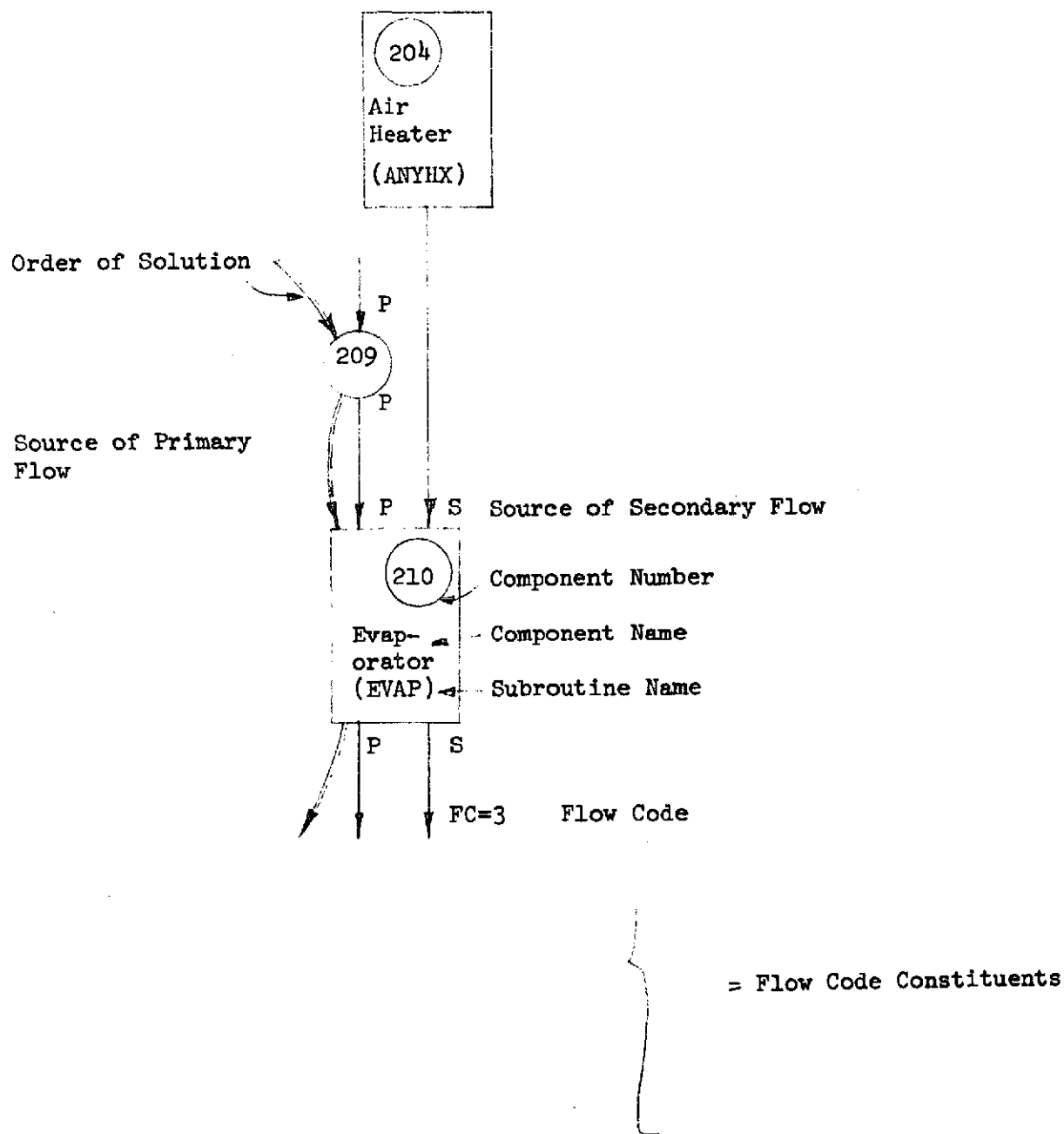


Figure 3.1 G-189 Schematic Representation



### 3.2.1 Low Temperature Heating Loop

The schematic of the Low Temperature Heating Loop (LTHL) is shown in Figure 3.3. The loop provides heat for the evaporator, five potable water storage tanks, water and air heaters, and the flush water tank. The coolant is heated in the isotope heater, component 100, and is then pumped by one of two pumps (104, 105) to the user components. Only one pump is used at any one time. The other pump is switched on upon failure of the first. Flow splits 108, 112, 113, 114, 115, 116, 117, 120, and 121 distribute heating fluid to the evaporator, the five storage tanks, the water heater, air heater and the warm flush water tank. Components 126, 127, 128, 129, 130, 131, 132, 133, and 134, simulated by subroutine FL/MIX, collect return flow from these components and return the flow to the radio-isotope heater.

Prior to the return to the heater, the fluid flows through an emergency heat exchanger, component 137. The heat exchanger is activated when loss of cooling in the loop drives the fluid temperature at the exit of the isotope heater higher than 190°F. The coolant supply, component 135, provides coolant flow to the by-pass temperature control system, components 136 and 139. When the radioisotope coolant outlet temperature reaches 190°F, the coolant flow is directed by flow split 136 to the heat exchanger.

Components 109 and 110 form a by-pass control loop to regulate the evaporator vapor outlet temperatures. When the vapor temperature reaches 105°F the heating fluid is bypassed around the evaporator. Flow to the evaporator is re-established when the vapor temperature drops to 104°F.

Components 124 and 125, form a bypass loop to control the warm flush water tank temperature to  $90 \pm 1^\circ\text{F}$ .

Similarly, components 122 and 123, form a bypass loop around the air heater to control the outlet air temperature to 110°F.

The detail GPOLY1 and GPOLY2 control logic which perform the functions detailed above is given in Table 3.1.

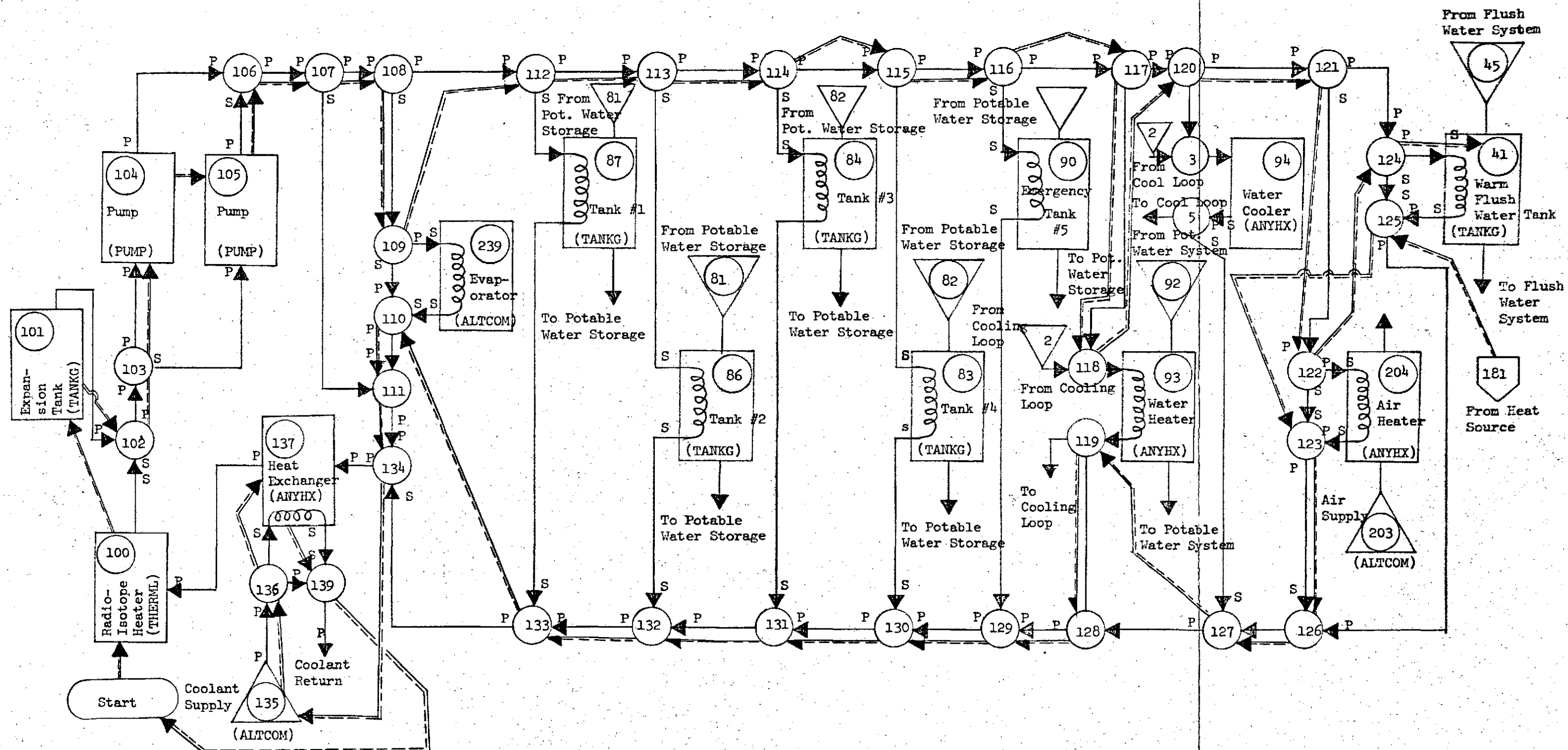


FIGURE 3.3 LOW TEMPERATURE HEATING LOOP G-189A SCHEMATIC

Table 3.1

## GPOLY LOGIC FOR THE LOW TEMPERATURE HEATING LOOP

Order of Solution - START, 100, 102, 103, 104, 105, 106, 107, 108, 109, 112, 113, 114, 115, 116, 117, 118,  
120, 121, 122, 124 to Flush Water Loop

From Heat Pipe, 125, 123, 126, 127, 119, 128, 129, 130, 131, 132, 133, 110, 111, 134,  
136, 137, 139

Component 100, Radioisotope Heater

Subroutine: THERMAL

GPOLY1

The inlet temperature and flows are transferred to the nodal network.

The inlet temperature A(2) is transferred to R(82). The inlet wC<sub>p</sub> is entered in R(75).

Component flow and properties passed to the output R-Array

GPOLY 2

The outlet flow temperature calculated by the nodal network in R(69) is transferred to the output R-array.

Component 104 Coolant Loop

Subroutine: PUMP

GPOLY 2

Pump power is set proportional to flow rate. If pump 1 fails switch valve 103 to pump 2.

```

C      GPOLY1 LOGIC FOR LOW TEMP. HEATING LOOP
C
C      INTERFACE OF THERMAL C100 TO C139
C
IF(N,NE,100) GO TO 100
NLAST=209
R(82)=A(2)
R(75)=CPA*A(1)
DO 1100 J=1,19
1100 R(J)=A(J)
100 CONTINUE
C
C      GPOLY2 LOGIC LOW TEMPERATURE HEATING LOOP
C
IF(N,NE,100) GO TO 100
C      INTERFACE OF THERMAL C100 TO C139
R(2)=R(69)
C
100 CONTINUE
C
C
C      COOLANT PUMP NO.1 COMPONENT 104
C
IF(N,NE,104) GO TO 104
R(70)=0.0151*A(1)
104 CONTINUE

```

Component 109, Evaporator Control Valve

Subroutine SPLIT

GPOLY1

If the evaporator outlet vapor temperature greater than 105°F bypass flow around evaporator. Reestablish evaporator flow when vapor temperature drops back to 104°F.

Component 122, Air Heater Control Valve

Subroutine SPLIT

GPOLY1

Bypass heating fluid around air heat exchanger when outlet air temperature greater than 110°F.

Component 124, Flush Tank Control Valve

Subroutine SPLIT

GPOLY1

When tank liquid temperature greater than 96°F, bypass heating flow. Reestablish flow when temperature drops to 94°F.

Component 136, Emergency HX Controller

Subroutine SPLIT

GPOLY1

When the isotope heater outlet temperature reaches 190°F start coolant flow to emergency heat exchanger.

~~C~~ ~~EVAPORATOR COOLING LIQUID CONTROL VALVE (COMPONENT 109)~~~~C~~~~IF(N,NE,109) GO TO 109~~~~IF(STEADY) GO TO 109~~~~TEMPS=VV(210,2)~~~~IF(TEMPS.GT.105)R(65)=1.0~~~~IF(R(65).GT.0.0.AND,TEMPS.LT.104.)R(65)=0.0~~~~IF(TEMPS.LT.104.)R(65)=0.0~~~~109 CONTINUE~~~~AIR HEATER CONTROL LOGIC COMPONENT 122~~~~IF(N,NE,122)GO TO 122~~~~AIRTEP=VV(204,2)~~~~R(65)=0.0~~~~IF(AIRTEP.GT.110.0)R(65)=1.0~~~~122 CONTINUE~~~~WARM FLUSH WATER TANK CONTROL LOGIC~~~~IF(N,NE,124)GO TO 124~~~~IF(STEADY) GO TO 124~~~~TTANK=VV(41,7)~~~~IF(TTANK.LT.94.0)R(65)=0.0~~~~IF(TTANK.GT.96.0)R(65)=1.0~~~~IF(R(65).GT.0.0.AND,TTANK.LT.94.0)R(65)=0.0~~~~124 CONTINUE~~~~LTHL HX CONTROLLER (COMPONENT 136)~~~~IF(N,NE,136)GO TO 136~~~~RITET=VV(100,2)~~~~R(65)=0.0~~~~IF(RITET.GT.190.0)R(65)=1.0~~~~136 CONTINUE~~



#### 3.2.1.1 Low Temperature Isotope Heater Nodal Network

The simulation of the low temperature heater and shield tank is accomplished using the G-189A generalized thermal analysis subroutine THERML. The low temperature heater consists of a plutonium capsule and electrical heater generating a total of 5120 BTU/hr, circumvented by a heat transfer plate which dissipates heat to the heating fluid (water at about 15.0 psi). The capsule with the heat dissipation plate are immersed in a water tank which not only acts as a radiation shield, but also serves to dissipate the heat from the radioisotope in the event the heating fluid flow stops. Figure 3.4 is a nodal model of the low-temperature heating system. It consists of a water tank, a heater exchanger through which the heat from the radioisotope is transferred to the heating fluid, and the radioisotope core.

#### 3.2.2 Flush Water Loop

The G-189A schematic for the flush water loop is shown in Figure 3.5. The first component in the solution path of the flush loop is the warm flush water tank. The flush water requirements for the urinal, blender, and the wash shredder are given in computer input Table Data 6, 7 and 8 respectively. (Appendix B) These requirements are imposed on the tank component and the flow is pumped to the corresponding valve (components 46, 48 or 50) leading to the user load. The GPOLY logic which performs this function is listed in Table 3.2.

Makeup water is added to the tank from the ECS condensate system. A constant water flow of 0.824 lb/hr at 70°F is entered through alternate component 51.

The TANKG subroutine which simulates the warm flush water tank has been modified to include internal and external convective heating as well as being able to perform steady state heat balance. The subroutine documentation for TANKG is included in Appendix A. A mass and energy balance schematic for the tank illustrating key tank parameters is shown in Figure 3.6. The schematic should be useful in readily selecting key tank data from the program output. The schematic is valid for all the tanks included in the RITE simulation.

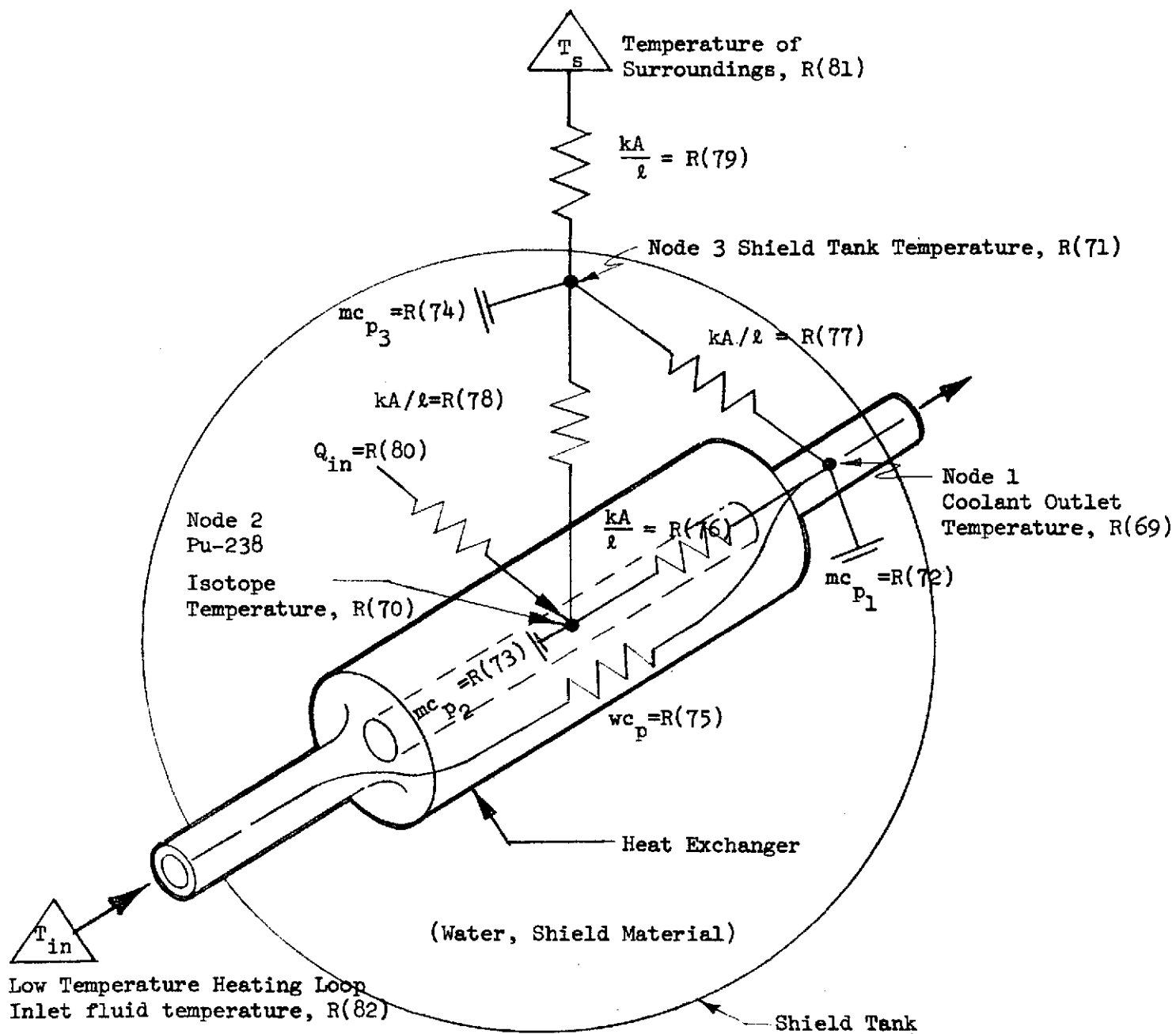


FIGURE 3.4 ISOTOPE HEATER THERMAL NODAL NETWORK

Flow Definition = 4

R(1) = { R(15) = urea  
R(16) = solids  
R(17) = water

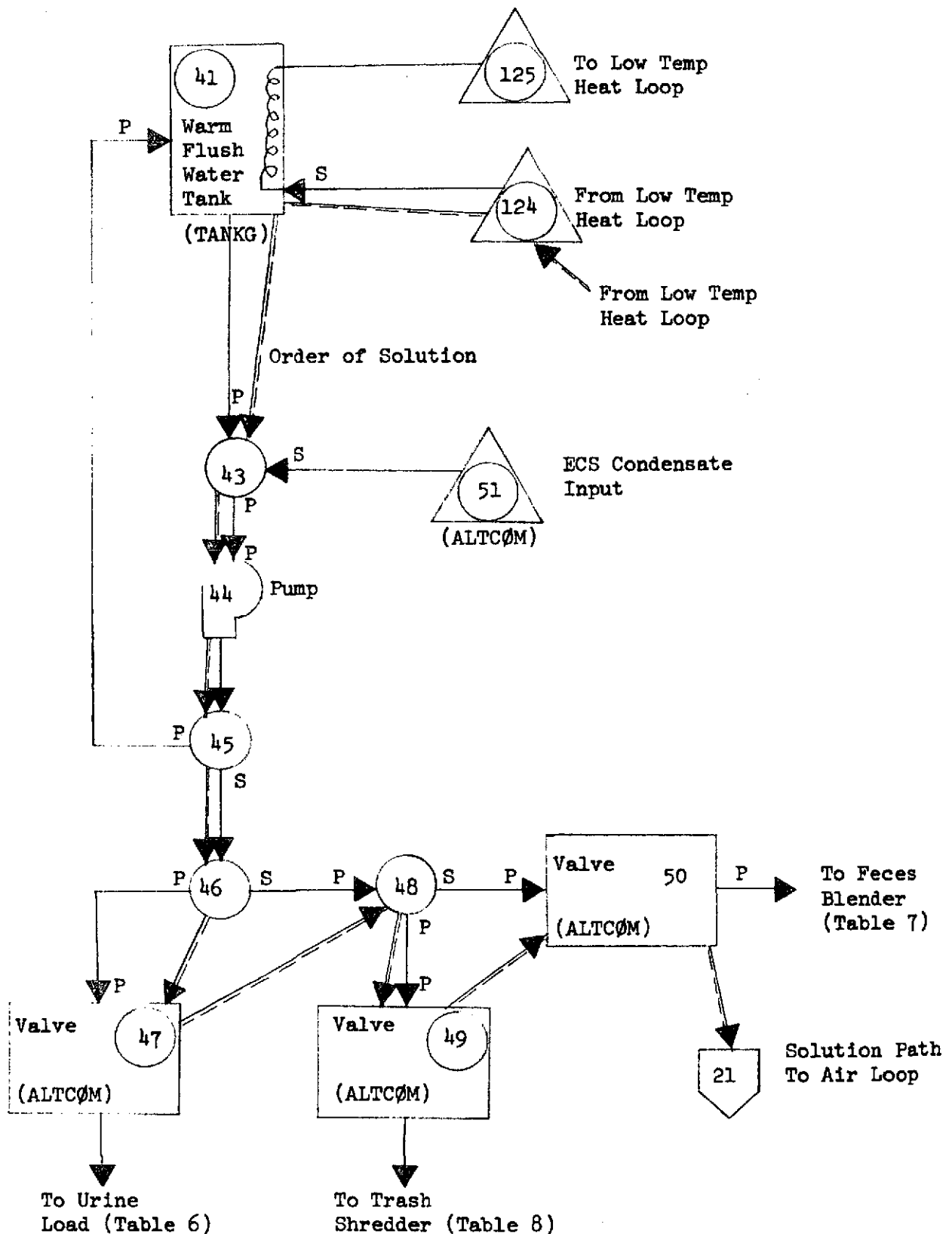


FIGURE 3.5 FLUSH WATER LOOP (FLOW CODE 4)

Table 3.2

## GPOLY LOGIC FOR THE WATER LOOP

Order of Solution From Low Temperature Heating Loop, 41, 43, 44, 45, 46, 47, 48, 49, 50, to Air Loop

Component 41, Warm Flush Water Tank

Subroutine: TANKG

## GPOLY1

The output of the tank is equal to the flush water requirements of the urinal,  $w_u$ , shredder,  $w_s$ , and blender,  $w_f$ . These are read from tables 6, 8 and 7 respectively (see below)

The split ratio of 45 is then calculated

$$SR = \frac{w_c}{w_c + w_u + w_s + w_f}$$

where  $w_c$  is the input condensate flow

The split ratio of 46 is then calculated

$$SR = \frac{w_s + w_f}{w_s + w_f + w_u}$$

The split ratio of 48 is then calculated

$$SR = \frac{w_f}{w_s + w_f}$$

These values are stored in their respective component data.

Component 44, Pump

Subroutine: PUMP

## GPOLY2

Pump electrical power calculated from the following relationship:

$$P_p = 0.0527 + 0.0473 w$$

where  $w$  is the flow rate

```

C
C      GPOLY1  LOGIC  FOR  FLUSH WATER LOOP
C
C
C
C
C
C      COMPONENT  41  WARM FLUSH WATER TANK
C      IF(N,NE,41) GO TO 41
C      WU=VALUE(6,CYCLE(3600,0,0,0),0,0)
C      WS=VALUE(8,CYCLE(21000,0,0,0),0,0)
C      WF=VALUE(7,CYCLE(21000,0,0,0),0,0)
C      R(17)=WU+WF+WS
C      R(1)=R(17)
C      W51=VY(51,17)
C      A(1)=W51
C      A(2)=70,0
C      A(17)=A(1)
C      SPLT45=0,0
C      IF(R(17),NE,0,0)SPLT45=R(17)/(R(17)+W51)
C      CALL SV(SPLT45,45,65)
C      SPLIT46=0,0
C      IF(R(17),NE,0,0)SPLIT46=(WS+WF)/R(17)
C      CALL SV(SPLIT46,46,65)
C      SPLIT48=0,0
C      IF(WS+WF,GT,0,0)SPLIT48=WF/(WS+WF)
C      CALL SV(SPLIT48,48,65)
C      41 CONTINUE

```

```

C
C      FLUSH WATER  PUMP=COMPONENT  44
C      CALCULATION OF PUMP POWER
C
C
C      IF(N,NE,44) GO TO 44
C      R(70)=0,0
C      IF(R(1),GT,0,0)R(70)=0,0527+0,0473*R(1)
C      44 CONTINUE

```

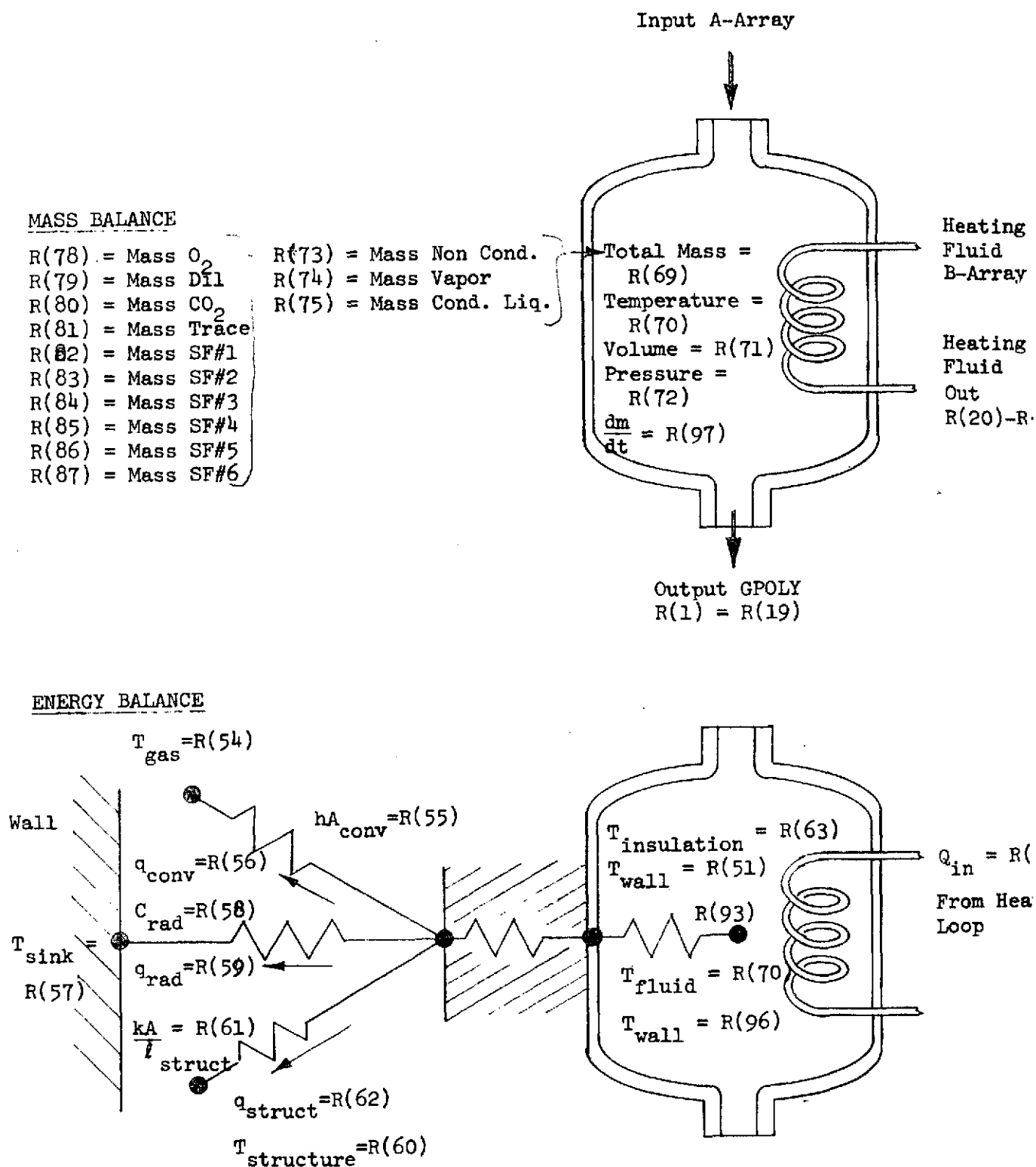


FIGURE 3.6 TANKG MASS AND ENERGY G-189 OUTPUT REPRESENTATION

### 3.2.3 Air Loop

The G-189A schematic for the air loop is shown in Figure 3.7. Urine and drying air enters the system through component 21, the urinal, as defined by the schedule entered in computer input Table 9 (air supply) and Table 10 (urine supply). Urinal flush water enters the loop through component 22. The combined water, urine and air flow enters the liquid-gas separator, component 23, where the liquids are separated and directed to the evaporator. The gases are blown by the fan through the air sterilizer and then returned to the cabin. The control logic for the loop is shown in Table 3.3.

#### 3.2.3.1 Air Sterilizer Thermal Model

The air sterilizer consists of a high temperature sterilization chamber and a counterflow heat exchanger. It operates during the commode use and serves to assure that viable organisms do not escape from the WM/WS via transport air flow. A secondary function is to provide thermal control of the RITE capsule in the heat block should the heat pipe malfunction. A nodal schematic of the air sterilizer is shown in Figure 3.8.

The simulation of the air sterilizer is accomplished using the G-189 component subroutine THERML, the thermal analyzer subroutine. Subroutine THERML was used to determine heat balance and heat loss due to air sterilizer operation and to predict the overall performance of the air sterilizer.

### 3.2.4 Solid Waste Loop

The G-189 schematic for the solid waste loop is shown in Figure 3.9. The loop serves to direct all liquid and solid wastes into the evaporator. Component 201 is used to enter feces in accordance to the schedule entered in Computer Support Table 16 (see Appendix B). Flush water for defecation enters from the flush water loop from component 50. Trash is supplied from component 50 in accordance to the schedule noted in Computer Table 13. The corresponding trash flush water supply enters through flush loop valve component 49. A constant supply of one pound per hour wash water enters the system from component 219. All the solids and liquids are directed to the primary side of the evaporator.

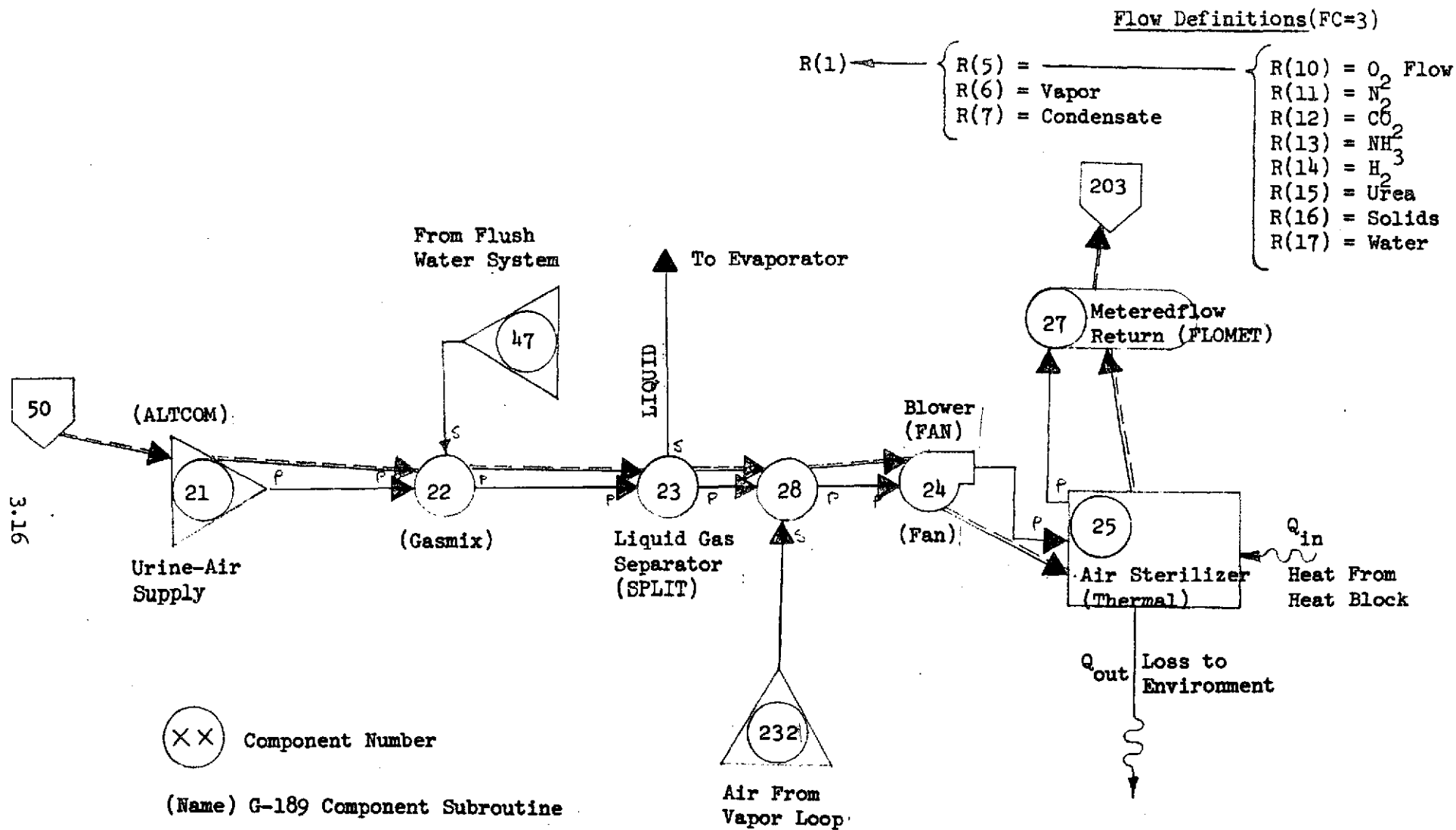


FIGURE 3.7 G-189 COMPONENT DEFINITION DIAGRAM FOR THE AIR LOOP (FLOW CODE 3)

Table 3.3

## GPOLY LOGIC FOR THE AIR LOOP

Component order of solution - From Flush Water Loop 21, 22, 23, 28, 24, 25, 27 to Solid Waste Loop

Component 21, Urine Air Supply

Subroutine: ALTCOM

GPLOY2

Air supply schedule taken from Table 9.  
Air is assumed to be composed of the following constituents:

Water Vapor	1.05%
Oxygen	26.3 %
Nitrogen	72 %
Carbon Dioxide	65 %

Urine input schedule from Table 10.  
Urine is assumed to be composed of the following constituents:

Water	94.2%
Urea	2.8%
Urine Solids	3.0%

Urine temperature set at 98.6°F  
Urine pressure set at 15 psia

Component 22 - Flush Water Supply

Subroutine: GASMIX

GPOLY1

Modification to compensate for flow  
code 4 of flush loop  
Set flush water  $c = 10^{-10}$   
Set molecular wt  $\bar{M} = 10^{-10}$   
This is required to use a gas mix  
subroutine with flush water

C COMPONENT 21 - URINE AIR SUPPLY

IF(N,NE,21) GO TO 21

C

C STANDARD AIR INPUT=14.7 PSI 70 DEG= F

C WATER VAPOR 1.05 PERCENT

C OXYGEN 26.3 PERCENT

C NITROGEN 72.0 PERCENT

C CARB-DIOXIDE 65.65 PERCENT

WAIR=VALUE(9,CYCLE(360,0,0.0),0.0)

R(6)=0.0105\*WAIR

R(10)=0.263\*WAIR

R(11)=0.72 \*WAIR

R(12)=0.0065\*WAIR

C

R(2)=98.6

C

INPUT= MICTURATION

WUR=VALUE(10,CYCLE(360,0,0.0),0.0)

R(3)=15.0

R(4)=15.0

R(17)=0.942\*WUR

R(15)=0.028\*WUR

R(16)=0.03\*WUR

21 CONTINUE

C

C

C

GPOLY1 LOGIC FOR AIR LOOP

C

C

C

IF(N,NE,22) GO TO 22

C

C

B(3)=15.

B(4)=15.

B(8)=1.0

B(9)=1.0E+10

22 CONTINUE



Table 3.3 GPOLY LOGIC FOR THE AIR LOOP (Continued)

Component 24, FanSubroutine: FANGPOLY2

The pump power is calculated using the following relationship:

$$P_p = 117. + 0.91 w$$

where w is the air flow

Component 25, Air SterilizerSubroutine: THERMALGPOLY1

The nodal network is initiated from the R-array

The inlet temperature is set to R(116)  
 $R(116) = A(2)$

All flow in fluid nodes are set to  
 $w_c = A(1) * CPA$

The heat block temperature which transmits heat to the sterilizer is called from heat block component 180  
 $R(118) = VV(180, 76)$

GPOLY2

Reset outlet temperature of R-array from nodal network

C AIR LOOP FAN - COMPONENT 24 -FAN ELECTRICAL POWER

~~C~~

C

IF(N,NE.24)GO TO 24

C

R(72)=0.0

IF(A(1).GT.0.0)R(72)=117.0\*0.91\*A(1)

~~24 CONTINUE~~

IF(N,NE.25)GO TO 25

C

THE FOLLOWING CARDS INTERFACE AIR STE  
 TO THE SYSTEM

C

C

FLOW CONVECTION TERMS

DO 1025 J=1,19

~~1025~~

~~R(J)=A(J)~~

~~R(116)=A(2)~~

~~R(106)=A(1)\*CPA~~

~~R(104)=A(1)\*CPA~~

~~R(102)=A(1)\*CPA~~

~~R(99)=A(1)\*CPA~~

~~R(96)=A(1)\*CPA~~

~~R(93)=A(1)\*CPA~~

C

HEAT BLOCK TEMPERATURE AT NODE 6 HTNG TO AIR STERILI

~~R(118)=VV(180,76)~~

~~25 CONTINUE~~

C

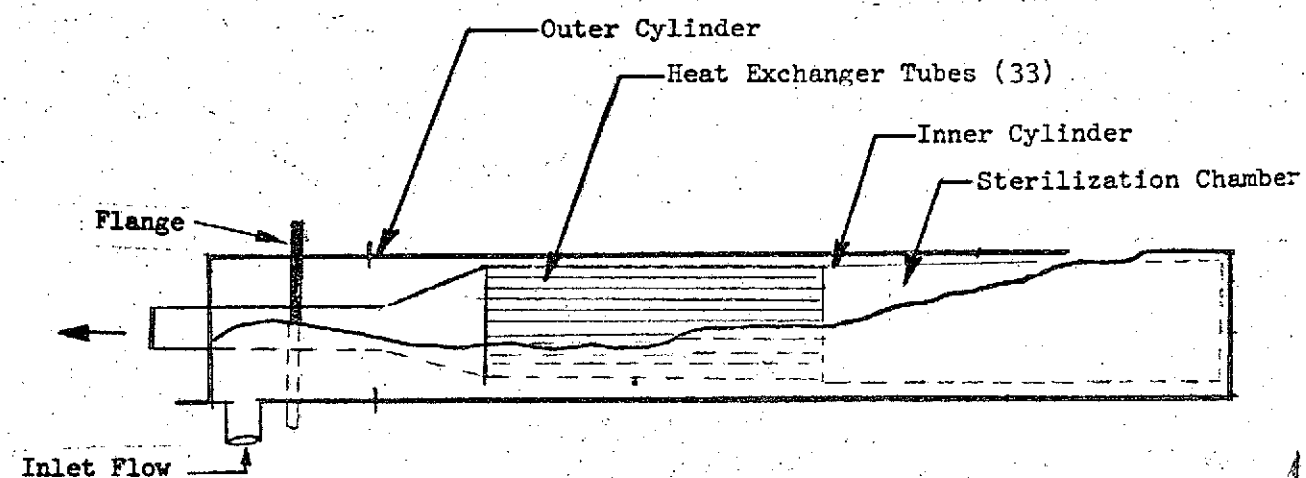
IF(N,NE.25)GO TO 25

~~C~~

~~INTERFACE COMPONENT 25 TO 27~~

~~R(2)=R(80)~~

25 CONTINUE



- Node 1: Outer Cylinder Inlet
- Node 2: Outer Cylinder Circumventing The Counterflow Hx
- Node 3: Outer Cylinder In Contact With Heat Block
- Node 4: Inner Cylinder Defining The Sterilization Chamber
- Node 5: 33 Tubes Defining the Heat Exchanger
- Node 6: Inner Tube Manifold Flow Through Hx to the Outlet
- Node 7: Air Regime Bounded by Nodes 1 and 6
- Node 8: Air Regime Bounded by Nodes 2 and 5, Flow Around Hx Tubes
- Node 9: Air Regime Bounded by Nodes 3 and 4
- Node 10: Air Regime Defining the Air Sterilization Chamber
- Node 11: Air Regime Defining the Return Flow Through Hx Tubes
- Node 12: Air Regime Defining the Outlet Section of Air Sterilizer

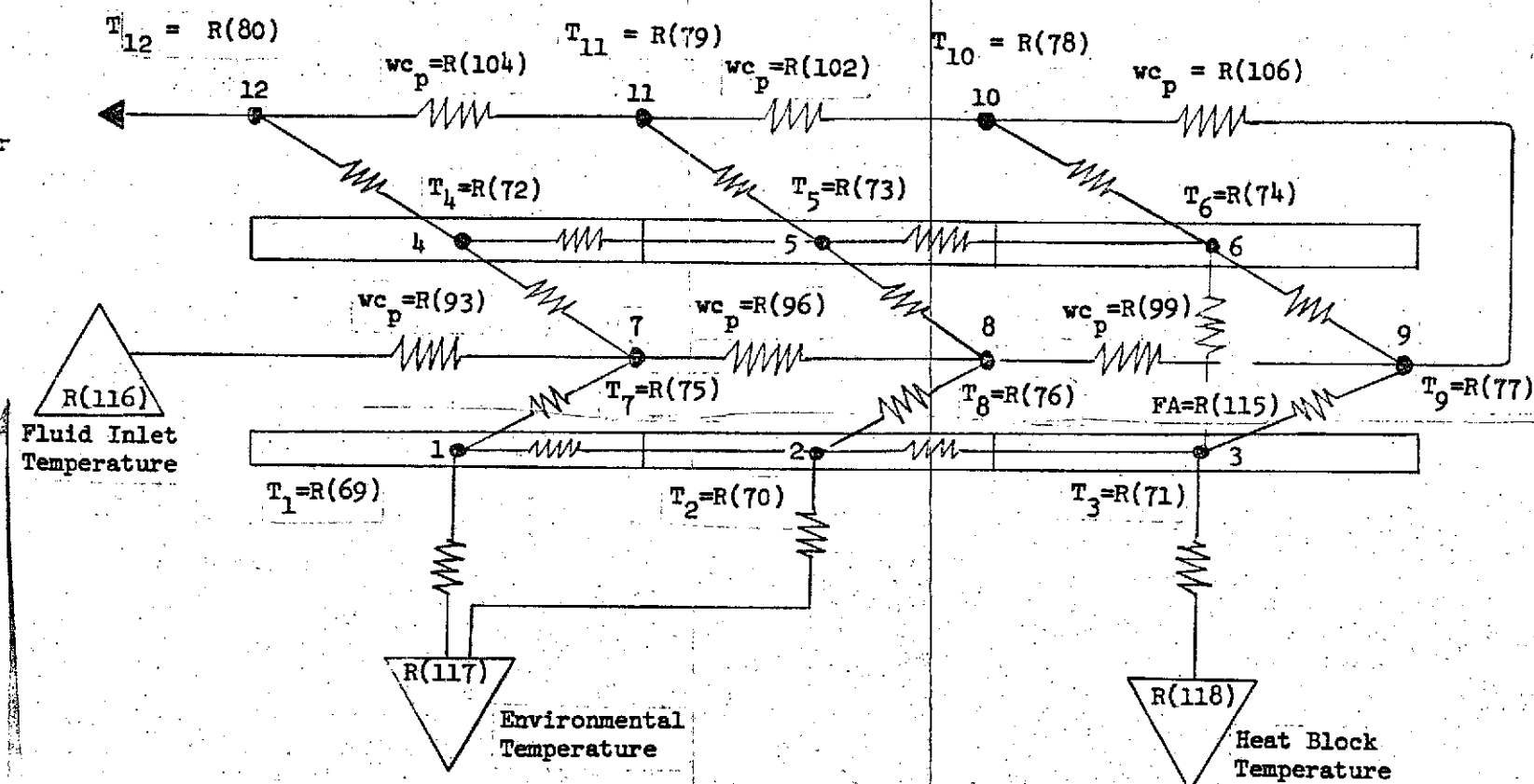


FIGURE 3.8 AIR STERILIZER SCHEMATIC DEFINING THERMAL NODES

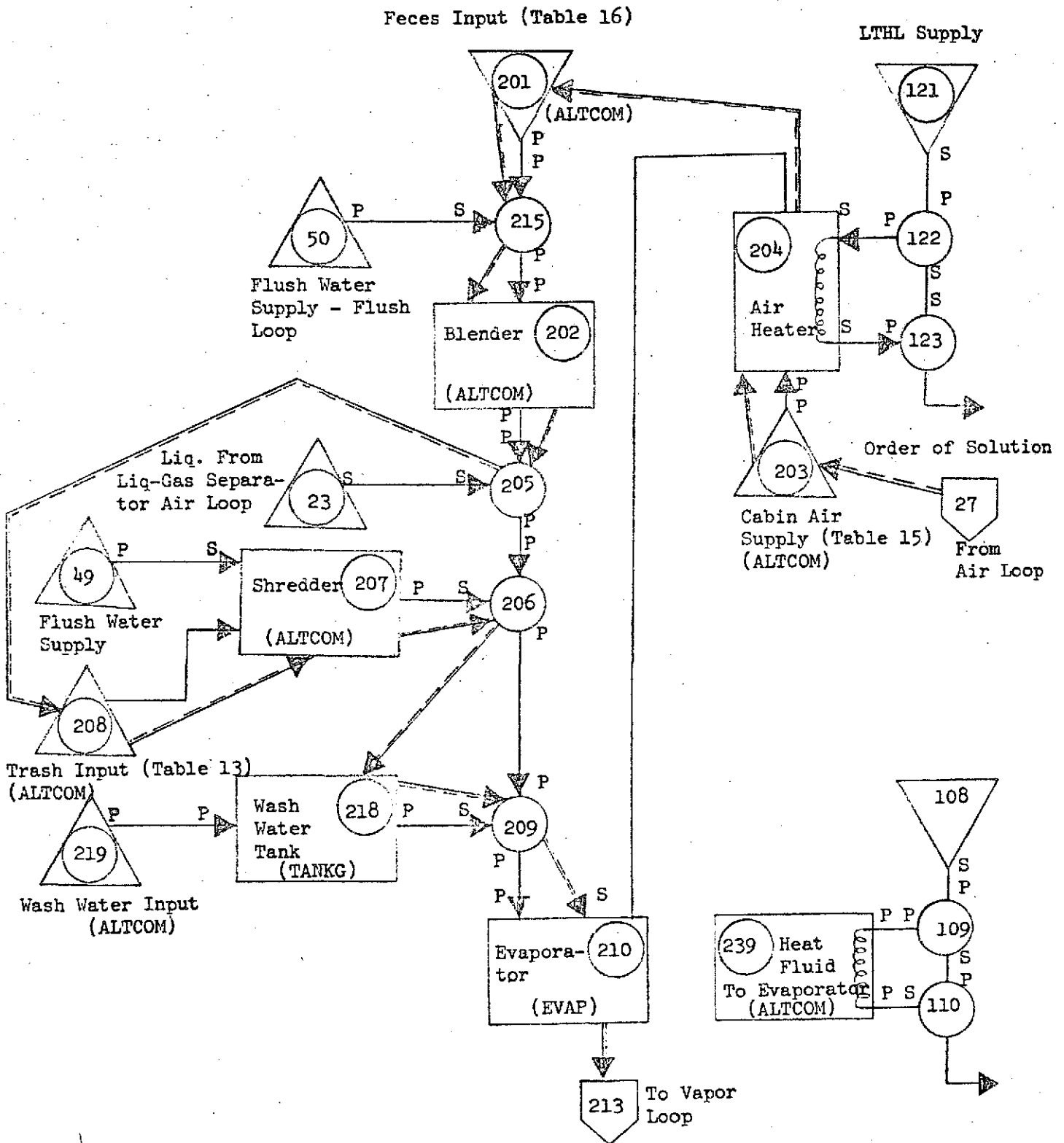


FIGURE 3.9 G-189A SCHEMATIC OF THE SOLID WASTE LOOP

Drying air for defecation enters from component 203 as scheduled by Computer Table 15. The air is heated by the low temperature heating loop prior to entering the evaporator.

The GPOLY logic for the solid waste loop is shown in detail in Table 3.4.

### 3.2.5 Vapor Loop

The vapor loop is comprised of those components through which the vapor stream flows, originating at the evaporator and terminating (condensing) at the condenser. The G-189A schematic of the system is shown in Figure 3.10.

All solid and gaseous inputs enter the evaporator from the solid waste loop. In the evaporator the solids are separated and directed to the incinerator. The water is evaporated and the vapor is directed to the pyrolysis units where ammonia and other products mixed with the vapor are oxidized. The neutralized vapor is then directed to the condenser where the vapor is condensed and the gaseous products of the pyrolysis reactions are vented to vacuum.

The principal problem encountered with the simulation of the vapor loop was the selection of the computational time increment during transient solution. The vapor loop requires a very small computational time increment for stability purposes because of very low evaporator and condenser contained gas weight to gas weight flow rate ratio. A scheme was implemented via GPOLY logic where a stable computing time increment smaller than the system computational time increment was computed at the beginning of the evaporator calculations and the program executed the required number of passes through the vapor loop before proceeding with the larger system time increment to the components that follow the vapor loop. The scheme for computing the stable time increment and performing the computations for the vapor loop components is shown in Table 3.5

#### 3.2.5.1 GPOLY Logic For Vapor Loop Components

The GPOLY logic for key vapor loop components is described below:

Table 3.4

## GPOLY LOGIC FOR SOLID WASTE LOOP

Order of Solution - Start from Air Loop Component 27

203, 204, 201, 215, 202, 205, 208, 207, 206, 218, 209, Vapor Loop Component 210

Component 203, Blender Air Supply

Get total air flow from tube 15

Constituents of air are

Water vapor, R(6) = 1.05%

Oxygen, R(10) = 26.3 %

Nitrogen, R(11) = 72 %

Carbon Dioxide R(12) = .65%

C GPOLY2 LOGIC FOR SOLID WASTE LOOP

C

C

C

COMPONENT 203 BLENDER AIR SUPPLY

IF(N,NE,203) GO TO 203

~~WAIRB=VALUE(15,CYCLE(21000,0,0,0,0),0,0)~~

R(6)= 0.0105\*WAIRB

R(10)= 0.263 \*WAIRB

R(11)= 0.72 \*WAIRB

R(2)=60.0

R(12)= 0.0065\*WAIRB

203 CONTINUE

Component 201, Solids InputSolids input into blender from  
Table 14.

C

C

~~COMPONENT 201--SOLIDS INPUT~~

IF(N,NE,201) GO TO 201

R(2)=98.6

~~R(16)=VALUE(14,CYCLE(21000,0,0,0,0),0,0)~~

201 CONTINUE

C

Component 202, BlenderThe power level of the blender when in use  
is 372

C

C

C

BLENDER--COMPONENT 202 POWER CALCULATION

R(65) IS SET EQUAL TO BLENDER ELECTRICAL POWER

IF(N,NE,202)GO TO 202

R(65)=0.0

~~IF(A(1),0,0,0)R(65)=372.0~~

202 CONTINUE

C

Table 3.4

## GPOLY LOGIC FOR SOLID WASTE LOOP

(continued)

Component 208, Trash Supply

Trash input is determined from Input Data  
Table 13.

```

C      COMPONENT 208 TRASH SUPPLY
C
C      IF(N,NE,205)GO TO 208
C      R(16)=VALUE(13,CYCLE(21000,0,0.0),0.0)
C      R(2)=80.0
C      208 CONTINUE
C

```

Component 207, Shredder

If shredder on the power level is 977 watts.

```

C
C      SHREDDER -- COMPONENT 207 ELECTRICAL POWER
C
C      IF(N,NE,207)GO TO 207
C      R(65)=0.0
C      IF(A(1),GT,0.0,JP,R(1),GT,0.0)R(65)=977.0
C      207 CONTINUE
C

```

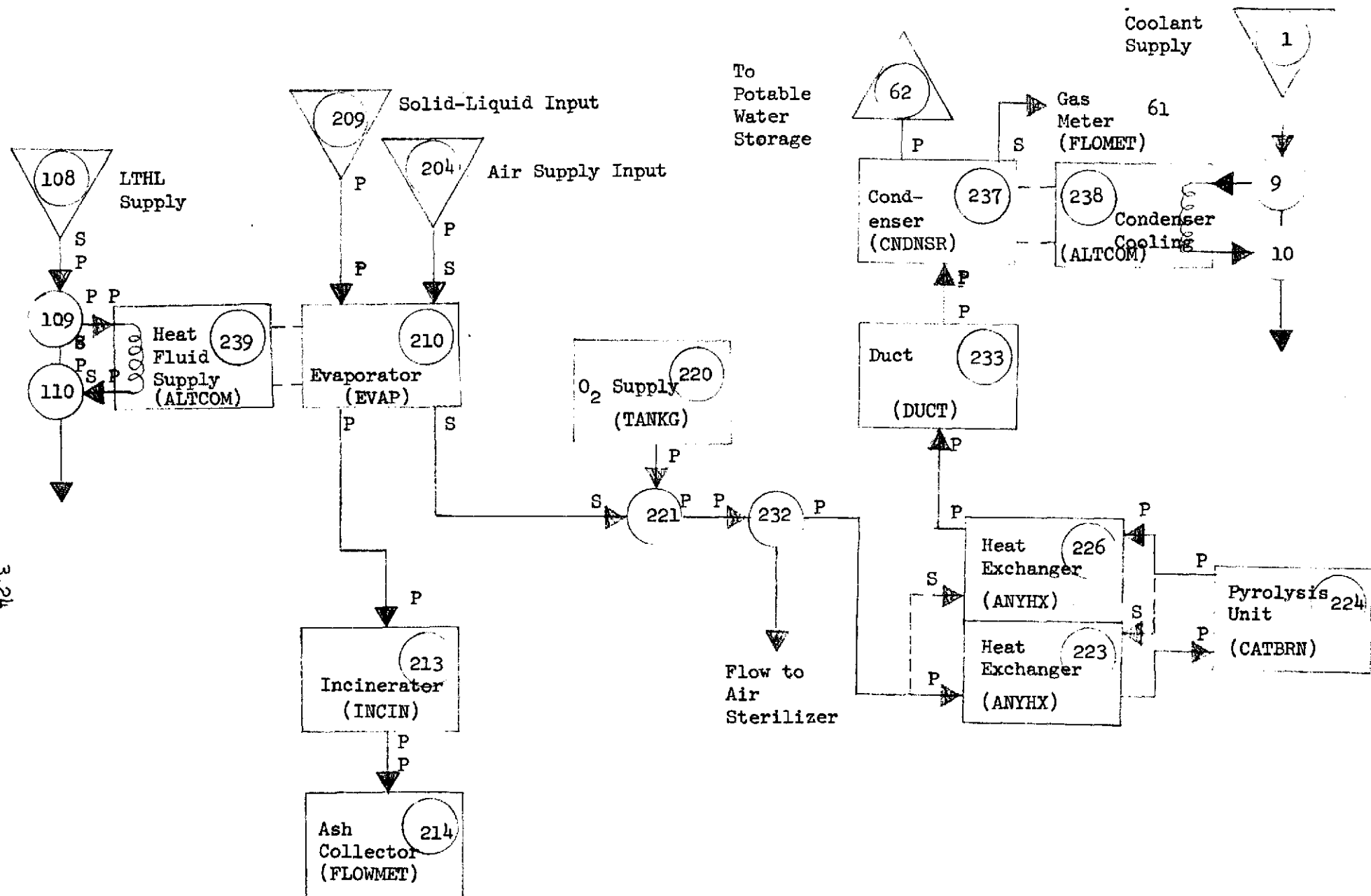


FIGURE 3.10 G-189 COMPONENT AND FLOW SCHEMATIC FOR THE VAPOR LOOP

Table 3.5

## GPOLY CONTROL LOGIC FOR COMPUTING/IMPLEMENTING STABLE TIME INCREMENT (VAPOR LOOP)

Component #210 EvaporatorGPOLY Logic

At the evaporator, the system time increment is saved, DELTAT. If NCOUNT > 0, then we do not compute vapor loop time increment. If NCOUNT=0, compute time increment, and number of passes per system pass required for the vapor loop.

TIME is incremented to maintain proper timing of system.

```

IF(STEADY) GO TO 210
***** INTERNAL TIME INCREMENT ***** (VAPOR LOOP)
SET SMALL TIME INCREMENT FOR OUTER LOOP
DELTAT=60.
IF ( NCOUNT ,GT, 0 ) GO TO 2103
TIMING=.1*VV(237,91)/R(20)*3600.
MAXCNT =IFIX( DTIME / TIMING ) +1
TIMING = DTIME/ FLDTAT(MAXCNT )
DTIME = TIMING
WRITE(6,666) DTIME
666 FORMAT(10,'THE VALUE OF DTIME IS EQUAL TO * ,F9.6 ///
2103 CONTINUE
NCOUNT = NCOUNT + 1
IF(NCOUNT ,LE, 1) TIME=TIME-DELTAT
TIME=TIME+TIMING
210 CONTINUE

```

Component #237 - CondenserGPOLY Logic

During transient saturation of the condenser a test for number of passes through vapor loop is made. If value is less than required the NEXT component to be solved is set to #210. Otherwise NCOUNT is reset to zero and system time step is reset to the original value.

```

IF(N ,NE, 237) GO TO 237
IF(STEADY) GO TO 237
C
C ***** INTERNAL TIME INCREMENT ***** (VAPOR LOOP)
IF( NCOUNT ,GT, MAXCNT ) GO TO 2375
NEXT = 210
GO TO 2373
2375 CONTINUE
DTIME = DELTAT
NCOUNT = 0
2373 CONTINUE
R(1)=R(69)
237 CONTINUE

```



Table 3.5

GPOLY CONTROL LOGIC FOR COMPUTING/IMPLEMENTING STABLE TIME INCREMENT (VAPOR LOOP)  
(Continued)

Component #100 - Isotope HeaterGPOLY Logic

To provide for correct time values and not to impede the system initialization process, the evaporator is made the last component once system starts executing.

C

IF (H, HE. 100) GO TO 100  
NLAST=200

## Component 210 - Evaporator

A subroutine capable of simulating the evaporator in the General Electric RITE system was not available in the G-189A program library. Therefore a new routine, EVAP, was written. The subroutine documentation, which is contained in Appendix A, includes schematics, mathematical modelling, and operational procedures.

The EVAP subroutine flow chart is shown in Figure 3.11. The subroutine performs mass and thermal balances for either steady state or transient operation. EVAP includes the capability of separating (filtering) and storing wet solids in a reservoir so as to allow their subsequent removal from the system by a solids pump. In addition the subroutine simulates biochemical reactions that occurs within the evaporator. Due to the complex nature of the mass balance in the evaporator a mass balance schematic is included (Figure 3.12) which is helpful in reading the computer output.

The General Electric Summary Report, Reference 1, modified and reproduced as Table 5-2 of this report was used as the reference for predicting the chemical reactions. The weight flowrate of gases was not given; therefore a prediction was made that the vent flowrates would be according to the following assumptions:

1. Flow rates of product gases out of the condenser were assumed to be in the same weight proportion as reported in the G.E. vent analysis. Chemical reactions in the system were predicted based on these resultant vent gas constituents.
2. Two principal non-condensable gases generated in the evaporator are  $\text{CO}_2$  and  $\text{NH}_3$ . They are generated from the biochemical decomposition of urea and from ammonium carbonate/bicarbonate during vaporization of the liquid.
3. Half of the urea is removed from the evaporator by the solids filter while the other half biochemically decomposes to yield  $\text{NH}_3$  and  $\text{CO}_2$ .
4. The amounts of  $\text{CO}_2$ ,  $\text{H}_2$ ,  $\text{O}_2$ ,  $\text{NO}_2$ ,  $\text{NO}$  and  $\text{N}_2$  vented are based on the amount of  $\text{NH}_3$  generated.

The GPOLY logic which was implemented to simulate the chemical reactions in the pyrolysis unit and to predict the amount of  $\text{O}_2$  required to meet the vent requirements is presented in Table 3.6.

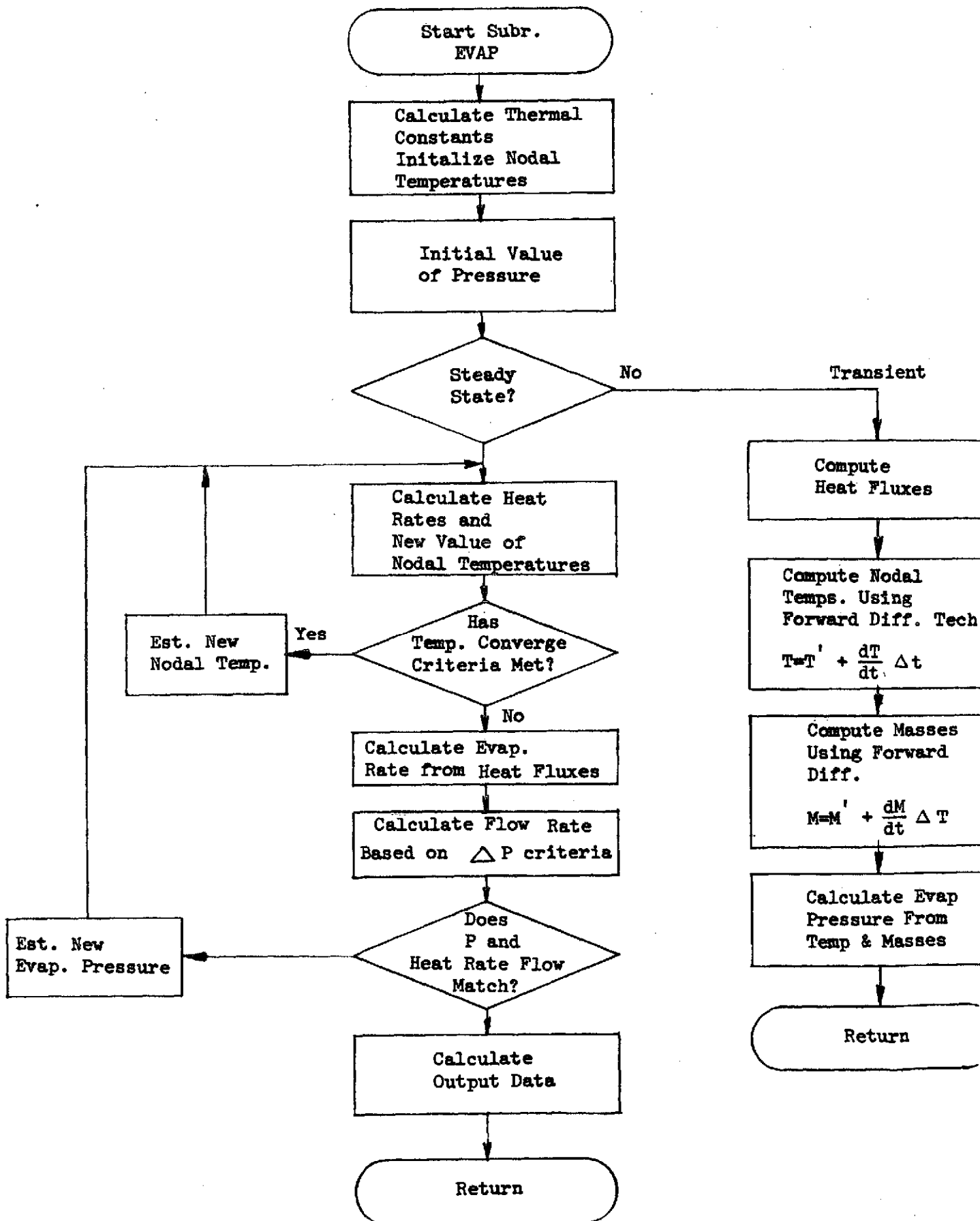


FIGURE 3.11 EVAP SUBROUTINE FLOW CHART

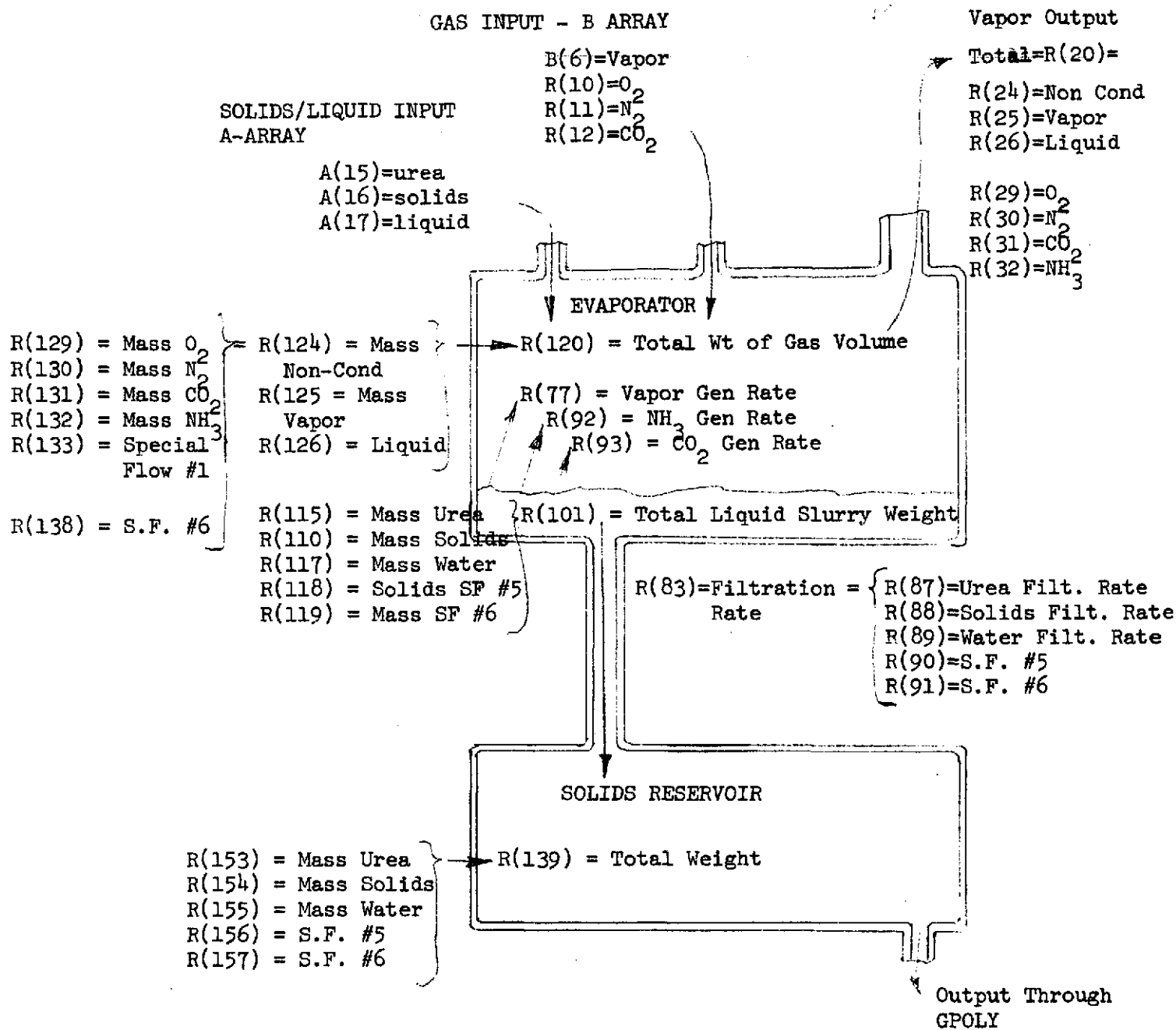


FIGURE 3.12 MASS BALANCE IN THE EVAPORATOR

TABLE 3.6 GPOLY LOGIC TO ATTAIN MASS BALANCE IN VAPOR LOOP

Order of Solution - From Solids Loop - 210, 221, 232, 223, 224, 226, 233, 237, 213, 214, to Potable Water Storage Loop

Component #221-TEEGPOLY1 Logic

The primary flow which defines the  $O_2$  flow injected into gas stream is defined as

$$w_{O_2} = .7651 * w_{NH_3}$$

where  $w_{NH_3}$  is  $NH_3$  flow.

Remaining code updates the storage tanks, weight and pressure.

~~IF (TIME - 221) GO TO 221~~

~~A(1)=B(13)\*.7651~~

~~A(5)=A(1)~~

~~A(10)=A(1)~~

~~A(2)=70.~~

~~W02TNK=VV(221,69)~~

~~A(8)=.22~~

~~A(9)=32.~~

~~W02TNK=W02TNK-A(1)\*DTIME~~

~~A(3)=W02TNK\*1545./32\*530./144.~~

~~A(4)=A(3)~~

~~CALL SV(W02TNK,220,69)~~

~~CALL SV(W02TNK,220,77)~~

~~CALL SV(W02TNK,220,78)~~

~~CALL SV(W02TNK,220,60)~~

~~CALL SV(A(3),220,3)~~

~~CALL SV(A(3),220,4)~~

~~CALL SV(A(3),220,70)~~

221 CONTINUE

Component #224 - Pyrolysis UnitGPOLY1 Logic

$H_2$  and  $N_2$  is formed from disassociation  $NH_3$ .  $NH_3$  flow is set to zero.

In the Bosch reaction the amount of carbon formed is function of  $CO_2$  generated. The reaction coefficient .98491 was derived from vent analysis presented in Reference 1.

C PYROLYZE  $NH_3 = 1/2 N_2 + 3/2 H_2$

~~A(14)=A(13)\*1.5\*(WTH(14)/WTMTC)~~

~~A(11)=A(13)\*0.5\*(WTH(11)/WTMTC)~~

~~A(13)=0.~~

C

~~C SIMULATE BOSCH REACTION  $2H_2 + CO_2 = C + 2H_2O$~~

C

~~CRRNDR=A(12)\*12./44.\*.98491~~

~~A(14)=A(14)+.98491\*A(12)\*4./44.~~

~~A(12)=.015809\*A(12)~~

224 CONTINUE

Table 3.6

## GPOLY LOGIC TO ATTAIN MASS BALANCE IN VAPOR LOOP

(continued)

Component #244-Pyrolysis UnitGPOLY2 Logic

Test to see if there is sufficient  $N_2$  to generate  $NO_x$  gases. The value of the reaction coefficient<sup>x</sup> was derived from vent gas analysis.

```

C      IF(N, NE. 224) GO TO 224
C      PREDICT AMOUNT NOX GENERATE IN THE PYRO
C      SPECIAL FLOW 2 (A(15))=NO2
G      SPECIAL FLOW 3 (A(16))=NO
      IF(A(11) .LT. 0.06*A(10))GO TO 2241
      A(15)=.00294 *A(10)*46./32.
      IF(R(11) .LT. 0.06*R(10)) GO TO 2241
      R(15)=.0032*R(10)*46./32.
      R(16)=.0032*R(10)*30./16.
      R(11)=R(11)-R(15)*44./46.-R(16)*14./30.
      R(10)=R(10)-R(15)*32./46.-R(16)*16./30.

2241      CONTINUE
224      CONTINUE

```

#### Component 220 Tee-Oxygen Supply

Component 220 supplies the oxygen required to generate the mass flows defined in Section 5. The oxygen is assumed to be available from a tank and the weight of the oxygen in the tank is continually updated.

#### Component 224 Pyrolysis Units

The mass balances for the pyrolysis reactions are implemented in the pyrolysis unit GPOLY1 and GPOLY2 logic. In GPOLY1 the disassociation of ammonia and the Bosch reaction are simulated. In GPOLY2 the oxidation of nitrogen to form NO and NO<sub>2</sub> are simulated. The logic is described in detail in Table 3.6.

#### Component 237 Condenser

A new subroutine, CNDNR, was written to simulate the RITE system condenser. The subroutine documentation includes a system schematic, mass and heat balance modeling and a description of the operation of the subroutine. The CNDNSR subroutine flow chart is shown in Figure 3.13.

The subroutine is capable of simulating condensing of water vapor entering the condenser, and venting of gases to vacuum. The condensate is removed by GPOLY logic. Due to the complex nature of the routine a mass balance schematic is included in Figure 3.14 to help identify key output parameters.

#### Component 213 Incinerator

The solid products removed by the solids filter are transferred to the incinerator for disposition. The waste products are processed by vacuum drying, chemical decomposition at high temperature, and venting of resultant gases. The incinerator is loaded in a batch process. The vent to vacuum is opened and the solids are dried. The vent to vacuum is closed and a

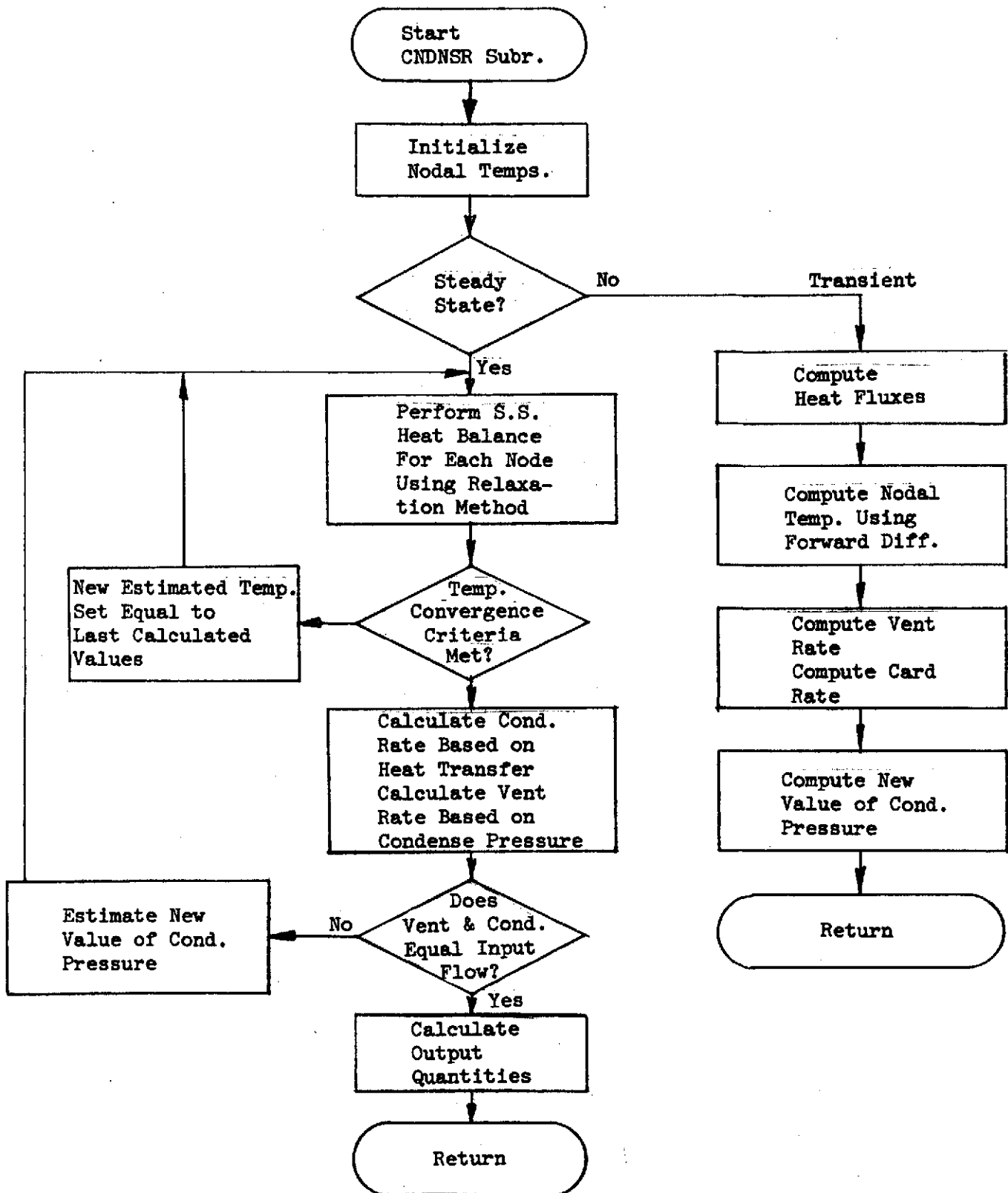
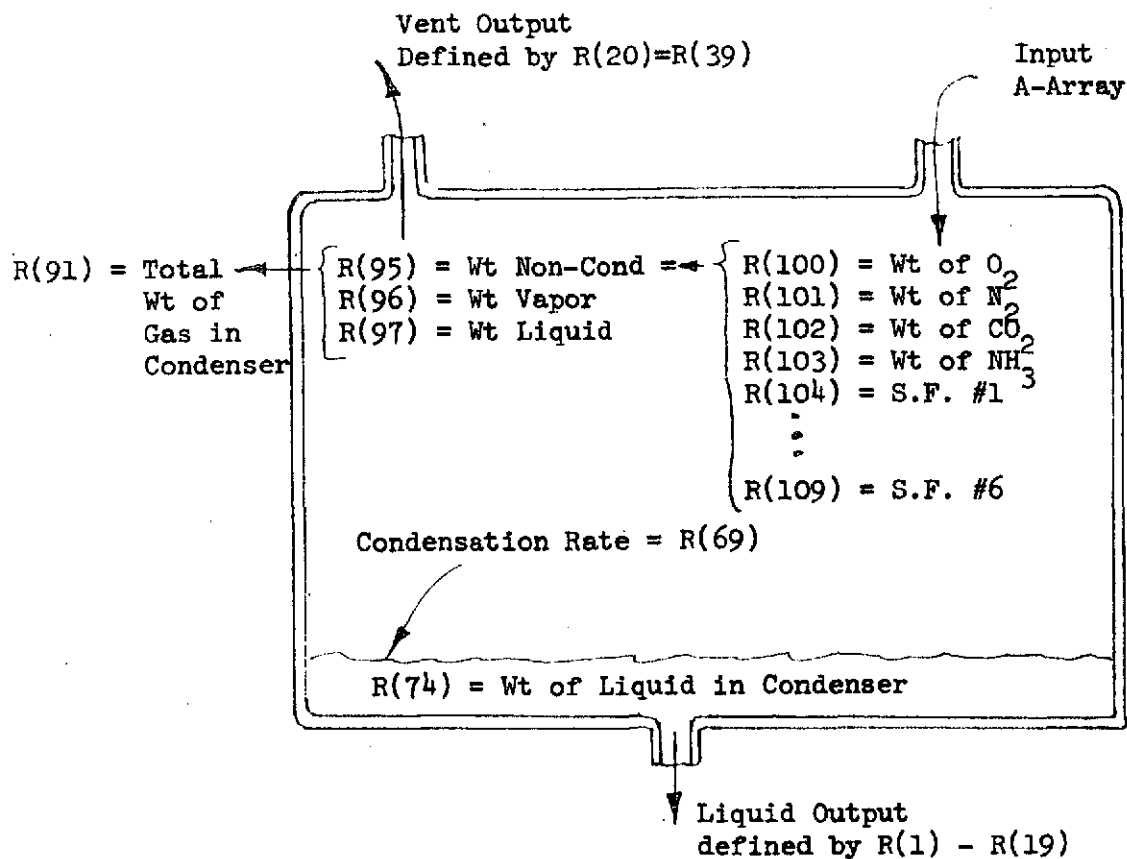
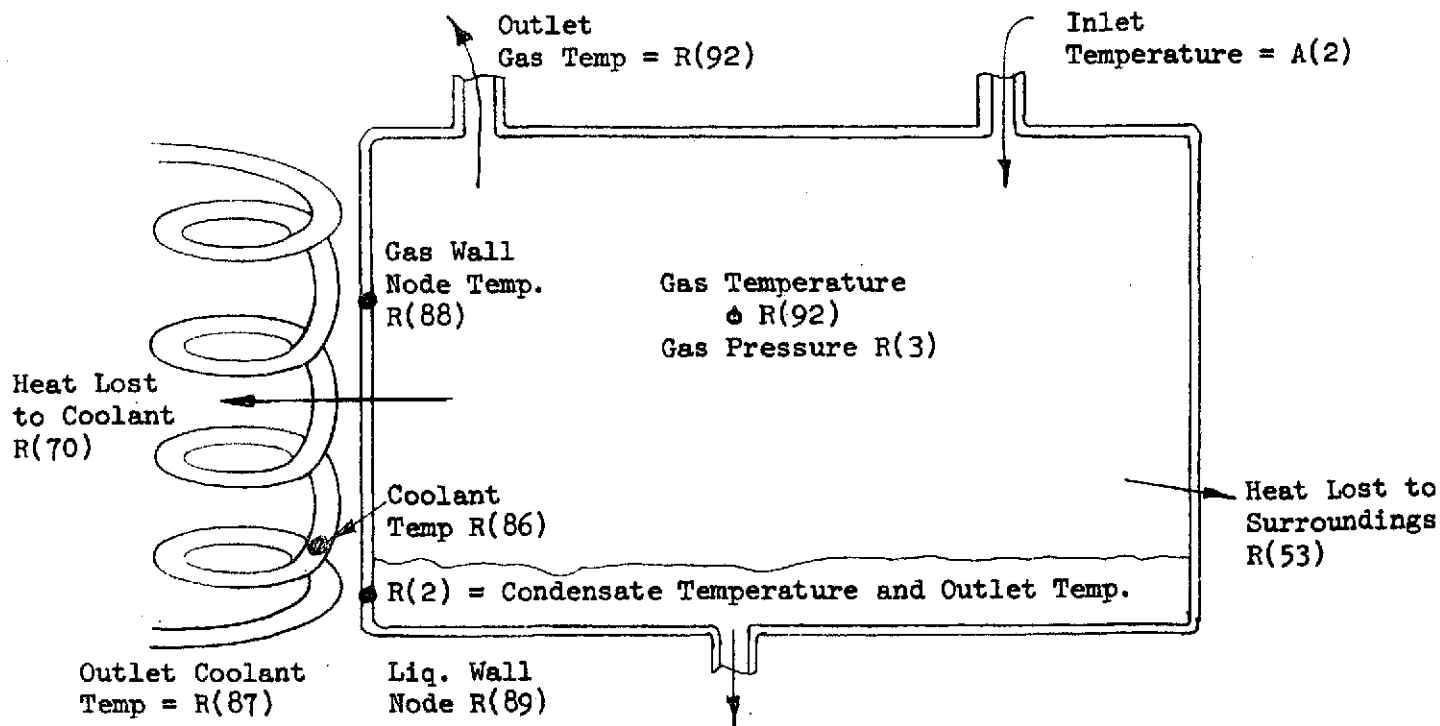


FIGURE 3.13 CNDNSR SUBROUTINE FLOW CHART





CONDENSER MASS BALANCE



CONDENSER ENERGY BALANCE

FIGURE 3.14 CONDENSER SUBROUTINE G-189 OUTPUT REPRESENTATION

small amount of oxygen is introduced with the solids and the oxidation reaction proceeds. The product gases are then vented to vacuum. The oxygen cycle is repeated until all the solids are incinerated. At the end of the cycle, the remaining ash is blown out by a nitrogen purge. Approximately 0.5 pounds of solids are processed per batch. The cycle duration is approximately three hours.

A G-189A library subroutine for the incinerator was not available. Therefore a new subroutine INCIN was written. The subroutine documentation, which includes a complete description of the cycle including mass and thermal balances is provided in Appendix A.

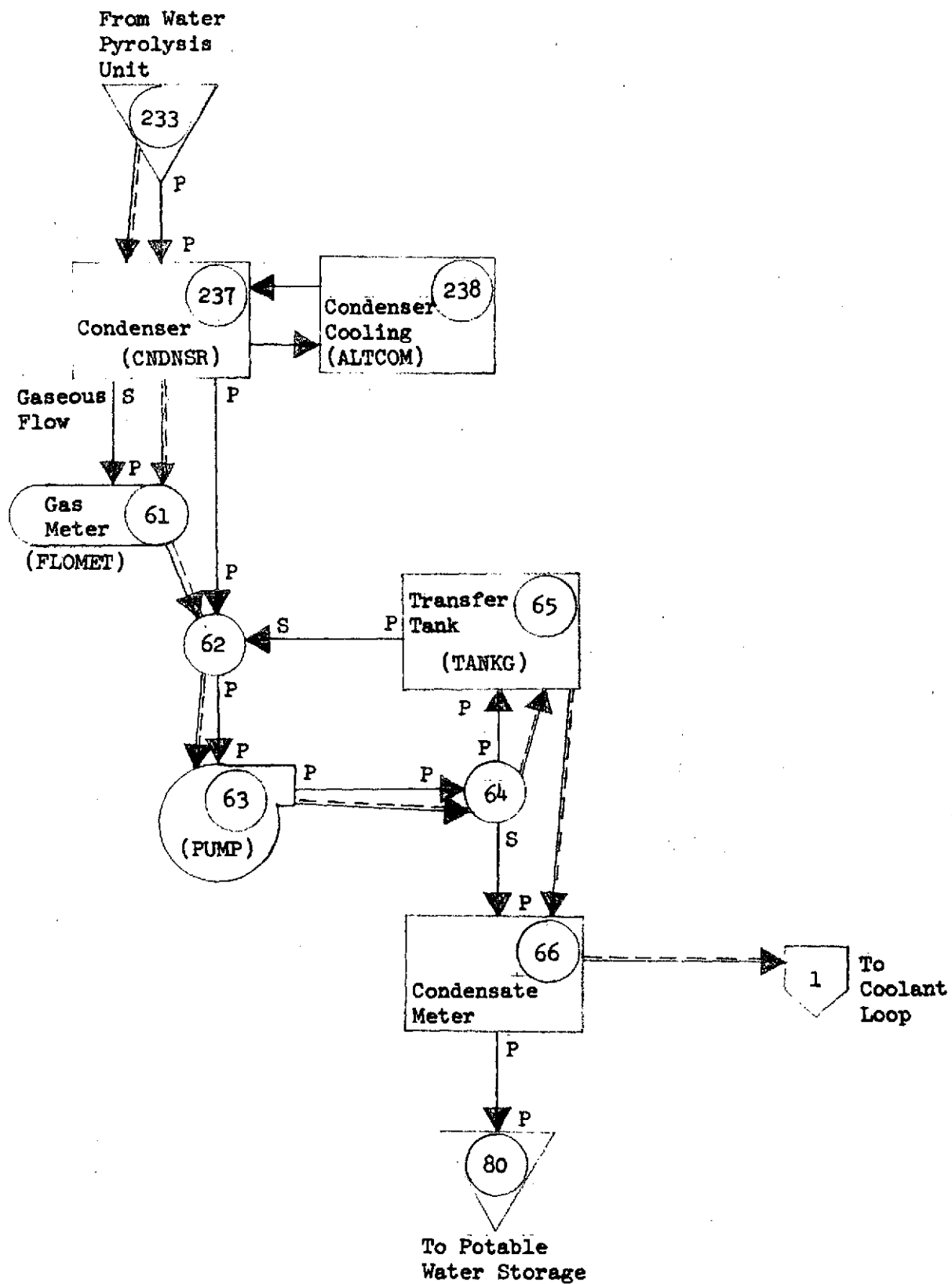
The subroutine requires a heat source temperature for waste processing. The corresponding heat block nodal temperature is entered into the subroutine through GPOLY1. The GPOLY1 logic is noted in Table 3.6.

#### 3.2.6 Potable Water Storage Loop

The G189A schematic for the potable water storage loop is shown in Figures 3.15 and 3.16. The potable water system collects condensate water from the condenser and stores it in four potable water storage tanks. A fifth tank, initially full, is available for emergency use, however this tank cannot be filled from the condenser.

When the condenser reaches the 15 percent level, the transfer pump, component 63, draws half of the condenser condensate and stores it in an intermediate transfer tank, component 65. The condenser outlet is then closed and the same pump is then used to transfer the water in the transfer tank to the potable water tanks.

The tanks are drained upon demand as specified in Table 16 (see Appendix B); the cold water demand, and Table 17, hot water demand. Currently the tanks are drained in order of 4, 3, 2, 1. The GPOLY logic for the potable water storage loop is shown in Table 3.7.



3.15 G-189A SCHEMATIC OF THE CONDENSATE COLLECTION LOOP

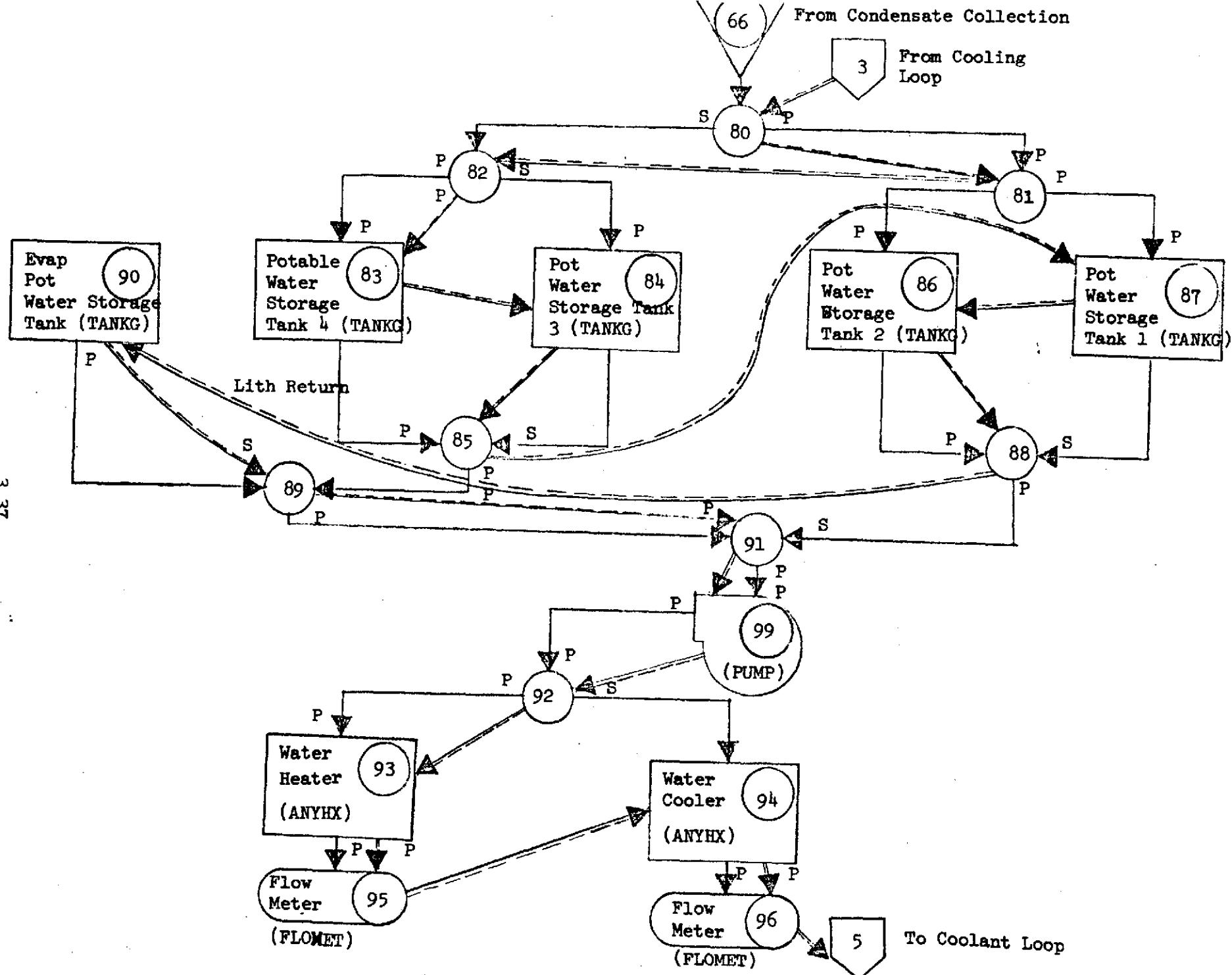


FIGURE 3.16 G-189A SCHEMATIC OF THE POTABLE WATER LOOP

Table 3.7

## GPOLY CONTROL LOGIC FOR POTABLE WATER STORAGE LOOP

Order of Solution - Start from component 3 from cooling loop. 80, 81, 82, 83, 84, 85, 87, 86, 88, 90, 89, 91, 99, 92, 93, 95, 94, 96. Return to cooling loop.

Filling of Potable Water Storage TankComponent 80 Flow split to tanks

GPOLY1

If tanks 83 and 84 are full  
bypass flow to component 81.

If not full flow to component 82

Component 81 Flow split to tanks  
1 and 2

GPOLY1

3.38 Flow to tank 86 if not full  
otherwise flow to tank 87

Component 82 Flow split to  
tanks 3 and 4

GPOLY1

If tank 83 full bypass flow to  
tank 84

Draining of Potable Water Storage TanksComponent 83 - Tank No. 4

GPOLY2

Outlet flow is the sum of cold water flow  
(tube 16) and hot water flow  
(tube 17). If there is enough  
water in tank 83 drain that tank.

```

C   GPOLY1 LOGIC FOR POTABLE WATER STORAGE LOOP
C
C   THE POTABLE WATER TANKS WILL BE FILLED IN THIS ORDER 83,84,87
C
C   COMPONENT SPLIT 80 == SPLIT TO TWO TANK GROUPS
C
C   IF(N,NE,80)GO TO 80
C   R(65)=1.0
C   A(1)=100.0
C   A(2)=100.0
C   IF(VV(83,69).GT.41.50-A(1)*DTIME/3600.0.AND.VV(84,69).GT.41.5
C   1)*DTIME/3600.0)R(65)=0.0
C   80 CONTINUE
C
C
C   COMPONENT SPLIT 81=SPLIT TO TANKS 86 AND 87
C
C   IF(N,NE,81)GO TO 81
C   R(65)=0.0
C   IF(VV(86,69).LT.41.50-A(1)*DTIME/3600.0)R(65)=1.0
C   81 CONTINUE
C
C
C   COMPONENT 82 SPLIT TO TANKS 83 AND 84
C
C   IF(N,NE,82)GO TO 82
C   R(65)=0.0
C   IF(VV(83,69).GT.41.50-A(1)*DTIME/3600.0)R(65)=1.0
C   82 CONTINUE
C
C   GPOLY2 LOGIC FOR POTABLE WATER STORAGE LOOP
C
C   COMPONENT 83 TANK NO. 1
C
C   IF(N,NE,83)GO TO 83
C   R(1)=0
C   WTOT=VALUE(16,TIME,0.0)+VALUE(17,TIME,0.0)

```

Table 3.7 GPOLY CONTROL LOGIC FOR POTABLE WATER STORAGE LOOP (Continued)

Component 84 - Tank No. 3

GPOLY2

If tank 83 full, drain tank 84  
if there is adequate water.

Component 87 - Tank No. 1

GPOLY2

If tank 83 and 84 drained, use  
tank 87 if there is adequate  
water supply.

Component 86 - Tank No. 2

GPOLY2

If previous tanks empty, drain  
tank 86 if there is adequate water  
supply

Component 90 - Emergency Tank

GPOLY2

If no other tanks available use  
emergency tank.

```

C
C COMPONENT 84 TANK NO. 2
C
  IF(N,NE:84)GO TO 34
  R(1)=0,0
  WTOT=VALUE(16,TIME,0,0)*VALUE(17,TIME,0,0)
  IF((VV(83,1),LE,0,0),AND,(WTOT,GT,0,0),AND,(R(69),GT,WTOT*DTIME/
13600,0)) R(1)=WTOT
  84 CONTINUE

C
C COMPONENT 87 TANK NO. 3
C
  IF(N,NE:87)GO TO 97
  R(1)=0,0
  WTOT=VALUE(16,TIME,0,0)*VALUE(17,TIME,0,0)
  IF((VV(85,1),LE,0,0),AND,(WTOT,GT,0,0),AND,(R(69),GT,WTOT*DTIME/
13600,0)) R(1)=WTOT
  87 CONTINUE

C
C COMPONENT 86 TANK NO. 4
C
  IF(N,NE:86)GO TO 86
  R(1)=0,0
  WTOT=VALUE(16,TIME,0,0)*VALUE(17,TIME,0,0)
  IF((WTOT,GT,0,0),AND,(VV(85,1),LE,0,0),AND,(VV(87,1),LE,0,0),AND,
1(R(69),GT,WTOT*DTIME/3600,0)) R(1)=WTOT
  86 CONTINUE

C
C COMPONENT 90 EMERGENCY TANK
C
  IF(N,NE:90) GO TO 90
  R(1)=0,0
  WTOT=VALUE(16,TIME,0,0)*VALUE(17,TIME,0,0)
  IF(WTOT,GT,0,0,AND,VV(85,1),LE,0,0,AND,VV(88,1),LE,0,0)R(1)=WTOT
  90 CONTINUE
  RETURN
  END

```

### 3.2.7 Cooling Loop

The G189A schematic of the cooling loop is shown in Figure 3.17. The loop provides cooling liquid for the condenser and the water cooler. A constant supply of ethylene glycol coolant (280 lb/hr, 350°F) is provided by component 1. Ninety-five percent of the fluid is directed to the cooler in order to kill any growth in bacteria which may have formed in the water cooler while it was operating at the lower temperatures. The switch in coolant direction is directed by data in Table 11 (See Appendix B), the water cooler supply requirements table. When the table indicates a hot water requirement the GPOLY logic directs the cooling flow to the secondary side of component 2. Alternately, the heating fluid is directed to the water cooler, which then serves to supply heated water to component 94. The GPOLY logic required to accomplish the change in roles is described in Table 3.8.

### 3.2.8 High Temperature Heat Source

The high temperature heat source consists of a heat block and a heat pipe for emergency heat rejection. The heat block provides heat for the pyrolysis units, incinerator and the air sterilizer. The inert gas controllable heat pipe is imbedded in the heat block to provide emergency cooling when the temperature reaches 1300°F. The heat pipe has the capability of rejecting 100 to 800 Btu/hr within the temperature range of 1300 to 1380°F. The component definition diagram for the high temperature heat source is shown in Figure 3.18.

#### Component 180 Heat Block

The heat block serves to distribute thermal energy from the radioisotope to the various components including the heat pipe, pyrolysis units, incinerator, and air sterilizer. The heat block assembly, simulated by subroutine THERML, is comprised of three principal components, the radioisotope heater, the heat block, and the insulation jacket. The radioisotope heater located in the middle of the heat block generates the thermal

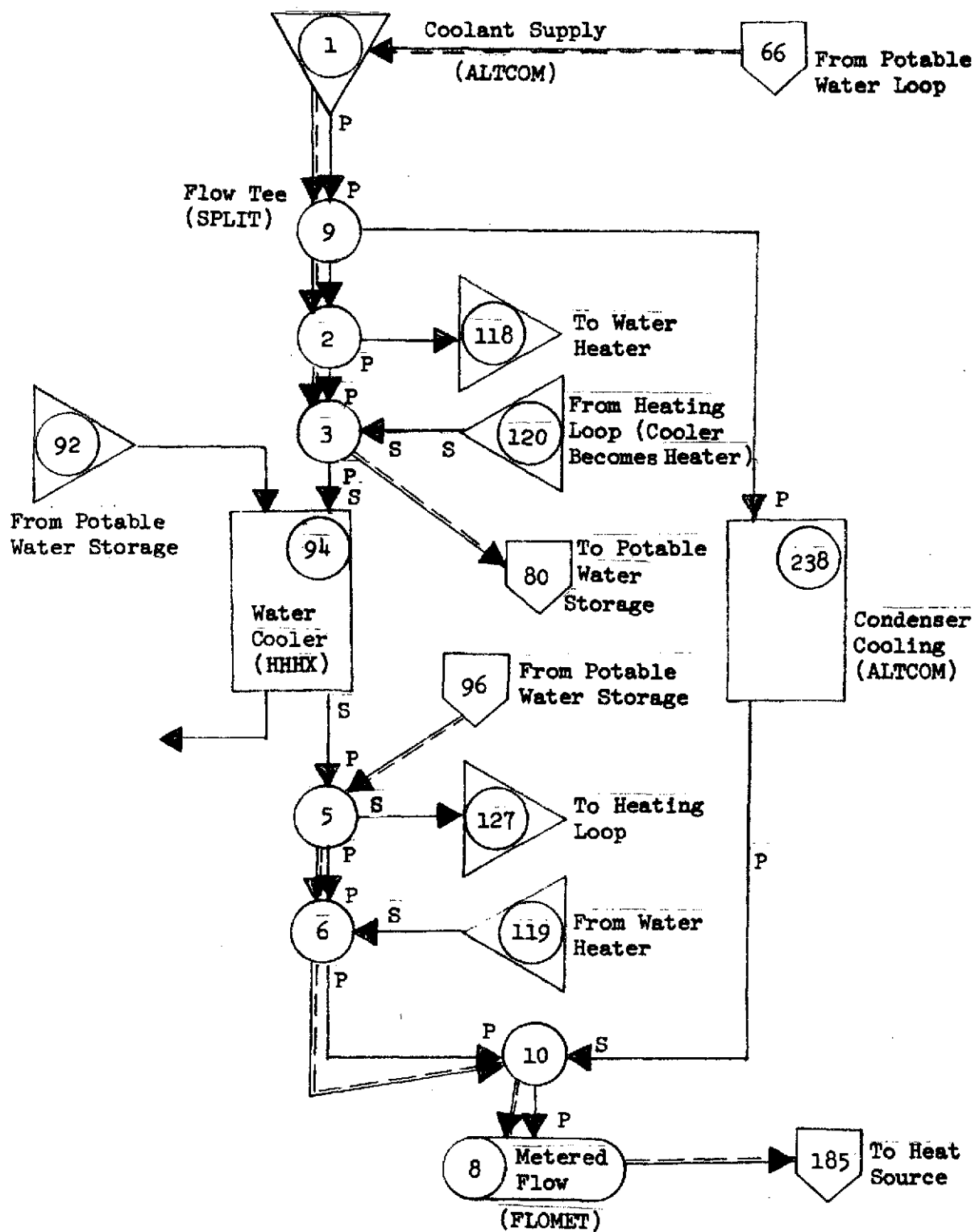


Figure 3.17 G-189 Schematic of the Cooling Loop



Table 3.8

## GPOLY LOGIC FOR THE COOLING LOOP

Order of Solution: 1, 9, 2, 3, to Potable Water Storage, from Potable Water Storage to 5, 6, 10, 8

Component 2, Tee to Water Heater

Subroutine: SPLIT

GPOLY1

Table 11 is read to determine quantity and type of water flow for component 94.

If chilled water is required from component 94 the coolant flow is directed to that component. If warm water is required from component 94, coolant flow is bypassed to component 118 and then to component 93, the other heat exchanger which now serves as the water cooler. When component 94 serves as the water heater, heating fluid flow is directed to component 3, and to the heat exchanger from the low temperature heating loop at component 120.

The heating fluid is returned to the low temperature heating loop by diverting flow at component 5 to the secondary side. This leads to the heating loop.

```

C  GPOLY1 LOGIC FOR COOLING LOOP
C
C  COMPONENT 2 FLOW SPLIT TO WATER HEATER
IF(N,NE,2)GO TO 2
IF(WCOOL.GE.0.0)GO TO 2000
R(65)=0.0
IF(WCOOL.GT.0.0) GO TO 2000
R(65)=1.0
CALL SV(1,0,5,65)
GO TO 2

```

```

2000 CALL SV(0,0,5,65)
2 CONTINUE

```

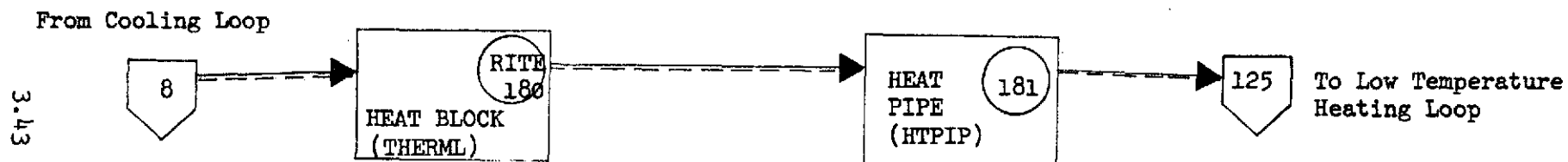


FIGURE 3.18 G-189A SCHEMATIC OF THE HIGH TEMPERATURE HEAT SOURCE LOOP

energy required for the incineration, air sterilization, and pyrolysis processes. The insulation jacket, a double walled vessel with the evacuated space between the walls containing high temperature multifoil insulation, contains the heat block and other purifier components and serves the function of minimizing heat loss from the purifier assembly to the environment. The heat block in the purifier assembly provides heat conduction to the purifier components and also serves as the positioning and holding fixture for the components within the insulation jacket.

A schematic of the heat block, showing the relative locations of the components is shown in Figure 3.19. The corresponding nodal network is shown in Figure 3.20.

Heat transfer from the heat block is calculated for the individual components. The heat loss values are then entered into the nodal network prior to calculation of the resultant nodal temperatures. The GPOLY logic required to perform this operation is shown in Table 3.9.

#### Component 181 Heat Pipe

An inert gas controllable heat pipe is used as an emergency heat rejection system for the high temperature isotope for a wide range of heat loads. The evaporator section of the heat pipe is imbedded into the heat block. If the temperature of the heat block reaches to a predetermined value, corresponding to the phase change pressure of the sodium vapor working fluid, heat is absorbed, and a phase change from liquid to vapor takes place. Due to the rise in temperature, pressure increases; this causes vapor to flow to the condenser region of the pipe where the process is reversed and heat is rejected. Very small temperature drops are coupled with the transfer of large quantities of heat in the phase change from liquid to vapor to liquid. This property provides the capability for transporting large quantities of thermal energy in a near-isothermal system. Following condensation, the cycle is completed with the return of the liquid to the vaporator region.

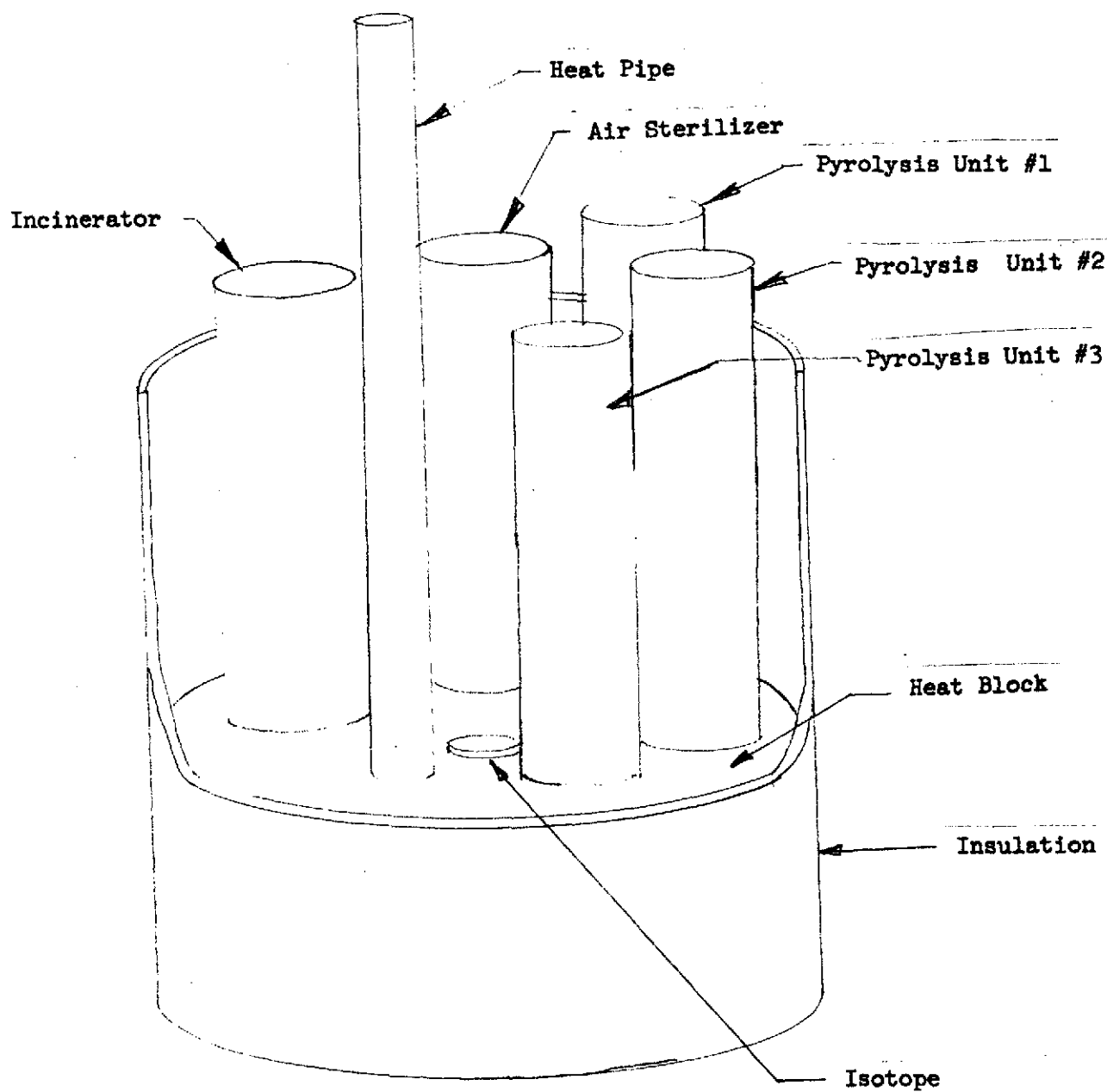


FIGURE 3.19 HEAT BLOCK MODEL

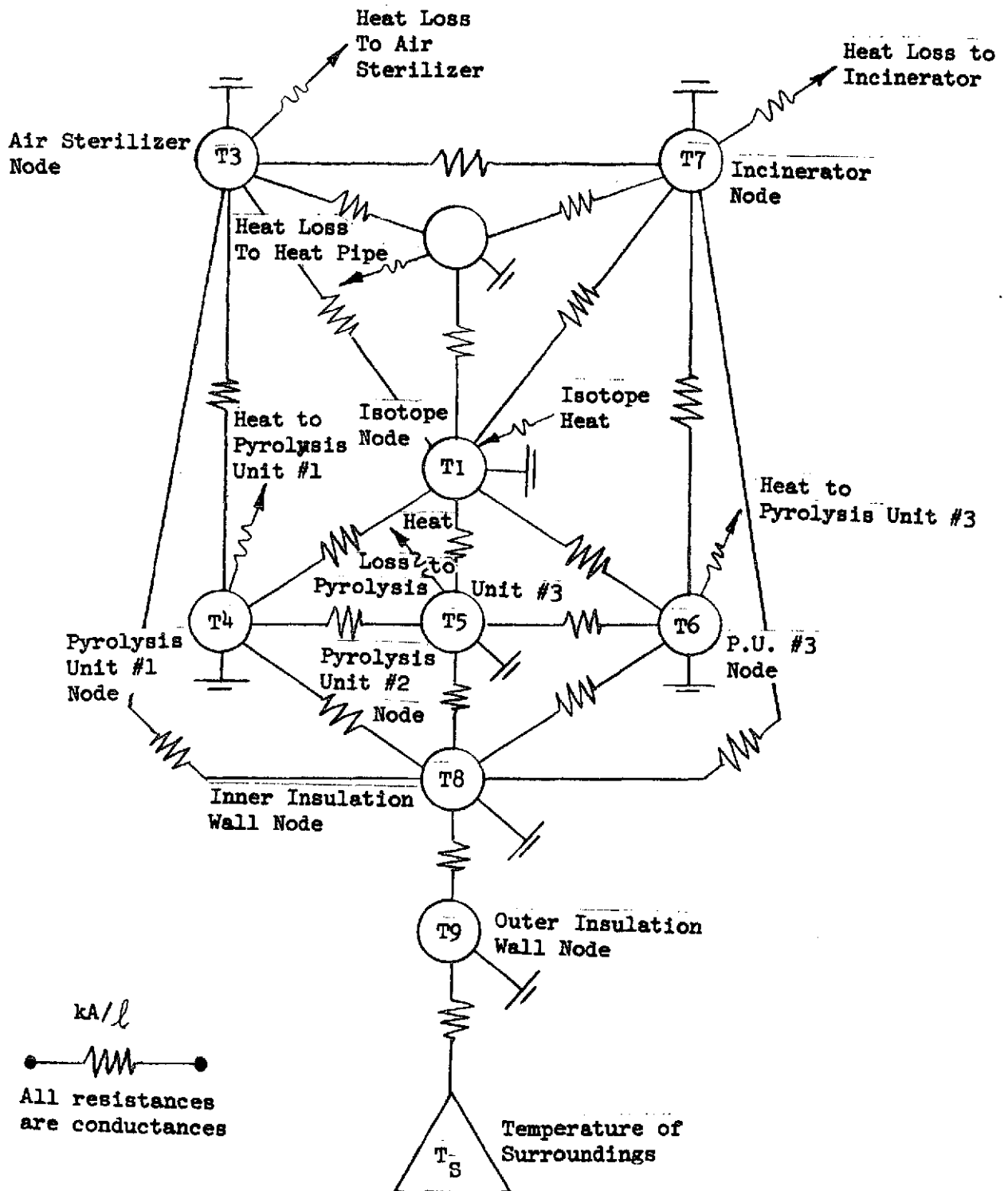


FIGURE 3.20 HEAT BLOCK NODAL NETWORK

Table 3.9

## GPOLY LOGIC FOR THE HIGH TEMPERATURE HEAT SOURCE

Order of Solution - From Cooling Loop, 180, 181, to Low Temperature Heat loop

Component 180 Heat Block

GPOLY1

Enter into the heat block thermal network all  
heat loss terms to the heat pipe components

R(88) is heat loss to heat pipe

R(90), R(91), R(92) are the heat loss terms  
to the pyrolysis units

R(93) is the heat loss term to the incinerator

```

IF(IN, NE, 180) GO TO 180
C      SET HEAT BLOCK INTERFACES
C      HEAT LOSS TO HEAT PIPE
      R(88)=VV(181,95)
C      HEAT LOSS TO AIR STERILIZER
      R(89)=VV(25,93)*( VV(25,116)-VV(25,80) )=(VV(25,100)
      -VV(25,117)))=(VV(25,107)-(VV(25,70)-VV(25,117)))
C      HEAT LOSS TO PYROLYSIS UNIT NO1
      R(90)=72.0
C      HEAT LOSS TO PYROLYSIS UNIT NO2
      R(91)=R(90)
C      HEAT LOSS TO PYROLYSIS UNIT NO3
      R(92)=R(90)
C      HEAT LOSS TO INCINERATOR
      R(93)=VV(213,66)
180 CONTINUE

```

Component 181 Heat Pipe

GPOLY1

The heat pipe interface temperature is deter-  
mined from the heat block.

```

C
C
C      IF(IN, NE, 181) GO TO 181
C      SET INTERFACE TEMP HEAT BLOCK TO HEAT PIPE
      R(66)= VV(180,70)
181 CONTINUE
C

```

A G189A library heat pipe subroutine was not available, therefore, a new subroutine was written. The subroutine documentation, which includes mass and thermal balances and a discussion of subroutine operation is included in Appendix A.

Successful heat pipe subroutine operation requires that a heat source temperature, corresponding to a heat block node temperature, be made available to it. The GPOLY logic for this operation is noted in Table 3.9.

## Section 4

### CORRELATION OF EXPERIMENTAL AND ANALYTICAL DATA

The procedure used in correlating experimental data from tests of individual RITE system components with G189A program computed data from corresponding analytical models generally consisted of the following steps:

1. General Electric test data in Reference 1, provided in letter form, or provided informally were reviewed.
2. Component descriptive data and drawings in Reference 1 were reviewed.
3. Component heat transfer parameters were computed from the data in items 1. and 2. For example, heat exchanger overall conductances, UA, and effectivenesses were computed from experimental temperature profiles and component thermal masses were computed from dimensional, density, and specific heat data.
4. Component parameters and boundary conditions were specified as input data and computer runs were obtained.
5. Discrepancies between experimental data and computed data were reconciled in terms of modifications to the component parameter data.
6. These modifications were made and additional computer runs were obtained. This iterative process was continued until satisfactory correlation was achieved.

Discussions of the correlations for individual components and the correlative data for these components now follows.



#### 4.1 Pyrolysis Units

The pyrolysis unit consists of a regenerative heat exchanger and a catalytic bed containing ruthenium catalyst. The unit was simulated using the ANYHX and CATBRN subroutines. The output data correlation was based on steady state heat transfer tests performed by General Electric on one pyrolysis unit. Thermocouples were attached at five locations as designated by C13, C14, B4, B5, and B6 on Figure 4.1. Figure 4.1 shows the resulting temperatures at these 5 locations for both the test and the RITE simulation. As seen from Figure 4.1 the ANYHX and CATBRN subroutines provide inlet and outlet temperatures which can be directly compared with the test data. The vapor flow as 0.807 lb/hr and the inlet temperature was 200°F.

The vapor temperature increased to 1180°F in the regenerative heat exchanger. The temperature further increased to 1240°F in the catalyst bed. The heat added in the catalyst bed was provided by the heat block. The computed outlet temperature was somewhat lower than the experimental data. This discrepancy is in the direction expected since the thermocouple was attached on the outside of the outlet pipe rather than being installed in the gas stream.

#### 4.2 Air Sterilizer

A transient heat transfer test was performed by General Electric on the air sterilizer. The normal commode use has been assumed to require 5 minutes per event. To simulate a worst case, the sterilizer was operated for 10 minutes. This represents two successive uses. The transient test data for the thermocouples located in the sterilization chamber and at the inlets and outlets for the counterflow regenerative heat exchanger are shown on Figures 4.2 and 4.3. The test was started with no flow and the sterilization chamber at 1280°F. A constant flow of 17 Std Cfm was then imposed for the 10 minute period.

The air sterilizer was simulated using the thermal analyzer subroutine THERML. Figure 4.4 shows the thermal model for the air sterilizer. Figures 4.2 and 4.3 also show the computed temperature data corresponding to the experimental data. Thermocouples 1 and 2 on Figure 4.2 are considered to be attached to the air sterilizer structure in the vicinity of gas nodes 10 and 9, respectively.

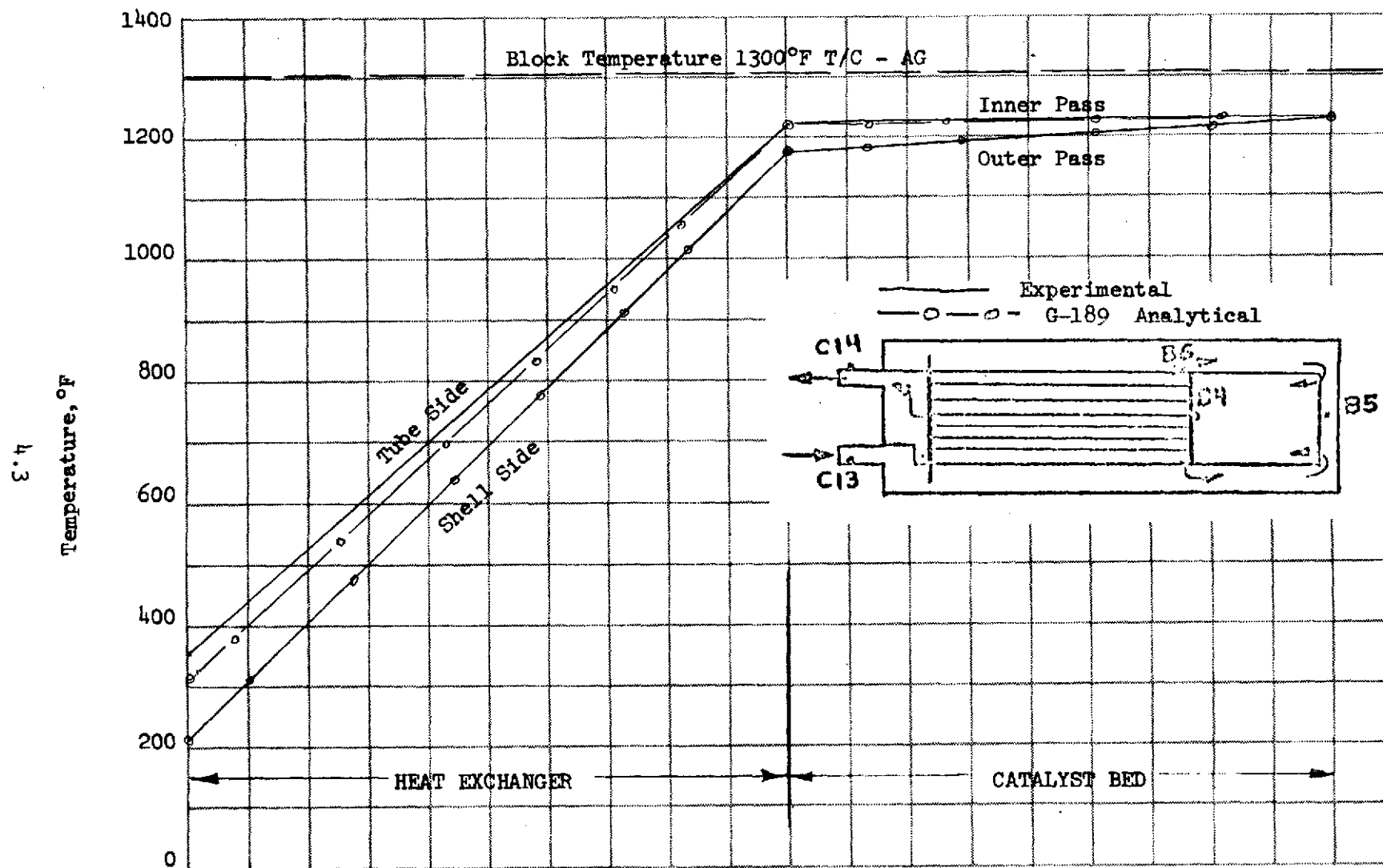


FIGURE 4.1 PYROLYSIS UNIT TEMPERATURE PROFILE

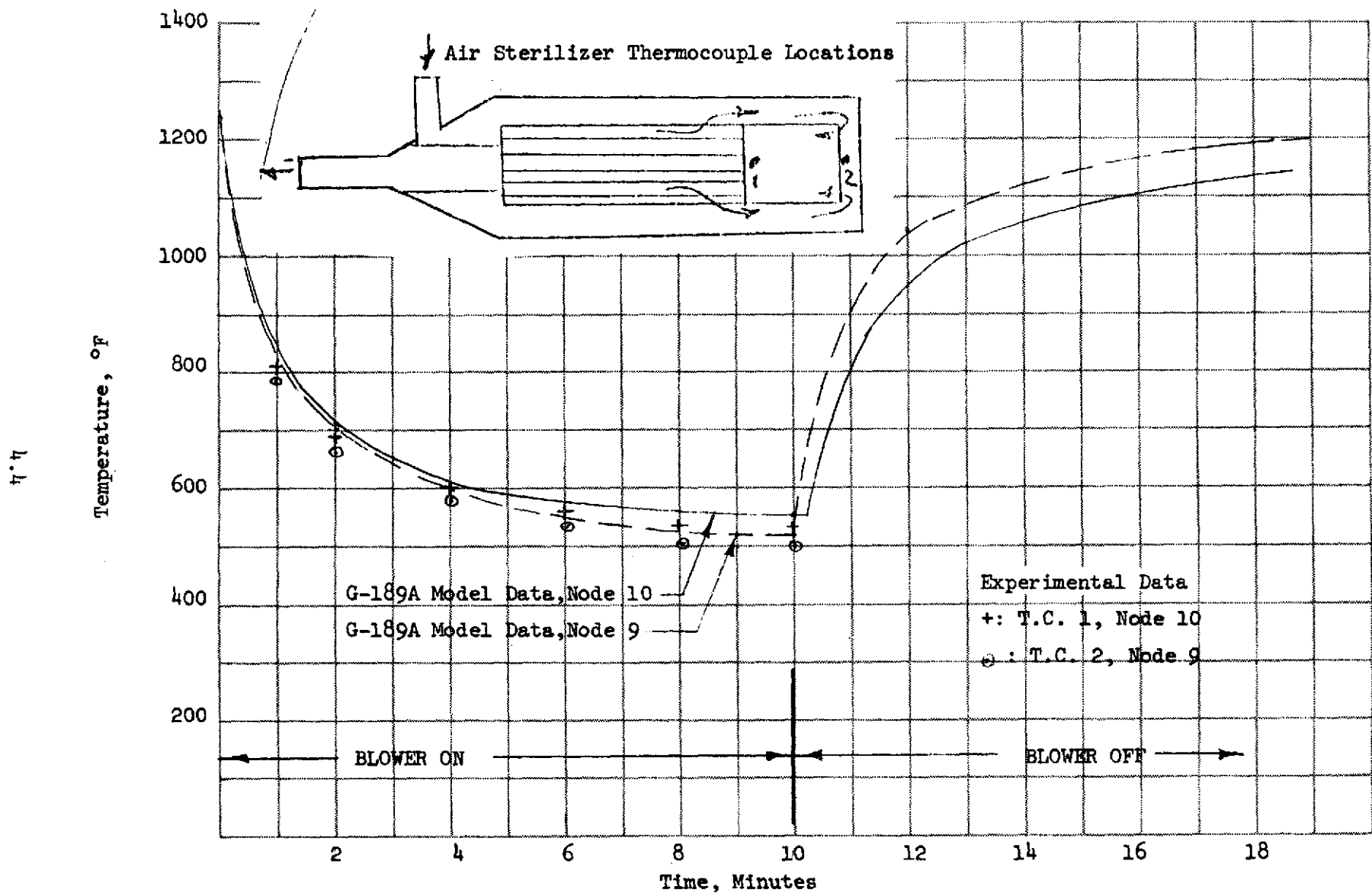


FIGURE 4.2 TIME-TEMPERATURE DATA, G-189 MODEL/EXPERIMENTAL FOR THE AIR STERILIZER

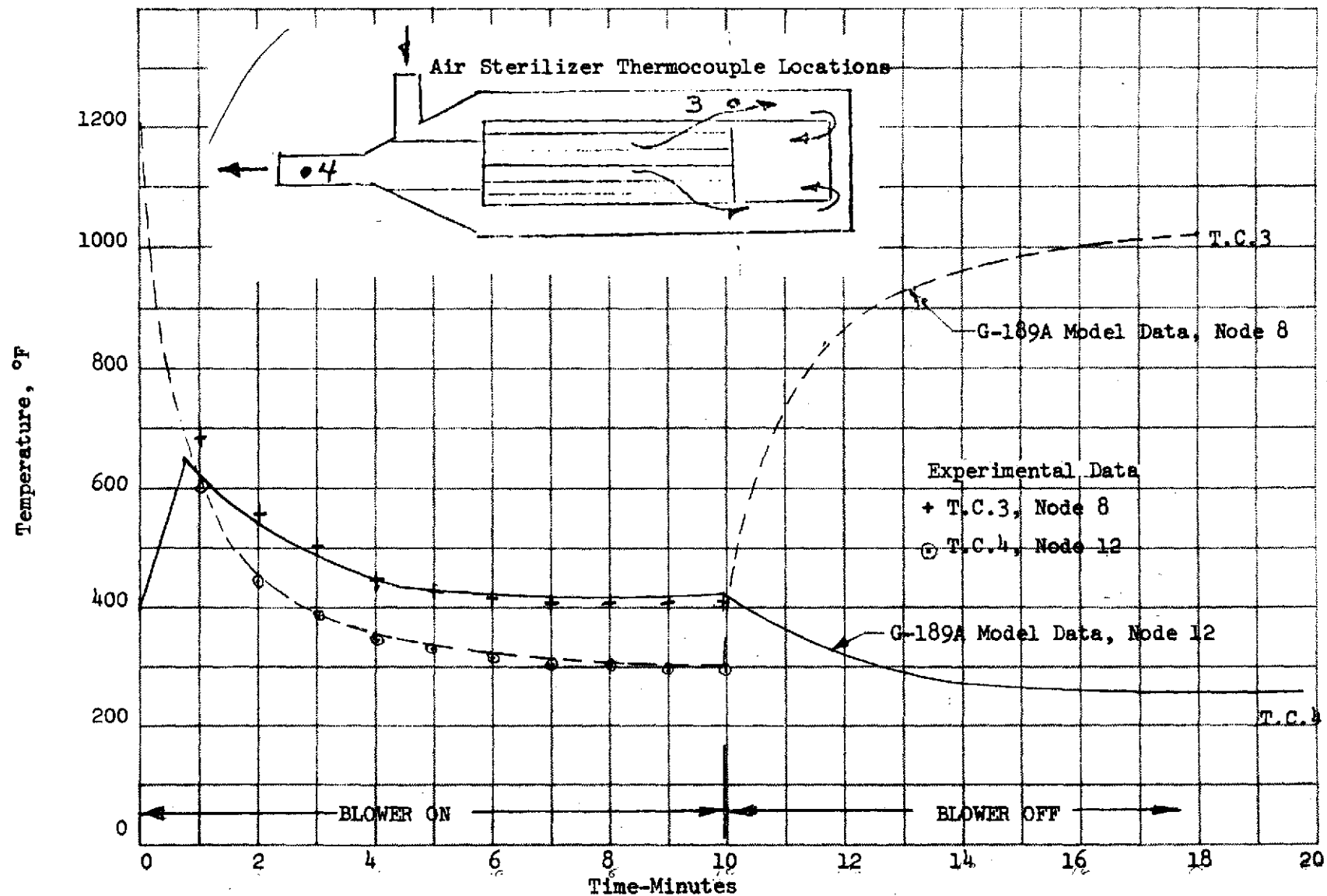


FIGURE 4.3 TIME-TEMPERATURE CURVE G-189A MODEL/EXPERIMENTAL

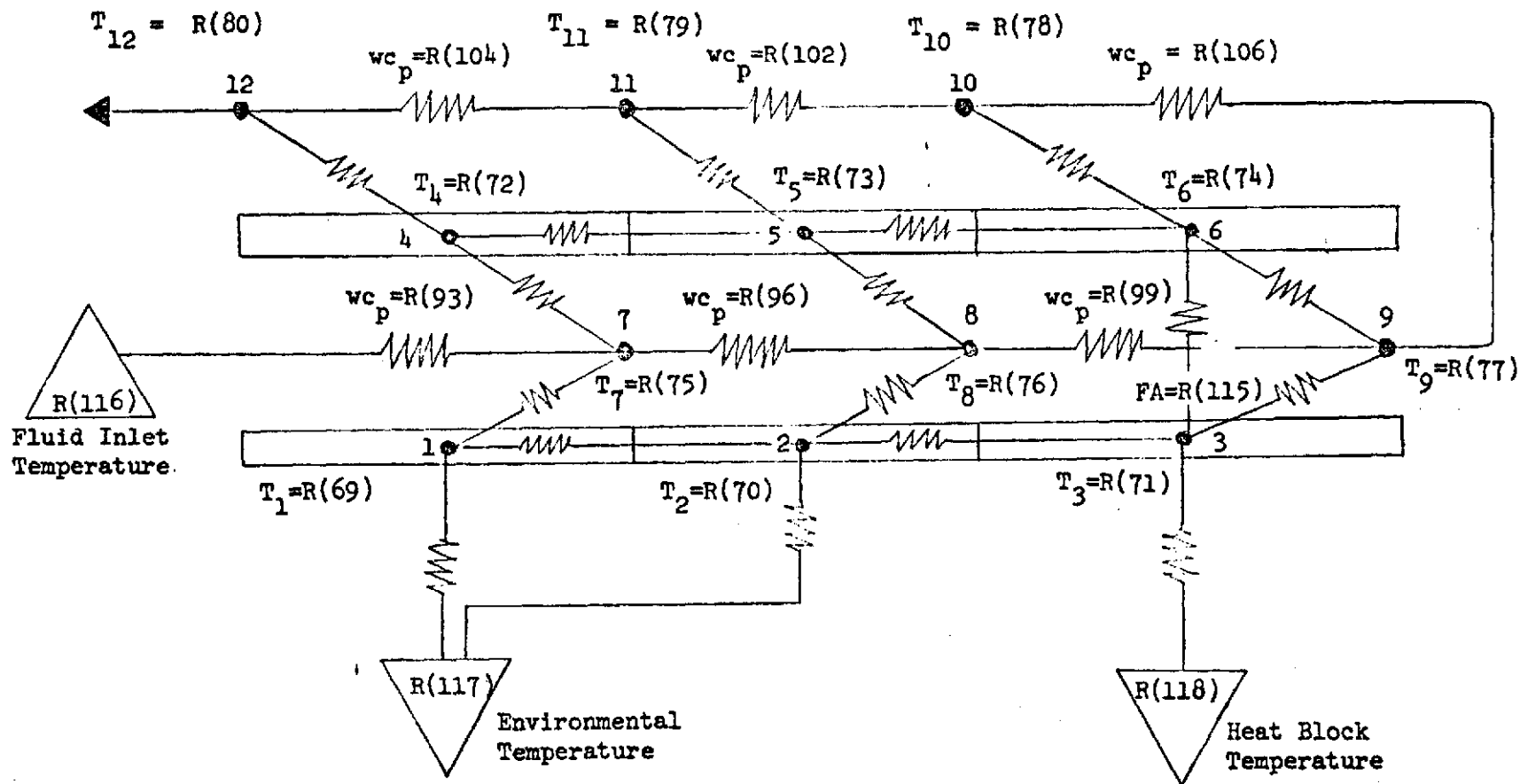


FIGURE 4.4 AIR STERILIZER THERMAL MODEL

Similarly thermocouples 3 and 4 on Figure 4.3 are related to gas nodes 8 and 12 respectively. The gas nodes should be cooler than the nearby thermocouples when the blower is on since heat is being transferred to the gas. This is seen to be the case on Figure 4.2 and 4.3. The correlation for the case when the blower is on is considered to be satisfactory in terms of the agreement between experimental and computed temperatures and also in terms of the amount of energy removed from the heat block.

When the blower is off the amount of heat removed from the heat block is far less than when the blower is on. With the blower off heat is only conducted axially in the air sterilizer outer shell away from the heat block. The heat is then transferred from this outer shell through an insulation blanket to the local environment. This heat is far less than the amount removed by blowing air through the air sterilizer.

When the blower is off it is seen from Figure 4.2 and 4.3 the temperatures at thermocouples 1, 2, and 3 rise rather rapidly due to their close proximity to the heat block. Thermocouple 4 shows the cooling effect of the ambient environment.

The gas nodes in the thermal model do not have representative temperatures for the nearby thermocouples when the blower is off. During these conditions gas nodes 7, 8, and 9 go to steady state at the average temperature for the two metal nodes which each are thermally connected to. Gas nodes 10, 11, and 12 go to the temperature of the surface that each is connected to. Thermal gradients in the metal nodes in this thermal model are quite severe due to the large difference in boundary heating/cooling conditions within

the heat block and due to ambient conditions. Therefore to accurately predict individual temperatures, at individual thermocouples with the blower off would require many more metal nodes than used here and would require more detailed information concerning the construction of the air sterilizer and the locations of the thermocouples than has been provided.

The important point is that with the blower on a good correlation has been achieved and that with the blower off the air sterilizer does not contribute significantly to the energy balance for the heat block and the detailed modelling required to obtain good temperature correlations is not justified.

#### 4.3 Heat Pipe

Steady state performance data for the heat pipe have been provided by General Electric. The two data points relating heat rejection rate and heat pipe temperature are shown on Figure 4.5. Thermophysical property data were specified for the heat pipe subroutine (subroutine HTPIP) used to simulate this component. These included dimensional data, heat pipe axial thermal conductance, condensible fluid and inert gas thermodynamic data, and external thermal conductance coefficients for the evaporator and condenser sections of the heat pipe. The heat pipe axial thermal conductance and external conductance coefficients were varied so as to achieve the correlation with test data shown in Figure 4.5. The heat pipe data is included in the RITE simulation Input Data of Appendix B.

Figure 4.5 clearly illustrates the three operational modes of the heat pipe. The control range of the heat pipe encompasses a temperature range of 1303 to 1380°F. In this range the heat rejection rate varies between 100 BTU/hr and 790 Btu/hr. Below 1303°F the heat pipe behaves like a fin and heat is rejected by conduction along the pipe and natural convection to the air. At temperatures greater than 1380°F, the condenser section of the heat pipe is at a maximum area and increased heat rejection rate is possible only by increasing the total heat pipe temperature.

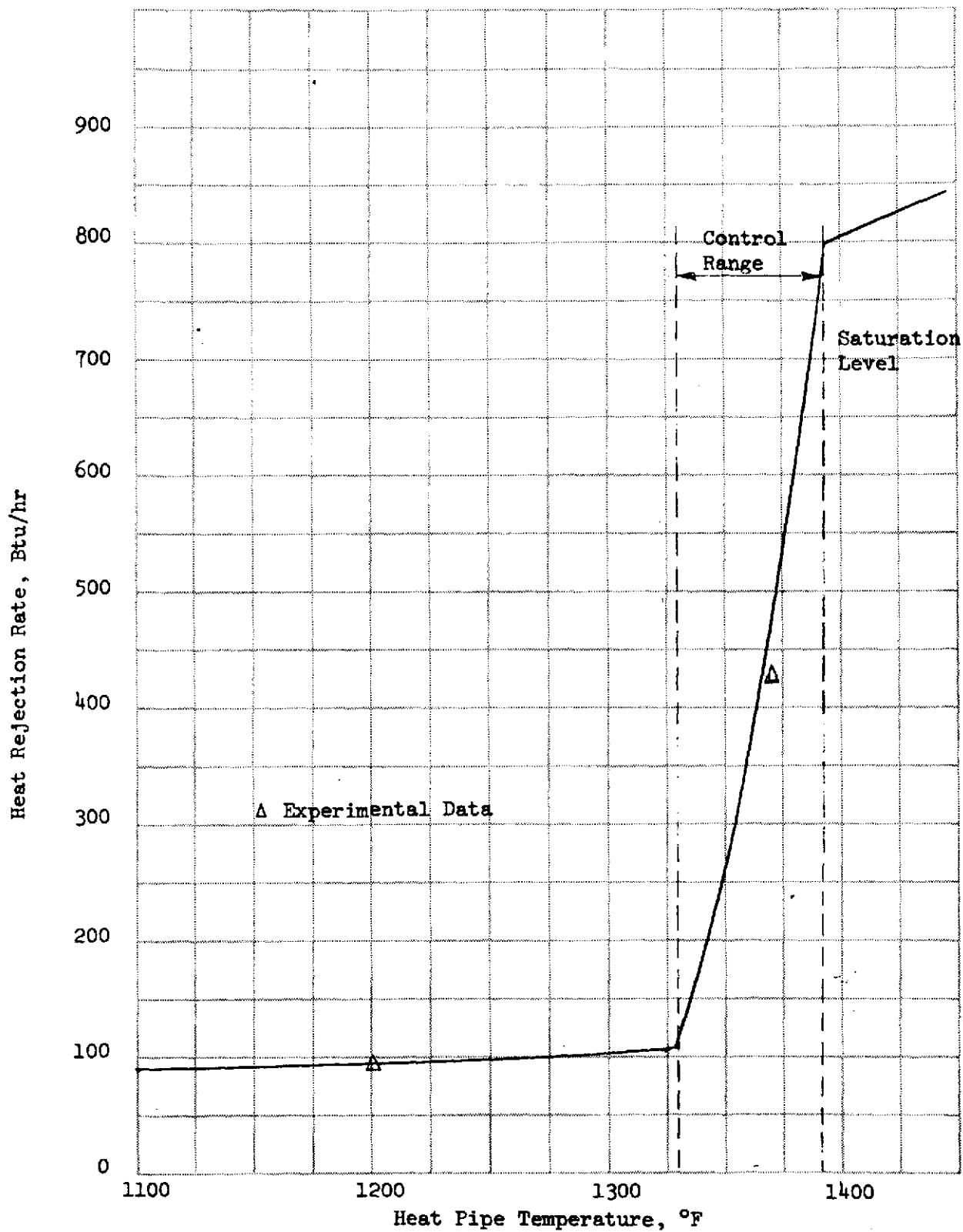


FIGURE 4.5 HEAT REJECTION RATE VS. SOURCE TEMPERATURE



#### 4.4 Heat Block

The heat block was simulated using the thermal analyzer subroutine THERML. Transient heating curves for the heat block were obtained by General Electric. (Reference 1) Different input power levels to the electrical heater mounted in the heat block were supplied. Figure 4.6 shows experimental data and the computed data obtained from the G189A program model for the heat block for a heating condition. Temperature variations within the heat block are quite small; indicative of good heat transfer by conduction within the heat block. Figure 4.7 shows the nodal network for the heat block. The thermal mass for the heat block was basically distributed at nodes which interface with the components which are installed in the heat block: pyrolysis units, air sterilizer, heater, incinerator, and heat pipe. The thermal conductances which interconnect these nodes have large values. This is in keeping with the good conduction heat paths provided in the heat block. Each temperature profile shown on Figure 4.6 is therefore essentially the temperature throughout the heat block in the vicinities of the installed components. The computed temperature profile was obtained for 420 watts heater power rather than for the experimental power levels since that is essentially the power level that is currently used. Because of the consistency of the experimental and computed data, it was considered that the correlation was adequately demonstrated without obtaining computed data for either or both of the experimental power levels.

#### 4.5 Low Temperature Isotope Heater, Evaporator, and Condenser

Experimental data used in the correlation studies for these components were obtained during the 180 day test of the complete RITE system. As such the data are representative of normal operating conditions for the system. The computed data from the G189A program models for the components were generated by setting boundary value data and then obtaining steady state computer runs for the individual components. In the case of the evaporator and condenser the two components were coupled for these computer runs by the flow of steam from the evaporator to the pyrolysis units and then to the condenser.

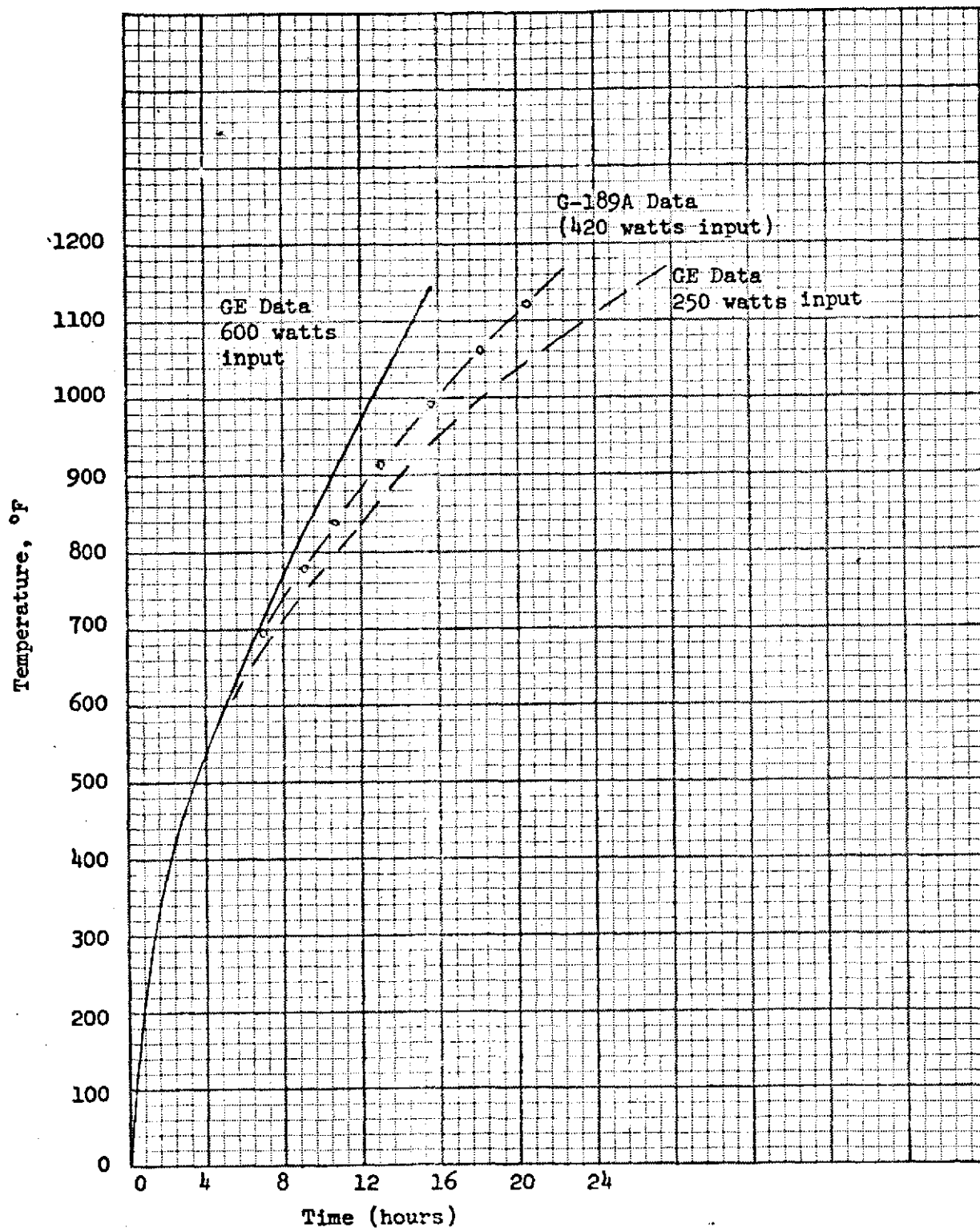


FIGURE 4.6 HEAT BLOCK HEATING PROFILE CORRELATION

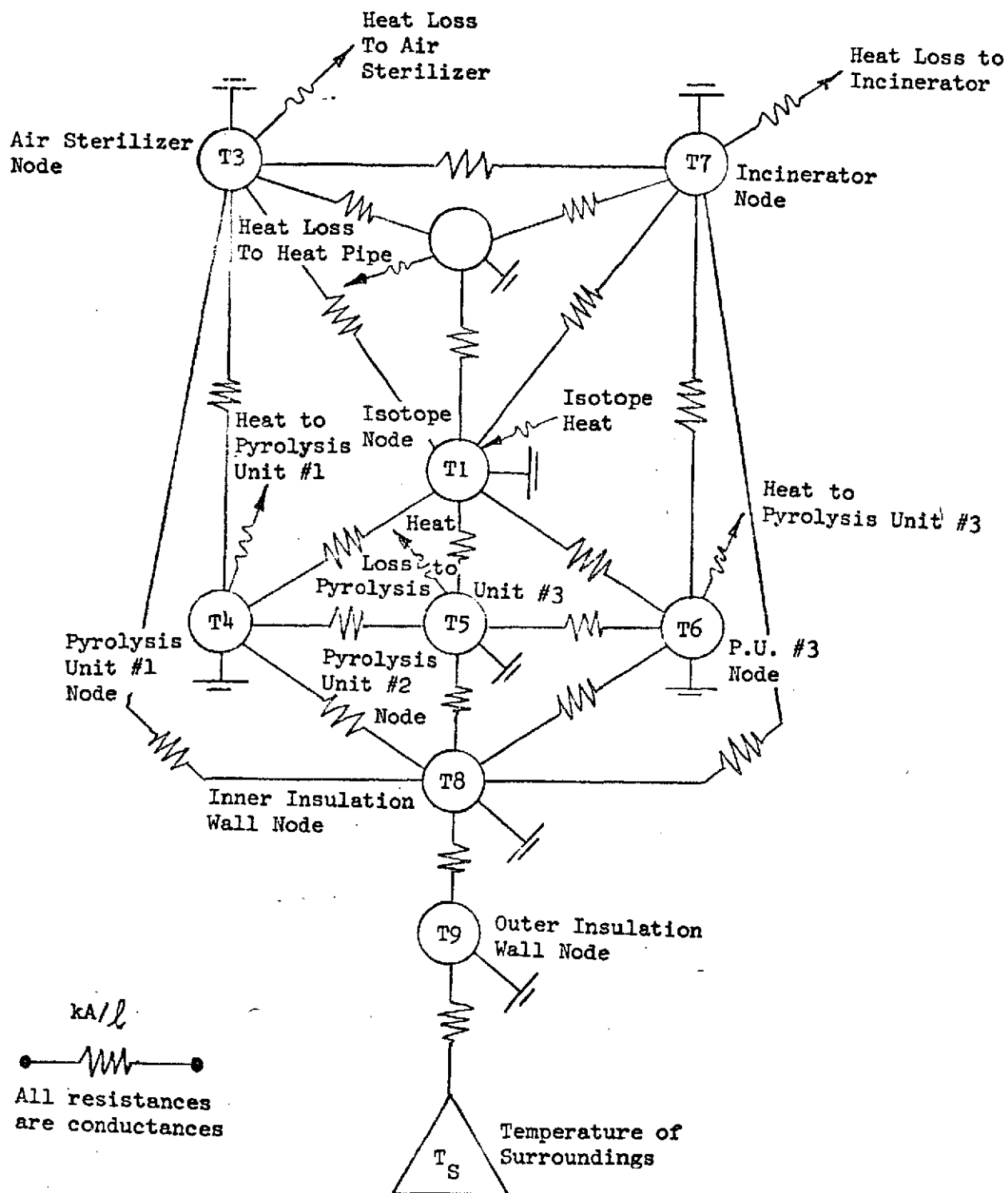


FIGURE 4.7 HEAT BLOCK NODAL NETWORK

The boundary conditions specified were the heating or cooling fluid temperatures and flow rates at the component inlets. The comparative data from the 180 day test and the computer runs are presented in Table 4.1 . It is seen that there is generally good agreement between the experimental data and the computed data. The computed condenser steam inlet temperature is seen to be about 10°F less than the experimental value. This is due to modelling the duct supplying the steam to the condenser such that excessive heat transfer from the ducted steam to the environment occurs. This simulation detail is considered to be relatively unimportant in the total simulation and has only recently been corrected.

The steam temperature into the condenser is higher than the steam temperature out of the evaporator due to the heat added in the pyrolysis units. Likewise the increase in steam flow into the condenser compared to the steam flow out of the evaporator is due to water generation in the pyrolysis units.

The analytical modeling and the data correlation performed on the key RITE system components provide assurance that the model is able to predict component performance for the operating condition to be studied. These components can now be combined with remaining components of their loop and loop performance can be investigated. Finally, the analytical models for each of the loops are combined for complete RITE system operation.

Table 4.1

## CORRELATIONS FROM COMPLETE SYSTEM OPERATION

Note: N/A = Not Available

<u>Low Temperature Isotope Heater</u>	<u>GE Data</u>	<u>G-189 Data</u>
Power level (Btu/hr)	5120	5120
Isotope Temperature (°F)	640	640
Inlet Fluid Temperature (°F)	179	179
Outlet Fluid Temperature (°F)	185	185.11
 <u>Evaporator</u>		
Heating fluid inlet temperature, °F	181.	181.2
Heating fluid outlet temperature, °F	123	123.03
Wall temperature, °F	116	115.55
Liquid temperature, °F	105	105.63
Gaseous outlet temperature, °F	105	105.6
Evaporation rate, lb/hr	2.3	2.36
NH <sub>3</sub> generation rate, lb/hr x 10 <sup>-3</sup>	N/A	4.62
CO <sub>2</sub> generation rate, lb/hr x 10 <sup>-3</sup>	N/A	5.98
Heat loss to surroundings	N/A	115.55
Pressure, psia	1.016	1.12
 <u>Condenser</u>		
Coolant inlet temperature, °F	32	35
Coolant outlet temperature, °F	47	48.3
Steam into condenser temperature, °F	127	126.7
Steam into condenser flow, lb/hr	2.3	2.367
Condensation rate, lb/hr	2.277	2.332
Efficiency, water recovery, %	98+	98+
Steam vented, lb/hr	.023	.035
Condensate wall temperature	75-82	75.4
Condensate temperature	N/A	55.6
Condenser pressure	N/A	.63

## Section 5

### CHARACTERISTICS OF THE RITE SYSTEM

The results of the G189-A simulation of the RITE system for the nominal RITE 180-day test conditions are presented in this section.

First, the simulation system performance averaged over a 24-hour period is denoted and compared with General Electric 180-day test data. Performance data are presented for each of the RITE subsystems. The transient effects caused by periodic micturition and defecation are then discussed. Finally the operational envelope of the RITE is developed. The operational envelope yields the potential performance of the RITE system when operated at various off-design conditions.

#### 5.1 RITE System Performance Characteristics

A computer run was made to simulate the nominal RITE operational conditions as reported in the General Electric RITE 180-day endurance test. The performance data for each of the system components was averaged over a 24-hour period. The results are presented in terms of mass and energy balance diagrams for each of the subsystems and detail component data are presented in table form.

##### 5.1.1 Low Temperature Heating Loop

The low temperature heating loop supplies heating fluid (water) to the evaporator, water storage tanks, flush water tank, hot water heater and the air heater. The energy and mass balance diagram for the loop are shown in Figure 5.1. Detail component data are then given in Table 5.1 along with comparative GE test results when available.

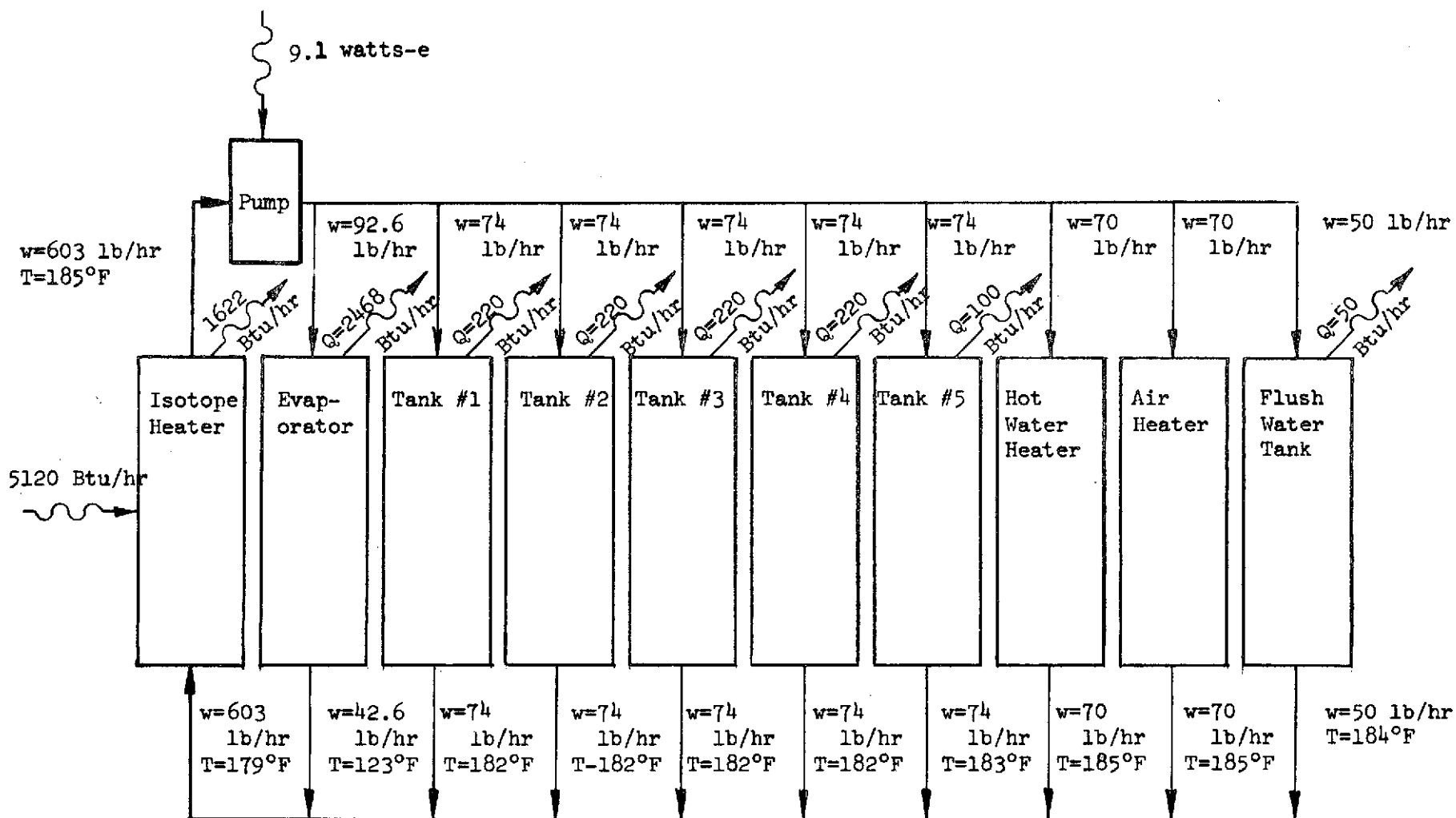


FIGURE 5.1 MASS AND HEAT BALANCE FOR LOW TEMPERATURE HEATING LOOP

TABLE 5.1

## RITE SYSTEM CHARACTERISTICS

NOTE: N/A = Not Available

## LOW TEMPERATURE HEATING LOOP

<u>Component Number</u>	<u>Component</u>	<u>G-189A Data</u>	<u>G.E. Test Data</u>
100	<u>Radioisotope Heater</u>		
	Heater Power Level	5120.0	5120.0
	Isotope Temperature (°F)	640.0	640.0
	Outlet Fluid Temperature (°F)	185.11	185.0
	Heat loss to tank and environment (Btu/hr)	1760.0	N/A
	Flow Rate (lb/hr)	603.0	N/A
104, 105	<u>Pumps</u>		
	Type	Centrifugal Series 400 Dynapump	
	Pump Power (watts)	9.1	N/A
108	<u>Split to Evaporator</u>		
	Flowrate (lb/hr)	42.6	N/A
109, 110	<u>Evaporator Liquid Control Valve</u>		
	Control Temperature	Evaporator Steam Outlet	
	Bypass on Temperature (°F)	105	105
	Bypass off Temperature (°F)	104	104
	Coolant Outlet Temperature (°F)	123.3	123
112, 113, 114, 115, 116	<u>Splits to Potable Water Storage Tanks</u>		
	Flowrate to Each Tank (lb/hr)	74.08	N/A
	Average Outlet Temperatures (°F)	182	N/A
117	<u>Split to Water Heater</u>		
	Flowrate (lb/hr)	70.0	N/A
121	<u>Split to Air Heater</u>		
	Flowrate (lb/hr)	70.0	N/A



TABLE 5.1 (cont)

<u>Component Number</u>	<u>Component</u>	<u>G-189A Data</u>	<u>G.E. Test Data</u>
124	<u>Warm Flush Water Tank Controller</u>		
	Flowrate (lb/hr)	50	N/A
	Control Temperature (°F)	Tank fluid	
	Bypass - on (°F)	96	96
	Bypass - off (°F)	94	94
	Outlet Temperature (°F)	184	N/A
135	<u>Emergency Coolant Supply</u>		
	Flowrate (lb/hr)	60	N/A
	Temperature (°F)	80	80
136,139	<u>Emergency Heat Exchanger Controller</u>		
	Control Temperature	Isotope Coolant Outlet	
	Flow-on Temperature (°F)	190	190
137	<u>Emergency Heat Exchanger</u>		
	Flowrate (lb/hr)	603	N/A
	Effective UA (Btu/hr-°F)	16.0	N/A
	Outlet Temperature (°F)	179	179
41	<u>Flush Water Tank</u>		
	Tank Capacity (lb)	40.0	40.0
	Heat Exchanger UA (Btu/hr-°F)	3.0	N/A
	LTHL Inlet Temperature (°F)	185	N/A
	LTHL Outlet Temperature (°F)	184	N/A
	Tank Outlet Temperature (°F)	98	N/A
44	<u>Pump</u>		
	Type	Centrifugal	
	Pump Power (watts)	1.04	N/A
51	<u>Condensate Water Supply Tee</u>		
	Flowrate (lb/hr)	.834	N/A
	Temperature (°F)	70.04	N/A

TABLE 5.1 (cont)

## AIR LOOP (CONDITIONS FOR MICTURITION)

<u>Component Number</u>	<u>Component</u>	<u>G-189A Data</u>	<u>G.E.Test Data</u>
21	Urine - Air Supply		
	Urea Flow Rate (lb/hr)	.392	N/A
	Urine Solids Flow Rate (lb/hr)	.42	N/A
	Urine Water Flow Rate (lb/hr)	13.2	N/A
	Urine Temperature (°F)	98.6	N/A
22	Flush Water Supply Tee		
	Water Flow (lb/hr)	20.0	N/A
	Water Temperature (°F)	98.0	N/A
24	Air Blower		
	Type	AMETEC Model No. E-4698-1F 15 v. 60 cy.	N/A
	Fan Power (watts)	167.98	N/A
	Flowrate (lb/hr)	56.0	N/A
25	Air Sterilizer		
	Flowrate (lb/hr)	56.0	N/A
	Inlet Temperature (°F)	98.6	N/A
	Outlet Temperature (°F)	362.8	N/A
	Heat Input (Btu/hr)	3674.0	N/A
	Heat lost to Environment (Btu/hr)	94.5	N/A

## VAPOR LOOP

210	Evaporator		
	Heating fluid inlet temperature (°F)	181.2	181.
	Heating fluid outlet temperature (°F)	123.03	123
	Wall temperature (°F)	115.55	116
	Liquid temperature (°F)	105.63	105
	Gaseous outlet temperature (°F)	105.6	105
	Evaporation rate (lb/hr)	2.3615	2.3
	NH <sub>3</sub> generation rate (lb/hr X10 <sup>-3</sup> )	4.62	N/A
	CO <sub>2</sub> generation rate (lb/hr-X10 <sup>-3</sup> )	5.98	N/A
	Heat loss to surroundings (Btu/hr)	115.55	N/A
	Pressure (psia)	1.085	1.016

TABLE 5.1 (continued)

<u>Component Number</u>	<u>Component</u>	<u>G-189A Data</u>	<u>G.E. Test Data</u>
223, 224 226	<u>Pyrolysis Units (includes integrated heat exchanger)</u>		
	Inlet temperature	105.63	125
	Outlet temperature	172.3	N/A
	NO <sub>2</sub> flowing out (lb/hr-X10 <sup>-6</sup> )	5.96	N/A
	NO <sub>2</sub> flowing out (lb/hr-X10 <sup>-6</sup> )	7.77	N/A
	CO <sub>2</sub> flowing out (lb/hr-X10 <sup>-3</sup> )	9.45	N/A
	H <sub>2</sub> flowing out (lb/hr-X10 <sup>-8</sup> )	1.12	N/A
	O <sub>2</sub> flowing out (lb/hr-X10 <sup>-3</sup> )	1.29	N/A
	N <sub>2</sub> flowing out (lb/hr-X10 <sup>-3</sup> )	3.8	N/A
	Heat transferred from heat block (Btu/hr)	206	N/A
	Pyrolysis unit temperature (°F)	1259.7	1200
237	<u>Condenser</u>		
	Coolant inlet temperature °F	35	32
	Coolant outlet temperature °F	48.3	47
	Steam into condenser temperature °F	126.7	127
	Steam into condenser flow lb/hr	2.3615	2.3
	Condensation rate lb/hr	2.3354	2.277
	Efficiency, water recovery %	99	99
	Steam vented lb/hr	.0261	.023
	Condensate wall temperature	75.4	75-82
	Condensate temperature	55.6	N/A
	Condenser pressure	.63	N/A
COOLING LOOP			
1	<u>Coolant Supply</u>		
	Flowrate (lb/hr)	280.0	280.0
	Coolant temperature (°F)	32.0	32.0
94	<u>Water Cooler</u>		
	Coolant flowrate (lb/hr)	21.28	N/A
	Coolant input temperature (°F)	35.0	N/A
238	<u>Condenser Cooling</u>		
	Flowrate (lb/hr)	258	258.72
	Inlet Temperature (°F)	35°F	35.0
	Outlet Temperature (°F)	48	48.0
10	<u>Return Flow Tee</u>		
	Flowrate (lb/hr)	280	N/A
	Temperature (°F)	47	N/A

TABLE 5.1 (continued)

## POTABLE WATER STORAGE SYSTEM

<u>Component Number</u>	<u>Component</u>	<u>G-189A Data</u>	<u>G.E. Test Data</u>
83, 84, 86, 87, 90	<u>Storage Tanks</u>		
	Average Inlet Flow (lb/hr)	224	N/A
	Inlet Flow Temperature (°F)	55	N/A
	Tank Capacity (gal)	5	5
	Liquid Temperature (°F)	150.25	N/A
	Tank Wall Temperature (°F)	148.15	145.0
	Heat Loss to Environment (Btu/hr)	200.0	N/A
	Heating Fluid Inlet Flow (lb/hr)	70.08	N/A
	Heating Fluid Inlet Temperature (°F)	185.	N/A
	Heat Fluid Exit Temp. (°F)	182.41	N/A
	Heat Input (Btu/hr)	200.0	N/A
99	<u>Pump</u>		
	Average Inlet Flow (lb/hr)	2.24	N/A
	Pump Power (watt)	1.07	N/A

Approximately 5120 Btu/hr enter the loop through the isotope and electrical heaters. The evaporator requires 2468 Btu/hr; 220 Btu/hr are required by each of the potable water storage tanks; approximately another 100 Btu/hr are required by the flush tank and water and air heaters. The specific heat consumption rate of the tanks and air heater depend on the fill and drain schedule for the operational mode in use at the time. The results noted in the Figure represent 24-hour averaged values during nominal 180-day test conditions. The remaining heat is lost to the cabin environment from heat leaks in the lines, tanks, evaporator walls, and to the isotope shield tank.

#### 5.1.2 Flush Water Loop

The mass and energy balance diagram for the flush water loop is shown in Figure 5.2. The loop delivers flush water upon demand to the urinal, trash shredder and the blender. The flush water flow rates are given in the figure. Water for the flush water loop is collected from the Environmental Control System condensate and stored in the warm flush water tank. The water is heated to 98°F by internal tank coils through which LTHL fluid is pumped. Temperature control is accomplished by a heating fluid bypass controller.

#### 5.1.3 Air Loop

The air loop directs flush water and drying air to the urinal during micturition. The micturition schedule is shown in Figure 5.6. The liquids are separated and directed to the evaporator while the exhaust air is sterilized in the air sterilizer. As the air passes through the sterilizer over 3600 Btu/hr are removed from the heat block. The air returns to the cabin at a temperature of over 360°F. The air loop mass and energy balance is shown in Figure 5.3.

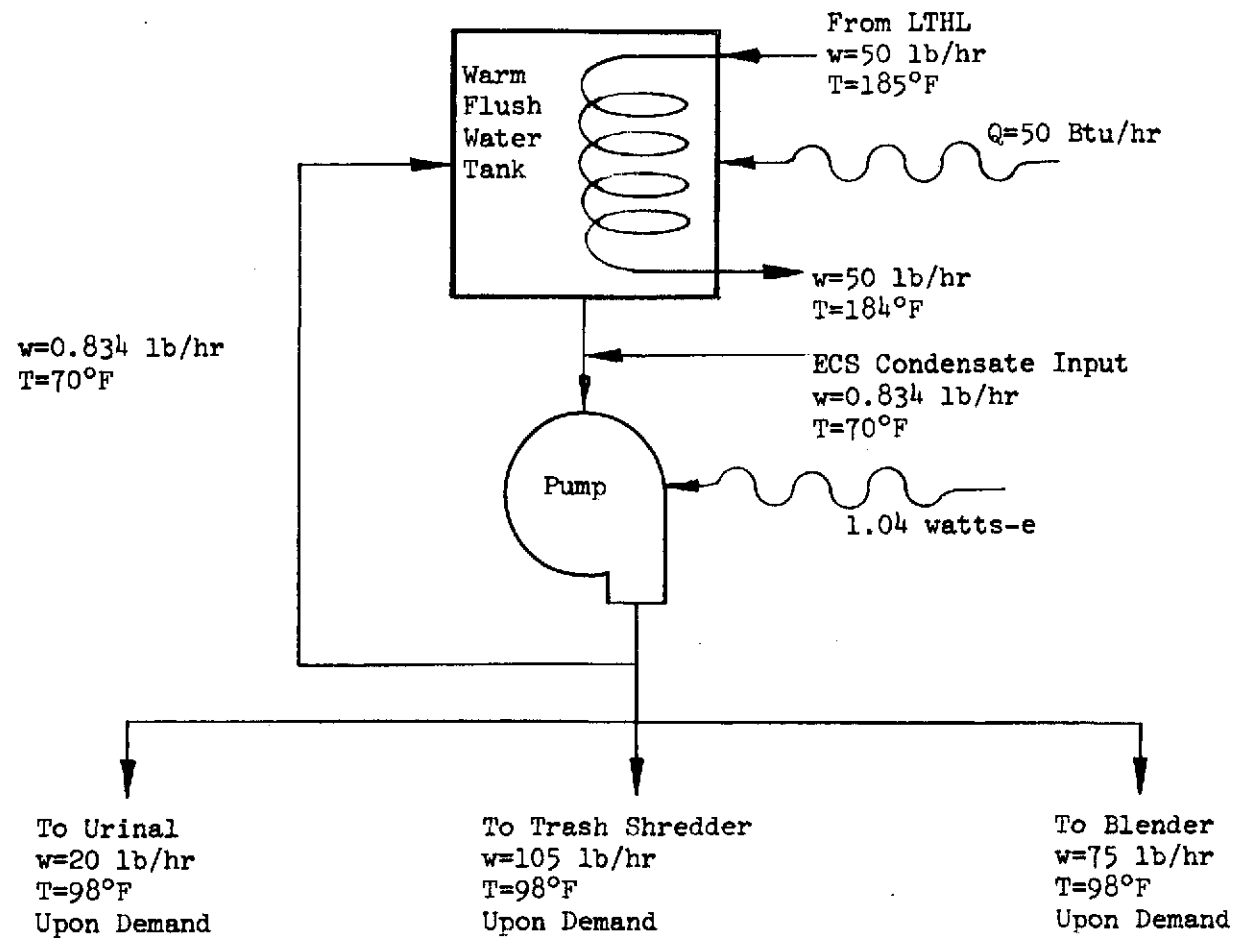


FIGURE 5.2 MASS AND HEAT BALANCE FOR FLUSH WATER LOOP

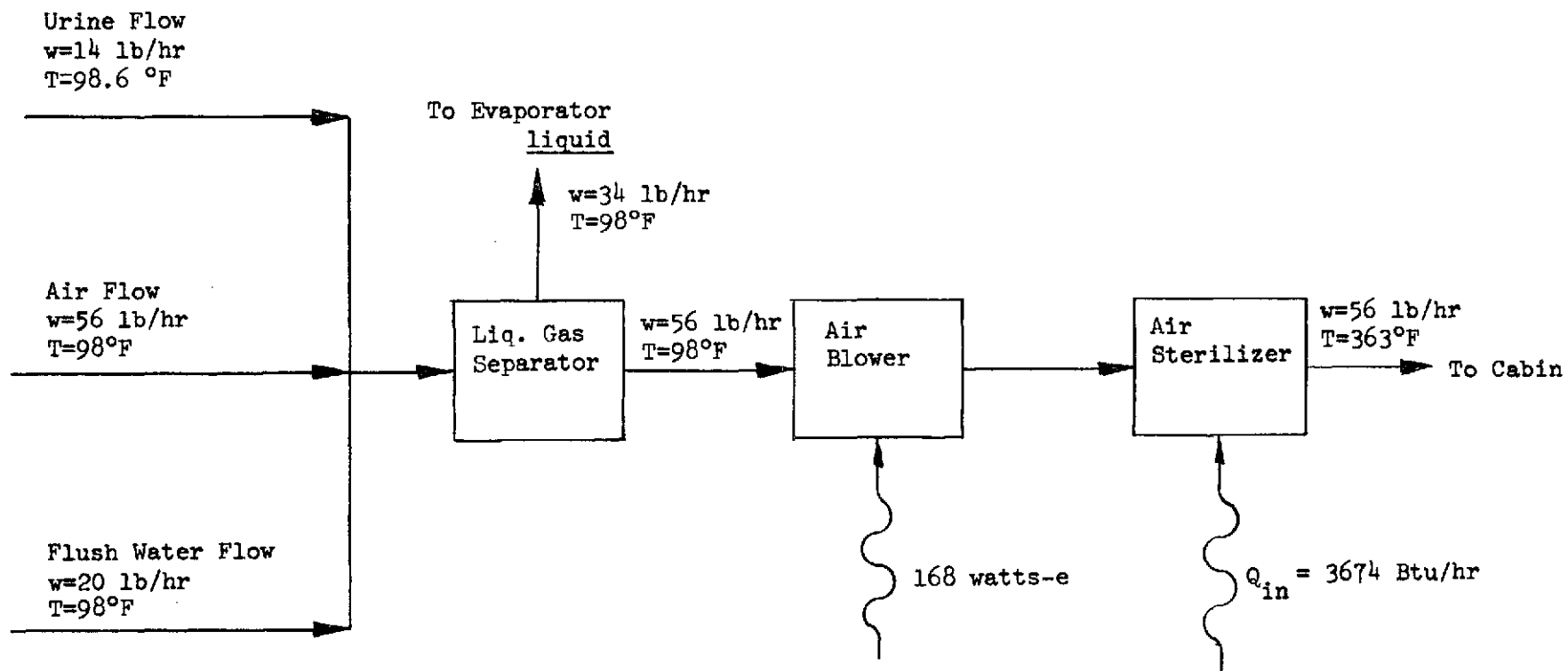


FIGURE 5.3 AIR LOOP MASS AND HEAT BALANCE DURING MICTURITION

#### 5.1.4 Vapor Loop

The vapor loop purifies the waste water to potable water by an evaporation/condensation phase change process. The loop is comprised of five principal components, the evaporator, the three pyrolysis units and the condenser. The evaporator collects all the solid wastes and water, removes the solids by filtration and evaporates the water. The energy used in evaporating the liquid is transferred from the heating fluid supplied by the LTHL. The pyrolysis units neutralize any gases generated in the evaporator by oxidation at high temperature. The vapor flow is condensed in the condenser and the gasses are vented to vacuum.

The mass and heat balance for the vapor loop is shown in Figure 5.4. Over 96% potable water recovery is achieved. The sources of water loss are the condenser vent and water absorbed by the solids and removed by the solids pump to the incinerator. A chemical analysis of the gas reaction processes occurring in the vapor loop was attempted in order to reproduce the identical gas mass ratio reported by G.E. in their condenser vent gas analysis. The types and mass ratios of the output vent gases were reported by G.E. in Reference 1 and are repeated in Table 5.2. In order to reproduce the gas mass properties reported by G.E. it was postulated that there had most likely been a Bosch reaction in the pyrolysis unit. In this reaction hydrogen combines with carbon dioxide yielding carbon and water vapor. The possibility exists that the carbon is deposited in the catalyst and after a period of time reduces the effectiveness of the pyrolysis units. Due to the importance of water conservation and the complex nature of the processes taking place in this loop the mass balance will be discussed in detail. The chemical reactions and the resulting product generation rates taking place in the vapor loop are shown in detail in Figure 5.5.

##### 5.1.4.1 Flow Input to the Vapor Loop

The vapor loop collects and processes wastes for a four-man crew for a 180-day simulated spacecraft mission. The quantities of solid and liquid wastes processed per day are summarized below:



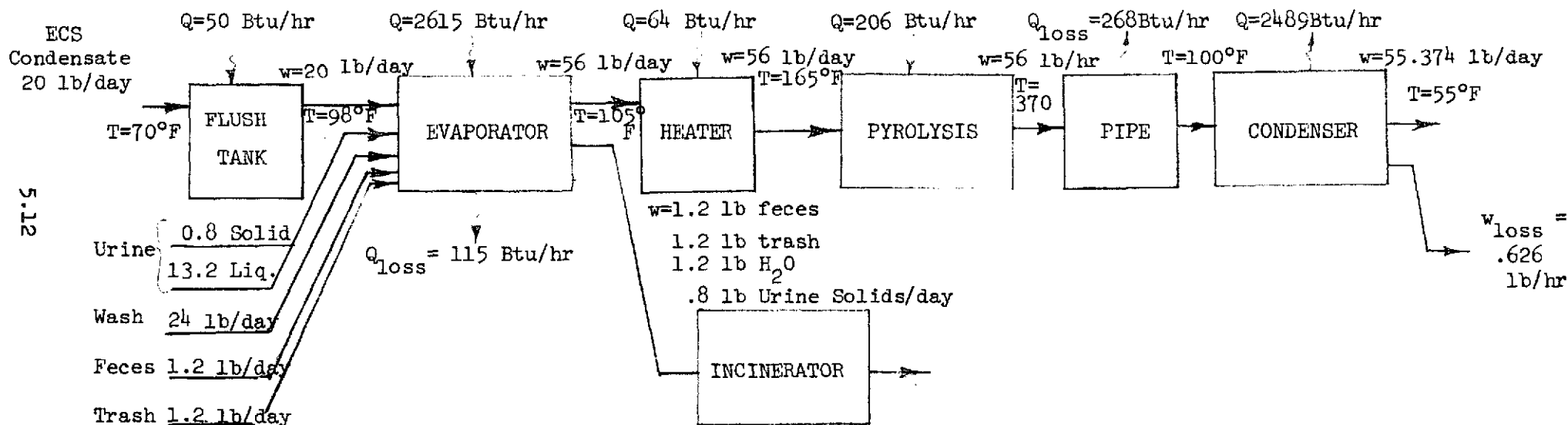


Figure 5.4 MASS AND HEAT BALANCE FOR THE VAPOR LOOP

Table 5.2

## PYROLYSIS VENT GAS ANALYSIS

Note: Data are from Reference 1

COMPONENTS	PERCENT (WATER FREE)	PERCENT (LESS WATER AND ARGON)	FLOW RATES CONDENSER VENT (lb/hr)
O <sub>2</sub>	12.2	12.6	$6.456 \times 10^{-4}$
N <sub>2</sub>	81.9	84.6	$3.7941 \times 10^{-3}$
CO <sub>2</sub>	1.3	1.34	$9.46 \times 10^{-5}$
NO	.5	.53	$1.9854 \times 10^{-5}$
NO <sub>2</sub>	.2	.2	$1.52 \times 10^{-5}$
H <sub>2</sub>	.7	.73	$2.3 \times 10^{-6}$
Ar	3.2		

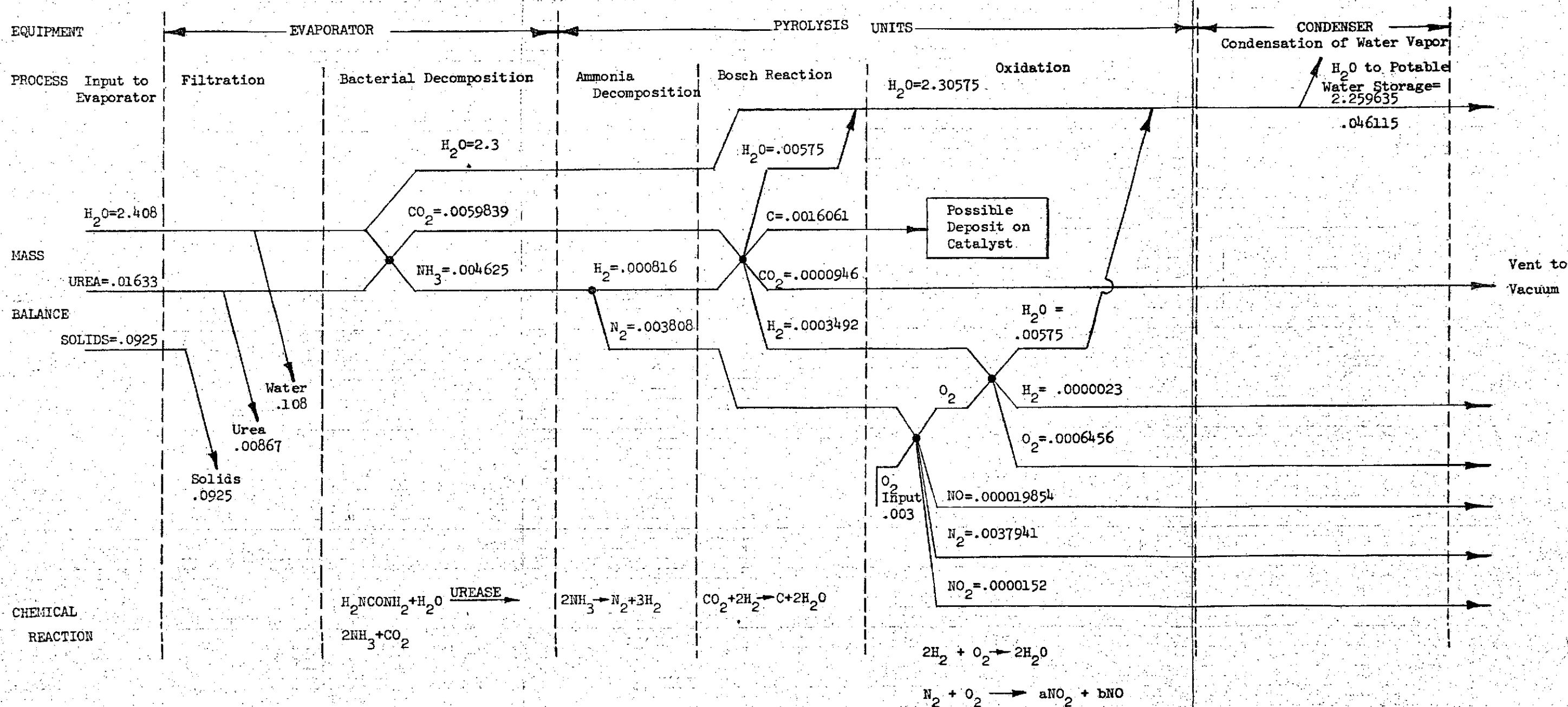


Figure 5.5 RITE System Mass Balance (ALL FLOW RATES IN LB/HR)  
 FIGURE 5.5 RITE SYSTEM CHEMICAL REACTIONS AND MASS BALANCE (ALL FLOW RATES IN LB/HR)

- (a) Feces: 1.2 pounds per day for 4 defecations
- (b) Urine: 14.0 pounds per day from approximately 24 micturitions.
- (c) Respiration and Perspiration: 20.0 pounds per day at a continuous flow rate from the environmental control system.
- (d) Wash Water: 24.0 pounds per day.
- (e) Trash: 1.2 pounds per day - food, packets, wipes, and paper.

The wastes enter the evaporator from the solids loop and the air loop. The time sequence in which these products are input to the evaporator is shown in Figures 5.6 to 5.8 for the urine, feces, and trash input respectively. Micturition occurs once per hour and the solids input occur an average of once per six hours.

#### 5.1.4.2 Evaporator Operation

The evaporator collects all solids and liquids that constitute waste, centrifugally separates the solids from the liquid, and evaporates the liquid. Some of the urea solids in the evaporator decomposes into ammonia and carbon dioxide and are vented with the water vapor.

#### Filtration

The pumping action of the impeller in the evaporator provides the means for the circulation of the liquid slurry thru an external perforated metal filter. The waste solids collect in the filter. When the filter (defined as the solids reservoir) fills with solids, the solids pump moves the solids to the incinerator. The filtration rate of the solids is proportional to solids concentration in the mixture.

As shown in Fig. 5.5, half of the urea that flows into the evaporator and all of the remaining solids that constitute the influent are assumed to be removed by the filter. For each pound of solids removed, approximately 1.17 pounds of water is also removed from the reservoir through filtration. This

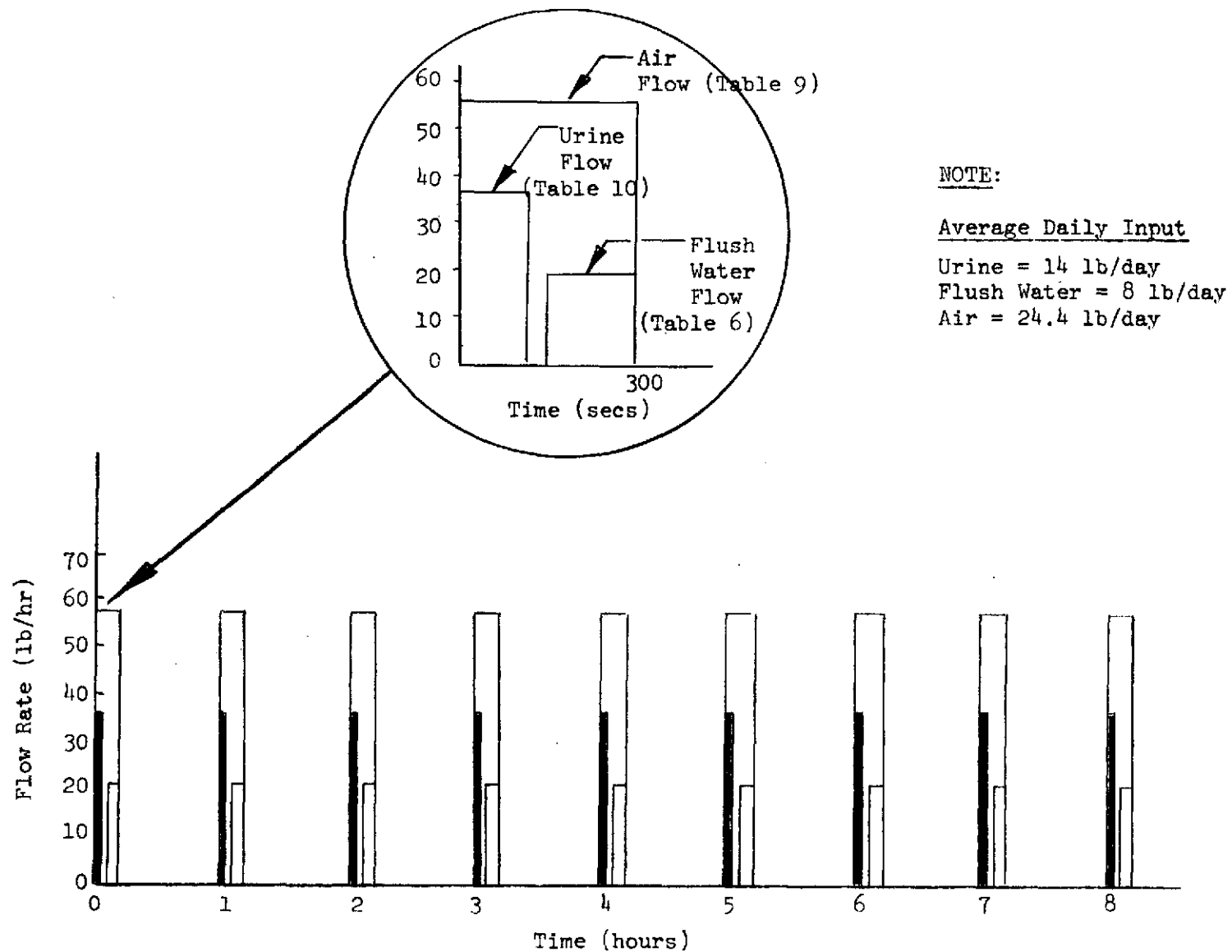


FIGURE 5.6 URINAL INPUT FLOW CYCLE

(Average Daily Input: Solids = 1.2 lb/day  
Flush Water = 5 lb/day  
Air = 8.7 lb/day)

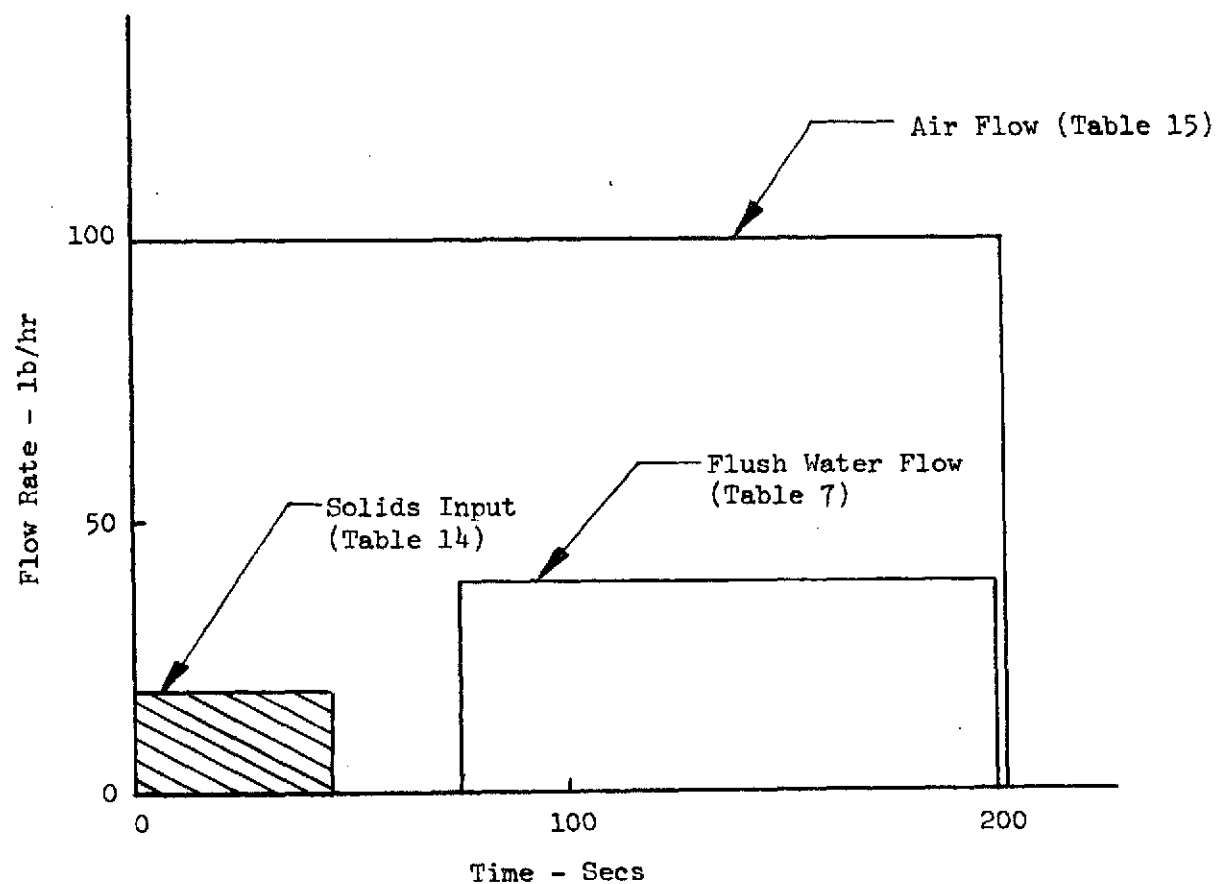


FIGURE 5.7 BLENDER INPUT CYCLE (one/6 hrs)

(Average Daily Input: Trash = 1.2 lb/day)  
Flush Water = 7 lb/day)

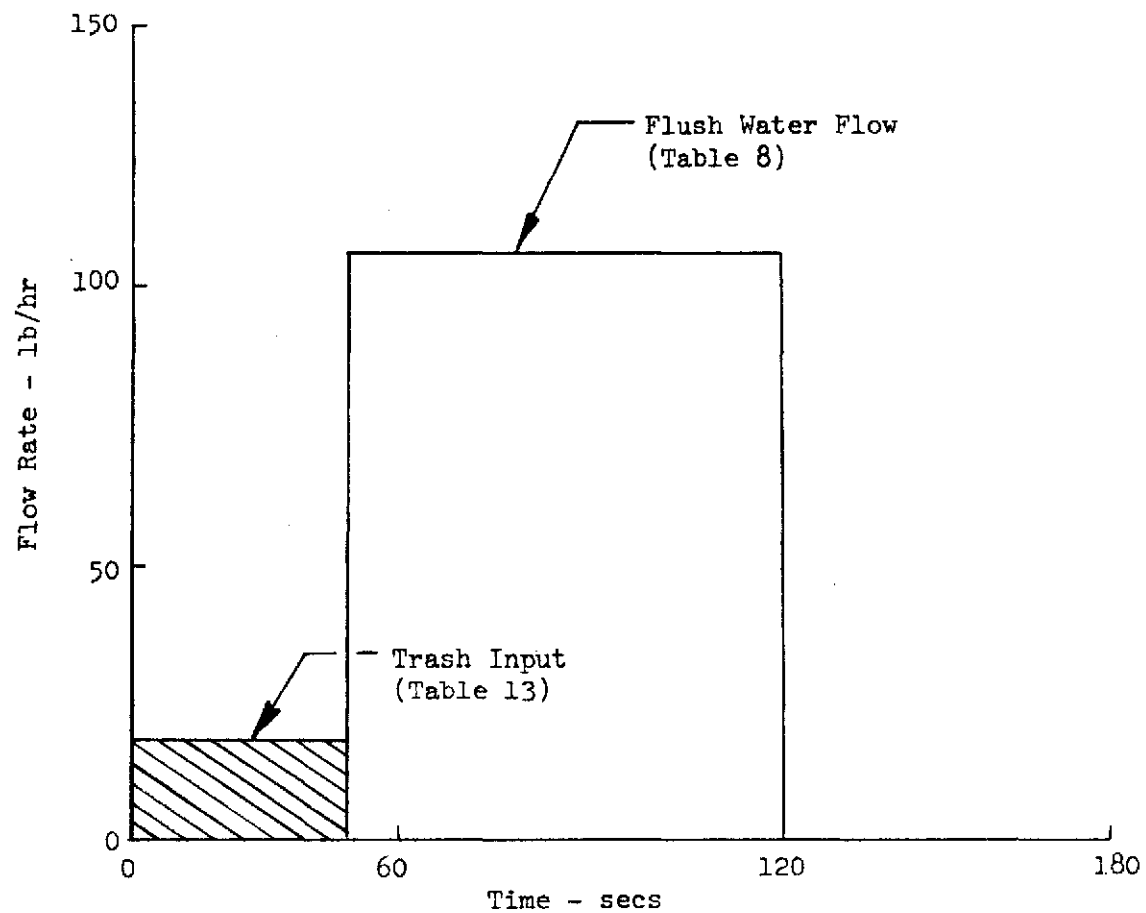


FIGURE 5.8 TRASH INPUT CYCLE (ONE/6 hrs)

water is added to the total solids in the reservoir and is also transferred to the incinerator. This represents a serious water loss to the system. A detailed discussion of the filtration process is included in the EVAP subroutine writeup.

#### Decomposition of Urea

The remaining urea is assumed to decompose biochemically/chemically generating  $\text{NH}_3$  and  $\text{CO}_2$ . Urine contains a certain amount ( $\sim 50 \text{ mg/l}$ ) of ammonium carbonate/bicarbonate which will liberate ammonia and carbon dioxide when it is vaporized. In addition, urea which is the principal component of urine ( $\sim 28000 \text{ mg/l}$ ) may decompose thermally when the temperature is above  $150^\circ\text{F}$  to yield  $\text{NH}_3$  and  $\text{CO}_2$ . Urea also decomposes in the presence of the enzyme urease which is provided from bacteria derived from the feces. Once the bacteria is introduced it will multiply while decomposing most of the urea within 12 to 48 hours. Since no germicidal chemicals are added to the evaporator it is assumed that biochemical decomposition of urea will occur. The urea decomposes to  $\text{NH}_3$  and  $\text{CO}_2$  according to the following reaction.



Since the bacterial decomposition of urea was assumed to be at  $0.008167 \text{ lb/hr}$ , the ammonia generation rate was  $0.004625 \text{ lb/hr}$  and the carbon dioxide generation rate was  $0.00594 \text{ lb/hr}$ .

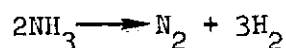
#### Vaporization of Water

The nominal design point steady state vaporization of water is  $2.3 \text{ lb/hr}$  at a vapor temperature of  $105^\circ\text{F}$  and vapor pressure of  $1.1 \text{ psia}$ . The Corresponding wall temperature is  $115^\circ\text{F}$ . A detail discussion of the evaporation heat transfer characteristics is included in the EVAP subroutine writeup.



#### 5.1.4.3 Pyrolysis Unit

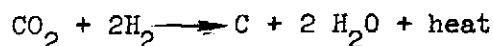
The pyrolysis units initially disassociate the ammonia to yield nitrogen and hydrogen



The product generation rates are given in Figure 5.5.

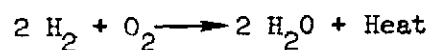
Oxygen is added to the vapor stream, upstream of the pyrolysis units, to oxidize the hydrogen to prevent ammonia from reforming. The oxygen flow to the vapor loop is determined by measuring the pH of the condensate in the condenser, where a basic solution would require an increase in  $\text{O}_2$  flow to the vapor stream while an acidic solution would result in the decrease of  $\text{O}_2$  flow to the vapor stream. The philosophy supporting the forementioned approach for  $\text{O}_2$  flow control is that excess  $\text{O}_2$  generates excess  $\text{NO}_x$  gasses in the pyrolysis units yielding an acidic condensate while insufficient  $\text{O}_2$  in the vapor stream causes nitrogen and hydrogen to reform causing the condensate to be basic. A balance of ammonia and nitrates/nitrites yields a neutral solution; but, possibly an unacceptable solution due to a high level of nitrate/nitrite salts in the condensate. The water analysis results presented in Reference 1, indicate the existance of nitrate/nitrite salts in the product and the possibility of a problem in this area.

The presence of hydrogen, carbon dioxide, high temperature ( $\sim 1250^\circ\text{F}$ ), and a catalyst in the pyrolysis unit yields an ideal condition for a BOSCH reaction. In the BOSCH process, carbon dioxide is hydrogenated to water and carbon according to the reaction

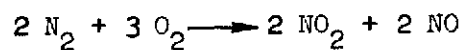


The water is recovered in the condenser while the carbon is unaccounted for. It is assumed that the carbon is deposited in the pyrolysis units, thus degrading the operation of the pyrolysis units. The rate of carbon deposition may be derived from the difference between CO<sub>2</sub> generated in the evaporator and the CO<sub>2</sub> vent rated from the condenser.

The remaining hydrogen (99.95%) that is not reacted in the BOSCH process is reacted with oxygen to generate more water as follows:



The excess oxygen reacts with nitrogen to form NO and NO<sub>2</sub> gasses as follows:



The product generation rates for all these processes are noted in Figure 5.5.

#### 5.1.4.4 Condenser

The condenser is the final component in the vapor loop. There the non-condensable gases are vented and the vapor is condensed. The condensation rate in the condenser was calculated to be 2.260 lb/hr and since the original water input to the system was 2.357 lb/hr the overall conversion efficiency is 96 percent.

#### 5.1.5 Cooling Loop

The cooling loop provides the coolant flow to the condenser and the water cooler. Approximately 2489 Btu/hr is added at the condenser increasing the coolant temperature from 35°F to 47°F. The loop schematic is shown in Figure 5.9.

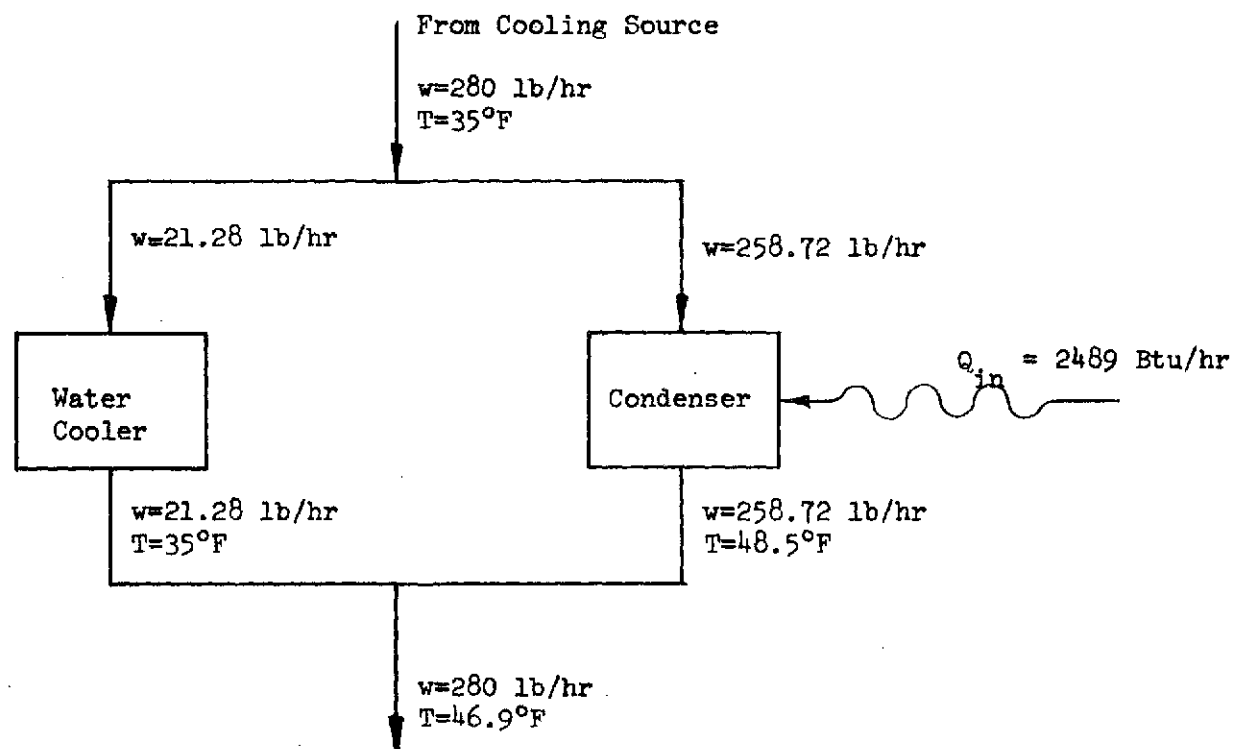


FIGURE 5.9 MASS AND HEAT BALANCE FOR THE COOLING LOOP

#### 5.1.6 Potable Water Loop

The potable water loop collects and stores purified water from the condenser. The system also distributes hot and cold water as required by users. A heat balance for a typical tank is shown in Figure 5-10. Approximately 210 Btu/hr are used to heat the water.

### 5.2 Transient Behavior

The G-189A model of the RITE system has been used to predict the performance for several transient events which occur during normal RITE operation. These events are micturition, defecation and incineration.

#### 5.2.1 Micturition

During micturition a slug of urine and flush water mixture suddenly enters the evaporator. The input schedule is shown in Figure 5.6. Urine is assumed to enter the evaporator at a body temperature of 98.6°F. The flush water is heated to  $99 \pm 1^\circ\text{F}$  in the flush tank prior to use. Urinal drying air is then blown through the air loop to the air sterilizer and returned to the cabin.

The response of the evaporator to the sudden entry of liquid is shown in Figures 5.11 and 5.12. The input liquid drops the average temperature of the slurry approximately  $2^\circ\text{F}$ . Since some of the heat is diverted into heating the liquid the evaporation rate drops from 2.3 lbs/hr to a minimum of 2.15 lb/hr. The pressure in the evaporator drops to the saturation pressure at that temperature. Since the temperature of the input liquid is close to that of the liquid in the evaporator the transient effects are minimal and disappear rapidly once the input flows cease.

The response of the heat block to the sudden surge of urinal air is shown in Figure 5.13. The temperature of the heat block node associated with the air sterilizer drops rapidly. The remaining nodes show only a small temperature drop due to the large thermal capacity of the heat block. This indicates that the operation of the incinerator and pyrolysis units are not jeopardized while the air sterilizer is being used.

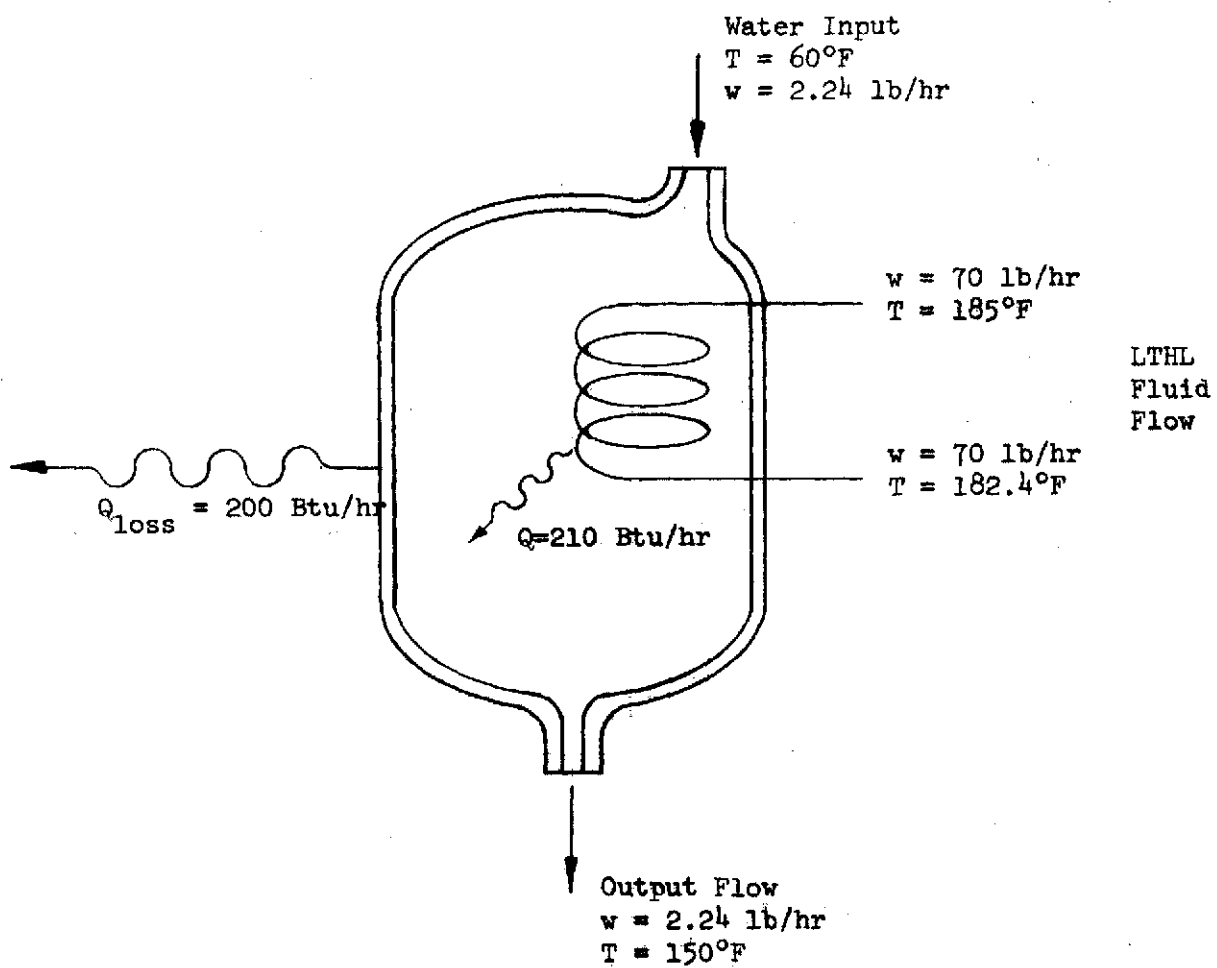


FIGURE 5.10 TYPICAL POTABLE WATER STORAGE TANK

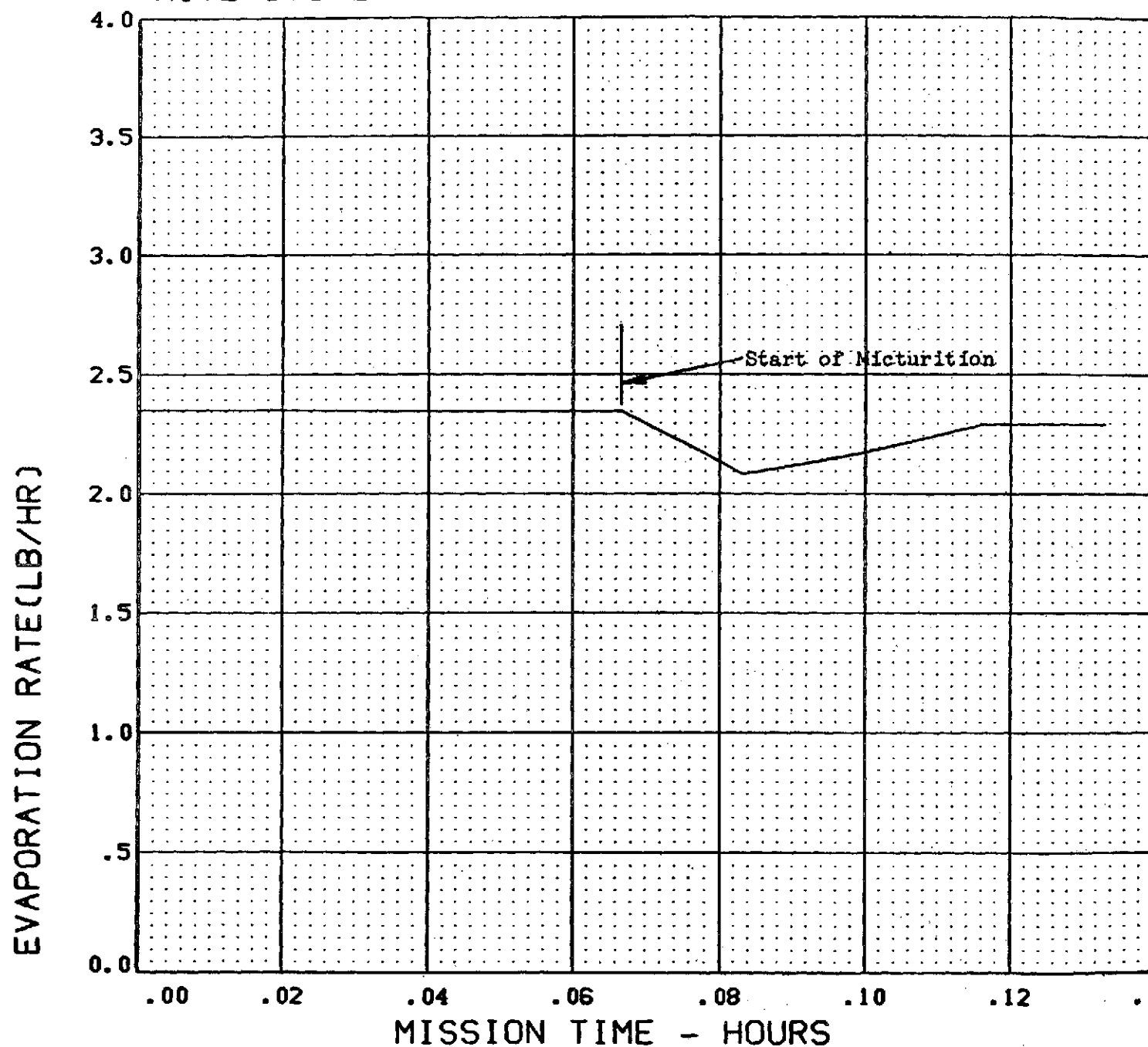


FIGURE 5.11 EVAPORATOR EVAPORATION RATE DURING MICTURITION

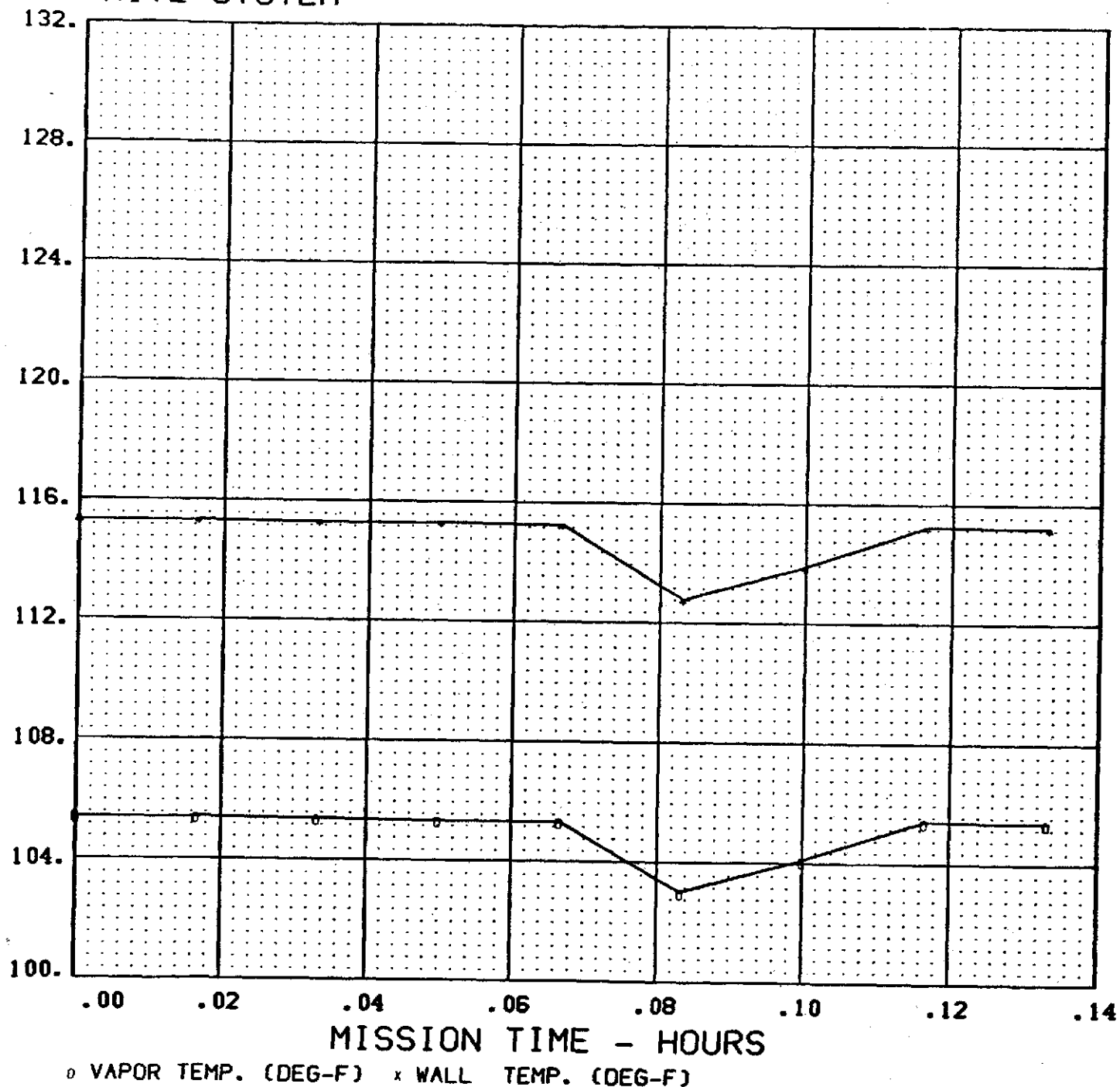
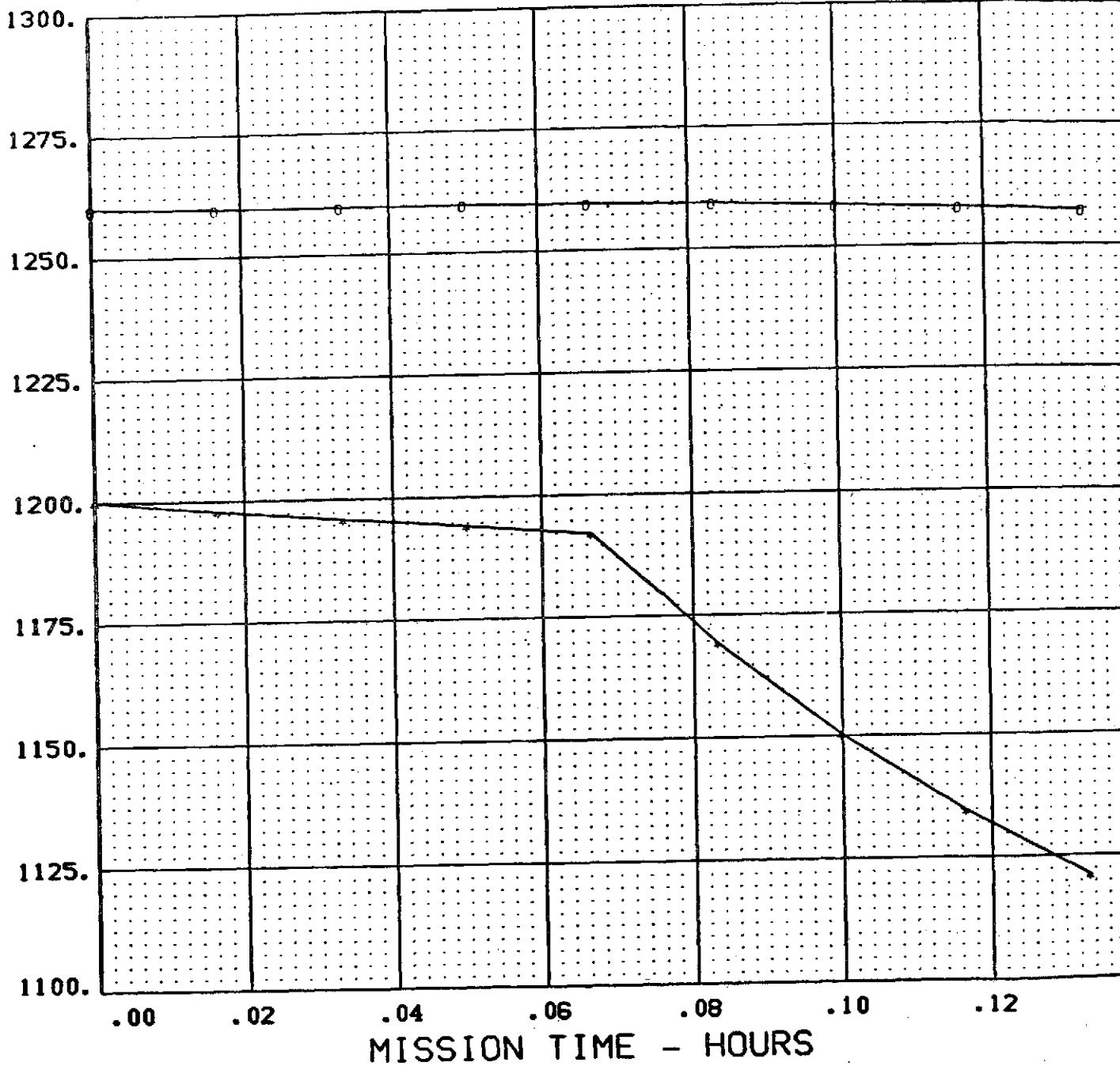


FIGURE 5.12 EVAPORATOR TEMPERATURE RESPONSE TO MICTURITION

G189 CASE 1

## RITE SYSTEM

HEAT BLOCK TEMPERATURES (DEG-F)



o ISOTOPE TEMPERATURE x NODE WITH AIR STER.

FIGURE 5.13 HEAT BLOCK TEMPERATURE RESPONSE TO MICTURITION



### 5.2.2 Defecation

During defecation, the normal low pressure operation of the evaporator is interrupted. The evaporator is exposed to cabin atmosphere and the output line is diverted to the air loop. Feces and flush water enter the evaporator through the blender. A stream of drying air is then blown from the blender, through the evaporator to the air loop and returned via the air sterilizer back to the cabin. During this time any liquid which is vaporized is lost to the cabin with the air stream. The vapor will then most likely be collected by the cabin's environmental control system and returned to the evaporator by way of the flush water loop.

The response of the RITE system to the disturbance caused by defecation is shown in Figures 5.14 and 5.15. The water recovery rate drops to zero since all vaporized liquid is returned to the cabin.

The evaporator temperature increases to 106°F since at the higher pressure boiling does not occur and the heat from the LTHL is used to heat the fluid. The temperature increase of the evaporator is limited to 105°F by the LTHL bypass controller which bypasses fluid around the evaporator when this temperature is reached.

The temperature response of the air sterilizer due to the sudden influx of the air (~37 cfm) is shown in Figure 5.16.

### 5.2.3 Incineration

The liquid-solid slurry in the evaporator is continuously being pumped through a circular metal screen filter. The solids remain behind and the liquid is returned to the evaporator. Periodically, an air driven piston compresses the solid concentrate inside the filter. The compressed solids are a sixty-fourth mixture of solids and liquid. When the filter is full (0.5 lbs) the solids concentrate is transferred mechanically to the incinerator.

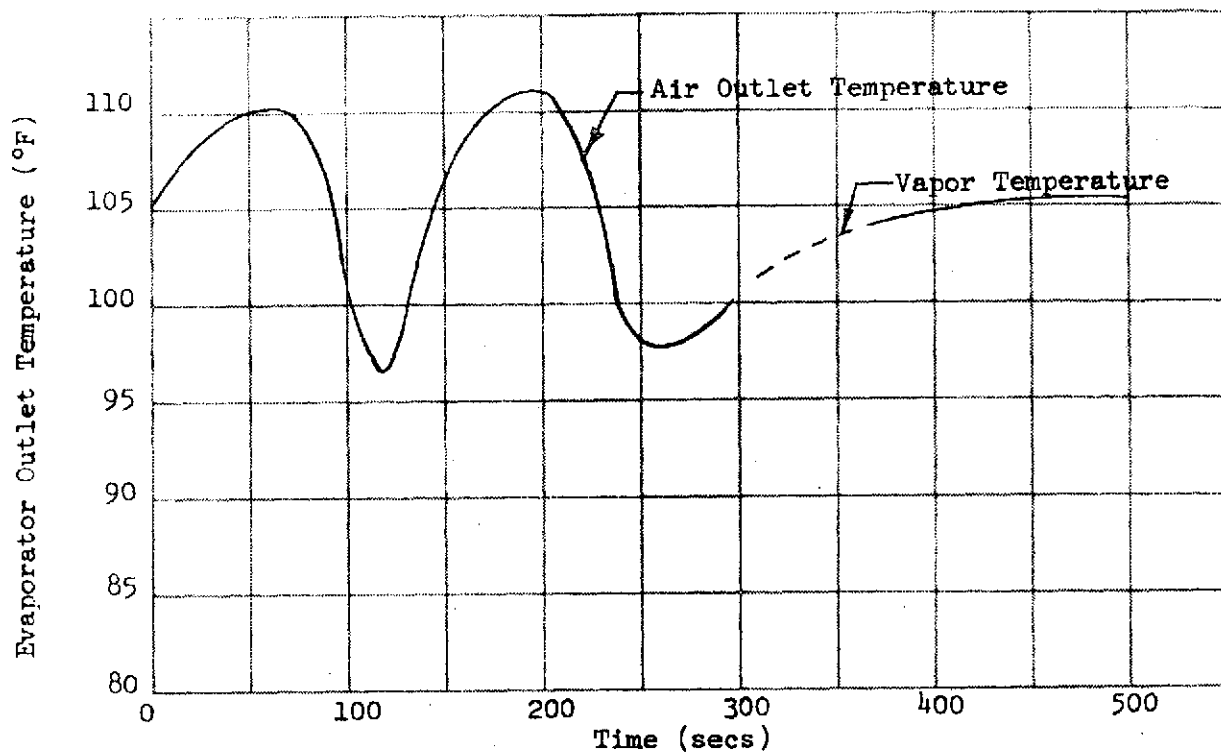


FIGURE 5.14 EVAPORATOR TEMPERATURE RESPONSE TO DEFECATION

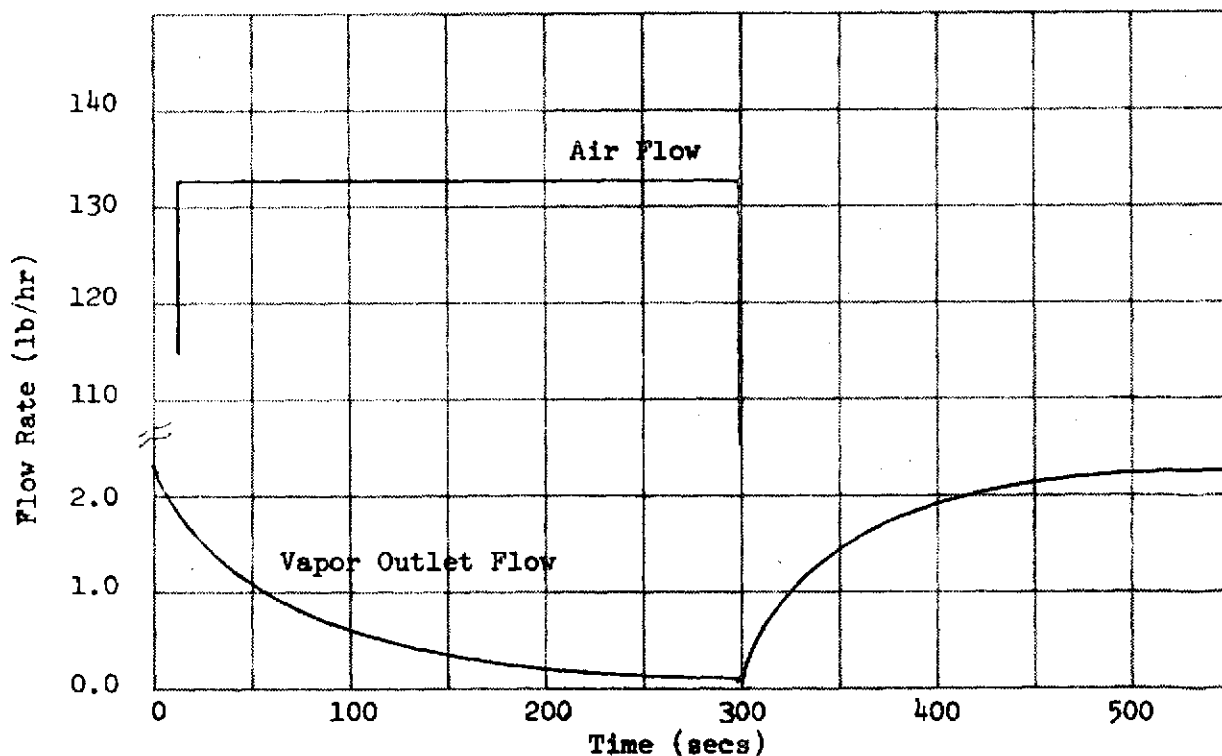


FIGURE 5.15 FLOW RATES IN EVAPORATOR DURING DEFECATION

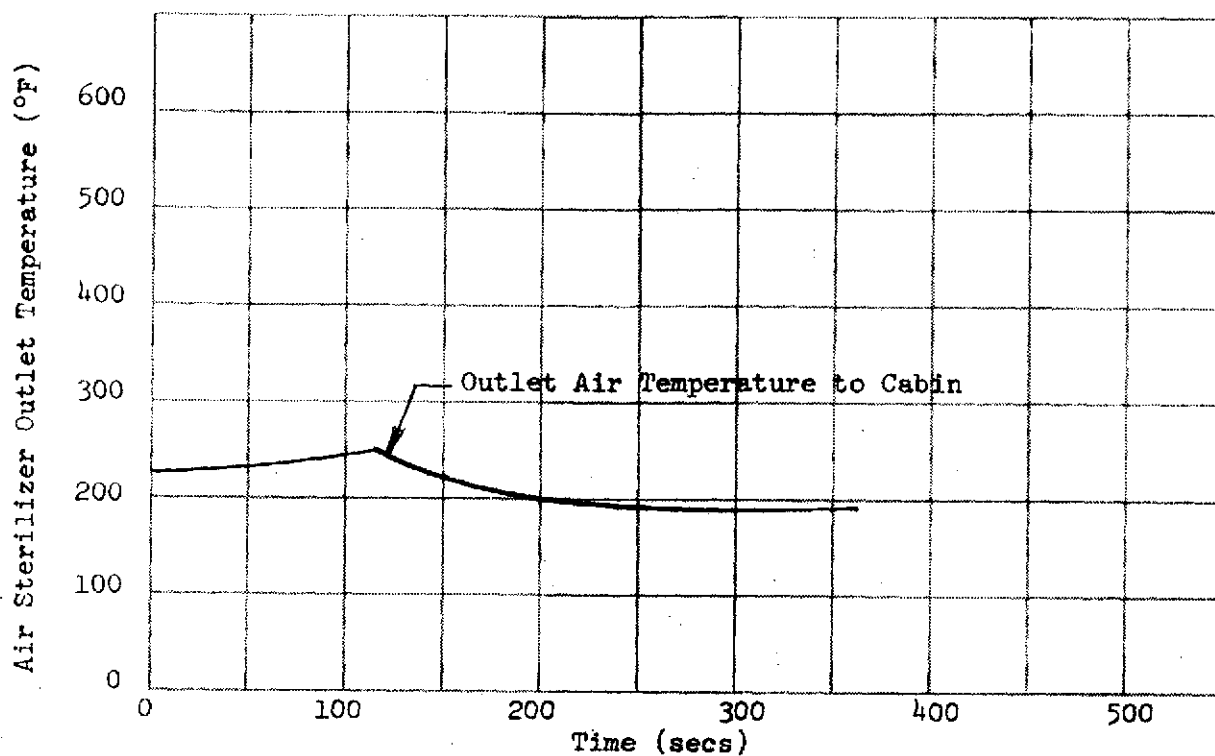


FIGURE 5.16 TEMPERATURE RESPONSE OF THE AIR STERILIZER DUE TO DEFECATION AIR

The solid waste is processed by vacuum drying, chemical decomposition at high temperature, and venting of resultant gasses. The temperature of the vented wastes of the incinerator are plotted in Figure 5.17.

During vacuum drying, heat from the heat block is used to evaporate the liquid. The vapor then flows to space through the open vent. Approximately 95% of the liquid is removed in the two hour drying period.

In the next step the vent to vacuum is closed and a small amount of oxygen is introduced with the solids and the oxidation reaction proceeds. Since the reactions are exothermic, there is a sudden rise in temperature. The chemical reactions are described in detail in the INCIN subroutine documentation contained in Appendix A. The product gasses are then vented to vacuum. The oxygen cycle is repeated until all the solids are incinerated. Approximately fourteen cycles are needed for full incineration. At the end of the cycle, the remaining ash is blown out by a nitrogen purge and stored in an ash collector. The volume of the remaining ash is only 1% of the original solid waste volume.

### 5.3 RITE System Operational Envelope

The RITE system operational envelope as generated by G-189A computer simulation is shown in Figure 5.18. The figure shows the variation in performance of the current RITE configuration as the boundary conditions, evaporator heat source temperature and condenser back pressure, are varied. The data shows excellent correlation with the General Electric 180-day run operating test point.

For any condenser pressure, increasing the LTHL fluid temperature increases water recovery rate until a vapor flow and temperature is reached which is beyond the condensation capacity of the condenser. The condenser removes only the sensible heat of the steam and the vapor escapes through the vent. The recovery rate rapidly drops to zero.

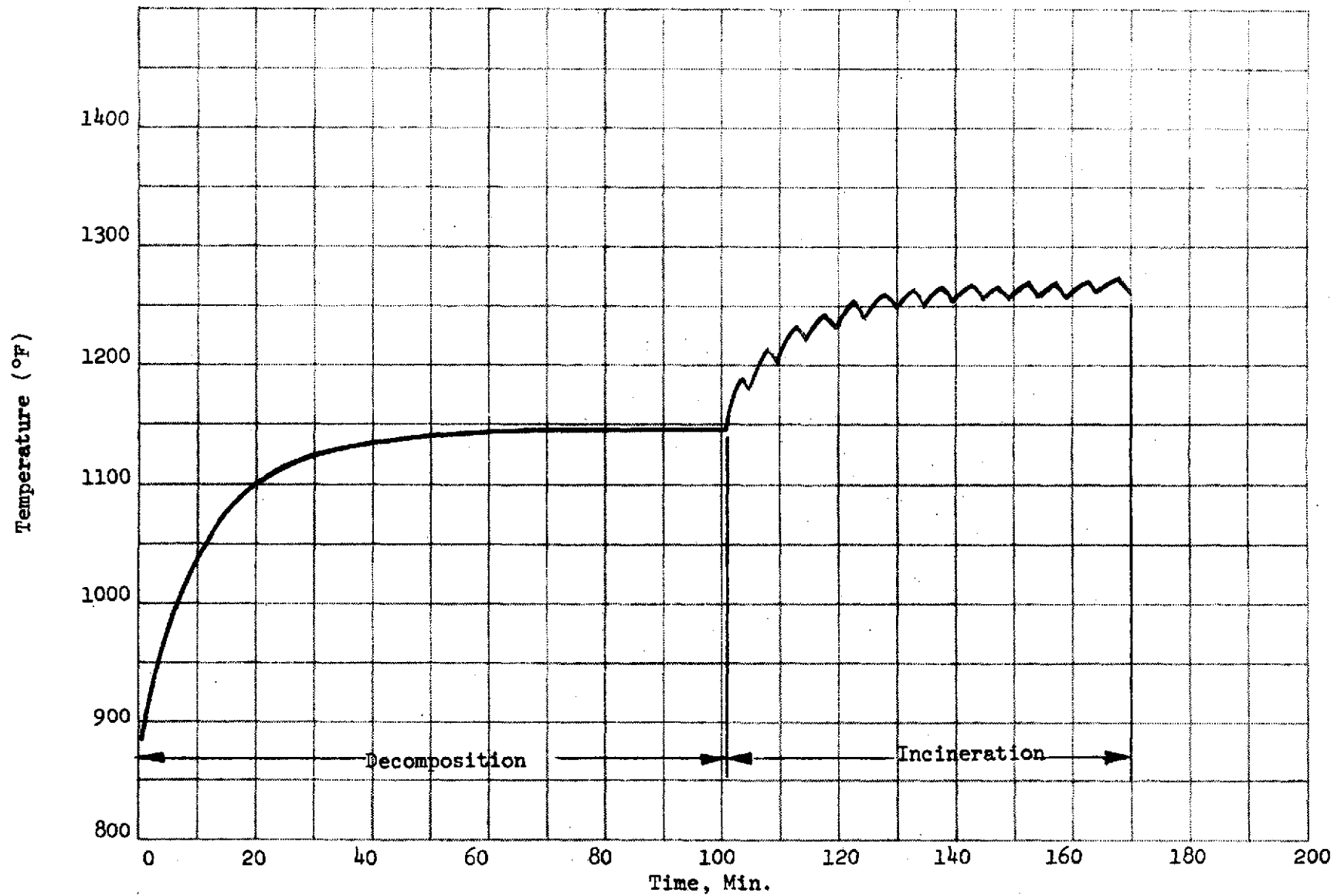


FIGURE 5.17 INCINERATOR TEMPERATURE CYCLE

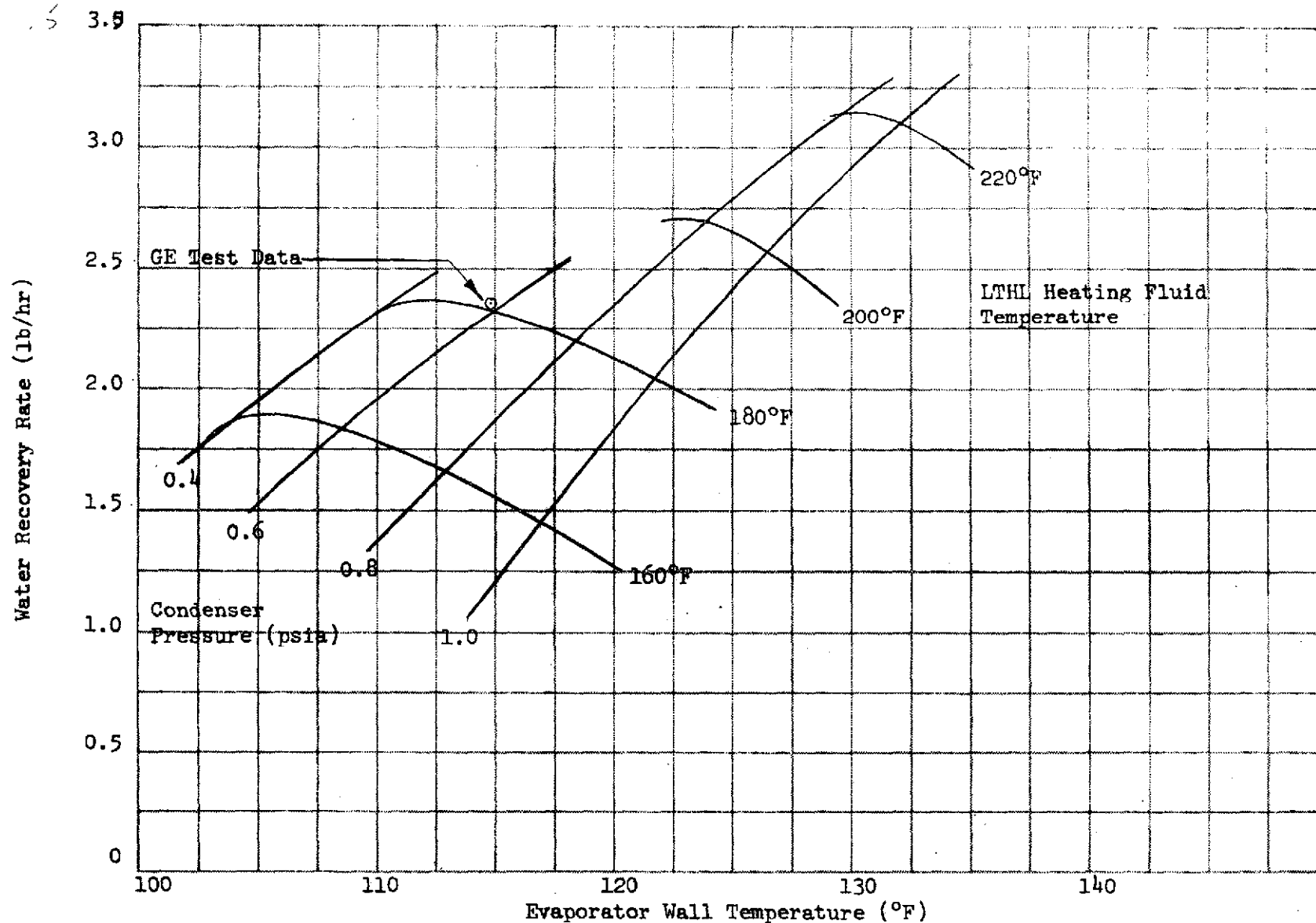


FIGURE 5.18 RITE SYSTEM OPERATIONAL ENVELOPE  
(CONDENSER COOLANT TEMPERATURE 35°F)

The performance of the RITE system as the condenser back pressure is varied with constant LTHL temperature is shown in Figures 5.19 and 5.20. The 180°F LTHL evaporator inlet temperature corresponds to the General Electric operating point. A peak condensation or recovery rate is reached at a condenser pressure of 0.63 psia. Below this pressure the saturation temperature of the liquid approaches the temperature of the condenser wall and therefore the condensation rate falls. The vapor then escapes out the vent which results in a low conversion efficiency. Increasing the condenser back pressure increases the condensation efficiency; however, the pressure head driving force from the evaporator to the condenser is reduced resulting in a lower evaporator output. A tradeoff must therefore be made between recovery efficiency and total recovery rate. The results indicate that General Electric has selected as its operating point the peak recovery rate and a recovery efficiency greater than 98%.

The evaporator heat input requirements corresponding to the water generation rates of Figure 5.18 are shown in Figure 5.21. The G.E. instrumentation does not yield evaporator heat requirements; however, the heat load generated by the G189A corresponding to the G.E. operating point is noted in the figure. The 2500 Btu/hr used by the evaporator is less than half of the 5120 Btu/hr available in the isotope. Assuming that 900 Btu/hr are used to heat the water tanks, there is a heat loss to the environment of approximately 1500 Btu/hr in the low temperature heating loop. If this heat loss were reduced, the system water recovery rate of 2.3 lb/hr could be improved by a factor of 1-1/2 or more. Alternately, if only 2.3 lb/hr were required, the isotope inventory could be reduced from 1500 watts to 1060 watts.

Figures 5.22 and 5.23 show evaporator vapor temperature and pressure as a function of evaporator wall temperatures. The G.E. test point is 105°F vapor temperature at 1.1 psia vapor pressure.

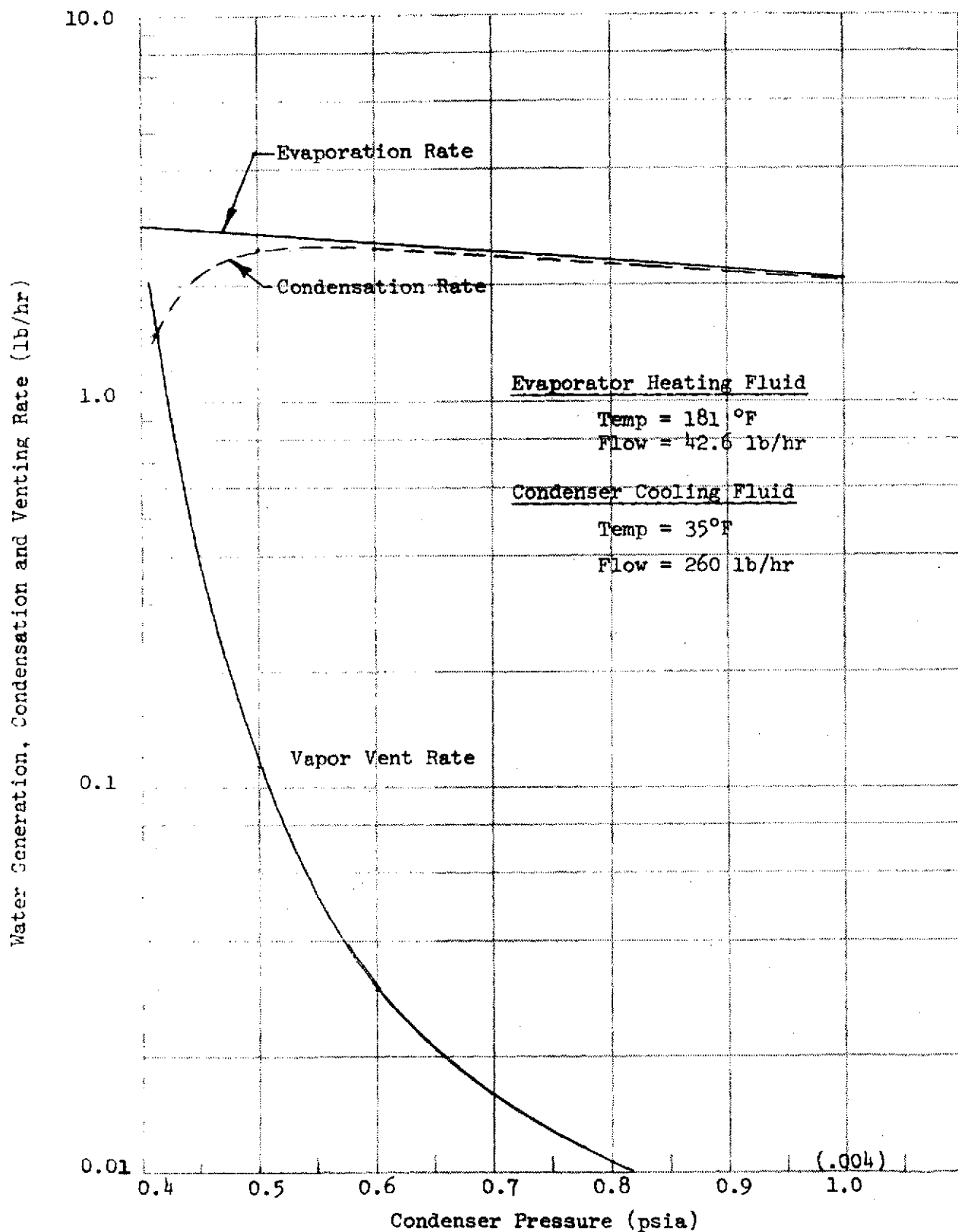


FIGURE 5.19 EVAPORATOR AND CONDENSER PERFORMANCE EVALUATION  
(at G.E. Operating Point of 181°F LTHL Temperature)



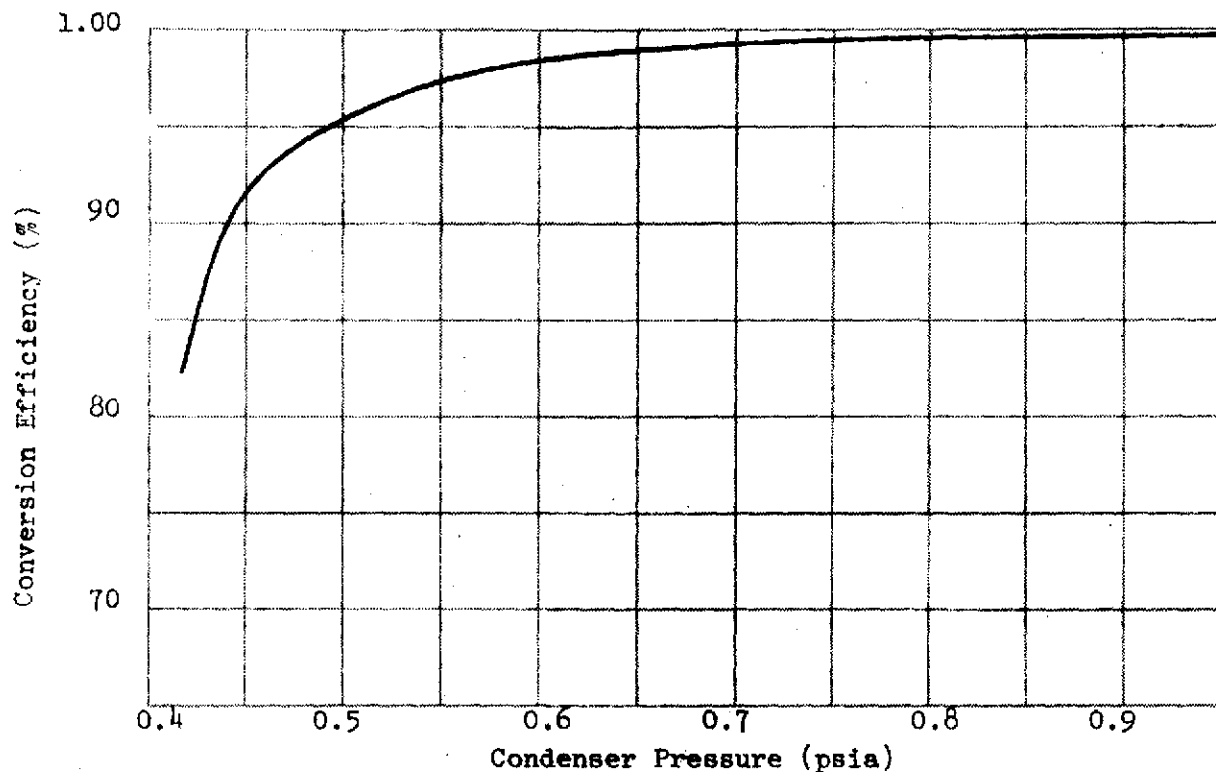


FIGURE 5.20 RITE SYSTEM WATER PURIFICATION EFFICIENCY

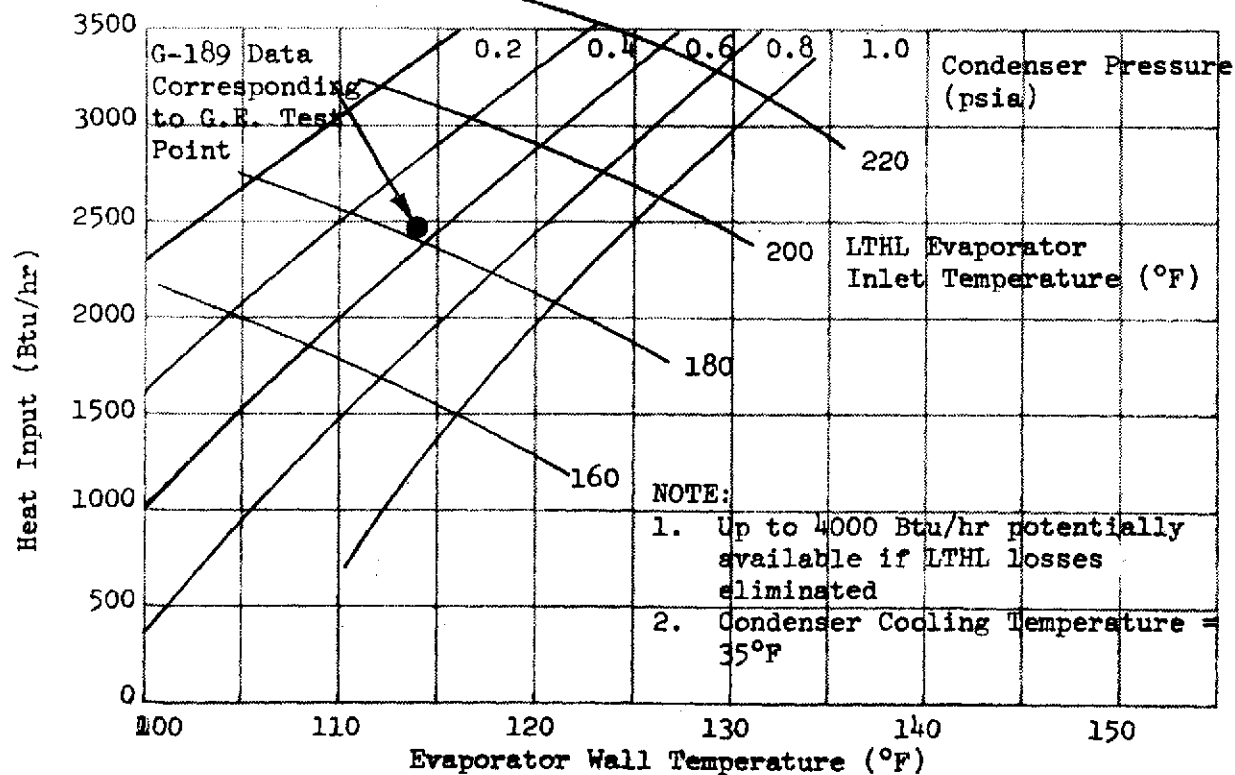


FIGURE 5.21 HEAT INPUT TO EVAPORATOR

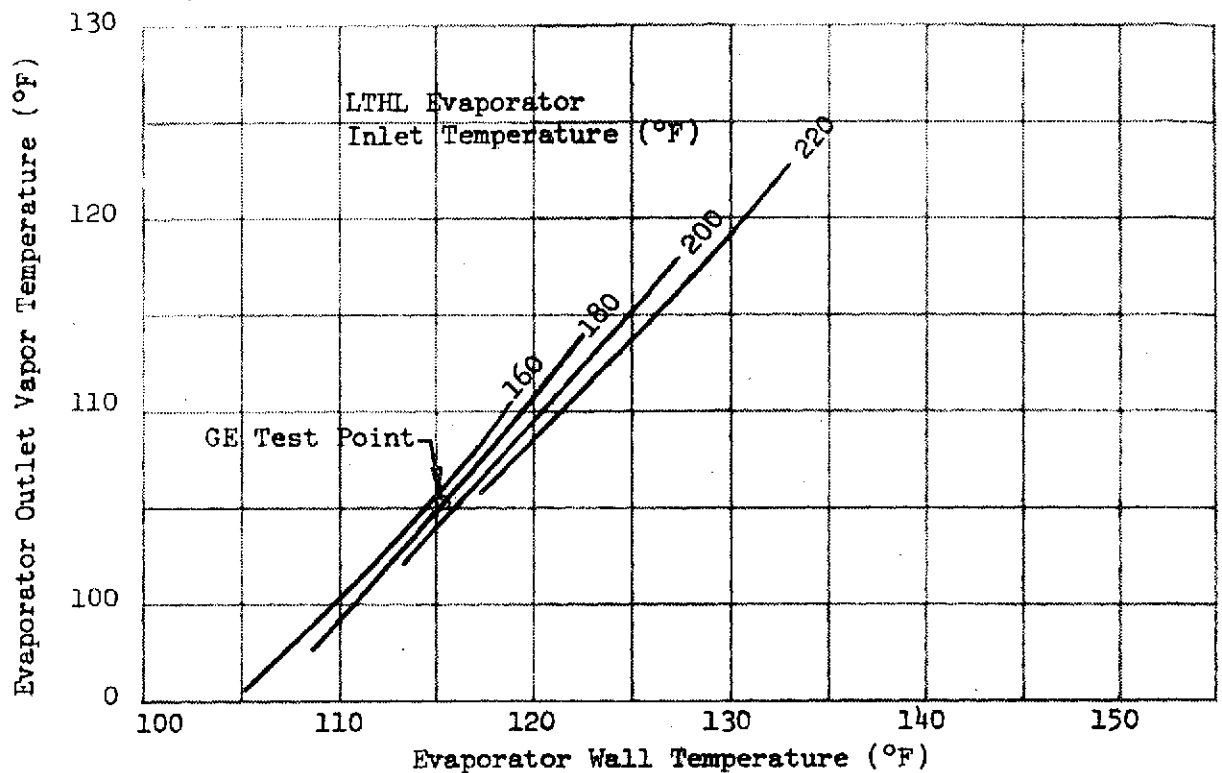


FIGURE 5.22 EVAPORATOR OUTLET TEMPERATURE AS A FUNCTION OF WALL TEMPERATURE

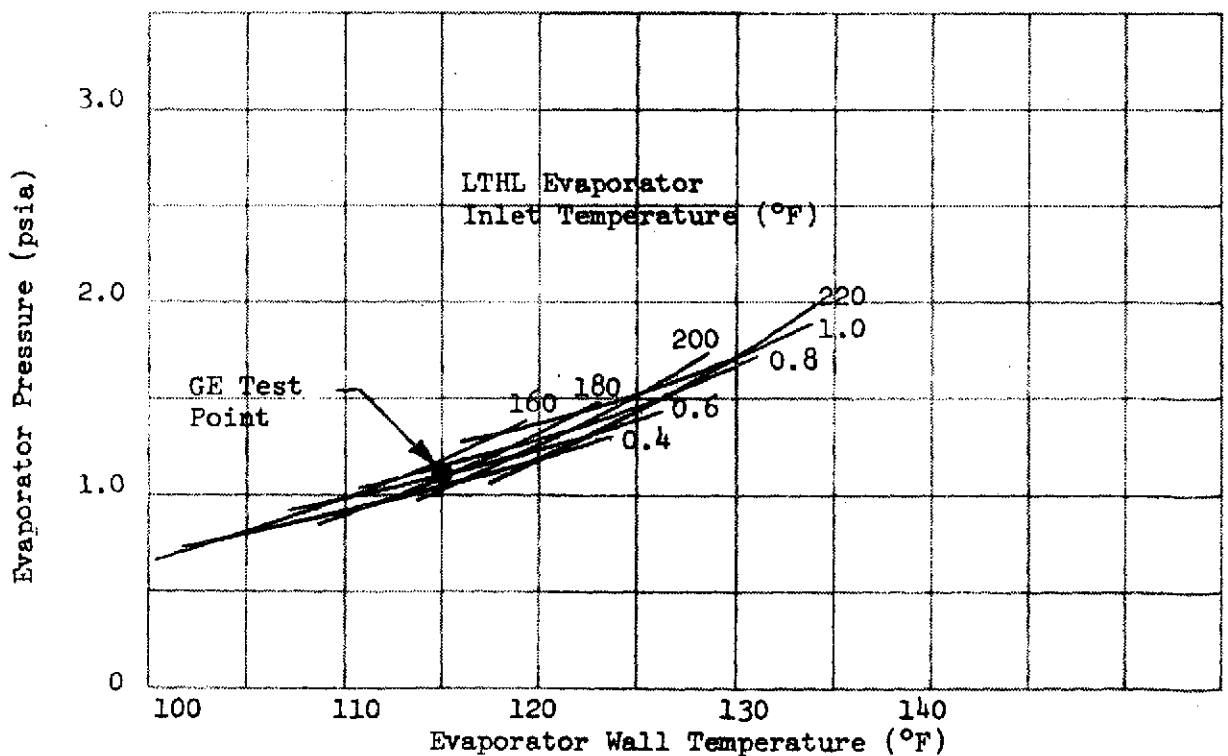


FIGURE 5.23 EVAPORATOR PRESSURE AS A FUNCTION OF WALL TEMPERATURE

## Section 6

### RITE Simulation Design Applications

Once the basic configuration of the system under study has been simulated by use of the G-189A Program, the simulation can be used in support of both detail component design, system design and system integration. Examples of G-189 program RITE simulation applications to RITE system design analyses and investigations are listed below:

1. Effect of Vehicle Equipment Limitations on RITE Performance
  - a. Available vehicle radiator coolant temperature level and heat load capability.
  - b. Amount and temperature level of heat available
  - c. Electrical power limitations
  - d. Oxygen supply limitation
  - e. Size and dimensional limitations for RITE
2. Effect of component design on RITE Performance
  - a. Effect of RITE component operating temperature levels
  - b. Thermal effectiveness of evaporator
  - c. Thermal effectiveness of condenser
  - d. Size and efficiency of other components
3. System Integration
  - a. Effect of venting gases to cabin
  - b. Heat loss to cabin
  - c. Nuclear safety
  - d. Timeline for crew activities in water/waste management events
  - e. Mission life
  - f. Controls and instrumentation

Analysis of each of these problems is beyond the scope of this study. Therefore, to illustrate the simulation design potential, the effects of available coolant temperature on system effectiveness and the study of potential design changes in the evaporator and condenser were chosen for study.

## 6.1 Condenser Coolant Inlet Temperature

The radiator outlet coolant temperatures available to the subsystems aboard advanced manned vehicles such as Space Shuttle and Sortie Lab will generally range from 40°F to over 100°F. The current RITE system is operating at a coolant temperature of 32°F which is clearly optimistic. Since other subsystems such as the batteries will make demands for the lower coolant temperatures, it is possible that the lowest coolant temperature available to the RITE would be in the 50 to 60°F range.

Figures 6.1, 6.2, and 6.3 detail the operational envelope of the RITE system for coolant temperatures of 35, 40 and 50°F. The system output is plotted as functions of evaporator wall temperature, evaporator heating fluid (LTHL heating fluid) inlet temperature and condenser pressure.

For all coolant temperatures, increasing the LTHL temperature with constant condenser pressure improved the system output, until a condensation limit is reached. At a coolant temperature of 50°F and the LTHL temperature of 220°F, the resultant vapor output temperature of the evaporator was too high for the condenser to be operational and as a result all the vapor was lost through the vent.

The system output at the baseline LTHL temperature of 180°F was cross-plotted in Figure 6.4 to show clearly the effect of condenser coolant temperature. As the condenser temperature increases the partial pressure of water required for condensation increase and therefore the total pressure of the condenser must increase. Increased condenser pressure, however, limits evaporator output due to a higher pressure drop or flow resistance in the vapor loop. Therefore, the optimum output of the system as the condenser temperature is increased from 35 to 40 to 50°F drops from 2.3, to 2.2 and to 2.13 lb/hr. Table 6.1 shows the key system parameters for the three condenser temperatures for the same condenser total pressure of 0.63 psia. In each case the evaporator output is the same, however, the ability of the condenser to condense the flow, which is defined as the system efficiency, drops as condenser temperature increases. Finally, at 50°F the condenser wall temperature of 83.2°F and the corresponding water partial pressure of 0.58 psi is too high for effective condensation.

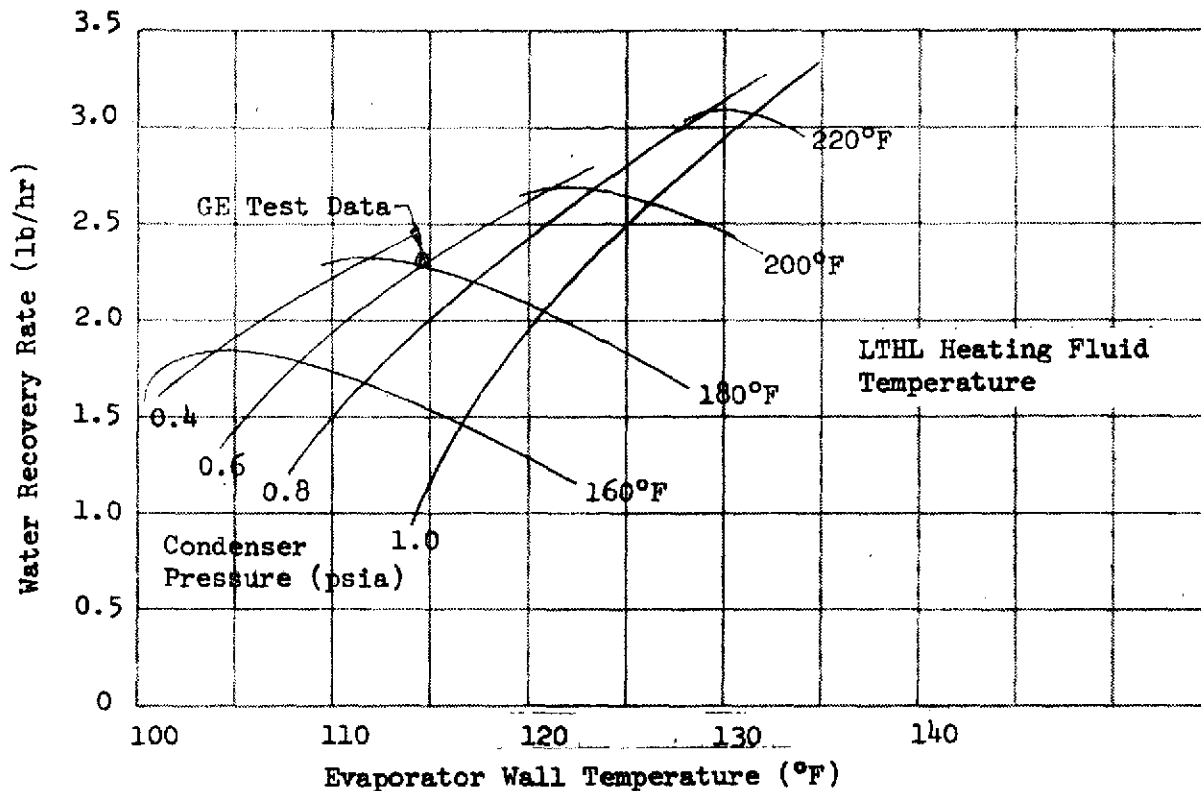


FIGURE 6.1 RITE SYSTEM OPERATIONAL ENVELOPE CONDENSER COOLANT TEMPERATURE 35°F

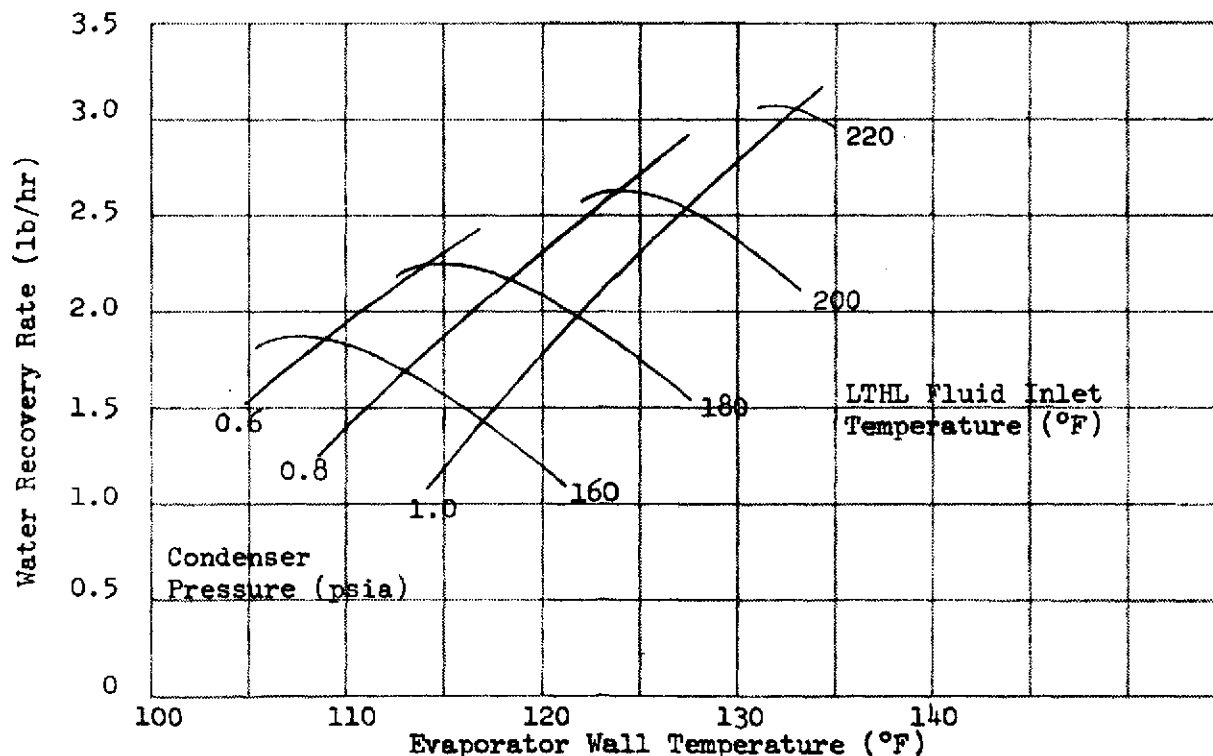


FIGURE 6.2 RITE OPERATIONAL ENVELOPE WITH CONDENSER COOLANT TEMPERATURE EQUAL TO 40°F

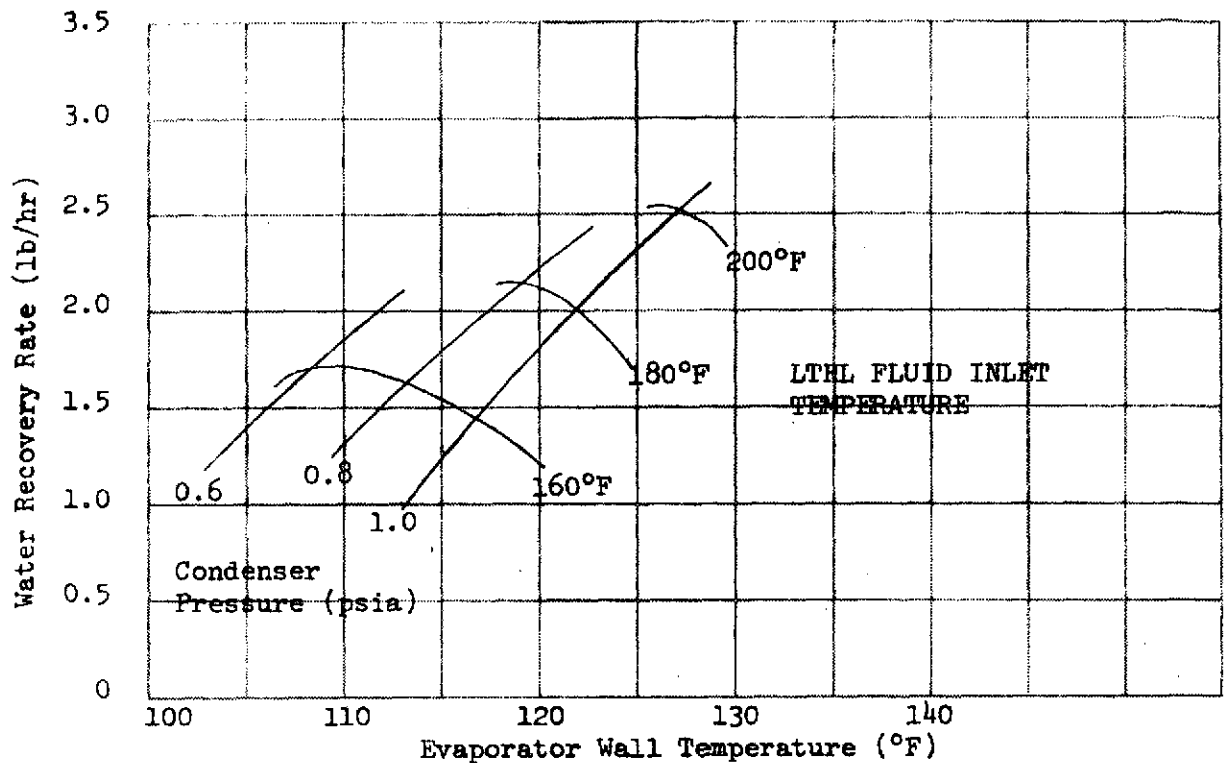


FIGURE 6.3 RITE OPERATIONAL ENVELOPE WITH RADIATOR COOLANT TEMPERATURE EQUAL TO 50°F

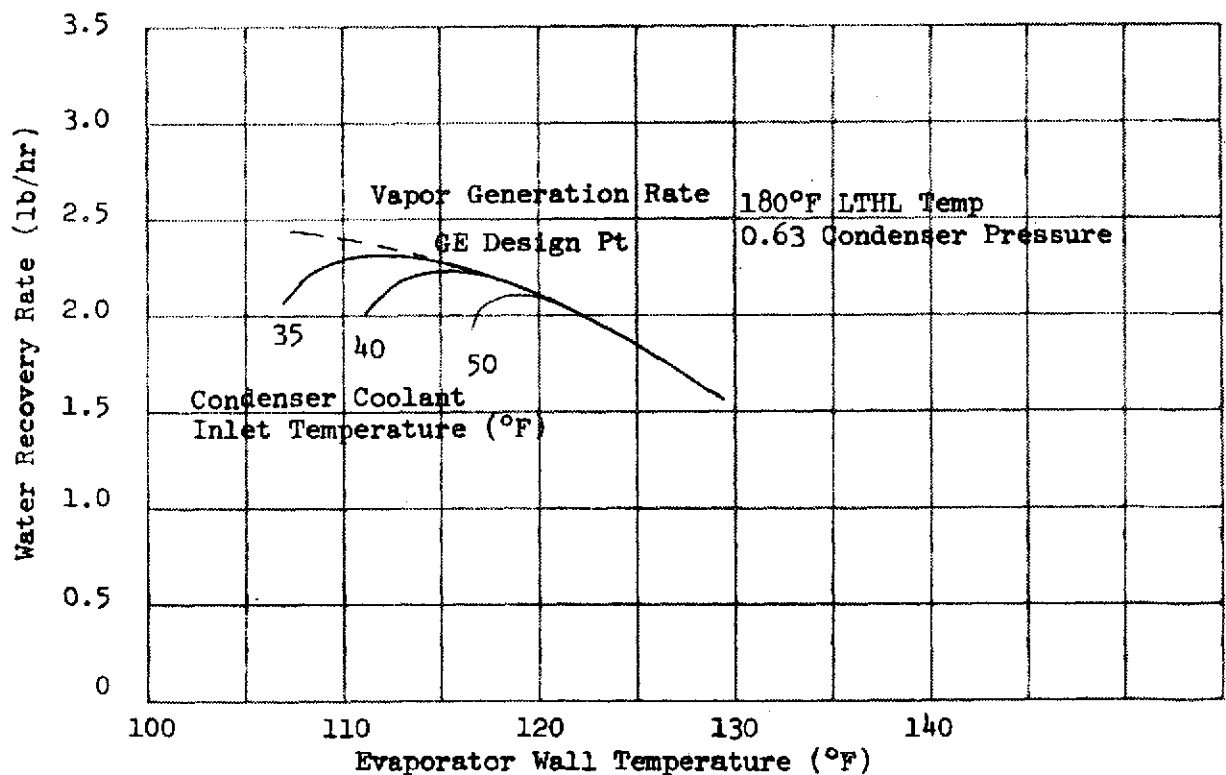


FIGURE 6.4 WATER RECOVERY RATE AS A FUNCTION OF CONDENSER COOLANT TEMPERATURE

TABLE 6.1

KEY SYSTEM PARAMETERS FOR RADIATOR COOLANT TEMPERATURES  
OF 35, 40 and 50°F

Condenser Temperature (°F)	35**	40	50
System Output (lb/hr)	2.301	2.207	0
System Efficiency*			
$\left( \frac{\text{condensation rate}}{\text{evaporator rate}} \right)$	$\frac{2.301}{2.3297} = .9877$	$\frac{2.207}{2.3297} = .947$	$\frac{0}{2.3297} = 0$
Condenser Wall Temperature (°F)	77.629	80.76	83.3
Condenser Total Pressure (psia)	0.63	0.63	0.63
Partial Pressure of Water Vapor (psia)	0.46	0.51	0.58
Evaporator Vapor Temp. (°F)	104.39	104.39	104.39
Evaporator Pressure (psia)	1.085	1.085	1.085
Evaporator Heat Input (Btu/hr)	2412	2412	2412

\* Efficiency as defined here does not include water losses to the incinerator.

\*\* Current system design point.

If in future systems the available coolant temperature is 50°F or higher, the vapor pressure and, therefore, the total pressure in the condenser, and the evaporator LTHL temperature level must all be increased in order to achieve a system with a specified output of 2.3 lbs./hr.

## 6.2 Component Design

The key parameters which control the effectiveness of the system components are the component operating temperatures, pressures and thermal effectiveness. The effect of varying these parameters for the key system components, the evaporator and condenser, will be investigated in this section.

### 6.2.1 Evaporator

The computed system output for overall thermal conductance, UA, values at the design point value of 92.6 Btu/hr °F and for 75 and 150 Btu/hr-°F are shown in Figures 6.5, 6.6 and 6.7. Increasing the effectiveness of the evaporator has the result of increasing the average wall temperature to a point closer to the LTHL heating fluid inlet temperature. The increased temperature improves the heat transfer to the evaporator bulk liquid and increases the evaporator pressure. The higher pressure in turn provides a larger driving force to improve vapor throughput.

The output of the RITE system as a function of evaporator UA for the current design conditions is summarized in Table 6.2. Increasing UA from 92.6 Btu/hr°F, the design point, to 150 BTU/hr °F results in an increase in wall temperature from 114.38°F to 116.19°F and an increase in water output from 2.3 lb/hr to 2.42 lb/hr.

The evaporator effectiveness can be improved by increasing the overall thermal conductance, UA, between the heating fluid and the vapor. This can be accomplished by placing the fluid tubes on the inside in direct contact with the liquid rather than on the outside wall of the evaporator. Other possibilities include increasing the number of tubes, use of better conducting materials such as copper, modifying structural dimensions, or increasing evaporator size.



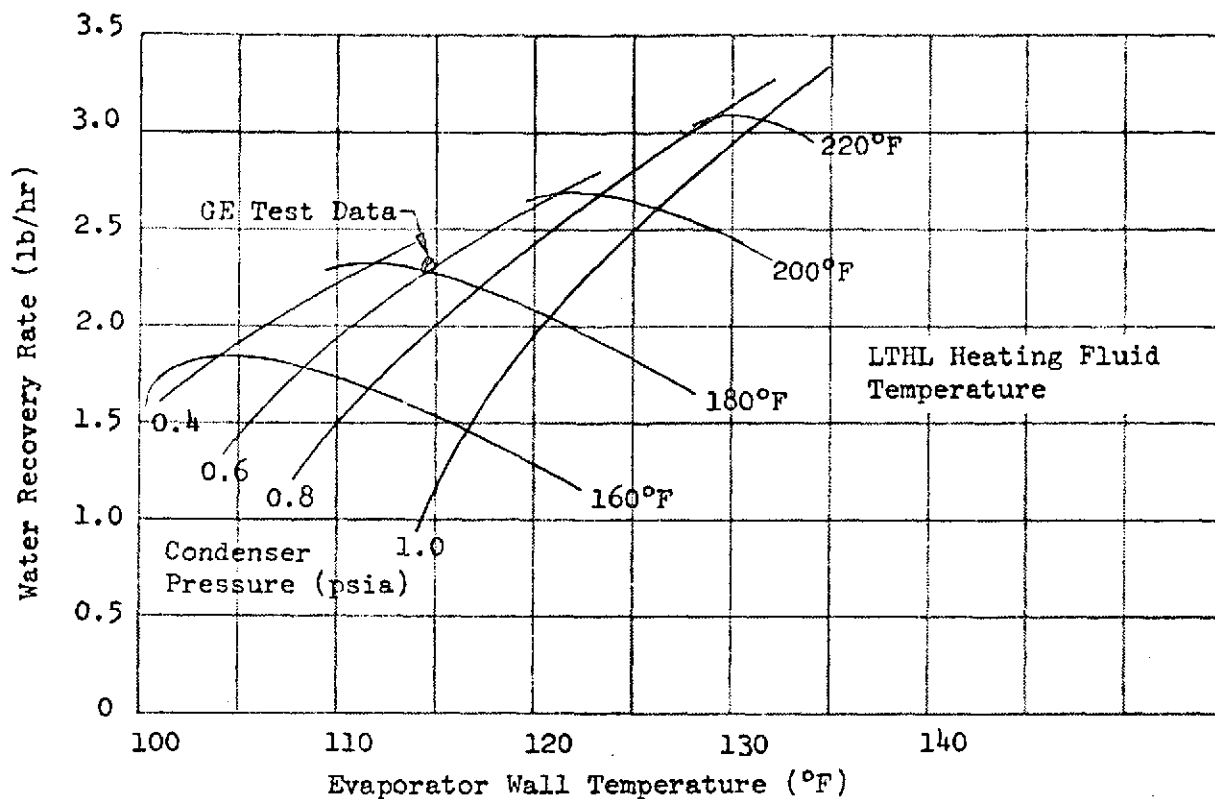


FIGURE 6.5 RITE OPERATIONAL ENVELOPE WITH EVAPORATOR  
 $UA = 92.6 \text{ Btu/hr-}^{\circ}\text{F}$

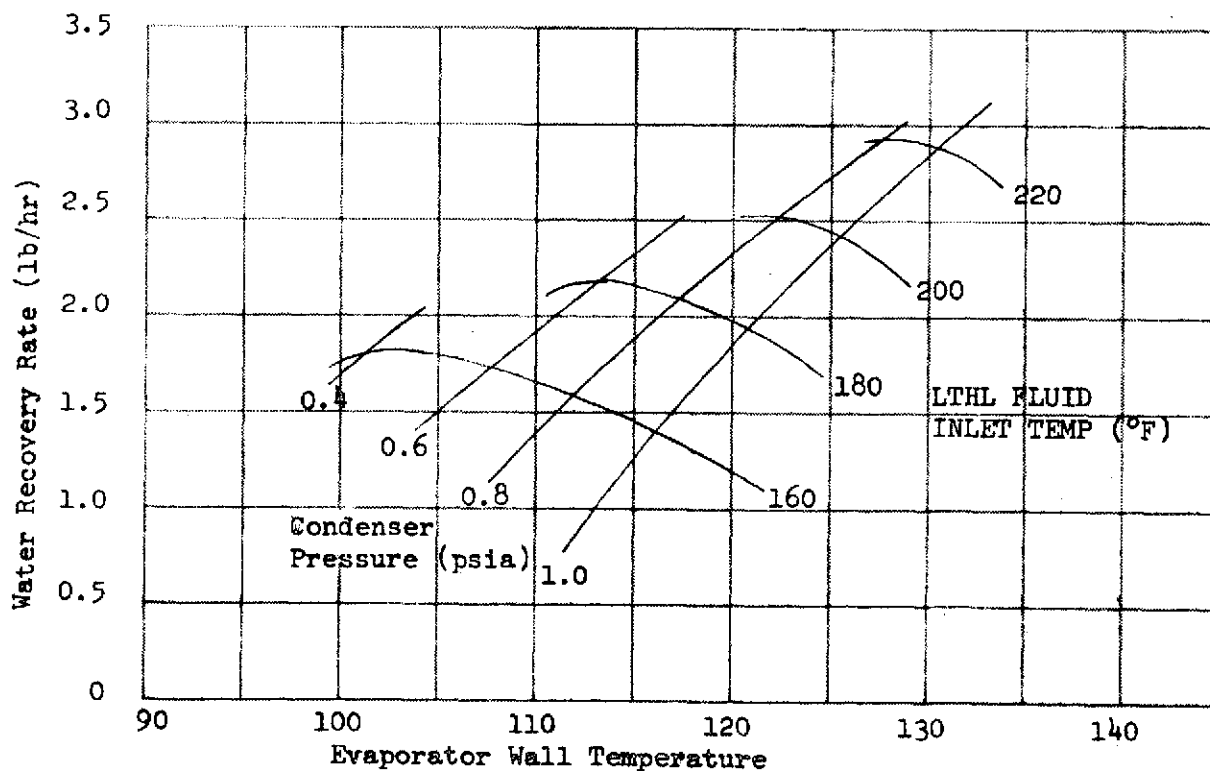


Figure 6.6 RITE OPERATIONAL ENVELOPE WITH EVAPORATOR  
 $UA = 75 \text{ Btu/hr-}^{\circ}\text{F}$

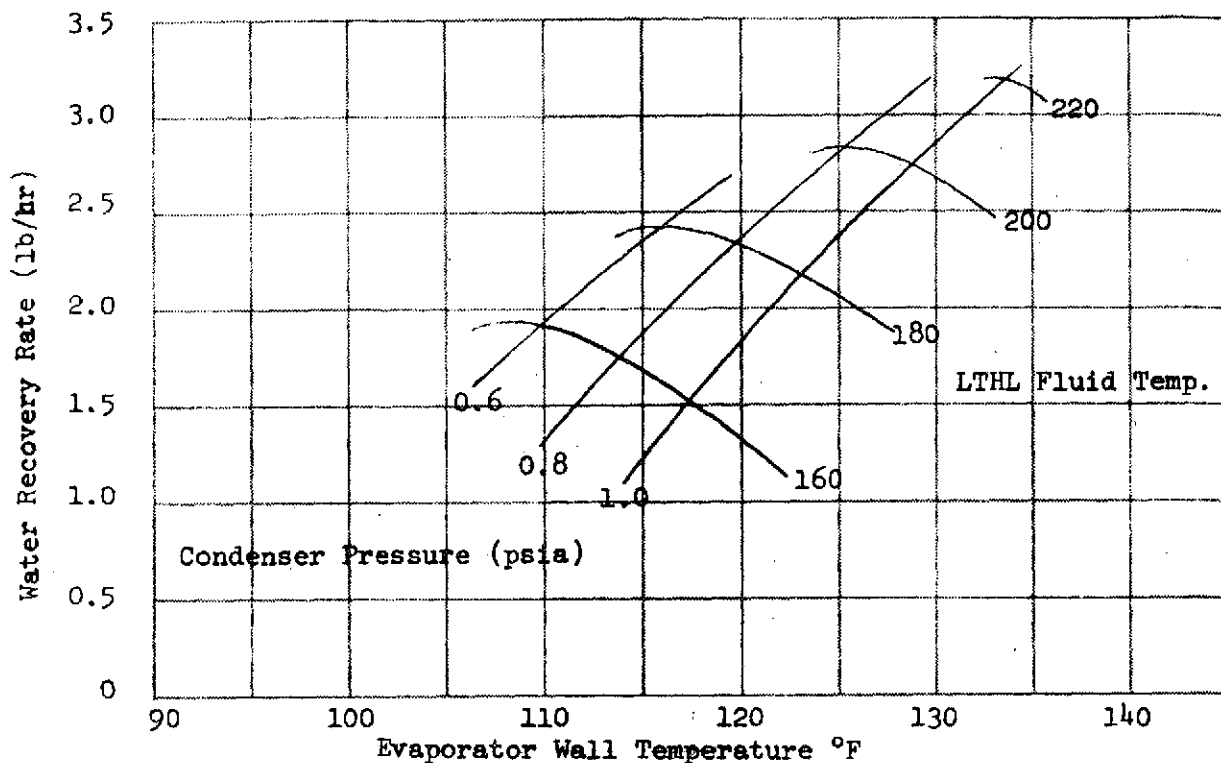


FIGURE 6.7 RITE OPERATIONAL ENVELOPE WITH EVAPORATOR  
UA=150 Btu/hr-°F

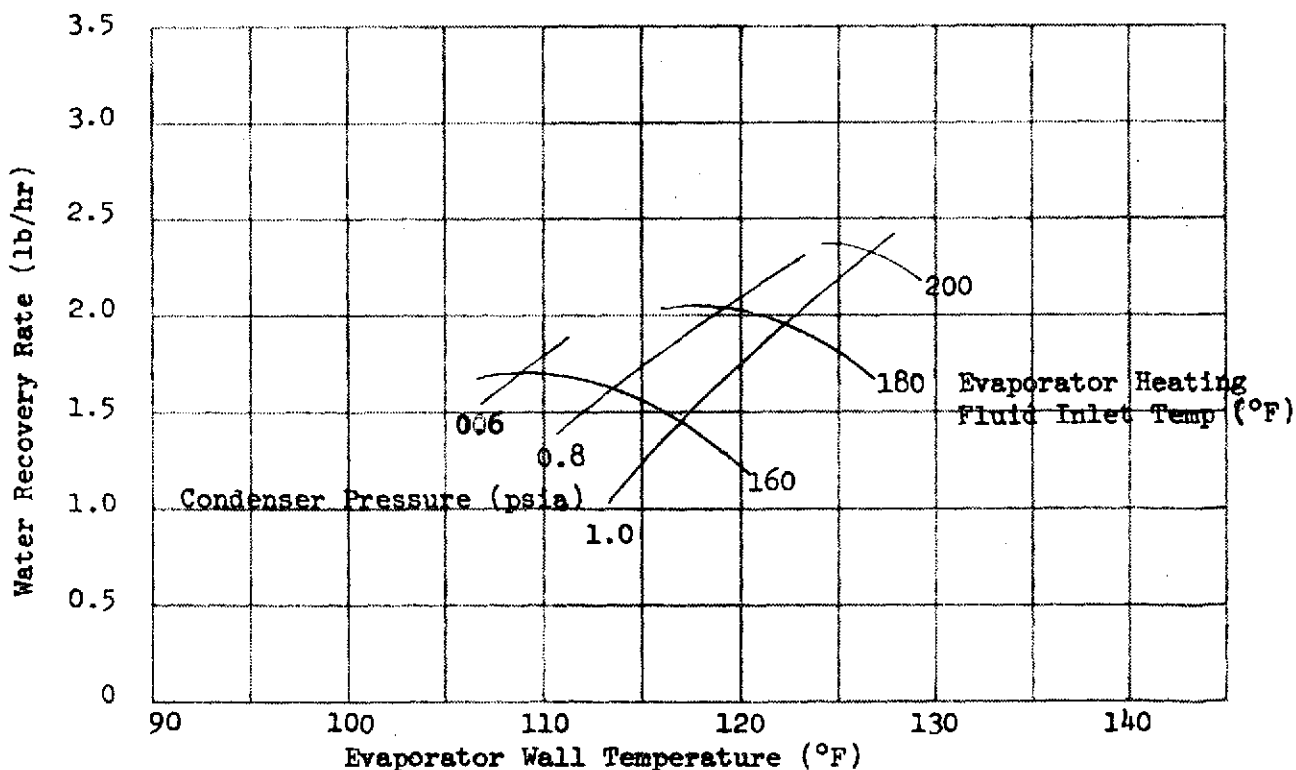


FIGURE 6.8 RITE OPERATIONAL ENVELOPE WITH CONDENSER  
UA=75 Btu/hr-°F

TABLE 6.2

SYSTEM OUTPUT FOR EVAPORATOR UA'S OF 75, 92.6

AND 150 BTU/HR °F

Evaporator UA (Btu/hr °F)	75	92.6**	150
System Output (lb/hr)	2.2342	2.301	2.4235
System Efficiency*			
$\left( \frac{\text{condensation rate}}{\text{evaporation rate}} \right)$	$\frac{2.2342}{2.238} = .998$	$\frac{2.301}{2.3297} = .9877$	$\frac{2.435}{2.4707} = .9856$
Condenser Total Pressure (psia)	0.63	0.63	0.63
Evaporator Vapor Temp. (°F)	103.62	104.39	106.19
Evaporator Pressure (psia)	1.0606	1.085	1.1433
Evaporator Heat Rate (Btu/hr)	2314	2412	2556

\* Efficiency does not include water loss to the incinerator

\*\* Current design point

### 6.2.2 Condenser

The operational envelope for the RITE system with condenser UAs of 117, 150, and 75 Btu/hr°F are shown in Figures 6.8, 6.9 and 6.10. Improving condenser effectiveness has a similar effect as does decreasing the condenser coolant inlet temperature. A lower wall temperature results in a lower partial pressure for water and therefore a greater condensation rate. Table 6.3 shows the system output for the three UAs at the system design point. Increasing UA beyond the design point of 117 results in a small improvement in efficiency. Decreasing the UA to 75 Btu/hr-°F results in a wall temperature of 83°F and therefore a partial pressure of water vapor higher than that required for condensation.

Improved condenser thermal design can be achieved using techniques similar to those suggested for the evaporator.

### 6.3 Design Study Summary

The recent prototype RITE 180-day operation has verified the basic RITE design principles. The RITE system simulation was used to test the application of these principles to various spacecraft applications and under various spacecraft restraints. The design section has shown that the current RITE configuration is able to accommodate water recovery requirements of up to 3.0 lb/hr by increasing LTHL temperatures to 220°F. Higher recovery rates can be achieved by increasing the UA of the condenser and the evaporator.

The available condenser coolant temperatures effect the system efficiency by reducing the condensation rate. Most spacecraft radiator coolant temperatures are in the range of 40°F to 60°F. At these temperatures the recovery rate drops 10 percent from the value at the prototype operating temperature of 32°F.

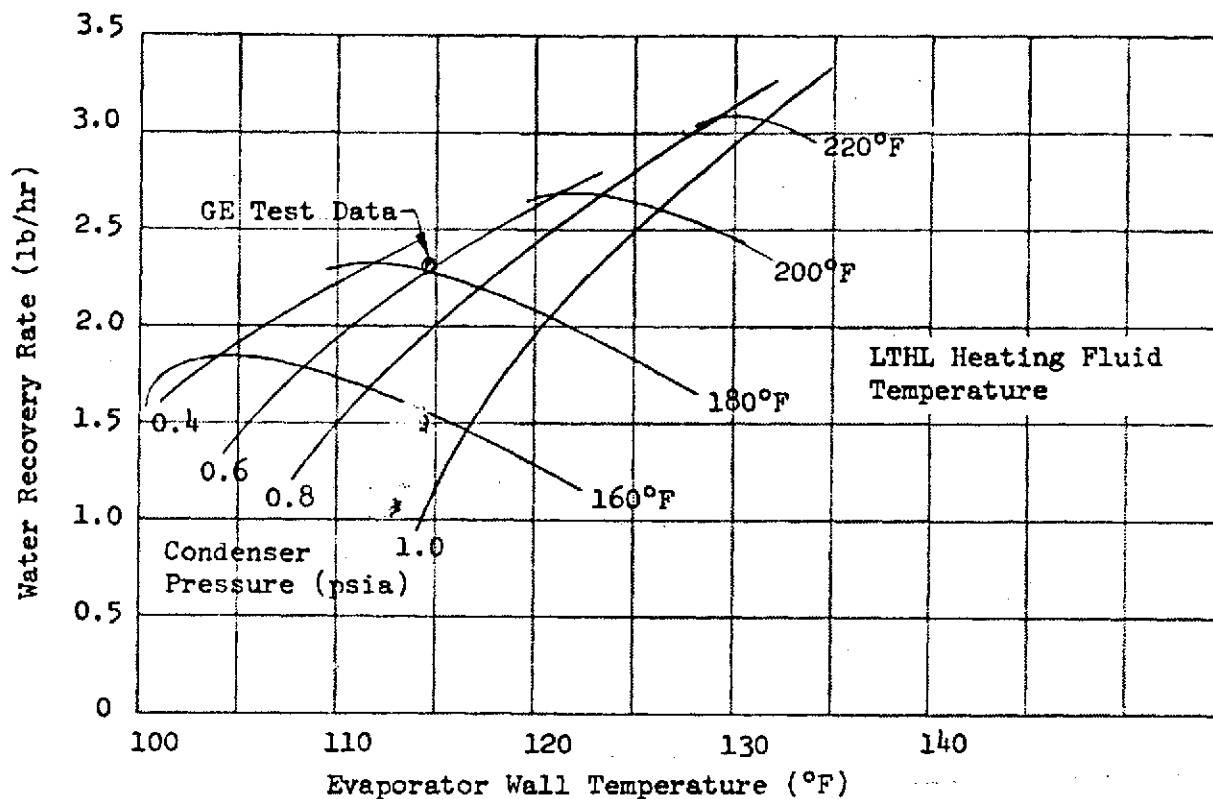


FIGURE 6.9 RITE OPERATIONAL ENVELOPE WITH CONDENSER  
 $UA = 117.0 \text{ Btu/hr-}^\circ\text{F}$

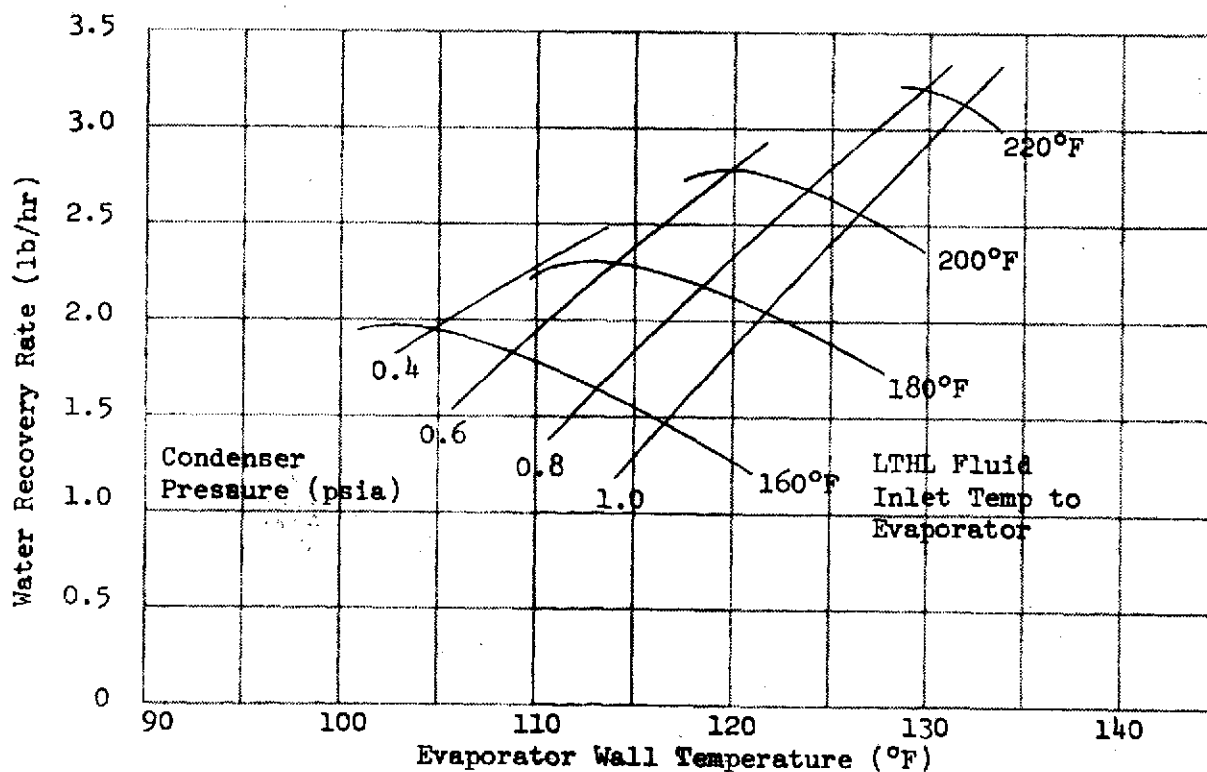


FIGURE 6.10 RITE System Operational Envelope with Condenser  
 $UA=150 \text{ Btu/hr-}^\circ\text{F}$

TABLE 6.3

KEY SYSTEM PARAMETERS FOR CONDENSER UA'S OF  
75, 117 AND 150 Btu/hr °F

Condenser UA (Btu/hr °F)	75	117**	150
System Output (lb/hr)	0	2.301	2.3175
System Efficiency*	0	2.301	2.3175
$\left( \frac{\text{condensation rate}}{\text{evaporation rate}} \right)$	$\frac{0}{2.3297} = 0.0$	$\frac{2.301}{2.3297} = .9877$	$\frac{2.3175}{2.3297} = .995$
Condenser wall temperature, (°F)	83	77.63	70.36
Condenser total pressure (psia)	0.63	0.63	0.63
Condenser coolant temperature (°F)	35	35	35

\* Efficiency does not include losses to the incinerator

\*\* Current system design point

## Section 7

### FAILURE MODE ANALYSES

One of the more productive applications of the G-189A RITE simulation is to predict the RITE response to proposed system failures. The failure modes analyzed can be divided into two categories. In the first category are failures which would only result in reduced system efficiency or at worst system shutdown. The second type of failures have potential impact on the integrity of the nuclear isotopes and therefore would threaten the safety of the crew.

Example of the first type of failures include: leakage in the vapor, flush water or potable water loop; pump failures; leakage or loss of flow in coolant loop, and solids transfer equipment failures. These failures can most likely be repaired in orbit, by resupply, or at worst would result in premature ending of the mission.

Failures that impact the low temperature nuclear isotope include all failures which effect the operation of the low temperature heating loop. Failures which effect the high temperature isotope capsule include cooling failure or a failure which would continuously add heat to the capsule. The isotope cooling failure could be the result of the loss of the heat pipe. Continuous heating of the isotope could result if oxidation of solids in the incinerator was to occur without periodic venting of the product gasses.

The failures simulated and discussed in this section are the cooling loop failure, the incinerator continuous oxidation failure, and the low temperature heating loop failure.

## 7.1 Cooling Loop Failure

Cooling loop failure could be the result of a loss of flow due to a pump or controller failure, or a loss of coolant from the loop. Each of these failures would result in the loss of condensing capability in the condenser. The water vapor generated in the evaporator would then be lost to space through the vent unless the system was shut down.

The response of the system to the cooling loop failure is summarized in Figures 7.1 and 7.2. The temperature of the condenser wall rises until it approaches the temperature of the incoming vapor. The higher wall temperature results in a loss of condensing capability and the water recovery rate rapidly drops to zero. The incoming vapor is lost to space through the vent. The evaporator vapor generation rate drops since the vapor accumulated in the condenser builds up the condenser pressure and therefore limits the pressure driving force from the evaporator to the condenser. The evaporator would tend to increase in temperature and pressure, however, the low temperature heating loop controller bypasses the heating fluid at a vapor temperature of 105°F and therefore limits the pressure to the corresponding saturation pressure of 1.1 psia.

The net result of the cooling failure is that the water recovery capability of the system is lost. Unless the system is shut down the evaporator low signal will automatically feed water to the evaporator from the wash water system. The water would be evaporated and then lost to the condenser vent. When the system is shut down the evaporator cooling load is transferred to the emergency heat exchanger keeping the LTHL in thermal equilibrium. Heat from the LTHL can be added to the evaporator and water recovery may proceed once the condenser problem is repaired.

## 7.2 Incinerator Vent Valve Failure

The most serious incinerator failure mode identified to date is the failure of the cycle to switch from the oxidation mode to the vent mode. This failure



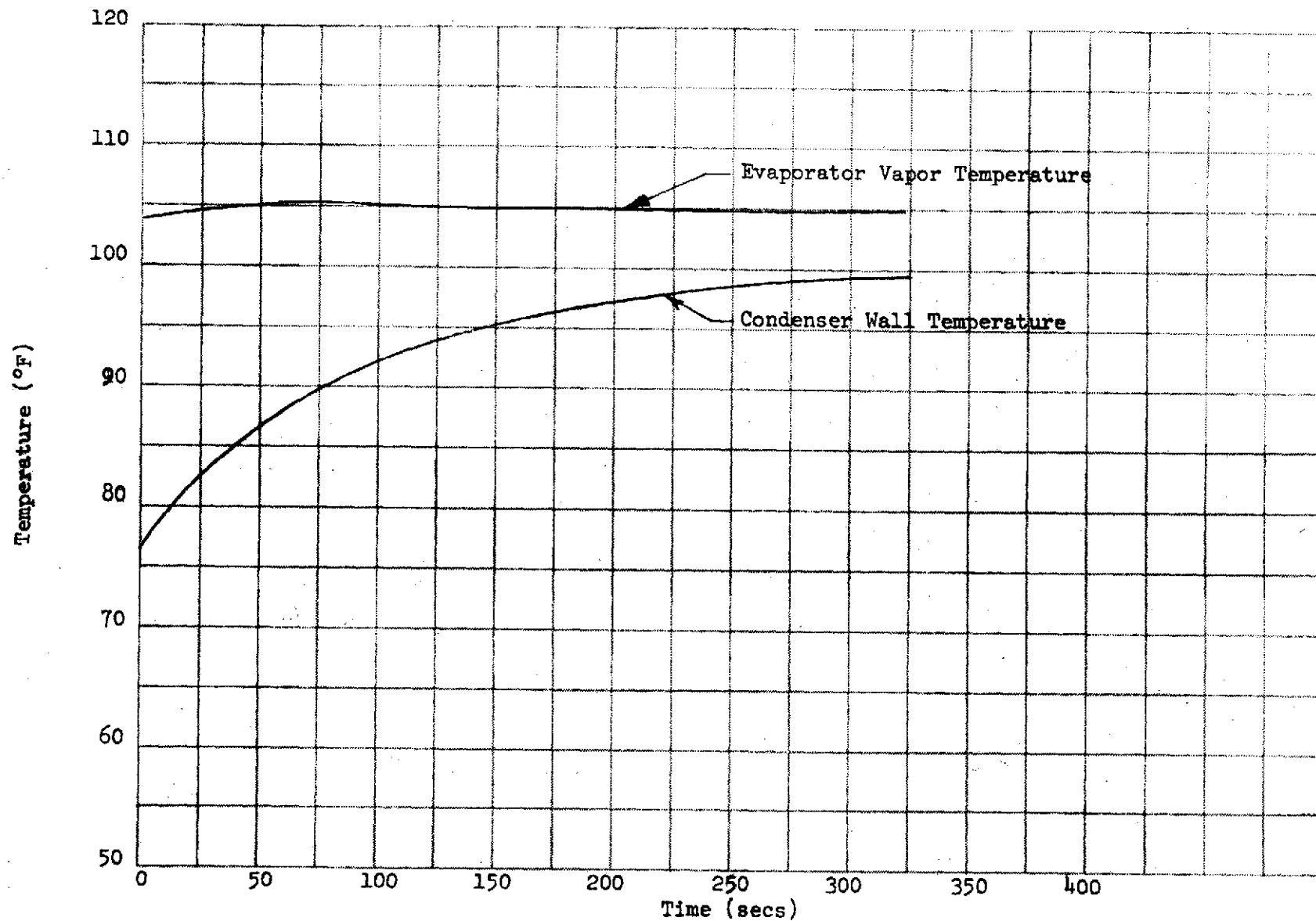


FIGURE 7.1 TEMPERATURE RESPONSE TO LOSS-OF-COOLING ACCIDENT

7.2

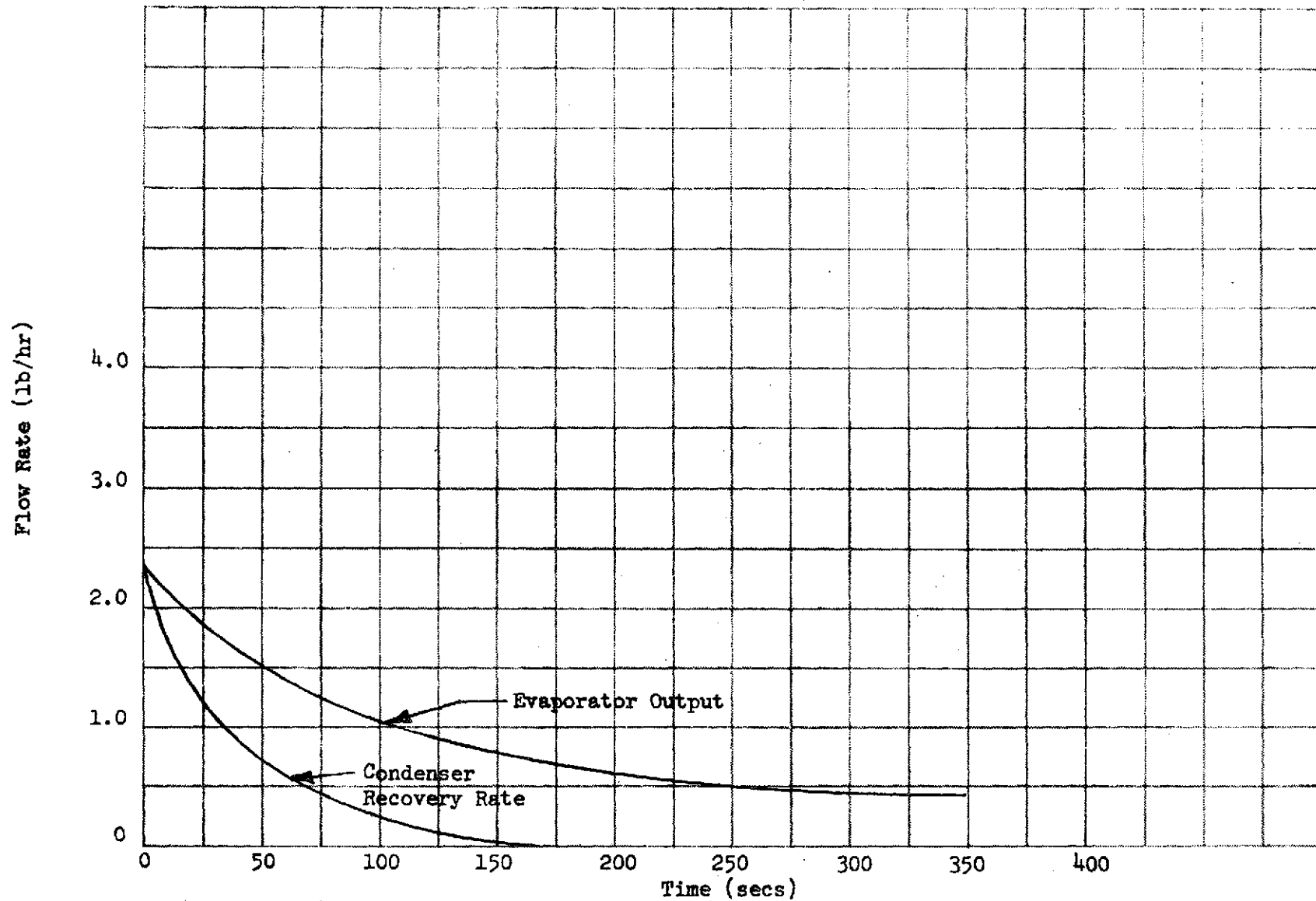


FIGURE 7.2 CONDENSER AND EVAPORATOR FLOW RATES AS A RESULT OF LOSS-OF-COOLING ACCIDENT

could occur if the vent valve to vacuum failed to open. Oxygen would continue to react with the solids and generate approximately 1100 Btu/hr. The possibility exists that this heat input would increase the temperature of the isotope in the heat block beyond its design point. Failure of the isotope capsule would then release the extremely toxic Pu-238 isotope to the cabin atmosphere.

The temperature response of the incinerator and the heat block as a result of the vent failure is shown in Figure 7.3. The temperature of the incinerator reaches only 1450°F before all solids in the shuttle load are burned. This is only 150°F higher than its normal operating point. The temperature rise is limited since heat generated by the chemical reaction is readily absorbed by the heat block. The isotope and heat block temperatures drop as they give up heat to the incinerator in the drying cycle. When the failure occurs the heat block absorbs the excess heat to return to its normal operating point sooner than normal. All the solids are burned and the heat generation rate drops to zero at the end of the oxidation cycle.

The increment of heat imposed on the heat block as a result of the vent failure is absorbed by its thermal mass and is in turn rejected to the environment by the heat pipe. The isotope capsule design point temperature is not exceeded.

### 7.3 Low Temperature Heating Loop Failure Analyses

Failure analysis of the Low Temperature Heating Loop (LTHL) is especially important due to potential nuclear hazards posed by the isotope heat source. A complete nuclear safety analysis from ground handling, launch and recovery would be required before approval is given for use of the nuclear capsule in a space system. The thermal analysis by the G-189A program of the LTHL is a preliminary effort to identify potential failures and corresponding effects prior to application design development.

The LTHL provides heat to the evaporator, potable water storage tanks, flush water tank, and the water heater. The heat source, a 1500 watt Pu-238 isotope, is enclosed in a large diameter water tank which serves as a

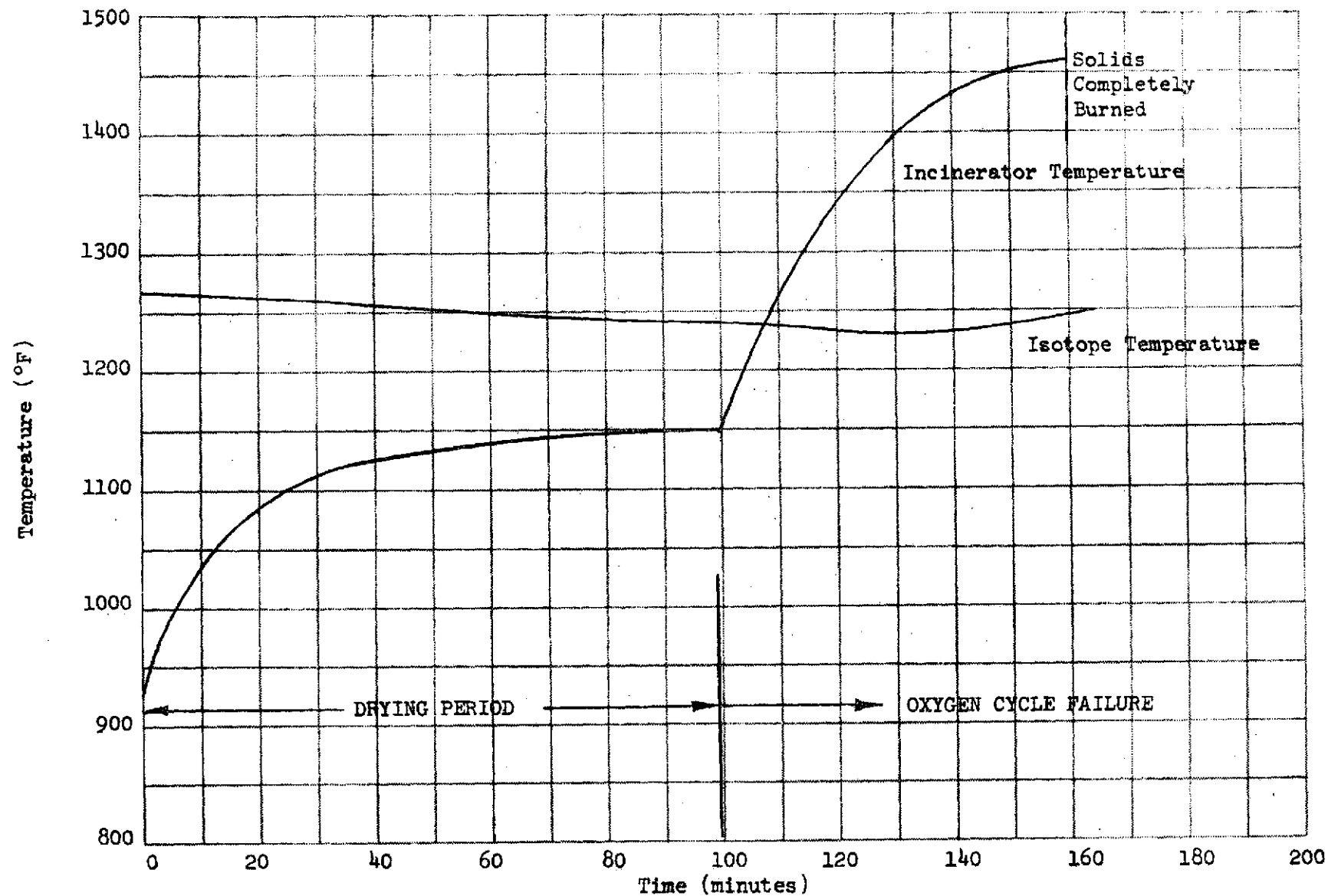


FIGURE 7.3 TEMPERATURE RESPONSE TO THE INCINERATOR OXYGEN CYCLE FAILURE

nuclear shield. The water tank, which will be discussed later, also serves to dampen isotope temperature excursions as a result of any system failures. Out of the total 5120 Btu/hr generated by the isotope approximately 2700 is used by the evaporator, 900 Btu/hr is used to heat the tanks and 1500 Btu/hr is lost through line losses and through the shielding tank wall.

Emergency cooling for the isotope is provided through an on line liquid-liquid heat exchanger located in the LTHL just upstream from the isotope heat source. A water coolant loop, with bypass control around the heat exchanger, provides the coolant supply. The bypass controller senses the LTHL fluid temperature at the heat source exit and at 190°F will direct flow to the emergency heat exchanger.

The potential failures identified for the LTHL are listed below:

1. Loss of coolant flow due to pump loss or line blockage.
2. Loss of coolant due to leakage.
3. Loss of evaporator heat removal (bypass on or evaporator drained).
4. Emergency coolant supply failure (pump or bypass valve) in conjunction with any of the above failures.
5. Shield tank drained in conjunction with evaporator or coolant supply failure.

The results of the loss of coolant flow and loss of coolant due to leakage are shown in Figure 7.4. When coolant remains in the isotope heater a conduction path is provided to the shield tank and then to the outside environment. Due to the large thermal capacity of the tank little change in temperature is recorded for the isotope and tank. When a loss of coolant occurs the conduction path to the tank is reduced and heat transfer is mainly by radiation. Temperature of the isotope then begins to rise rapidly raising the possibility of a nuclear accident.

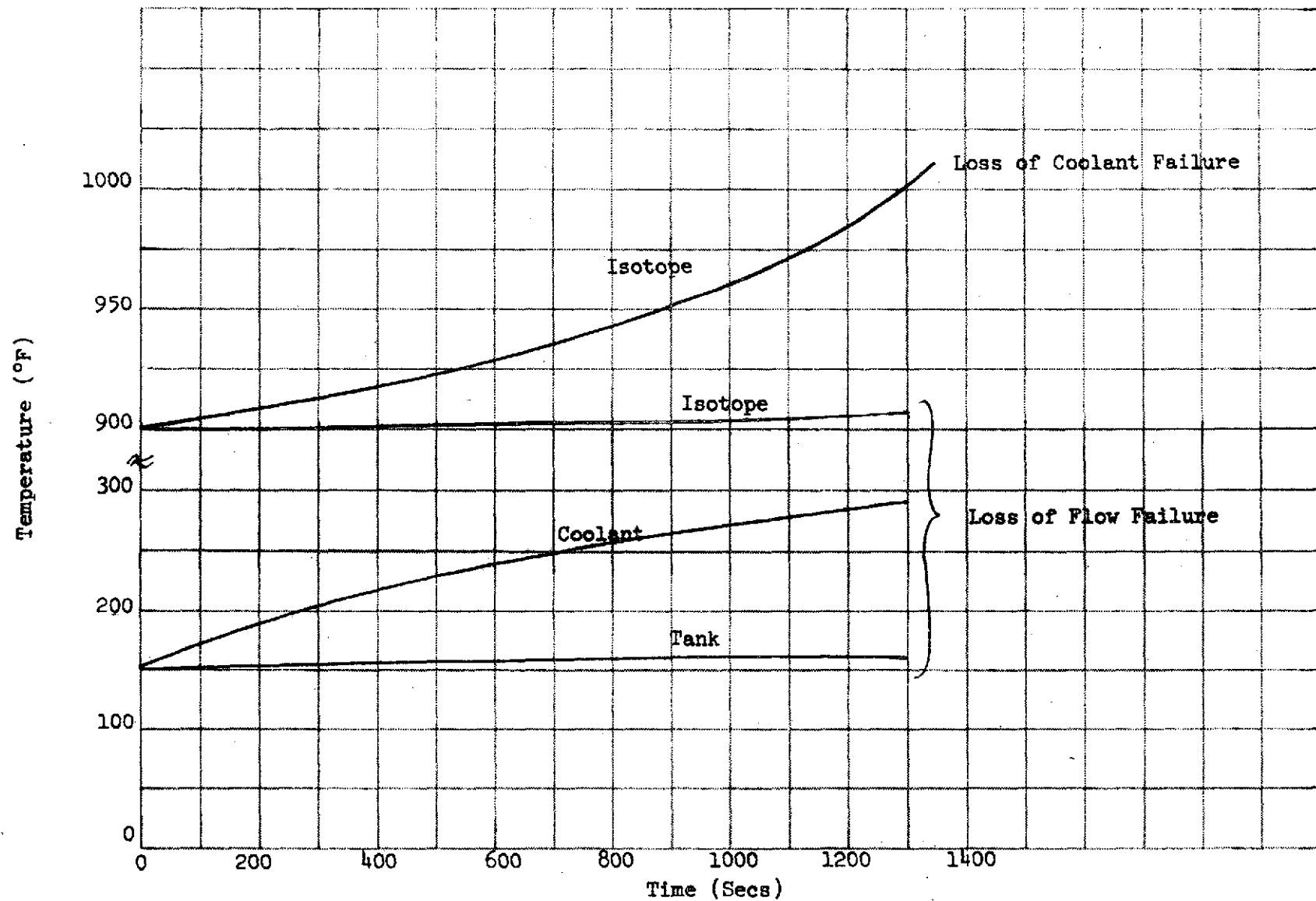


FIGURE 7.4 TEMPERATURE RESPONSE TO LOSS-OF-FLOW AND LOSS-OF-COOLANT FAILURES

Loss of evaporator heat removal capability can be caused by controller failure in the bypass mode or draining of the evaporator liquid. The result of this type failure on the LTHL is shown in Figure 7.5. Due to lack of definitive emergency heat exchanger thermal data, hand calculations were made that indicated that the heat exchanger UA was the 15-60 Btu/hr °F range. Three runs were made in this range to test LTHL coolant response. At time = 0 an evaporator bypass failure was assumed and in 220 seconds the LTHL coolant temperature reached the 190°F limit which triggered the emergency coolant heat exchanger flow. For UA less than 50 Btu/hr °F, the emergency heat exchanger did not stabilize the LTHL temperatures at less than 190°F. The increase of coolant temperature and therefore pressure could result in loop valve or fitting leakage which in turn would precipitate the loss of coolant failure described above. At a UA of 50 Btu/hr °F the temperature cycles within the range of 189 to 198°F. Therefore, it is recommended that the emergency heat exchanger UA should be specified to be greater than 50 Btu/hr-°F.

The next set of computer runs were made with the assumption that the shield tank was drained. For applications in space a clean isotope would be used and the shield tank would not be required. An evaporator bypass failure (in the full bypass position) was additionally imposed and the results are shown in Figure 7.6. The emergency cooling system was able to control LTHL coolant temperature to 210°F and the isotope temperature rose from 990 to 995°F. When a loss of emergency flow failure was run in conjunction with a drained shield tank and a failed evaporator bypass valve (in the full bypass position), the lack of thermal capacity in the shield tank water results in a rapid increase in coolant and isotope temperature. The data shown in Figure 7.6 indicate a coolant temperature greater than 350 degrees within 20 minutes. The corresponding saturation pressures of greater than 150 psi at these temperatures could result in loop leakage.

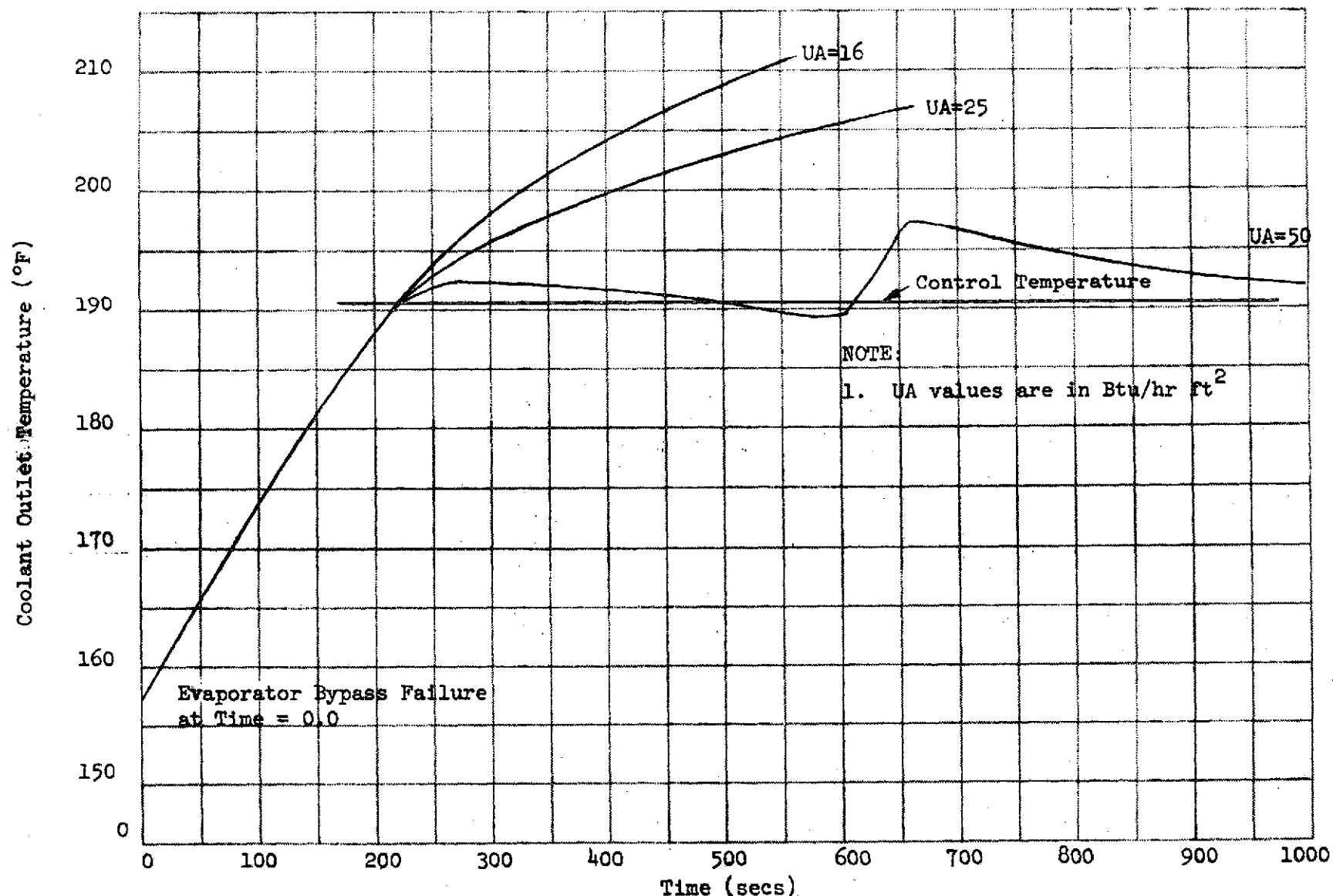


FIGURE 7.5 EVAPORATOR BYPASS FAILURE -  
ISOTOPE COOLANT OUTLET TEMPERATURES WITH VARIOUS EMERGENCY HEAT EXCHANGER UA VALUES



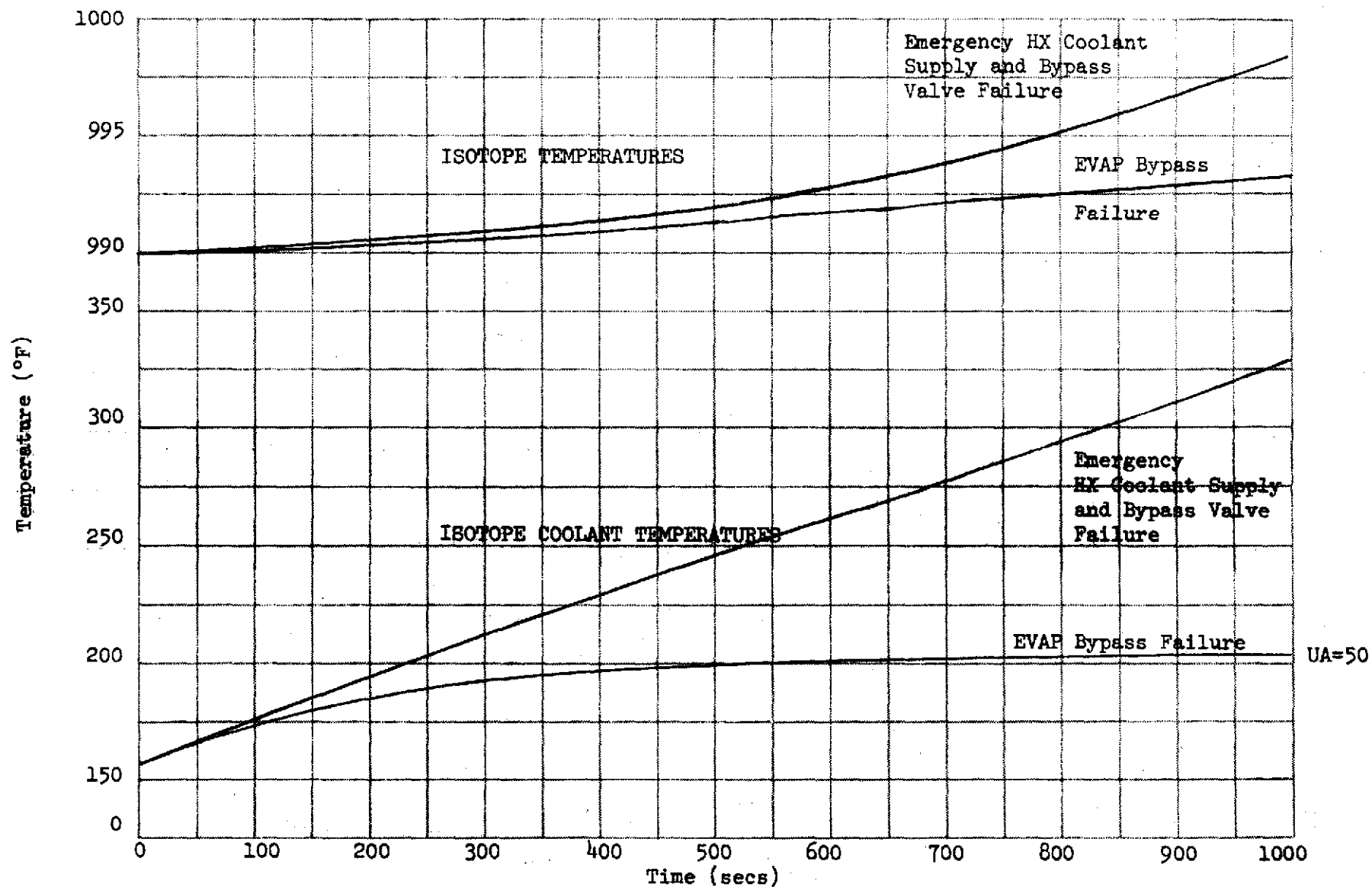


FIGURE 7.6 EFFECTS OF EVAPORATOR BYPASS AND EMERGENCY HEAT EXCHANGER COOLANT SUPPLY FAILURES ON ISOTOPE AND ISOTOPE COOLANT TEMPERATURES (SHIELD TANK DRAINED)

The preliminary failure analysis discussed above indicates that, with the exception of the LTHL, the RITE system can compensate for the failure modes discussed. To eliminate the possibility of a nuclear accident, the LTHL pumps should be made failure independent, and a secondary, static, emergency cooling mode should be provided for the isotope.

## Section 8

### REFERENCES

1. Schelkopf, J. D., Witt, T. J., and Murray, R. W. Summary Report Integrated Waste Management - Water System Using Radioisotopes for Thermal Energy. General Electric Report NYO-4104-1, September 1970.
2. Murray, R. W., Shivers, R. W., Ingelfinger, A. L., and Metzger, C. A. Integrated Waste Management - Water System Using Radioisotopes for Thermal Energy. ASME Paper 71-Av-4, Presented to the SAE/ASME/AIAA Life Support and Environmental Control Conference, July 1971.
3. Barker, R. S., Blakely, R. L., Hamill, T. D., and Nicol, S. W. G189A Generalized Environmental/Thermal Control and Life Support Systems Computer Program. McDonnell Douglas Astronautics Company, Western Division, Report MDAC-G2444, September 1971.
4. Vaughan, R. L., Barker, R. S., and Stephens, H. M. Generalized Environmental Control and Life Support System Fortran Programs. McDonnell Douglas Astronautics Company, Western Division, Report SM-49403, Volumes I through III, May 1966.

## Appendix A

### SUBROUTINE DOCUMENTATION

EVAPQ

CNDNSR

INCIN

HTPIP

TANKG

EVAP

## Component Subroutine No. 59 - Evaporator for Waste Management/Water Recovery System

1. Subroutine Description

This subroutine simulates an evaporator such as the one used in the RITE Waste Management/Water Recovery System developed by General Electric at Valley Forge, Pennsylvania. See Reference 1. The evaporator collects all solids and liquids that constitute waste, centrifugally separates the solids from the liquid, and evaporates the liquid. The vaporized liquid (steam) in the RITE system is ducted through a purifier assembly and into a condenser where pure water is recovered.

The evaporator accepts both primary and secondary flows. The primary inlet and outlet flows define the liquid and solid flow through the evaporator and use flow codes 0 or 4. The secondary inlet and outlet flows define the gaseous flow through the evaporator and employ flow codes 2 or 3. Heating to the evaporator is provided by either electrical heaters attached to the evaporator wall, or by heating fluid flowing through the tubes encircling the evaporator, or by both. The evaporator component schematic is shown in Figure 1.

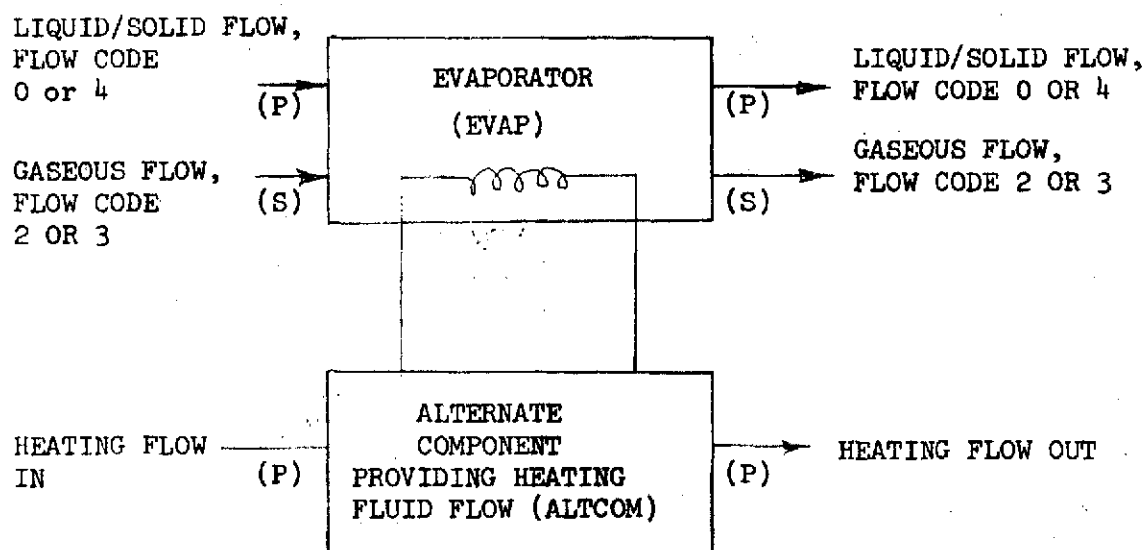


FIGURE 1. COMPONENT SCHEMATIC OF THE EVAPORATOR

The evaporator subroutine performs mass and thermal balances for either steady state or transient operation. It includes the capability of separating (filtering) and storing wet solids in a reservoir so as to allow their subsequent removal from the system by a solids pump. In addition, the subroutine may simulate, as an option, a biochemical reaction that may occur within the evaporator. This simulation includes the decomposition of urea, defined by primary inlet/outlet special flow #2, R(15), into two gaseous components, ammonia, secondary flow trace contaminant, R(32), and carbon dioxide, R(31). The evaporator schematic is shown in Figure 2.

## 2.0 SUBROUTINE DATA

### 2.1 General Notes

1. The primary inlet and outlet flows define the liquid/solid flows through the evaporator.
2. The secondary inlet and outlet flows define the gaseous flows through the evaporator.
3. The phase separator is assumed to be 100% effective and zero entrained liquid flow is assumed in the evaporator secondary loop.
4. The solids filtration rate, removing the solids from the evaporator to the solids reservoir, is a product of the solids concentration in the evaporator and an input value defined as the filtration rate. If the solids reservoir attains a maximum value (input value) the filtration process ceases and an appropriate flag is set.
5. The biochemical reaction rate in the evaporator is a product of weight of the decomposing solid (always special flow #2 on the primary side) in the evaporator and the decomposition constant. The products of the decomposition are carbon dioxide R(31) and ammonia which is provided in the trace contaminant location, R(32), for the secondary outlet flow.
6. The initial weights of liquids and solids in the evaporator must be input; if the removal rate of the solids is not equal to the solids influent or the water removal/vaporization rates do not equal the water inlet flow rate, appropriate flags are set.

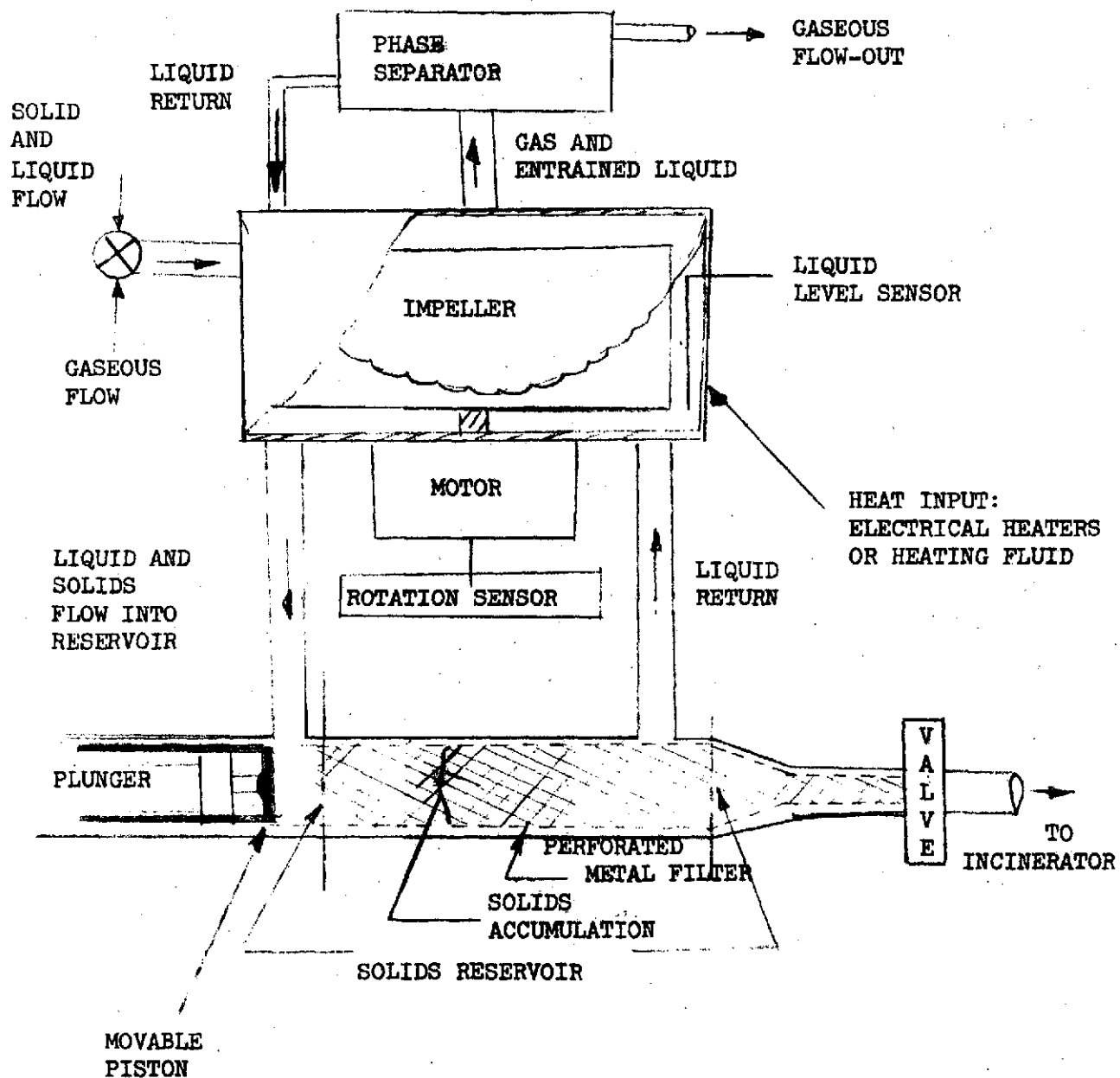


FIGURE 2 EVAPORATOR SCHEMATIC

7. The primary side special flows 2 and 4 have specific uses where:

Special Flow #2 is defined as urea when bacterial decomposition is considered.

Special Flow 4 is defined as water

## 2.2 Instruction Options

NSTR(1): Heating fluid availability

= 0 Heating fluid is available

= 1 Heating fluid is not available

NSTR(2): Not used

NSTR(3): Not used

NSTR(4): Operating pressure mode

= 0 Low pressure operating mode

= 1 High pressure operating mode

## 2.3 Heat Loss V-Array Data

<u>Reference Location</u>	<u>Description</u>	<u>Data Type</u>
51	Temperature of evaporator wall ( $^{\circ}\text{F}$ )	I(R)
52	Effective thermal conductance from evaporator wall to surroundings ( $\text{Btu/hr-}^{\circ}\text{F}$ )	0
53	Total heat loss to surroundings ( $\text{Btu/hr}$ ) $R(53) = R(56) + R(59) + R(62)$	0
54	Ambient gas temperature ( $^{\circ}\text{F}$ )	I(0)
55	Thermal conductance between surface of insulation and ambient gas ( $\text{Btu/hr-}^{\circ}\text{F}$ )	I(0)
56	Convective heat loss to ambient gas ( $\text{Btu/hr}$ )	0
57	Ambient wall temperature ( $^{\circ}\text{F}$ )	I(0)
58	Thermal radiation, $F_A$ , factor from surface of insulation to ambient wall ( $\text{ft}^2$ )	I(0)
59	Radiative heat loss to ambient wall ( $\text{Btu/hr}$ )	0
60	Structure temperature ( $^{\circ}\text{F}$ )	I(0)



<u>Reference Location</u>	<u>Description</u>	<u>Data Type</u>
61	Conductance (kA/x) between evaporator wall and structure (Btu/hr-°F)	I(0)
62	Conductance heat loss to structure (Btu/hr)	0
63	Insulation surface temperature, (°F)	0
64	Conductance (kA/x) between evaporator wall and outer surface of insulation, (Btu/hr-°F) (If R(64) = 0, there is no insulation)	I(0)

#### 2.4 Steady State K-Array Data

<u>Reference Location</u>	<u>Description</u>	<u>Data Type</u>
16	Packed failure flag data    6   5   4   3   2   1  Individual digits are set to one for a failed condition and set to zero for a passing condition.  Digit <u>Position</u> <u>Condition</u>  1            The solid reservoir has been filled, the filtration process has been stopped.  2            The solids filtration and decomposition rate does not correspond to the solids inlet rate in steady state.  3            The liquid removal and vaporization rate does not equal the liquid inlet rate in steady state.  4            Liquid level in the evaporator is above the maximum specified.  5            Liquid level in the evaporator is below the minimum allowable.  6            Not used	0
17	Component number of the alternate component providing the heating fluid flow	I(R) if NSTR(1)=0
18	Table number defining the heat flux, q/A, between the fluid in the evaporator and the evaporator wall as a function of temperature differential between fluid and wall and liquid pressure	

## 2.5 Steady State V-Array Data

<u>Reference Location</u>	<u>Description</u>	<u>Data Type</u>
65	Effective conductance between the evaporator wall and the heating fluid (Btu/hr-°F)	I(R) if NSTR(1)=0
66	Electrical heat input into the evaporator wall (Btu/hr)	I(O)
67	Internal height of the evaporator (ft)	I(R)
68	Internal diameter of the evaporator (ft)	I(R)
69	Effective height of the liquid slurry in the evaporator (ft)	0
70	Minimum liquid level allowable for normal operation (ft)	I(R)
71	Maximum liquid level allowable for normal operation (ft)	I(R)
72	Heat lost in vaporization of water to air flowing through the evaporator (Btu/hr)	0
73	Total effective gas space in the evaporator (ft <sup>3</sup> )	0
74	Heat flow from evaporator wall to liquid (Btu/hr)	0
75	Heat flux, $q/A$ , from wall to liquid (Btu/hr-ft <sup>2</sup> )	0*
76	Gas space in the evaporator that is not subject to liquid filling (ft <sup>3</sup> )	I(R)
77	Vaporization rate of water (lb/hr)	0
78	Bacterial/thermal decomposition factor for urea/ammonia carbonate-bicarbonate in the evaporator (1/hr)	I(O)
79	Decomposition rate of urea (lb/hr) (Always special flow #2 of the inlet primary flow)	0
80	Liquid used in thermal/biochemical decomposition reaction (lb/hr) removed from special flow #4 of the inlet primary flow	0
81	Ratio of liquid to solid in the solids reservoir (decimal)	I(R)
82	Filtration factor of the liquid/solid filter (lb/hr)	I(R)
83	Removal rate of solids from the liquid/solid stream by the filter (lb/hr)	0

\* Value is interpolated from Table Data specified in Table number 18.

<u>Reference Location</u>	<u>Description</u>	<u>Data Type</u>
84	In the high pressure mode, the fraction of liquid that is expected to vaporize to the air flowing through the evaporator as compared to maximum, assuming fully saturated air leaving the evaporator (decimal).	I(O)
85	The liquid/solid reservoir capacity (lb) (If not input then R(83) will be always zero.)	I(O)
86	Liquid lost due to filtration (special flow #4 in inlet primary flow (lb/hr)	0
87	Urea (special flow #2 in inlet primary flow) removal rate to solid reservoir (lb/hr).	0
88	Remaining solids (special flow #3 in inlet primary flow) removal rate to solid reservoir (lb/hr)	0
89	Water (special flow #4 in inlet primary flow) removal rate to solid reservoir (lb/hr) (same as R(86) )	0
90	Special flow #5, primary inlet flow constituent, removed to the solid reservoir (lb/hr)	0
91	Special flow #6, primary inlet flow constituent removed to the solid reservoir (lb/hr)	0
92	NH <sub>3</sub> , (located in the trace contaminant address, R(32) in secundar outlet flow) generation rate in the evaporator (lb/hr)	0
93	CO <sub>2</sub> , R(31), generation rate due to thermal/biochemical decomposition of urea in the evaporator (lb/hr)	0
94	Steady state convergence criterion for prediction of temperature (per cent)	0
95	Power required to drive the impeller (watts)	I(O)
96	Conductance between wall and liquid in evaporator (Btu/hr-°F)	I(O)
97	Not used	
98	Not used	
99	Effective orifice area between the evaporator and the source of downstream pressure (used only in the low pressure mode) (ft <sup>2</sup> )	I(R)
100	Downstream pressure, generally the condenser pressure, where the vaporized liquid is flowing (psia)	I(R)

<u>Reference Location</u>	<u>Description</u>	<u>Data Type</u>
The V-array locations 101 through 119 are a 19 member set of data defining the liquid slurry in the evaporator. These data must be initialized because the evaporator will not run dry.		
101	Total weight of liquid slurry in the condenser (lbs)	I(R), 0
102	Temperature of the liquid slurry in the condenser (°F)	I(R), 0
103	Pressure of liquid slurry in the evaporator (psia)	I(R), 0
104	Pressure of liquid slurry in the evaporator (psia)	I(R), 0
105-114	Not used	
115	Weight of urea in the evaporator. Urine in primary inlet/outlet flow is special flow #2 (lbs)	I(R), 0
116	Weight of solid defined by primary inlet/outlet special flow #3 in the evaporator (lbs)	I(R), 0
117	Weight of water, defined by primary inlet/outlet special flow #4 in the evaporator (lbs)	I(R), 0
118	Weight of solids component defined by primary inlet/outlet special flow #5 in the evaporator (lbs)	I(R), 0
119	Weight of solid component defined by primary inlet/outlet special flow #6 in the evaporator (lbs)	I(R), 0
The V-Array locations 120 through 138 are a 19 member set of data defining the initial gas mixture in the evaporator. If the run is executed for a transient case only, the initial values must be input. In steady state, values are computed which initialize the parameters for a subsequent transient run.		
120	Total weight of gas in the evaporator (lbs)	I(0), 0
121	Temperature of the gas in the evaporator (°F)	I(0), 0
122, 123	Pressure of the gas in the evaporator (psia)	I(0), 0
124	Weight of non-condensables in the evaporator (lbs)	I(0), 0
125	Weight of condensable gas (vapor) in the evaporator (lbs)	I(0), 0
126	Weight of entrained liquid (always zero) in the evaporator (lbs)	I(0), 0
127	Effective capacitance of non-condensable gas in the evaporator (Btu/lb-°F)	I(0), 0

<u>Reference Location</u>	<u>Description</u>	<u>Data Type</u>
128	Molecular weight of non-condensable gas in the evaporator (lb/lb-mole)	I(0), 0
129	Weight of O <sub>2</sub> in the evaporator (lbs)	I(0), 0
130	Weight of diluent, N <sub>2</sub> , in the evaporator (lbs)	I(0), 0
131	Weight of CO <sub>2</sub> in the evaporator (lbs)	
132	Weight of trace contaminant, NH <sub>3</sub> , in the evaporator (lbs)	I(0), 0
133	Weight of special flow #2 in evaporator (lbs)	I(0), 0
134	Weight of special flow #2 in evaporator (lbs)	I(0), 0
135	Weight of special flow #2 in evaporator (lbs)	I(0), 0
136	Weight of special flow #4 in evaporator (lbs)	I(0), 0
137	Weight of special flow #5 in evaporator (lbs)	I(0), 0
138	Weight of special flow #6 in evaporator (lbs)	I(0), 0

The V-array locations 139 through 157 are a 19 member set of data defining the liquid/solid mixture in the solid reservoir.

139	Total weight of solids in solid reservoir (lbs)	0
140	Temperature of solids in solid reservoir (°F)	0
141, 142	Pressure in solid reservoir	0
143, 152	Not used	
153	Weight of urine in solids reservoir (lbs)	0
154	Weight of solid, defined by primary inlet/outlet special flow #3 in the solids reservoir (lbs)	0
155	Weight of water, defined by primary inlet/outlet special flow #4 in the solids reservoir (lbs)	0
156	Weight of solid, defined by primary inlet/outlet special flow #5 in the solids reservoir (lbs)	0
157	Weight of solid, defined by primary inlet/outlet special flow #6 in the solids reservoir (lbs)	0

## 2.6 Transient V-Array Data

<u>Reference Location</u>	<u>Description</u>	<u>Data Type</u>
158	Thermal capacitance of evaporator shell (Btu/°F)	I(R)
159	Thermal capacitance of liquid in the evaporator (Btu/°F)	0

### 3.0 ANALYTICAL MODEL DESCRIPTION

Subroutine EVAP was designed to model an evaporator such as that which is an integral part of the Waste Management/Water Recovery system (RITE system) developed by General Electric at Valley Forge. Consequently, the model includes the simulation of solid filtering capabilities and biochemical/thermal decomposition of urea in addition to the mass and energy balance during steady state and transient analyses of the evaporator.

#### 3.1 Thermal Balance

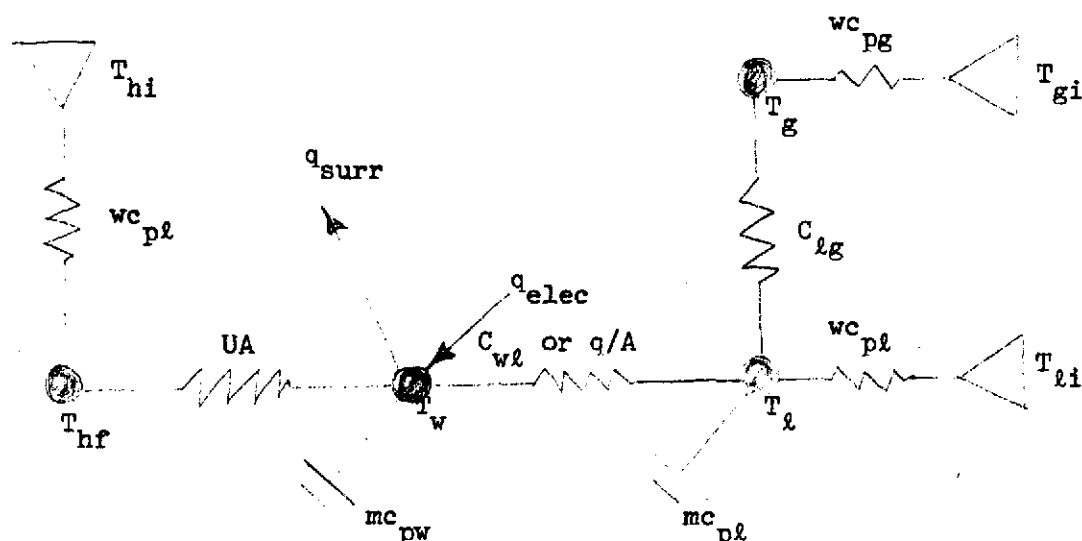
The energy balance for the fluid within the evaporator is developed by defining two modes of operation; hereafter, referred to as the high pressure mode and the low pressure mode. In the high pressure mode, the value of NSTR(4) is set to 1 and the following assumptions are made:

1. A relatively high gaseous flow prevails through the evaporator.
2. The pressure in the evaporator is equal to the inlet pressure of the gas stream.
3. The air stream leaving the evaporator may either be fully saturated; that is; air temperature equals the dew point temperature of the air, or it may vaporize relatively small amounts of liquid from the evaporator as it flows through it.

In the low pressure mode, the value of NSTR(4) is set to 0 and the following assumptions are made for the thermal analysis.

1. The gas temperature is equal to the liquid temperature.
2. The liquid temperature is equal to the boiling temperature, a function of the evaporator pressure.

The nodal thermal model developed for the energy balance of evaporator is shown below:



where:

- $T_{hi}$  = Heating fluid inlet temperature ( $^{\circ}\text{F}$ )
- $wc_{p,hf}$  = Heating fluid flow conductance ( $\text{Btu/hr-}^{\circ}\text{F}$ )
- $T_{hf}$  = Heating fluid outlet temperature ( $^{\circ}\text{F}$ )
- $UA$  = R(65) = Effective thermal conductance between heating fluid and evaporator wall ( $\text{Btu/hr-}^{\circ}\text{F}$ )
- $T_w$  = R(51) = Evaporator tank wall temperature ( $^{\circ}\text{F}$ )
- $q_{elec}$  = R(66) = Electrical heat input ( $\text{Btu/hr}$ )
- $q_{surr}$  = R(53) = Heat lost to the surroundings ( $\text{Btu/hr}$ )
- $C_{wl}$  = R(75) = Heat conductance wall to liquid ( $\text{Btu/hr-}^{\circ}\text{F}$ )
- $T_l$  = R(2) = Temperature of the liquid in the evaporator ( $^{\circ}\text{F}$ )
- $mc_{pl}$  = R(159) = Thermal capacitance of the liquid slurry in the evaporator ( $\text{Btu/}^{\circ}\text{F}$ )
- $mc_{pw}$  = R(158) = Thermal capacitance of the evaporator shell ( $\text{Btu/}^{\circ}\text{F}$ )
- $wc_{pl}$  = Fluid flow conductance of liquid slurry flowing into the evaporator ( $\text{Btu/hr-}^{\circ}\text{F}$ )

$$\begin{aligned}
T_g &= R(21) = \text{Temperature of the gas leaving the evaporator } (^{\circ}\text{F}) \\
w_{c_{pg}} &= B(1)*CPB = \text{Flow conductance of gas flowing through the} \\
&\quad \text{evaporator (Btu/hr-}^{\circ}\text{F)} \\
T_{li} &= A(2) = \text{Temperature of the liquid flowing into the evaporator } (^{\circ}\text{F}) \\
T_{gi} &= B(2) = \text{Temperature of the gas flowing through the evaporator } (^{\circ}\text{F}) \\
C_{lg} &= \text{Thermal conductance between liquid and gas flowing through} \\
&\quad \text{the system (Btu/hr-}^{\circ}\text{F)}
\end{aligned}$$

### Steady State

The steady state analysis in the high pressure mode has a provision for the vaporization of liquid into the air that is flowing through the evaporator. This is accomplished by predicting the saturation vapor pressure of air flowing through the evaporator and the amount of vaporized liquid that is required to attain saturation. The mathematical development of this portion of the model is presented below.

Saturation vapor pressure leaving the evaporator is derived from air temperature using the function PSAT.

$$P_{H_2O, s} = \text{PSAT}(T_{gi}) \quad (1)$$

$$w_{H_2O, s} = \frac{P_{H_2O}}{P_t} * \frac{18.0}{M_g} * w_g \quad (2)$$

$$w_{\text{vapr}} = K_v * (w_{H_2O, s} - w_{H_2O, i}) \quad (3)$$

where:

$$\begin{aligned}
P_{H_2O} &= \text{Saturation pressure of vapor (psia)} \\
w_g &= \text{Total gaseous flow through the system (lb/hr)} \\
w_{H_2O, s} &= \text{Assuming saturation vapor flow leaving evaporator (lb/hr)}
\end{aligned}$$



$w_{H_2O, i}$  = Vapor flow into the evaporator (lb/hr)

$w_{vapr}$  = Vaporization rate of liquid (lb/hr)

$K_v$  = Vaporization fraction of liquid factor (between 0. and 1.)

$P_t$  = Total pressure of gas in evaporator (psi)

$M_g$  = Molecular weight of gas entering evaporator (lb/lb-mole)

The thermal balance in the high pressure mode is accomplished by predicting heat flow to the wall from the heating fluid, heat flow from the wall to surroundings and to the liquid in the evaporator and cooling of the liquid due to evaporation of water into the gas stream. The nodal heat balance equations are given below:

Heating Fluid Node,  $T_{hf}$

Heat lost by fluid = Heat transfer to wall

$$w_{c_{pl}} (T_{hi} - T_{hf}) = w_{c_{pl}} (1 - e^{-UA/w_{c_{pl}}}) (T_w - T_{hi})$$

Evaporator Wall Node,  $T_w$

Heat transfer from heating fluid + Electrical heat input =

Heat lost to surroundings + Heat transfer to slurry

$$w_{c_{pl}} (1 - e^{-UA/w_{c_{pl}}}) (T_w - T_{hi}) + q_{elec} = q_{surr} + C_{wl} (T_w - T_l)$$

Liquid Node,  $T_l$

Heat transfer from wall = Heat transfer to entering liquid + Heat transfer to gas

$$C_{wl} (T_w - T_l) = w_{c_{pl}} (T_l - T_{li}) + C_{lg} (T_l - T_g)$$

Gas Node,  $T_g$ 

Heat transfer from liquid = Heat gain by gas flow

$$C_{lg} (T_l - T_g) = wc_{pg} (T_g - T_{gi})$$

The steady state heat balance is assumed accomplished when the individual heat balances for the heating fluid node, the wall node, and the liquid node are approximately numerically correct, meeting the steady state convergence criteria.

The steady state thermal analysis in the low pressure mode is accomplished in a similar fashion to that for the high pressure mode except that the heat flux from the wall to the liquid is predicted from an input table. The table relates the boiling heat transfer rates to the temperature differential between the wall and the liquid and the evaporator pressure. The nodal equations for the low pressure mode reduce to the following:

Heating Fluid Node,  $T_{hf}$ 

Heat lost by fluid = Heat transfer to wall

$$wc_{pl} (T_{hi} - T_{hf}) = wc_{pl} (1 - e^{(-UA/wc_{pl})}) (T_w - T_{hi})$$

Evaporator Wall Node,  $T_w$ 

Heat transfer from heating loop + Electrical Heat Input = Heat lost  
to surroundings + Heat transfer to liquid

$$wc_{pl} (1 - e^{(-UA/wc_{pl})}) (T_w - T_{hi}) + q_{elec} = q_{surr} + q/A (T_w - T_l)$$

Liquid Node,  $T_l$ 

Heat transfer from wall = Heat transfer to input liquid + Heat for  
evaporation

$$q/A (T_w - T_l) = wc_{pl} (T_l - T_{li}) + w_v h_{fg}$$

Since the temperature of the liquid is set equal to the dew point temperature corresponding to the evaporator pressure level, the heat balance is computed for only the heating fluid and wall nodes. The evaporator pressure is computed from the pressure drop between the evaporator and the downstream pressure, R(100). In this calculation, the vapor flow rate is equal to the current evaporation rate, R(77), and the flow resistance is due to the effective orifice area, R(99).

### Transient

The transient thermal analysis of the evaporator is accomplished using the forward difference techniques for the nodes that include thermal capacitance. These nodes are the wall node (capacitance input in R(158) ) and the liquid node. The liquid node's thermal capacitance is calculated from the amount of liquid in the evaporator and its specific heat. The value is stored in R(159).

The critical time step is computed for each node to insure against computations instability. The minimum critical time step is compared with the system time step; if the value of the minimum critical time step is below the system time step then the subroutine time step is set equal to the minimum critical time step. Each time the subroutine is called, successive time step passes are performed in the subroutine until the system time step is achieved. The critical time step is defined as:

$$\Delta t_c = \frac{\text{node capacitance}}{\Sigma \text{ conductances}}$$

### Mass Balance

The mass balance of the evaporator system, besides accounting for the amount of gases, liquids, and solids flowing through the system also has the capabilities of computing the biochemical/thermal decomposition of urea into  $\text{NH}_3$  and  $\text{CO}_2$  and the removal of solids from the evaporator to the solid reservoir by the solids filter.

Urine contains a certain amount of ammonium carbonate/bicarbonate which will liberate 500 mg/l of carbon dioxide when it is vaporized. In addition, urea which is the principal solute in urine (~ 28000 mg/l) may decompose thermally when the temperature is above 150°F to yield  $\text{NH}_3$  and  $\text{CO}_2$ . Urea also decomposes in the presence of the enzyme urease which is produced by bacteria contained in the feces. Once the bacteria is introduced they will multiply while decomposing most of the urea within 12 to 48 hours. Since no germicidal chemicals are added to the evaporator it is assumed

that biochemical decomposition of urea will occur. The urea decomposes to  $\text{NH}_3$  and  $\text{CO}_2$  according to the following reaction:



The reaction is modeled as follows:

$$w_{\text{ag}} = K_R * \frac{M_{\text{ag}}}{M_{\text{ur}}} * m_{\text{ur}} * 2 \quad (5)$$

$$w_{\text{CO}_2\text{g}} = K_R * \frac{M_{\text{CO}_2}}{M_{\text{ur}}} * m_{\text{ur}} \quad (6)$$

$$w_{\text{urd}} = \frac{M_{\text{ur}}}{(M_{\text{H}_2\text{O}} + M_{\text{ur}})} * (w_{\text{ag}} + w_{\text{CO}_2\text{g}}) \quad (7)$$

$$w_{\text{H}_2\text{O}u} = \frac{M_{\text{H}_2\text{O}}}{(M_{\text{H}_2\text{O}} + M_{\text{ur}})} * (w_{\text{ag}} + w_{\text{CO}_2\text{g}}) \quad (8)$$

where:

$w_{\text{ag}}$  =  $\text{NH}_3$  generation rate (lb/hr)

$w_{\text{CO}_2\text{g}}$  =  $\text{CO}_2$  generation rate (lb/hr)

$w_{\text{urd}}$  = Urea decomposition rate (lb/hr)

$w_{\text{H}_2\text{O}u}$  = Water used in decomposition of urea (lb/hr)

$M_{\text{ag}}$  = Molecular weight of ammonia (lb/lb-mole)

$M_{\text{ur}}$  = Molecular weight of the urea (lb/lb-mole)

$M_{\text{CO}_2}$  = Molecular weight of  $\text{CO}_2$  (lb/lb-mole)

$m_{\text{ur}}$  = Weight of urea in evaporator (lbs)

$K_R$  = Decomposition factor (1/hr)

Subroutine EVAP models the solids filtering capability of the evaporator. In the evaporator the pumping action of the impeller circulates the liquid slurry thru an external perforated metal filter. The waste solids collect in the filter. When the filter (defined as the solids reservoir) fills with solids, a means through GPOLY logic is provided to remove the solids from the solids reservoir. The mathematical development of the process is presented below:

$$m_{st} = \sum m_j \quad (9)$$

$$CONC_s = m_{st}/m_t \quad (10)$$

$$w_{sfr} = CONC_s * K_F \quad (11)$$

$$w_l = w_{sfr} K_{PL} \quad (12)$$

$$w_j = \frac{m_j}{m_{st}} * (w_{sfr}) \quad (13)$$

where:

- $m_{st}$  = Total weight of solids in the evaporator (lbs)
- $m_j$  = Weight of individual solid component in evaporator (lbs)
- $m_t$  = Total weight of liquid slurry in the evaporator (lbs)
- $CONC_s$  = Concentration of solids in evaporator (decimal)
- $w_{sfr}$  = Solids filtering rate (lb/hr)
- $K_F$  = Filtration constant (lb/hr)
- $w_l$  = Liquid filtration rate by filter (lb/hr)
- $K_{PL}$  = Ratio of liquid to solid in solids reservoir (decimal)
- $w_j$  = Removal rate of individual components from evaporator to solids reservoir (lb/hr)

The transient mass balance is accomplished by maintaining flows in and out of the evaporator, accounting for urea that may have biochemically reacted, solids and liquid that may be filtered out, and the liquid that is vaporized. The flow stability (the critical time step) for the low pressure node flow exiting the evaporator and flowing into downstream components must be determined by GPOLY logic within the EVAP subroutine. In this mode the evaporator pressure and flow is a function of the downstream (e.g., condenser) pressure. The mass balance for all components in the evaporator may be represented by the following general equation.

$$m' = m + (\pm w_{\text{reac}} + w_{\text{in}} - w_{\text{f}} - w_{\text{o}}) * \Delta t \quad (14)$$

where,

$\pm w_{\text{reac}}$  = Rate of generation of component in evaporator (lb/hr)

$w_{\text{in}}$  = Inlet flow of component (lb/hr)

$w_{\text{f}}$  = Outlet flow to solids filter (lb/hr)

$w_{\text{o}}$  = Outlet flow to vapor loop (lb/hr)

$m, m'$  = Old and new values for mass of constituent in evaporator (lb)

#### 4.0 LOWER LEVEL SUBROUTINES AND FUNCTIONS REQUIRED

HF

HG

ESTIM

QSURR

PSAT (T)

TSAT (P)

PROP

FLOARY

FLØWØR

LV

SV

#### 5.0 REFERENCES

1. Schelkopf, J. D., Witt, F. J., Murray, R. W.; Summary Report, Integrated Waste Management - Water System Using Radioisotopes for Thermal Energy, General Electric Report No. NYO-4104-1, September 8, 1970.
2. Putnam, D. F., Chemical Aspects of Urine Distillation, American Society of Mechanical Engineers, Paper No. 65-AV-24, March 1965.
3. Barker, R. S., et. al., G-189A Generalized Environmental/Thermal Control and Life Support Systems Computer Program, MDAC G2444, September, 1971.



CNDNSR

## Component Subroutine No. 60 - Condenser

1. Subroutine Description

This subroutine simulates a condenser such as that used by a Waste Management/ Water Recovery system, specifically the RITE system that has been developed by General Electric at Valley Forge, Pennsylvania. See Reference 1. The condenser is used to condense the water vapor discharged by the pyrolysis units and to store the condensate prior to its transfer to the collection tanks. The condenser component schematic is shown below.

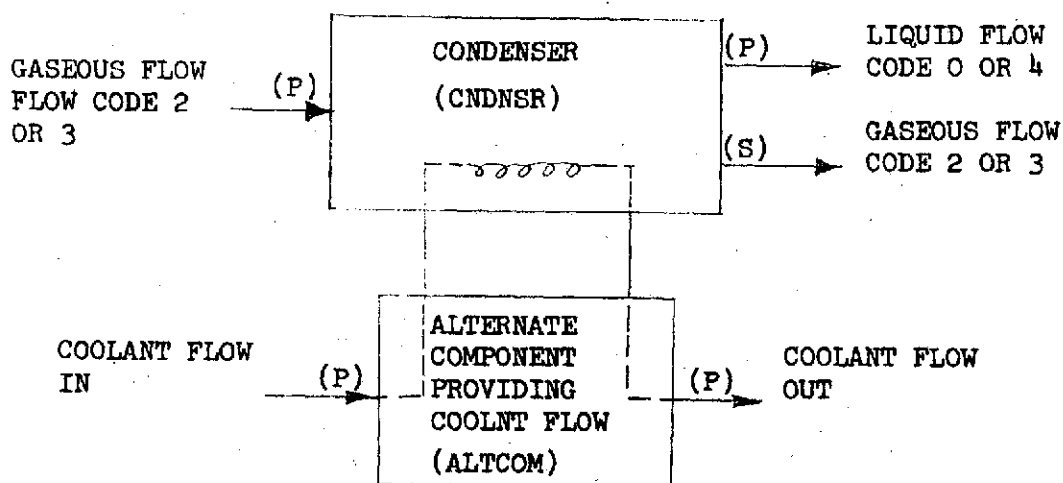


FIGURE 1. COMPONENT SCHEMATIC OF THE CONDENSER

An engineering sketch of the condenser is shown in Figure 2.

Water vapor and inert gases enter at the top of the condenser. The sensible and latent heat of the vapor is removed at the top part of the condenser wall. Cooling is provided by an alternate component providing the coolant flow through the tubes attached to the condenser. The non-condensable gases remain in the condenser until vented to vacuum as secondary flow. The condenser is

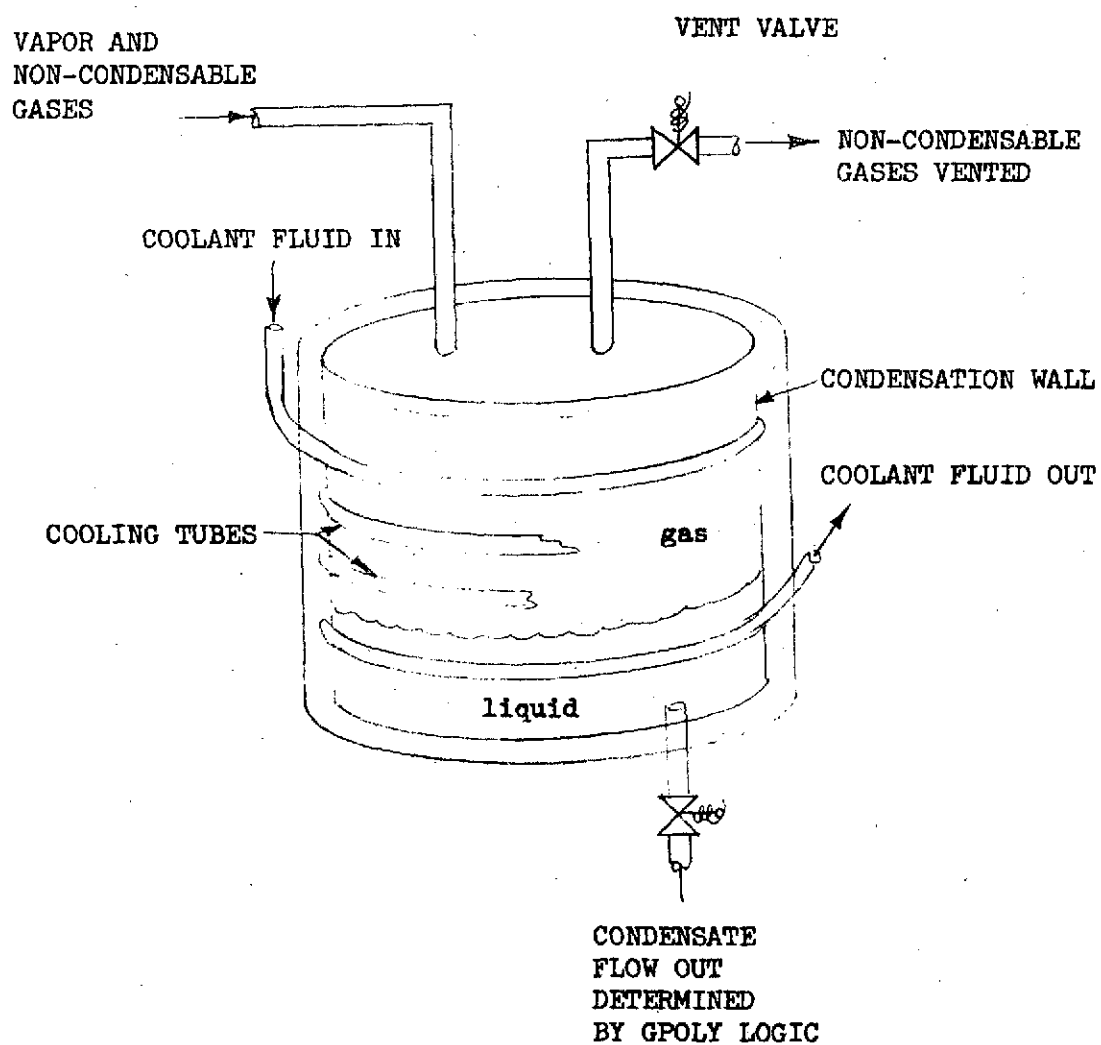


FIGURE 2. CONDENSER UNIT

designed to retain the condensate at the bottom either by the force of gravity or by capillary action in a zero-g environment. The condensate may be removed as primary flow by an external command through GPOLY1 logic.

## 2.0 Subroutine Data

### 2.1 General Notes

1. The primary inlet flow must be gaseous; that is, use flow codes 2 or 3.
2. There is no secondary inlet flow.
3. The primary outlet flow is liquid using flow code 0 or 4 and the rate is determined by GPOLY1 logic.
4. The secondary outlet flow is gaseous and is computed by the subroutine.
5. The R-Array data R(91) through R(109) are a 19-member set of variables defining the gas mixture in the condenser. Initial values are computed during steady state analysis to satisfy the steady state criteria. It is recommended that a steady state run be made prior to transient analysis in order to initialize these variables.

### 2.2 Instruction Options

NSTR(1) Type of Condenser

= 0 Cylindrical condenser

= 1 Conical condenser

NSTR(2) Criteria used for steady state mass balance

= 0 Fraction of vent gas defined by R(79) is used in obtaining the mass balance in steady state.

= 1 The total pressure in evaporator is defined and used as basis in obtaining mass balance

### 2.3 Steady State K-Array Data

Reference

Location

Description

Data Type

16

Component number of alternate component

I(R)

## 2.4 Steady State/Transient V-Array Data

<u>Reference Location</u>	<u>Description</u>	<u>Data Type</u>
	(See Figure 2 for Thermal Network)	
65	Total heat flux through the condenser wall (Btu/hr)	0
66	Effective orifice area of the valve used in venting gases out of the condenser (ft <sup>2</sup> ). The orifice is sized during steady state conditions according to the NSTR(2) criterion, inlet conditions, condensing rate, and downstream pressure R(67). During transient conditions the area can be adjusted by GPOLY logic to simulate vent valve operation.	
67	Downstream pressure to which the gases are venting (psia)	I(0)
68	Dew point temperature of gases in condenser (°F)	0
69	Water condensation rate in the condenser (lb/hr)	0
70	Heat loss to wall due to latent heat of condensation (Btu/hr)	0
71	Internal radius of condenser (ft)	I(R)
72	Internal height of condenser (ft)	I(R)
73	Height of the liquid level in the condenser (ft)	0
74	Mass of liquid in the condenser, (lbs) (an initial value may be specified)	I(0), 0
75	Effective thermal conductance between condenser wall and the coolant in the tubing which is attached to the condenser wall (Btu/hr-°F)	I(R)
76	Steady state convergence tolerance (decimal)	I(R)
77	Condenser total lumped thermal capacitance (Btu/°F)	I(R)
78	Condensate lumped thermal capacitance (Btu/°F)	0
79	Molar fraction of vent gas that is condensable vapor (decimal)	I(R) when NSTR(2) =
80	Conductance (KA/l <sub>1</sub> ) between the two condenser wall nodes (Btu/hr-°F)	I(R)
81	Effective total conductance between the condenser wall and the surroundings (Btu/hr-°F)	I(R)

## 2.4 Steady State/Transient V-Array Data (continued)

<u>Reference Location</u>	<u>Description</u>	<u>Data Type</u>
82	Effective temperature of the surroundings ( $^{\circ}\text{F}$ )	I(R)
83	Heat transfer convection coefficient between the gas and the wall and liquid nodes; the sensible heat transfer coefficient ( $\text{Btu/hr-}^{\circ}\text{F}$ )	I(R)
84	Conductance ( $KA/l_2$ ) between the wall node and the condensate node, ( $\text{Btu/hr-}^{\circ}\text{F}$ )	0
85	Heat transfer coefficient due to condensation of vapor on the condenser wall ( $\text{Btu/hr-}^{\circ}\text{F-Ft}^2$ )	0
86	Coolant outlet temperature at coolant node 1 ( $^{\circ}\text{F}$ )	0
87	Coolant outlet temperature at coolant node 2, ( $^{\circ}\text{F}$ )	0
88	Condenser wall node 1 temperature ( $^{\circ}\text{F}$ )	0
89	Condenser wall node 2 temperature ( $^{\circ}\text{F}$ )	0

The V-array locations 91 through 109 are a 19-member set of data defining the gas mixture in the condenser. If the run is executed for a transient case only, the initial values must be input. If a steady state case is executed, the values are computed by the subroutine and are not required to be input.

91	Total weight of gas in condenser, (lbs)	I(0), 0
92	Temperature of the gas in condenser, ( $^{\circ}\text{F}$ )	I(0), 0
93, 94	Pressure of the gas in condenser, (psia)	I(0), 0
95	Weight of non-condensables in condenser, (lbs)	I(0), 0
96	Weight of condensable vapor in condenser, (lbs)	I(0), 0
97	Weight of contained liquid (always zero), (lbs)	I(0), 0
98	Effective capacitance of non-condensable gas, ( $\text{Btu/lb-}^{\circ}\text{F}$ )	I(0), 0
99	Molecular weight of non-condensable gas, (lb/lb mole)	I(0), 0
100	Weight of $\text{O}_2$ in the condenser, (lbs)	I(0), 0
101	Weight of $\text{N}_2$ in the condenser, (lbs)	I(0), 0
102	Weight of $\text{CO}_2$ in the condenser, (lbs)	I(0), 0
103	Weight of $\text{NH}_3$ in the condenser, (lbs)	I(0), 0
104	Weight of special flow #1 in the condenser, (lbs)	I(0), 0

Reference Location	Description	Data Type
105	Weight of special flow #2 in the condenser, (lbs)	I(0), 0
106	Weight of special flow #3 in the condenser, (lbs)	I(0), 0
107	Weight of special flow #4 in the condenser, (lbs)	I(0), 0
108	Weight of special flow #5 in the condenser, (lbs)	I(0), 0
109	Weight of special flow #6 in the condenser, (lbs)	I(0), 0

### 3.0 Analytical Model Description

The modeling of the condenser by this subroutine includes both mass and thermal balances. Figure 3 is a schematic of the condenser thermal network. Thermal parameters for the condenser are shown. The mass flow constituents entering the condenser (from the pyrolysis units in the RITE system) include liquid water, water vapor, carbon dioxide, oxygen, nitrogen, and other non-condensable gases. The condensate is assumed to be stored in the condenser until pumped from the condenser by an external command, determined by GPOLY1 logic. The non-condensable gases and vapor that is not condensed are considered to be either contained in the condenser or vented to vacuum by opening the vent valve.

### 3.1 Thermal Balance

The energy balance of the water condenser is accomplished by a formulated 6-node thermal network with the wall nodes defined by the liquid level of the condensate. The lumped parameter thermal network is shown in Figure 3 below

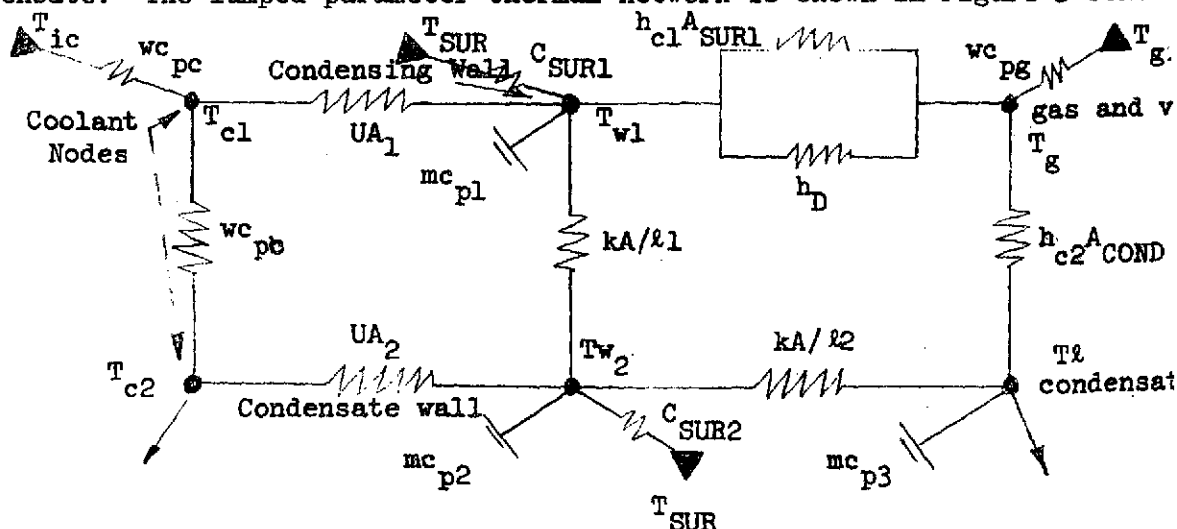


FIGURE 3. CONDENSER NODAL NETWORK

where:

- $T_{ic}$  = Inlet temperature of the coolant, ( $^{\circ}\text{F}$ )
- $T_{c1}$  = R(86) = Temperature of coolant at node 1, ( $^{\circ}\text{F}$ )
- $T_{c2}$  = R(87) = Temperature of coolant at node 2, ( $^{\circ}\text{F}$ )
- $T_{w1}$  = R(88) = Temperature of wall node 1, ( $^{\circ}\text{F}$ )
- $T_{w2}$  = R(89) = Temperature of wall node 2, ( $^{\circ}\text{F}$ )
- $T_g$  = R(21) = Temperature of gas in the condenser, ( $^{\circ}\text{F}$ )
- $T_l$  = R(2) = Temperature of the condensate in the condenser, ( $^{\circ}\text{F}$ )
- $T_{gi}$  = A(2) = Temperature of the gas flowing in, ( $^{\circ}\text{F}$ )
- $w_{pc}$  = Fluid flow conductance, coolant flow, (Btu/hr- $^{\circ}\text{F}$ )
- $w_{pg}$  = Fluid flow conductance gaseous inlet flow, (Btu/hr- $^{\circ}\text{F}$ )
- $T_{SUR}$  = Effective temperature of the surroundings, ( $^{\circ}\text{F}$ )
- $C_{SUR1}$  = Effective conductance to the surroundings, wall node 1 to surroundings, (Btu/hr- $^{\circ}\text{F}$ )
- $C_{SUR2}$  = Effective conductance to the surroundings, wall node 2 to surroundings, (Btu/hr- $^{\circ}\text{F}$ )
- $UA_1$  = Effective thermal conductance between wall node 1 and coolant node 1, (Btu/hr- $^{\circ}\text{F}$ )
- $UA_2$  = Effective thermal conductance between wall node 2 and coolant node 2, (Btu/hr- $^{\circ}\text{F}$ )
- $KA/\ell_1$  = R(80) = Conductance between wall node 1, and wall node 2, (Btu/hr- $^{\circ}\text{F}$ )
- $KA/\ell_2$  = R(84) = Conductance between wall node and liquid node, (Btu/hr- $^{\circ}\text{F}$ )
- $hc_1$  = R(83) = Convective heat transfer coefficient between wall node and gas node, (Btu/hr-Ft $^2$ - $^{\circ}\text{F}$ )
- $h_D$  = Heat transfer coefficient, diffusion, for condensation of vapor on the condenser wall, (Btu/hr- $^{\circ}\text{F}$ -Ft $^2$ )
- $mc_{p1}$  = Thermal capacitance of wall node 1, (Btu/ $^{\circ}\text{F}$ )
- $mc_{p2}$  = Thermal capacitance of wall node 2, (Btu/ $^{\circ}\text{F}$ )
- $mc_{p3}$  = R(78) = Thermal capacitance of liquid in the condenser, (Btu/ $^{\circ}\text{F}$ )
- $A_{SUR1}$  = Surface area of node 1 (ft $^2$ )
- $A_{COND}$  = Surface area of condensate (ft $^2$ )

Steady State

A steady state heat balance is performed about each node.

1. Coolant node associated with condensation wall,  $T_{c1}$

Heat gained by the coolant = Heat transfer from condenser wall node

$$wc_{pc}(T_{c1} - T_{ic}) = wc_{pc}(1 - e^{-(uA_1/wc_{pc})})(T_{w1} - T_{ic}) \quad (1)$$

2. Coolant Node associated with condensate storage wall node,  $T_{c2}$

Heat gained by the coolant = Heat transfer from wall,  $T_{c2}$

$$wc_{pc}(T_{c2} - T_{c1}) = wc_{pc}(1 - e^{-(uA_2/wc_{pc})})(T_{w2} - T_{c1}) \quad (2)$$

3. Wall node associated with condensing portion of wall,  $T_{w1}$

Heat lost to coolant + Heat lost to surrounding + Heat lost to lower wall = Sensible heat gained from gas node + Heat of condensation

$$wc_{pc}(1 - e^{-(uA_1/wc_{pc})})(T_{w1} - T_{ic}) + C_{SUR1}(T_{w1} - T_{SUR}) + kA/l_1(T_{w1} - T_{w2}) = h_{c1}A_{SUR1}(T_g - T_{w1}) + h_{D1}A_{SUR1}(T_v - T_{w1}) \quad (3)$$

where  $T_v = TSAT$  (Vapor Pressure)

4. Wall node associated with condensate storage part of wall,  $T_{w2}$

Heat lost to coolant + Heat lost to surroundings = Heat transfer from upper wall + Heat gained from condensate

$$wc_{pc}(1 - e^{-(uA_2/wc_{pc})})(T_{w2} - T_{c2}) + C_{SUR2}(T_{w2} - T_{SUR}) = kA/l_1(T_{w1} - T_{w2}) + kA/l_2(T_l - T_{w2}) \quad (4)$$

5. Gas/Vapor Node,  $T_g$

Heat lost by incoming gas vapor = Heat lost by condensation + Sensible heat loss = Connection heat lost to condensate

$$WC_{PG}(T_{gi} - T_g) = h_{c1}A_{SUR1}(T_g - T_{w1}) + h_{D1}A_{SUR1}(T_v - T_{w1}) + h_{c2}A_{COND}(T_g - T_l) \quad (5)$$



6. Condensate Node,  $T_e$ 

Heat gained from gas = Heat lost to wall

$$h_{c2} A_{COND} (T_g - T_l) = kA/l_2 (T_l - T_{w2}) \quad (6)$$

The energy balance of the thermal network in steady state is accomplished using the Gauss-Seidel iteration method. The heat balance equations are modified to yield six equations for the nodal temperatures.

$$T_{c1} = T_{ic} + C_1 * (T_{w1} - T_{ic}) \quad (7)$$

$$\text{where: } C_1 = 1. - e^{-(uA_1/wc_{pc})} \quad (7a)$$

$$T_{c2} = T_{c1} + C_4 * (T_{w2} - T_{c1}) \quad (8)$$

$$\text{where: } C_4 = 1. - e^{-(uA_2/wc_{pc})} \quad (8a)$$

$$T_{w1} = \frac{T_{ic} C_2 + \frac{KA}{l_1} T_{w2} + h_{c1} A_{SUR1} T_g + h_D A_{SUR1} T_v + (C_{SUR1} T_{SUR})}{C_2 + \frac{KA}{l_1} + h_{c1} A_{SUR1} + h_D A_{SUR1} + C_{SUR1}} \quad (3)$$

$$\text{where: } C_2 = wc_{pc} * C_1 \quad (9a)$$

$$T_{w2} = \frac{C_3 T_{c1} + \frac{KA}{l_1} T_{w1} + \frac{KA}{2} T_l + C_{SUR2} T_{SUR}}{C_3 + \frac{KA}{l_1} + \frac{KA}{2} + C_{SUR2}} \quad (10)$$

$$\text{where: } C_3 = C_4 * wc_{pc} \quad (10a)$$

$$T_g = \frac{h_{c1} A_{SUR1} T_{w1} + h_{c2} A_{COND} T_l + wc_{pg} T_{gi}}{hc_1 + hc_2 + wc_{pg}} \quad (11)$$

$$T_l = (h_{c2} A_{COND} T_g + \frac{KA}{l_2} T_{w2}) / (h_{c2} A_{COND} + \frac{KA}{l_2}) \quad (12)$$

After the new temperatures for the six nodes are computed, they are tested against previously computed temperatures, if all the values are within the convergence criterion, an input value, the thermal balance on the condenser is assumed to be attained for the given condenser vapor pressure. At each iterative pass the vapor pressure is determined from the function PSAT for the current vapor temperature.

### Mass Balance

The mass balance of the condenser is attained by making the following assumptions:

- 1) The constituency of the gases venting out of the condenser is the same as the constituency of gases in the condenser.
- 2) The total non-condensable flow and the constituency of non-condensable flow does not change between inlet and outlet.

From the condensing wall temperature,  $T_{wl}$ , and the dew point temperature,  $T_v$ , of the gas in the condenser, we compute the condensation rate,  $w_{cond}$ .

$$w_{cond} = h_D * (T_v - T_{wl}) / h_{fg} \quad (13)$$

where:  $h_{fg}$  = heat of vaporization, (Btu/lb)

Since the condensation rate cannot exceed the vapor inlet flow, a test is made to insure that the condensation rate is not greater than the vapor inlet flow. If vapor inlet flow is less than the condensation rate, then the partial pressure of vapor,  $pp_{H_2O}$ , is decreased and a new heat balance is computed using the new,  $T_v$ , saturation temperature of the gas.

The vented flow rate is derived from the partial pressure of vapor,  $pp_{H_2O}$ , in the condenser and the volumetric percentage of water vapor,  $V_{ratio} = 2R(79)$ , is vented with non-condensable gases by first predicting the total pressure in the condenser.

Total pressure:

$$P_t = PP_{H_2O} / V_{ratio} \quad (14)$$

The molecular weight of gas vented is derived from the following relationship

$$M_t = \frac{M_c * PP_{H_2O} + M_{nc} * (P_t - PP_{H_2O})}{P_t} \quad (15)$$

where:

- $M_t$  = Average molecular weight of gas in the condenser (same as vented gas) (lbs/lb mole)
- $M_c$  = Molecular weight of condensable gas (water vapor) (lbs/lb mole)
- $M_{nc}$  = Molecular weight of the non-condensable gas in the condenser, the value is equal to the molecular weight of non-condensable gas at inlet (lbs/lb mole)

The total flow vented is computed from the following relationship:

$$w_t = w_{nc} * \left( \frac{P_t}{P_t - PP_{H_2O}} \right) * \left( \frac{M_t}{M_{nc}} \right) \quad (16)$$

where:

- $w_t$  = Total flow vented, (lb/hr)
- $w_{nc}$  = Total non-condensable gas vented, (lb/hr)
- $w_c$  = Total condensable gas (vapor) vented, (lb/hr)

From total flow and non-condensable we compute the vapor flow vented:

$$w_c = w_t - w_{nc} \quad (17)$$

The vapor flow vented out of the condenser should equal the vapor flowing in less the condensation rate. If the vapor flow vented out is higher than the vapor flowing in less condensation rate, we assume that there was insufficient condensation and we increase the partial pressure of vapor to induce a higher condensation rate. If the vapor flow vented is lower than the difference between vapor flowing in less condensation rate, we decrease the  $pp_{H_2O}$  to decrease the condensation rates. In both cases a new steady state thermal balance is computed for the new  $T_v$  term.

If the vapor flow vented approximately equals the difference between vapor flow in less condensation rate, we assume a mass balance has been achieved. With the steady state flow computed, we determine the initial values for the gas components in the condenser by first computing the total weight of gas in the condenser and then the individual components.

$$m_t = (V * p_t * M_t) / (1545. * (T + 460.)) \quad (18)$$

where:

- $m_t$  = Weight of gas in the condenser, (lbs)
- $V$  = Gaseous volume in the condenser, (Ft<sup>3</sup>)
- $T$  = Temperature of gas in the condenser, (°F)
- $M_t$  = Molecular weight of gas in the condenser (lbs/lb-mole)

$$w_i = (m_i / m_t) * w_t \quad (19)$$

where:

- $m_i$  = Weight of constituent i in the condenser, (lbs)
- $w_i$  = Flow rate of constituent i vented (lbs/hr)

With the temperature, pressure, and weight of individual components calculated for the gas in the condenser, we compute the effective orifice area of the vent,  $A_v$ , that would yield the specified flow rate computed above. The value is derived using the choked flow formula as follows:

$$A_v = \frac{\left( \frac{w_t}{3600. * P_t} \right) \sqrt{\frac{1545.}{M_t} (T + 460)}}{\sqrt{r * g_c * \left( \frac{2}{r+1} \right) (r+1)/(r-1)}} \quad (20)$$

where:

$A_v$  = Effective orifice area of the vent, (ft<sup>2</sup>)

$g_c$  = 32.174 proportionality constant (lb<sub>m</sub>/lb<sub>f</sub> sec<sup>2</sup>)

$r$  = Ratio of specific heats (dimensionless)

### Transient Analysis

The nodal temperatures during transient analysis are computed through the use of forward difference techniques for nodes that include capacitance values, namely the two wall nodes, and the liquid condensate node. The two coolant nodes and the gas node are assumed to have negligible capacitance values. The temperatures for the two liquid coolant nodes and the gas node are computed from the following:

$$T_{c1} = T_{ic} + (T_{w1} - T_{ic}) * C_1 \quad (21)$$

$$T_{c2} = T_{c1} + (T_{w2} - T_{c1}) * C_4 \quad (22)$$

$$T_g = (h_{c1} A_{SUR1} T_{w1} + h_{c2} A_{COND} T_l + w_{pg} T_{gi}) / (h_{c1} A_{SUR1} + h_{c2} A_{COND} + w_{pg}) \quad (23)$$

From which we compute heat flux from all nodes to coolant nodes

$$Q_{out1} = C_2 * (T_{w1} - T_{ic}) \quad (24)$$

$$Q_{out2} = C_3 * (T_{w2} - T_{cl}) \quad (25)$$

where:

$Q_{out1}$  = Heat flow from wall node 1 to coolant node 1, (Btu/hr)

$Q_{out2}$  = Heat flow from wall node 2 to coolant node 2, (Btu/hr)

The total heat flows into the wall nodes are computed as follows:

$$Q_{tw1} = -Q_{out1} + C_{SUR1} * (T_{SUR} - T_{w1}) + \frac{KA}{l_1} * (T_{w2} - T_{w1}) \quad (26)$$

$$+ h_D A_{SUR1} * (T_v - T_{w1}) + h_{cl} A_{SUR1} * (T_g - T_{w1})$$

$$Q_{tw2} = -Q_{out2} + C_{SUR2} * (T_{SUR} - T_{w2}) + \frac{KA}{l_1} * (T_{w1} - T_{w2}) \quad (27)$$

$$+ \frac{KA}{l_2} * (T_l - T_{w2})$$

where:

$Q_{tw1}$  = Heat flux through wall node 1, (Btu/hr)

$Q_{tw2}$  = Heat flux through wall node 2, (Btu/hr)

The heat flow into the liquid node is computed by the following:

$$Q_L = \frac{KA}{l_2} * (T_{w2} - T_l) + h_{cl} A_{COND} * (T_g - T_l) \quad (28)$$

New values for constituent gas masses and the total gas mass are obtained. PROP is used to obtain the molecular weight for the total gas mixture and the water vapor partial pressure and the total pressure are computed as follows:

$$m_i' = m_i + w_i * \Delta t \quad (34)$$

$$m_t = \sum_i^n m_i \quad (35)$$

$$pp_{H_2O} = m_{H_2O} * 1545. * (T+460.) / (V * 18. * 144.) \quad (36)$$

$$p_t = m_t * 1545. * (T+460.) / (V * m_t * 144.) \quad (37)$$

where:

$$m_{H_2O} = \text{Weight of vapor in the condenser (lb.)}$$

$$V = \text{Gaseous volume in the condenser (ft}^3\text{)}$$

$$m_t = \text{Total weight of gases in the condenser (lb)}$$

#### 4.0 Lower Level Subroutines Called

PROP

FLOWOR

#### 5.0 References

1. Schelkopf, J. D., Witt, F. J., Murray, R. W., Summary Report, Integrated Radioisotopes for Thermal Energy, General Electric Report No. NYO-4104-1, September 8, 1970.
2. Barker, R. S., et al., G-189A Generalized Environmental/Thermal Control and Life Support Systems Computer Program, MDAC G2444, September 1971.

where:

$$Q_L = \text{Heat flux through liquid node, (Btu/hr)}$$

Before the transient temperatures are computed, the critical time step is computed for each node, and compared against the system time step; if the value of the minimum critical time step is below the system time step then the subroutine time step is set equal to the minimum critical time step. Each time the subroutine is called, successive time passes are performed in the subroutine until the system time step is achieved. The critical time step is defined as:

$$\Delta t = \frac{\text{node capacitance}}{\Sigma \text{conductances}} \quad (29)$$

The new temperatures for the liquid and wall nodes are computed from the following:

$$T_{w1}' = T_{w1} + Q_{tw1} * \Delta t / mc_{p1} \quad (30)$$

$$T_{w2}' = T_{w2} + Q_{tw2} * \Delta t / mc_{p2} \quad (31)$$

$$T_{\ell}' = T_{\ell} + Q_L * \Delta t / mc_{p3} \quad (32)$$

Where, the prime, denotes the predicted temperatures of the wall and liquid nodes at the end of the time step.

The mass balance during transient analysis is computed by first determining the total flow of vent gas using the function `FLOWOR`. This function computes choked or sonic flow through an orifice.

The flow rate of individual constituents through the vent valve is determined as follows:

$$w_i = w_t * \frac{m_i}{m_t} \quad (33)$$



## INCIN

### Component Subroutine No. 58 - Incinerator

#### 1.0 Subroutine Description

This subroutine simulates an incinerator in which the waste products are processed by vacuum drying, chemical decomposition at high temperature, and venting of resultant ash. Waste products such as paper and feces are loaded into the incinerator in a batch process. The vent to vacuum is opened and the solids are dried. The vent to vacuum is closed and a small amount of oxygen is introduced with the solids and the oxidation reaction proceeds. The product gasses are then vented to vacuum. The oxygen cycle is repeated until all the solids are incinerated. At the end of the cycle, the remaining ash is blown out by a nitrogen purge. The incineration cycle is summarized in Figure 1.

#### 2.0 Subroutine Data

##### 2.1 General Notes

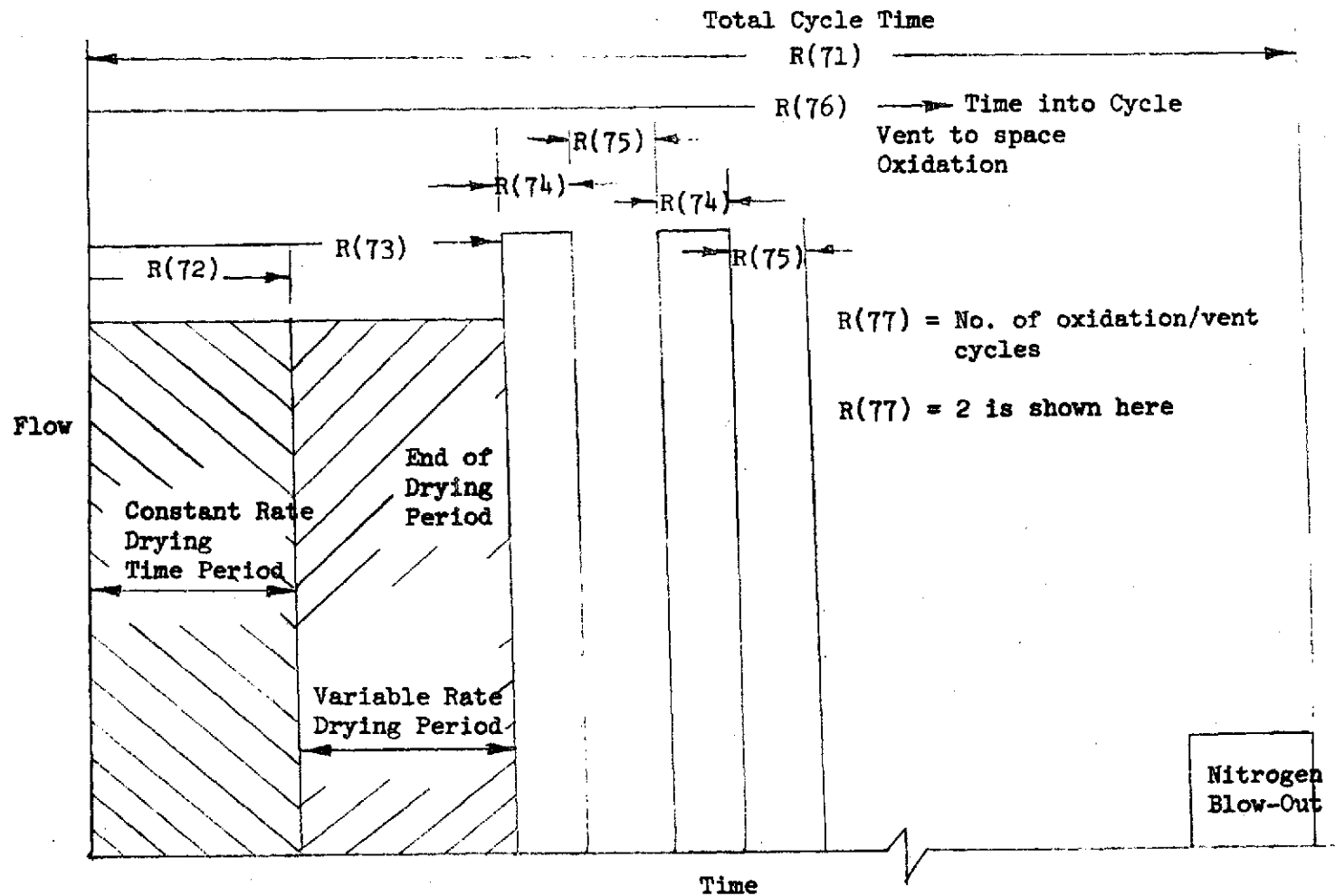
1. The input is an instantaneous batch process, therefore the solid and liquid mass is input through GPOLY logic.
2. The primary outlet flow is gaseous using flow code 3.
3. There is no secondary outlet flow.
4. During steady state the incinerator is assumed to be empty.

##### 2.2 Instruction Options

##### 2.3 Heat Loss V-Array Data

<u>Reference Location</u>	<u>Description</u>	<u>Data Type</u>
51	Temperature of incinerator ( $^{\circ}\text{F}$ )	0
52	Effective thermal conductance from incinerator to surroundings ( $\text{Btu/hr-}^{\circ}\text{F}$ )	0

# INCINERATOR CYCLE DEFINITION



<u>Reference Location</u>	<u>Description</u>	<u>Data Type</u>
53	Total heat loss to surroundings (Btu/hr) $R(53) = R(56) + R(62)$	0
54	Ambient gas temperature ( $^{\circ}\text{F}$ )	I (0)
55	Thermal conductance between surface of insulation and ambient gas (Btu/hr- $^{\circ}\text{F}$ )	I (0)
56	Convective heat loss to ambient gas (Btu/hr)	0
57	Ambient wall temperature ( $^{\circ}\text{F}$ )	I (0)
58	Thermal radiation FA factor from surface of insulation to ambient wall ( $\text{ft}^2$ )	I (0)
59	Radiative heat loss to ambient wall (Btu/hr)	0
60	Structure temperature ( $^{\circ}\text{F}$ )	I (0)
61	Conductance (kA/X) between incinerator and structure (Btu/hr- $^{\circ}\text{F}$ )	I (0)
62	Conductance heat loss to the structure (Btu/hr)	0
63	Insulation surface temperature ( $^{\circ}\text{F}$ )	0
64	Conductance (kA/X) between incinerator and outer surface of insulation (Btu/hr- $^{\circ}\text{F}$ ) (If $R(64) = 0$ there is no insulation)	I (0)

#### 2.4 Steady State/Transient V-Array Data

<u>Reference Location</u>	<u>Description</u>	<u>Data Type</u>
65	Total heat transfer in the incinerator (Btu/hr)	0
66	Heat input by conduction from the heat source (Btu/hr)	0
67	Evaporative heat loss (Btu/hr)	0
68	Heat generated by oxidation reaction (Btu/hr)	0
69	Heat of reaction (Btu/lb-oxygen)	I (0)
70	Stability time increment (seconds)	0

R-array locations R(71) to R(77) define the incineration cycle. See Figure 1.

71	Total cycle time (seconds)	I (R)
72	Time to constant rate drying period (seconds)	I (R)
73	Time to end of drying period (seconds)	I (R)

<u>Reference Location</u>	<u>Description</u>	<u>Data Type</u>
74	Oxidation time period (seconds)	I(R)
75	Gas venting time period (seconds)	I(O)
76	Time into cycle (seconds)	I(O)
77	Number of oxidation vent cycles (number of cycles)	I(R)
78	Mass of water in incinerator (lb)	I(O) or GPOLYL
79	Mass of solids in incinerator (lb)	I(O) or GPOLYL
80	Critical moisture content (lb)	0
81	Mass of carbon dioxide in incinerator (lb)	0
82	Mass of nitrogen oxide in incinerator (lb)	0
83	Mass of ash in incinerator (lb)	0
84	Mass of oxygen added per oxidation cycle (lb)	I(R)
85	Mass of ash in incinerator (lb)	0
86	Mass of nitrogen used for venting (lb)	I(R)
87	Saturation pressure (psia)	0
88	Environmental pressure for venting of exhaust products (psia)	I(R)
89	Heat source temperature (°F)	I(R)
90	Mass transfer coefficient (lb/hr-ft <sup>2</sup> -psi)	I(R)
91	Evaporation surface area (ft <sup>2</sup> )	I(R)
92	Heat of vaporization (Btu/lb)	0
93	Thermal conductance between heat source and incinerator (Btu/hr-°F)	I(R)
94	Effective thermal capacitance of shuttle (Btu/lb-°F)	I(R)

### 3.0 Analytical Model Description

#### Constant Rate Drying Period

During the constant-rate drying periods, drying proceeds by diffusion of vapor from the saturated surface of the material to the environment. The mass transfer rate is:

$$\left(\frac{dw}{dt}\right)_c = k A (p_s - p_e)$$

where:

w = moisture contents (lbs water/lb solids)

$\left(\frac{dw}{dt}\right)_c$  = drying rate, (lbs water/lb solids/hr)

k = mass transfer coefficient, (lbs water/lb solids/(hr-ft<sup>2</sup>-atms) )

A = area of evaporation surface (ft<sup>2</sup>)

p<sub>s</sub> = vapor pressure of water at solids surface temperature (atms)

p<sub>e</sub> = partial pressure of water vapor in incinerator environment (atms)

#### Falling Rate Period

The falling rate drying rate is proportional to the moisture contents of the solid.

$$\left(\frac{dw}{dt}\right)_f = -K w$$

K is a function of the constant rate as follows:

$$K = \left(\frac{dw}{dt}\right)_c / w_c$$

$\left(\frac{dw}{dt}\right)_f$  = falling drying rates (lbs water/hr)

w<sub>c</sub> = critical moisture contents (lb water/lb dry solid)

#### Energy Balance

During drying heat is input to the incinerator by conduction from the isotope heat block.

$$q_i = \frac{kA}{\ell} (T_{HB} - T_i)$$

where

$q_i$  = Heat conducted to incinerator from heat block (Btu/hr)

$\frac{kA}{\ell}$  = Thermal conductance (Btu/hr °F)

$T_i$  = Average incinerator temperature (°F)

$T_{HB}$  = Heat block temperature (°F)

The evaporative heat loss is given by:

$$q_e = \left( \frac{dw}{dt} \right) \cdot \Delta h_{fg}$$

where

$q_e$  = Evaporative heat loss, Btu/hr

$\Delta h_{fg}$  = latent heat of evaporation, Btu/lb

Heat is lost to the surroundings by conduction to the supporting structure,  $q_k$ , by convection to the cabin air,  $q_c$ , and by radiation to the cabin,  $q_r$ . These heat terms are calculated using subroutine QSURR. The analytical model for this is provided in subroutine QSURR's writeup. The sum of the three terms is set equal to  $q_{SURR}$ .

$$q_{SURR} = q_r + q_c + q_k$$

The temperature of the incinerator is solved using the forward difference technique.

$$T_i = T_i' + \frac{(q_i - q_{SURR} - q_e) \Delta t}{mc_p}$$

where

$T_i$  = Incinerator temperature at end of time step (°F)

$T_i'$  = Incinerator temperature at start of time step (°F)

$mc_p$  = Effective thermal capacity of incinerator (Btu/lb °F)

The stability time increment is computed as follows:

$$\Delta t_{\text{stab}} = \frac{\Sigma mc_p}{\Sigma c_i}$$

where

$\Sigma mc_p$  = Total heat capacitance of incinerator (Btu/°F)

$\Sigma c_i$  = Sum of conductances to incinerator (Btu/hr °F)

$\Delta t_{\text{stab}}$  = Stability time increment (hrs)

If the stability time increment is less than the system time increment, then the number of internal passes is set so as to accommodate the stability time increment.

### Incineration

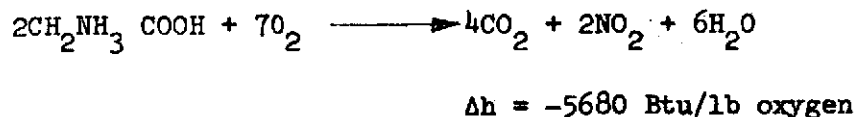
Incineration occurs in the incinerator shuttle when oxygen is introduced in the presence of isotope heat. Typical constituents of dried incinerator solids are listed in Table 1.

TABLE 1  
CONSTITUENTS OF INCINERATOR SOLIDS

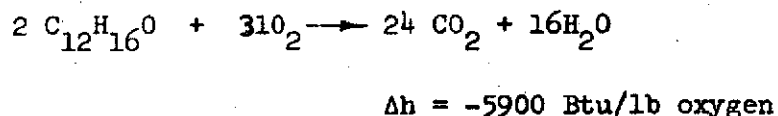
Constituent	% of Total Weight	Chemical Symbol
Bacteria	30	$(CH_2)_x NH_3 COOH$
Indigested protein	2-3	$(CH_2)_x NH_3 COOH$
Fat	10-20	$C_{12} H_{16} O$
Ruffage	30	$C H_2 O$
Inorganic Materials	20-30	Various

The oxidation reactions for each of these constituents are listed below along with the heats of reaction.

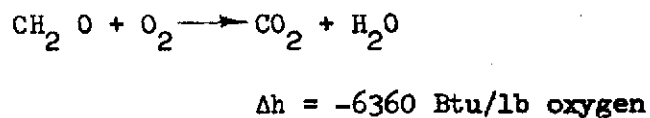
#### Bacteria and Protein



#### Fat



#### Ruffage



For a typical incinerator load, 1.89 pounds of oxygen is required for each pound of solids incinerated. The average heat of reaction is 5650 Btu/lb.

The heat generated by the reaction is determined from the first law of thermodynamics.

$$q = E_2 - E_1$$

where

$E_2$  = internal energy of the products (Btu/lb)

$E_1$  = internal energy of the reactants (Btu/lb)

This equation can be written as:

$$q = \sum m_p c_{v,p} (T_2 - T_B) - \sum m_R c_{v,R} (T_B - T_1) - \Delta h$$



where

$T_2$  = Product temperature ( $^{\circ}\text{F}$ )

$T_1$  = Initial reactants temperature ( $^{\circ}\text{F}$ )

$T_B$  =  $68^{\circ}\text{F}$ , Reference temperature of reaction for  $\Delta h$  ( $^{\circ}\text{F}$ )

$m_p$  = Mass of reactants (lb)

$m_R$  = Mass of reactants (lb)

The temperature of the incinerator then becomes

$$T_i = T'_i \left( \frac{q_i + q - q_{\text{SURR}}}{mc_p} \right) \Delta t$$

At the end of the reaction, the product gasses are vented to the environment.

#### 4.0 Lower Level Subroutines Called

PSAT

QSURR

# HTPIP

## Component Subroutine No. 38 - Heat Pipe

### 1.0 Subroutine Description

This subroutine simulates an inert gas controllable heat pipe that may be used for temperature control of a near-isothermal system for a wide range of heat loads. As shown in the heat pipe schematic, Figure 1, heat is absorbed in the evaporator region and a phase change from liquid to vapor takes place. The vapor flows to the condenser region of the pipe where the process is reversed, vapor is condensed, and heat is rejected. This provides the capacity for transporting large quantities of thermal energy in a near-isothermal system.

### 2.0 Subroutine Data

#### 2.1 General Notes

1. The source temperature is generally input via GPOLY logic.

#### 2.2 Steady State K-Array Data

<u>Reference Location</u>	<u>Description</u>	<u>Data Type</u>
16	Table number for boiling temperature of the heat transport medium as a function of pressure.	I(R)

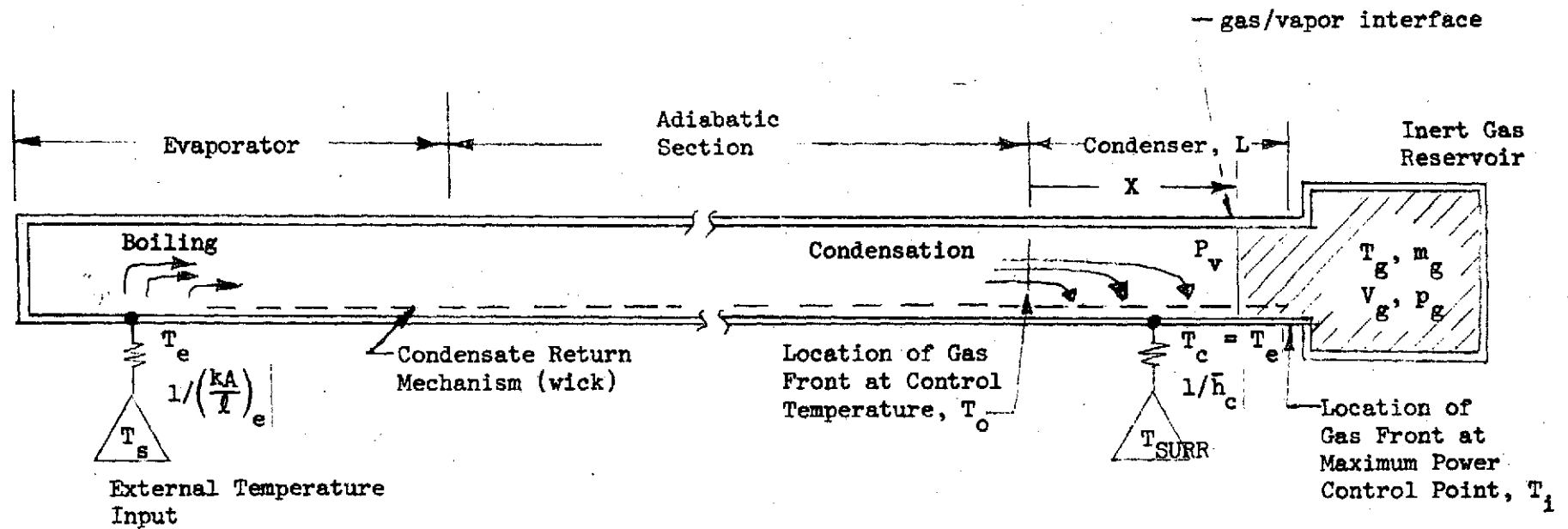
#### 2.3 Steady State V-Array Data

<u>Reference Location</u>	<u>Description</u>	<u>Data Type</u>
65	Heat pipe control temperature (°F)	I(R)
66	Temperature of source from which heat is rejected (°F) (Input via GPOLY1 logic)	I(R)
67	Conductance (KA/l) between the heat source and the heat pipe (Btu/hr-°F)	I(R)

FIGURE 1 - SCHEMATIC OF INERT GAS CONTROLLABLE HEAT PIPE

NOTE:

See Section 3.0 for definition of symbols



## 2.3 Steady State V-Array Data

<u>Reference Location</u>	<u>Description</u>	<u>Data Type</u>
68	Heat pipe diameter (ft)	I(R)
69	Length of the condensing region of the heat pipe (ft)	I(R)
70	Mass of the inert gas in the heat pipe (lb)	I(R)
71	Gas constant of the inert gas used (ft-lb/lb-°R)	I(R)
72	Heat pipe temperature at the cold end of heat pipe (°F) (Average temperature of gas).	I(R)
73	Volume of inert gas reservoir of heat pipe, (ft <sup>3</sup> )	I(R)
74	Heat pipe internal pressure (psfa)	0
75	Heat rejected by the heat pipe (Btu/hr)	0
76	Heat transfer coefficient between surface of heat pipe and ambient gas (Btu/hr-ft <sup>2</sup> -°F)	I(R)
77	Ambient temperature (°F)	I(R)
78	Temperature of the heat pipe in the evaporation region, (°F)	0
79	Not Used	
80	Convection heat transfer coefficient between surface of heat, pipe and ambient gas (Btu/hr-ft <sup>2</sup> -°F)	I(R)
81	Thermal conductivity of heat pipe material (Btu/hr-ft-°F)	I(R)
82	Heat pipe thickness (ft)	I(R)

### 3.0 Analytical Model Description

The mathematical model simulating the inert gas controllable heat pipe is simplified by making the following assumptions:

1. The inert gas obeys the ideal gas law.
2. An infinitely sharp interface exists between the inert gas and vapor.
3. Steady state conditions exist.

There are three modes of heat rejection rate by the heat pipe considered in this analysis. The first is when the source temperature is below the control temperature (boiling temperature) of the heat transport medium in the evaporator. The heat pipe in this case is treated as a fin of finite length protruding from a heat source. The heat-flow rate from the heat pipe to the surroundings can then be expressed by the following relationship.

$$\dot{q}_{HP} = \sqrt{PhkA} (T_S - T) \tanh \left( \sqrt{\frac{hP}{kA}} L \right) \quad (1)$$

where:

- $\dot{q}_{HP}$  = Heat flow from the heat pipe ( $\frac{\text{Btu}}{\text{hr}}$ )
- $P$  = Perimeter of heat pipe (ft)
- $k$  = Pipe conductance ( $\frac{\text{Btu}}{\text{hr-}^\circ\text{F-ft}}$ )
- $A$  = Gross sectional area of heat pipe ( $\text{ft}^2$ )
- $T_S$  = Temperature of heat source ( $^\circ\text{F}$ )
- $T$  = Temperature of surroundings ( $^\circ\text{F}$ )
- $L$  = Pipe length (ft)
- $h$  = Corrective heat transfer coefficient between heat pipe condenser and the cooling medium ( $\text{Btu/hr-ft}^2\text{-}^\circ\text{F}$ )

The second mode of heat transfer through the heat pipe occurs when the source temperature exceeds the lower temperature level of the control temperature band. Increasing the evaporator power level increases the vapor pressure by moving the interface further down the pipe exposing more condenser area and decreasing the amount of volume available for the inert gas. This process is shown schematically in Figure 1. The algorithm adopted for the solution of heat flow through a heat pipe for this mode of heat transfer is as follows:

1. The evaporator temperature is first assumed to equal the heat pipe control temperature; from which, heat flow to the heat pipe from the heat source is computed.

$$q_{HP} = \left( \frac{kA}{l} \right)_e (T_s - T_e) \quad (2)$$

where

$$\left( \frac{kA}{l} \right)_e = \text{Thermal conductance between heat pipe and heat source} \left( \frac{\text{Btu}}{\text{hr-}^\circ\text{F}} \right)$$

$$T_e = \text{Temperature of the evaporator } (^\circ\text{F})$$

$$T_s = \text{Temperature of the source } (^\circ\text{F})$$

2. From the initial heat flow computation,  $q_{HP}$ , the condenser area, specifically the length of the condensing region of the heat pipe is computed.

$$X = q_{HP} / (\pi D \bar{h}_c (T_e - T_{SURR})) \quad (3)$$

where

$$X = \text{Length of effective condensing region of heat pipe (ft.)}$$

$$D = \text{Diameter of heat pipe (ft.)}$$

$$\bar{h}_c = \text{Convective heat transfer coefficient} \left( \frac{\text{Btu}}{\text{hr-}^\circ\text{F-ft}^2} \right)$$

3. The pressure of the inert gas is computed using the ideal gas law

$$p_g = m_g R_g T_g / (V_g + \pi D(L-X)) \quad (4)$$

where

$V_g$  = Volume of inert gas reservoir ( $\text{ft}^3$ )

$m_g$  = Mass of inert gas in reservoir (lb.)

$R_g$  = Inert gas constant ( $\frac{\text{ft-lb}}{\text{lb-}^\circ\text{F}}$ )

$T_g$  = Temperature of inert gas (equals temperature of surrounding) ( $^\circ\text{F}$ )

$p_g$  = Heat pipe pressure (psfa)

A new value for evaporator boiling temperature is determined from the input table data relating evaporator boiling temperature to pressure.

4. The new value for X, the length of the condensing region of the heat pipe, is compared against the last computed value and if the difference exceeds 1%, steps 1, 2, and 3 are repeated until the two values become approximately equal.

The third mode of heat flux through the heat pipe occurs when the power rejection requirements exceed the limitations of the pipe. That is, when the condensing region of the pipe encompasses the entire surface allowable. The procedure used in determining the heat flux is as follows:

The evaporator temperature is first computed from the predicted pressure of the heat pipe.

$$p_g = m_g R_g T_g / V_g \quad (5)$$

$$T_e = f(p_g) \text{ (Table data)} \quad (6)$$

where

$p_g$  = Heat pipe pressure (psfa)

Initial heat flux is computed from the initial estimate for evaporator temperature,  $T_e$ .

$$q_{HP} = h_c \pi DL (T_e - T_{SURR}) \quad (7)$$

A new value for  $T_e$  is computed from the predicted heat flux and source temperature

$$T_e = T_s - q_{HP} / (kA/l) \quad (8)$$

where

$T_s$  = Temperature of the source ( $^{\circ}F$ )

$kA/l$  = Thermal conductance between source and heat pipe  
(Btu/hr- $^{\circ}F$ )

and then heat flux is recomputed using equation (7).

#### 4.0 Lower Level Subroutines and Functions Required

None.

#### 5.0 References

1. Kreith, F., Principles of Heat Transfer, International Textbook Company, Scranton, Pennsylvania, 1958.



## TANKG

### Component Subroutine No. 30 - Storage Tank

#### 1.0 Subroutine Description

This subroutine simulates a storage tank fitted with one inlet and one outlet on the primary side that accepts either liquid or gas flow. (Flow codes 0 or 4 for liquids; flow codes 1, 2, or 3 for gasses). In addition, the tank can include the capability of transferring heat from or to the tank by means of fluid flowing through coils or other heat transfer devices which may either be attached to the tank wall or be immersed in the fluid in the tank. The heating/cooling fluid comprises the secondary inlet and outlet flows of the tank. The schematic of the tank is shown below:

Primary Flow Inlet  
Flow Code 0, or 4 Dry  
For Liquids; 1, 2  
or 3 for Gasses

Heating/Cooling  
Flow:  
Secondary Flow Inlet

Heating/Cooling  
Flow out  
Secondary Flow

Primary Flow  
Out Determined  
By GPOLY Logic  
or Equals Inlet

The routine performs mass and thermal balances for either steady state or transient operation. It is capable of running adiabatic gas ventings and fillings as special cases. The tank is comprised of a tank shell, an optional layer of insulation, and the tank fluid. Two electrical heaters, one connected to the tank wall and the other immersed in the fluid, are provided for thermal control.

The tank exchanges heat with the environment according to the thermal model shown in Figure 2, which is a typical model for many G189A components.

## 2.0 Subroutine Data

### 2.1 General Notes

The inlet and outlet streams of principal concern are usually the primary flows. The secondary flow is used only as a heat transport medium which may be provided for thermal control of the tank. In some problems there may be no primary source component for the tank. For example, the tank may be a storage vessel with only a primary flow outlet capability. Even if there is no primary source component, the user must enter the correct flow code and special flow types on the KBAS card for the primary flow of the tank. The secondary flow data may be excluded if the heat addition/rejection capability in the tank is not provided. If the heating/cooling capability is included, the appropriate data must be entered in the KBAS and NSTR cards for the tank.

In most problems, the user will input data which specifies the initial total mass and composition inside the tank. The very nature of a tank (or storage device) is such that there is usually no relation between the flow out of the tank and the flow in. Inlet and exit flows are, in general, totally independent

The user, therefore, will set the outlet flow from the tank equal to a desired value in a GPØLY1 statement in most simulations. Only the total flow need be specified; the individual constituent flows will be determined by the routine from the known concentrations in the tank. The outlet flow could be keyed to demand schedules, evaluated from orifice equations, or determined by control logic.

## 2.2 Instruction Options

NSTR(1): Calculation of outlet flow during Steady State

- = 0 In steady state calculations the outlet flow is set equal to the inlet flow despite any coding in GPØLY1. In transient, the GPØLY1 coding will be executed.
- = 1 The outlet flow is determined by a GPØLY1 statement or the previous value if no GPØLY1 statement changes it.

NSTR(2): Steady State Heat Balance

- = 0 In steady state, tank fluid temperature is equal to temperature of inflowing stream.
- = 1 The tank fluid temperature is specified in R(68) and the heating or cooling load required to offset ambient heat losses and fluid expansion is calculated for steady state operation.
- = 2 The tank fluid temperature is calculated by iteration on the steady state heat balance.

NSTR(3): Transient Heat Balance

- = 0 Skip the transient heat balance and proceed according to NSTR(2)
- = 1 Perform transient heat balance.

NSTR(4): Initial Mass or Volume in Tank.

- = 0 Volume, pressure, and temperature are taken as correct in input data. Initial mass calculated by program.
- = 1 Mass, pressure, and temperature are taken as correct input data. Volume calculated and used in transient runs.

NSTR(5): The provision for heat rejection/addition through use of secondary flow as heat transport medium.

- = 0 No provision for heat rejection/addition through use of secondary flow as heat transport medium exists.
- = 1 The heating/cooling device for thermal control of the tank is attached to the tank wall.
- = 2 The heating/cooling device for thermal control of the tank is immersed in the tank.

## 2.3 Heat Loss V-Array Data

<u>Reference Location</u>	<u>Description</u>	<u>Data Type</u>
51	Temperature of tank wall, ( $^{\circ}\text{F}$ )	0
52	Effective thermal conductance from tank wall to surroundings (Btu/hr- $^{\circ}\text{F}$ )	0
53	Total heat loss to surroundings (Btu/hr) $R(53) = R(56) + R(59) + R(62)$	0
54	Ambient gas temperature ( $^{\circ}\text{F}$ )	I(R)
55	Thermal conductance between surface of insulation and ambient gas (Btu/hr- $^{\circ}\text{F}$ )	I(R)
56	Convective heat loss to ambient gas (Btu/hr)	0
57	Ambient wall temperature ( $^{\circ}\text{F}$ )	I(R)
58	Thermal radiation A factor from surface of insulation to ambient wall ( $\text{ft}^2$ )	I(R)
59	Radiative heat loss to ambient wall (Btu/hr)	0
60	Structure temperature ( $^{\circ}\text{F}$ )	I(R)
61	Conductance ( $kA/X$ ) between tank wall and structure (Btu/hr- $^{\circ}\text{F}$ )	I(R)
62	Conductance heat loss to the structure (Btu/hr)	0
63	Insulation surface temperature ( $^{\circ}\text{F}$ )	0
64	Conductance ( $kA/X$ ) between tank wall and outer surface of insulation (Btu/hr- $^{\circ}\text{F}$ ) (If $R(64) = 0$ there is no insulation)	I(R)

## 2.4 Steady State V-Array Data

<u>Reference Location</u>	<u>Description</u>	<u>Data Type</u>
65	Fluid heating or cooling load in steady state (Btu/hr)	I(0)
66	Tank wall heating or cooling load in steady state (Btu/hr)	I(0)
67	Effective heat transfer conductance (UA) between heating/cooling fluid and the tank wall or fluid Btu/hr - ( $^{\circ}$ F)	I(R) if NSTR(5)>
68	Maximum weight of liquid in tank. (Full conditions (lbs.))	I(R) if Flow co 0 or 4
69	Total fluid weight in tank (lb)	I(0)
70	Fluid temperature in tank ( $^{\circ}$ F)	I(0)
71	Fluid volume in tank (ft <sup>3</sup> )	I(0)
72	Fluid pressure in tank (psia)	I(0)
73	Weight of non-condensables in tank (lb)	I(0)
74	Weight of condensable vapor (lb)	I(0)
75	Weight of condensable liquid (lb)	I(0)
76	Non-condensable specific heat in tank (Btu/lb- $^{\circ}$ F)	I(0)
77	Non-condensable molecular weight (lb)	I(0)
78	Weight of oxygen in tank (lb)	I(0)
79	Weight of diluent in tank (lb)	I(0)

<u>Reference Location</u>	<u>Description</u>	<u>Data Type</u>
80	Weight of CO <sub>2</sub> in tank (lb)	I(O)
81	Weight of trace contaminants in tank (lb)	I(O)
82	Weight of special flow type 1 (lb)	I(O)
83	Weight of special flow type 2 (lb)	I(O)
84	Weight of special flow type 3 (lb)	I(O)
85	Weight of special flow type 4 (lb)	I(O)
86	Weight of special flow type 5 (lb)	I(O)
87	Weight of special flow type 6 (lb)	I(O)
88	Specific heat of fluid in tank (Btu/lb-°R)	0
89	Density of fluid in tank (lb/ft <sup>3</sup> )	0
90	Viscosity of fluid (lbm/ft-hr)	0
91	Molecular weight of fluid in tank (lb/lb-mole)	0

## 2.5 Transient V-Array Data

<u>Reference Location</u>	<u>Description</u>	<u>Data Type</u>
92	Thermal capacitance of tank shell (Btu/°F) (If zero, steady state logic will be used.)	I(R)
93	Thermal conductance between internal fluid and tank wall (Btu/hr-°F) (If R(91) = 0 process will be adiabatic.)	I(R)
94	Maximum allowable temperature change in one internal compute step to avoid instabilities in the heat balances (°F) (Typical value = 5°F)	I(R)
95	External heat load to fluid (Btu/hr)	I(R)
96	External heat load to tank wall (Btu/hr)	I(R)
97	Rate of change of mass in tank (lb/hr)	0

## 2.6 Special Input Feature

It will be noticed that reference locations 69 to 87 contain information analogous to what is contained in the A or B arrays. The difference is that pounds replace pounds/hr and R(71) contains the tank volume instead of a pressure. Ordinarily the user must input data values into most of these locations in order to define the initial contents of the tank.

There is a special option (which does not depend on flagging an NSTR variable) which simplifies inputting the initial contents of the tank. The user can either (a) input the actual initial mass of each constituent, or (b) input the percent or fraction or relative amount of each constituent. The value for the total mass in the tank R(69) will be taken as correct and the mass of the individual constituents will be calculated by a scaling or normalization procedure.

### 3.0 Analytical Model Description

#### 3.1 Mass Balance

The rate of change of the mass of species  $i$  within the tank is given by

$$\frac{d m_i}{dt} = w_{i,in} - w_{i,out} \quad (1)$$

#### 3.2 Energy Balance

An energy balance for the fluid within the tank is developed in this section. In words, the conservation of energy for an open system in the absence of mechanical work is

$$\begin{array}{l} \text{Rate of change of} \\ \text{internal energy} \end{array} = \begin{array}{l} \text{Rate of flow of} \\ \text{enthalpy in} \end{array} - \begin{array}{l} \text{Rate of flow of} \\ \text{enthalpy out} \end{array} + \text{Heat added}$$

In mathematical notation

$$\frac{dU}{dt} = \sum_{i=1}^n w_{i,in} h_{i,in} - \sum_{i=1}^n w_{i,out} h_{i,out} + q \quad (2)$$

where

- $U$  = Total internal energy of fluid in the tank (Btu)
- $w_i$  = Rate of flow of species  $i$  (lb/hr)
- $h_i$  = Specific enthalpy of species  $i$  (Btu/lb)
- $q$  = Rate of heat transfer into fluid (Btu/hr)
- $n$  = Number of species in the stream



The subscripts "in" and "out" refer to the inlet and outlet flows, respectively.

The total internal energy within the tank is given by

$$U = \sum_{i=1}^n m_i u_i \quad (3)$$

where

$m_i$  = Mass of species  $i$  in the tank (lb)

$u_i$  = Specific internal energy of species  $i$  (Btu/lb)

The time derivative of  $U$  is broken into two terms as follows:

$$\frac{dU}{dt} = \sum m_i \frac{du_i}{dt} + \sum u_i \frac{dm_i}{dt} \quad (4)$$

For incompressible liquids and ideal gases, the specific internal energy  $u_i$  is a function of temperature only, so that

$$\frac{du_i}{dt} = \frac{du_i}{dT} \frac{dT}{dt} = c_{v,i} \frac{dT}{dt} \quad (5)$$

where

$c_{v,i}$  = The specific heat at constant volume (Btu/lb-°R)

$T$  = Absolute temperature (°R)

Substituting for  $dU/dt$  into the energy equation, Equation 4, gives

$$\sum m_i c_{vi} \frac{dT}{dt} = \sum w_{i,in} h_{i,in} - \sum w_{i,out} h_{i,out} - \sum u_i \frac{dm_i}{dt} + q \quad (6)$$

Using the relation  $dm_i/dt = w_{i,in} - w_{i,out}$  and defining  $\sum m_i c_{vi} = mc_v$  the balance becomes

$$(mc_v) \frac{dT}{dt} = \sum w_{i,in} [h_{i,in} - u_i] - \sum w_{i,out} [h_{i,out} - u_i] + q \quad (7)$$

The specific internal energy can be written

$$u_i = h_i - p v_i \quad (8)$$

for ideal gases  $p v_i = R_{gi} T = (c_{pi} - c_{vi}) T$  whereas for liquids (except at high pressure)  $u_i = h_i$

Thus, both gases and liquid can be handled by the same equation if we set the  $c_v$  of a liquid equal to its  $c_p$ .

Substituting for  $u_i$  into the energy balance gives

$$\begin{aligned} m c_v \frac{dT}{dt} = & \int w_{i,in} (h_{i,in} - h_i) + \int w_{i,in} (c_{pi} - c_{vi}) T \\ & - \int w_{i,out} (c_{pi} - c_{vi}) T + q \end{aligned} \quad (9)$$

but

$$h_{i,in} - h_i = \int_T^{T_{in}} c_{pi} dT = c_{pi} (T_{in} - T) \text{ for constant } c_{pi} \quad (10)$$

Thus, the final energy balance employed in the subroutine is

$$\begin{aligned} m c_v \frac{dT}{dt} = & w_{in} c_{p,in} (T_{in} - T) + w_{in} (c_{p,in} - c_{v,in}) T \\ & - w_{out} (c_{p,out} - c_{v,out}) T + q \end{aligned} \quad (11)$$

where

$m$  = Total mass in tank (lb)

$c_v$  = See  $c_{v,out}$

$T$  = Absolute temperature of fluid in tank ( $^{\circ}R$ )

$w_{in}$  = Total rate of flow into tank (lb/hr)

$T_{in}$  = Absolute temperature of inlet stream ( $^{\circ}R$ )

$c_{p,in}$  = Constant pressure specific heat of inlet stream (Btu/lb- $^{\circ}R$ )

$c_{v,in}$  = Constant volume specific heat of inlet stream (btu/lb- $^{\circ}R$ )

$w_{out}$  = Total flow of outlet stream (lb/hr)

$c_{p,out}$  = Constant pressure specific heat of outlet stream (Btu/lb- $^{\circ}R$ )

$c_{v,out}$  = Constant volume specific heat of outlet stream (Btu/lb- $^{\circ}R$ )

$q$  = Rate of heat transfer into fluid (Btu/hr)

Three checks on the overall energy balance are provided below:

#### Special Cases

1. Adiabatic venting  $w_{in} = 0$   $q = 0$

$$m c_v \frac{dT}{dt} = -w_{out} (c_p - c_v) T \quad (12)$$

but  $w_{out} = -dm/dt$  so that

$$\frac{dT}{T} = \frac{c_p - c_v}{c_v} \frac{dm}{m} = (\gamma - 1) \frac{dm}{m} \quad (13)$$

$$\ln \frac{T_2}{T_1} = (\gamma - 1) \ln \frac{m_2}{m_1} \quad (14)$$

$$\frac{T_2}{T_1} = \left( \frac{m_2}{m_1} \right)^{\gamma - 1} \quad \text{Q.E.D.} \quad (15)$$

2. Adiabatic Filling  $w_{out} = 0$   $dm/dt = w_{in}$   $q = 0$

$$m c_v \frac{dT}{dt} = \frac{dm}{dt} c_p (T_{in} - T) + (c_p - c_v) T \quad (16)$$

$$= \frac{dm}{dt} c_v (\gamma T_{in} - T)$$

$$\frac{dT}{\gamma T_{in} - T} = \frac{dm}{m} \quad (17)$$

$$- \ln \frac{\gamma T_{in} - T_2}{\gamma T_{in} - T_1} = \ln \frac{m_2}{m_1} \quad (18)$$

$$\frac{T_2 - \gamma T_{in}}{T_1 - \gamma T_{in}} = \frac{m_1}{m_2} \quad \text{Q.E.D.} \quad (19)$$

3. Adiabatic draining of a liquid  $c_p = c_v$   $w_{in} = 0$   $q = 0$

$$m c_v \frac{dT}{dt} = 0$$

which implies that the temperature of the liquid stays constant, which is correct.

The external heat flow into the fluid is comprised of two terms

$$q = UA (T_{tank} - T) + q_{elec} \quad (20)$$

where

UA = Heat transfer conductance between fluid and tank wall (Btu/hr-°R)

$T_{\text{tank}}$  = Temperature of tank wall ( $^{\circ}\text{F}$ )

$q_{\text{elect}}$  = A heat source term that simulates an electrical heater immersed in the fluid (Btu/hr)

A thermal balance on the tank shell is also made in the subroutine, to wit:

$$(m c_p) \frac{dT_{\text{tank}}}{dt} = UA (T - T_{\text{tank}}) + q_{\text{elect}} - q_{\text{surr}} \quad (21)$$

where

$(m c_p)$  = Thermal capacitance of tank shell (Btu/ $^{\circ}\text{F}$ )

$q_{\text{elect}}$  = A heat source term which simulates electrical strip heaters mounted directly on the tank (Btu/hr)

$q_{\text{surr}}$  = Heat losses to the surroundings (Btu/hr) (calculated by a standard subroutine, QSURR)

The subroutine calculates a stable computing time based on thermal and fluid flow time constants. The stable computing time is then employed in a simple forward difference integration of the differential equations. For steady state, the thermal balance is solved by iteration.

#### 4.0 Lower Level Subroutines and Functions Required

QSURR

PRØP

#### 5.0 References

- (1) The general energy balances were based on Chapter 15: Macroscopic Balances for Nonisothermal Systems from

Bird, R. B., Stewart, W. E., Lightfoot, E. N.: Transport Phenomena, John Wiley and Sons, Inc. (1960)

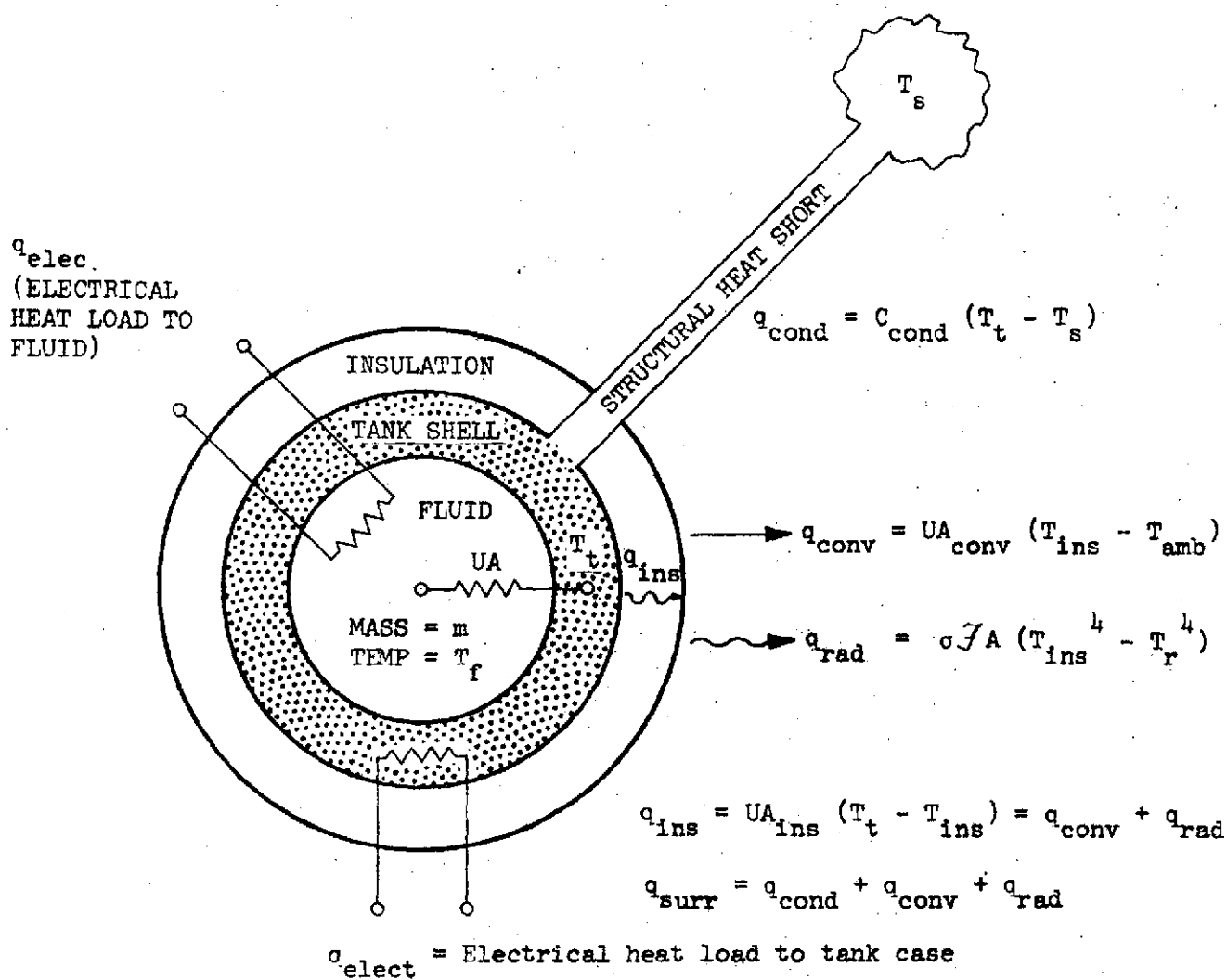


FIGURE 1 THERMAL MODEL FOR TANK SUBROUTINE

## Appendix B

### INPUT DATA CARDS DESCRIPTION

The complete set of G189A input data cards required to run the RITE system simulation are listed in this Appendix. The input cards may be classified into four functional groups, they are: Case Data, K & V Component Data, Table Data and Plot Data. A complete description of each card and options available to the programmer are available in the G189 Program Manual, Reference 3.

The Case Data group contains such data as necessary to specify the overall data storage arrangement, describe the model in general and control the execution of the case. Included are such items as:

- Number of components
- Start time
- Pressure drop flag
- Steady state and/or transient run flag
- Steady state or transient mode type
- Maximum number of iterations allowed
- Iteration tolerances
- Input/output flags
- Maximum/Minimum allowable values of system/component parameters

This group also contains the fluid property data (point values of heat capacity, ratio of heat capacities, viscosity, molecular weight and thermal conductivity) for each of the constituents of flow.

The K and V Component Data groups consists of all the data required to specify and describe each component. These data include the connection and option data entered in integer format and the numerical values



describing the physical properties of the component entered in floating point format. The former are referred to as K array values while the latter are referred to as V array values. Some of these items are:

#### K Array

- Component subroutine number
- Primary and Secondary flow code
- Primary and Secondary source flow comp. no's.
- Extra storage location requirements
- Table numbers for fluid property and pressure drop curves

#### V Array

- Physical dimensions (areas, lengths, diameters, etc.)
- Heat transfer conductance terms
- Heat transfer source temperatures
- Pressure loss coefficients

Each of the K and V array card contains a descriptive comment section which describes its function.

The Table Data group contains all the tables or curves referenced by the K and V Component Data. The form of the tables may be two dimensional (one independent variable) or three dimensional (two independent variables). These tables relate such variables as input flows as a function of time and heat transfer properties as a function of temperature.

The Plot Data group is included as the last portion of data in a case when SK-4060 graphical output is desired. As many as six variables may be plotted in one figure. The plot cards contain "self documenting" commentary feature.

```

TAPE
BASIC          1      239      139      1      YEA      NAY
CASE          RITE JOB USING SS SOLUTION FOR CONDENSER DURING TRANS
$CASE1
  KCHOUT=0, KPRNT=6, KPTINV(1)=1, MAXSLP=6, MINSSI=2, KRUN=1, DTIME=60.,
  TIMEMX=1200., MAXSSI=3, TMAX=2000., TMIN=-450., WTMAX=1.E4 $
$PROPI
CP(1)=1.,RHO(1)=62.4 ,VISC(1)= 3.6 ,WTM(1)=18.0 ,XK(1)=.325 ,
CP(2)=.722 ,RHO(2)=67.6 ,VISC(2)=.396 ,WTM(2)=25.2 ,XK(2)=.2276 ,
CP(3)=.175,RHO(3)=.25 ,VISC(3)= .036 ,WTM(3)=44.0 ,XK(3)=.007 ,
CP(14)=.20 ,RHO(14)=.01144,VISC(14)=.008 ,WTM(14)=2.0 ,XK(14)=.08 ,
CP(15)=.60 ,RHO(15)=81.744,VISC(15)=38. ,WTM(15)=60.1 ,XK(15)=.7 ,
CP(16)=.50 ,RHO(16)=76.0 ,VISC(16)=22. ,WTM(16)=120. ,XK(16)=0.2 ,
CP(5)=.25 ,RHO(5)=.1 ,VISC(5)=.008 ,WTM(5)=46. ,XK(5)=.08 ,
CP(6)=.25 ,RHO(6)=.1 ,VISC(6)=.008 ,WTM(6)=30. ,XK(6)=.08 ,
CP(17)=1. ,RHO(17)=62.4 ,VISC(17)=3.6 ,WTM(17)=18. ,XK(17)=.325 ,
CP(25)=.60 ,RHO(25)=81.744,VISC(25)=38. ,WTM(25)=1.0E+9,XK(12)=.7 ,
CP(26)=.50 ,RHO(26)=76.0 ,VISC(26)=22. ,WTM(26)=1.0E+9,XK(26)=0.2 ,
GAMGAS=1.4, VISGAS=0.44, WTMCON=18., WTMOIL=28., WTMTC=17.,
CPCONL=1., CPCONV=0.44, CPCO2=0.2, CPDIL=0.25, CPOXY=0.22, CPTC=0.5,
CPCO2 = .178, WTMCON=18. , XKLIO=.36 ,
XKGAS=0.146 $
ID** 1 COOLANT SUPPLY
NSTR 1 -0 NONE RORD
KBAS 1 49 0 2
VARY 1 1 280.0 COOLANT INPUT FLOW RATE(LB/HR)
VARY 1 2 35.0 COOLANT INLET TEMPERATURE
ID** 2 -0 SPLIT TO WATER COOLER OR WATER HEATER(NORMALLY TO COOLER)
KBAS 2 -0 10 9 0 2 0 2 3
NSTR 2 -00 USE UNIVERSAL SPLIT RATIO FOR ALL FLOWS
ID** 3 -0 LIQMIX FROM HEATING LOOP (NORMALLY COOLANT FLOW)
KBAS 3 -0 7 2 0 2 120 0 2 80
NSTR 3 -01 USE PRIMARY FLOW PRESS AS MIX PRESS
ID** 5 -0 SPLIT TO HEATING LOOP (NORMALLY COOLANT FLOW )
KBAS 5 -0 10 -94 0 2 0 2 6
NSTR 5 -00 USE UNIVERSAL SPLIT RATIO FOR ALL FLOWS
ID** 6 -0 LIQMIX FROM WATER HEATER(NORMALLY COOLANT FLOW )
KBAS 6 -0 7 5 0 2 119 0 2 10
NSTR 6 -01 USE PRIMARY FLOW PRESS AS MIX PRESS
ID** 8 -0 METERED OUTLET COOLANT FLOW
KBAS 8 -0 29 4 10 0 2 180
NSTR 8 -00 METER TOTAL FLOW ONLY.
ID** 9 -0 FLOW TEE TO PROVIDE FLOW TO WATR SUPPLY AND CONDENSER
KBAS 9 -0 10 1 0 2 2
NSTR 9 -00 UNIVERSAL SPLIT RATIO
VARY 9 1 20.0 COOLANT FLO BYPASSES CONDENSER LB/HR
VARY 9 2 35. TEMPERATURE OF THE COOLANT DEG-F
VARY 9 3 15. PRESS. PSI
VARY 9 4 15. PRESS. PSI
VARY 9 20 260. COOLANT FLOWRATE THROUGH CONDENSER LB/HR
VARY 9 21 35. TEMPERATURE OF THE COOLANT DEG-F
VARY 9 22 15.
VARY 9 23 15.
VARY 9 65 .924
ID** 10 -0 FLO JUNCTION ALLOW FOR RETURN FLOW FROM EITHER HX FOR H2O SUPPLI
KBAS 10 -0 7 6 0 2 -238 0 2 8
NSTR 10 -03 USE MAX PRESS
ID** 21 -0 COMPONENT PROVIDING AIR/URINE FLOW THROUGH THE COMODE.
KBAS 21 -0 49 3 142526171819 22

```

```

NSTR 21 -0
ID** 22 -0 COMMODE MIXING THE AIR FLOW WITH WATER.
KBAS 22 -0 6 21 3 142526171819 -47 3 142526171819 23
NSTR 22 -01 THE MIXED FLOW PRESSURE EQUALS PRIM SOURCE
ID** 23 -0 LIQUID-GAS SEPARATOR.
KBAS 23 -0 10 12 22 3 142526171819 3 142526171819 28
NSTR 23 -01 EACH OUTLET CONSTITUENT USES OWN SPLIT RATIO
VARY 23 73 1.0 SPIT RATIO FOR SF* 2 (UREA)
VARY 23 74 1.0 SPIT RATIO FOR SF* 3 (SOLIDS)
VARY 23 75 1.0 SPIT RATIO FOR SF* 4 (WATER)
ID** 24 1 AIR BLOWER 20 CFM ,STATIC PRESS=15. INCHES WATER, REQUIRING
ID** 24 2 APPROXIMATELY 120 WATTS. SUBROUTINE FAN
KBAS 24 -0 23 28 3 14 5 6 7 8 9 25
NSTR 24 -0003 ADD HEAT SPECIFIED IN R(91) ONLY.
ID** 25 -0 REGENERATIVE COUNTERFLOW HEAT EXCHANGER (AIR STERILIZER).
KBAS 25 -0 37 50 35 24 3 14 5 6 7 8 9 27
KARY 25 16 12 NUMBER OF NODES
KARY 25 17 35 NUMBER OF HEAT TRANSFER TERMS
KARY 25 19 0000001000 STORAGE TERM,NODE NO. 1
KARY 25 20 0000002000 STORAGE TERM,NODE NO. 2
KARY 25 21 0000003000 STORAGE TERM,NODE NO. 3
KARY 25 22 0000004000 STORAGE TERM,NODE NO. 4
KARY 25 23 0000005000 STORAGE TERM,NODE NO. 5
KARY 25 24 0000006000 STORAGE TERM,NODE NO. 6
KARY 25 25 0000007000 STORAGE TERM, NODE NO. 7
KARY 25 26 0000008000 STORAGE TERM, NODE NO. 8
KARY 25 27 0000009000 STORAGE TERM, NODE NO. 9
KARY 25 28 0000010000 STORAGE TERM, NODE NO. 10
KARY 25 29 0000011000 STORAGE TERM, NODE NO. 11
KARY 25 30 0000012000 STORAGE TERM, NODE NO. 12
KARY 25 31 3000007116 FLUID CONVECTION SOURCE TO 7
KARY 25 32 5000007004 WALL TO FLUID CONVECTION 4 TO 7
KARY 25 33 4000007001 WALL TO FLUID CONVECTION 1 TO 7
KARY 25 34 2000008007 FLUID CONVECTION ,7 TO 8
KARY 25 35 5000008005 WALL TO FLUID CONVECTION 5 TO 8
KARY 25 36 4000008002 WALL TO FLUID CONVECTION 2 TO 8
KARY 25 37 2000009008 FLUID CONVECTION ,8 TO 9
KARY 25 38 5000009006 WALL TO FLUID CONVECTION 6 TO 9
KARY 25 39 4000009003 WALL TO FLUID CONVECTION 3 TO 9
KARY 25 40 2000011010 FLUID CONVECTION 10 TO 11
KARY 25 41 4000011005 WALL TO FLUID CONVECTION 5 TO 11
KARY 25 42 2000012011 FLUID CONVECTION 11 TO 12
KARY 25 43 4000012004 WALL TO FLUID CONVECTION 4 TO 12
KARY 25 44 2000010009 FLUID CONVECTION 9 TO 10
KARY 25 45 4000010006 WALL TO FLUID CONVECTION 6 TO 10
KARY 25 46 6000001117 CONDUCTION SOURCE TO 1
KARY 25 47 6000002117 CONDUCTION SOURCE TO 2
KARY 25 48 6000003118 CONDUCTION SOURCE TO 3 HEAT FROM HEAT BLOCK
KARY 25 49 5000001002 NODE TO NODE CONDUCTION 1 TO 2
KARY 25 50 5000002003 NODE TO NODE CONDUCTION 2 TO 3
KARY 25 51 5000006005 NODE TO NODE CONDUCTION 6 TO 5
KARY 25 52 5000005004 NODE TO NODE CONDUCTION 5 TO 4
KARY 25 53 7000006003 NODE TO NODE RADIATION 6 TO 3
VARY 25 65 1.00 SOLUTION CONVERGENCE TOLERANCE (DEG F)
VARY 25 69 220.0 TEMPERATURE NODE NO. 1
VARY 25 70 400.00 TEMPERATURE NODE NO. 2
VARY 25 71 1280.0 TEMPERATURE NODE NO. 3
VARY 25 72 230.0 TEMPERATURE NODE NO. 4
VARY 25 73 400.00 TEMPERATURE NODE NO. 5

```

VARY	25	74	1250.0	TEMPERATURE	NODE	NO. 6	
VARY	25	75	1200.0	TEMPERATURE	NODE	NO. 7	
VARY	25	76	200.00	TEMPERATURE	NODE	NO. 8	
VARY	25	77	1200.0	TEMPERATURE	NODE	NO. 9	
VARY	25	78	1200.0	TEMPERATURE	NODE	NO.10	
VARY	25	79	1200.0	TEMPERATURE	NODE	NO.11	
VARY	25	80	1200.0	TEMPERATURE	NODE	NO.12	
VARY	25	81	0.268	CAPACITANCE	NODE	NO. 1	
VARY	25	82	0.252	CAPACITANCE	NODE	NO. 2	
VARY	25	83	0.126	CAPACITANCE	NODE	NO. 3	
VARY	25	84	0.1078	CAPACITANCE	NODE	NO. 4	
VARY	25	85	0.3895	CAPACITANCE	NODE	NO. 5	
VARY	25	86	0.1100	CAPACITANCE	NODE	NO. 6	
VARY	25	87	0.	CAPACITANCE	NODE	NO. 7	
VARY	25	88	0.	CAPACITANCE	NODE	NO. 8	
VARY	25	89	0.	CAPACITANCE	NODE	NO. 9	
VARY	25	90	0.	CAPACITANCE	NODE	NO. 10	
VARY	25	91	0.	CAPACITANCE	NODE	NO. 11	
VARY	25	92	0.	CAPACITANCE	NODE	NO. 12	
VARY	25	93	1.0	FLUID CONVECTION	SOURCE	TO 7	
VARY	25	94	3.31	WALL TO FLUID CONVECTION	4	TO 7	
VARY	25	95	3.0	WALL TO FLUID CONVECTION	1	TO 7	
VARY	25	96	1.0	FLUID CONVECTION	7	TO 8	
VARY	25	97	32.	WALL TO FLUID CONVECTION	5	TO 8	
VARY	25	98	3.0	WALL TO FLUID CONVECTION	2	TO 8	
VARY	25	99	1.0	FLUID CONVECTION	8	TO 9	
VARY	25	100	3.3				
VARY	25	101	3.0				
VARY	25	102	1.0	FLUID CONVECTION	10	TO 11	
VARY	25	103	32.				
VARY	25	104	1.0	FLUID CONVECTION	11	TO 12	
VARY	25	105	1.4	WALL TO FLUID CONVECTION	4	TO 12	
VARY	25	106	1.0	FLUID CONVECTION	9	TO 10	
VARY	25	107	2.3	WALL TO FLUID CONVECTION	6	TO 10	
VARY	25	108	0.7	CONDUCTION	SOURCE	TO 1	
VARY	25	109	0.7	CONDUCTION	SOURCE	TO 2	
VARY	25	110	2000.	CONDUCTION	SOURCE	TO 3	
VARY	25	111	0.0275	NODE TO NODE CONDUCTION	1	TO 2	
VARY	25	112	0.0275	NODE TO NODE CONDUCTION	2	TO 3	
VARY	25	113	0.1874	NODE TO NODE CONDUCTION	6	TO 5	
VARY	25	114	0.1874	NODE TO NODE CONDUCTION	5	TO 4	
VARY	25	115	.135	INLET TEMPERATURE (DEG F)			
VARY	25	116	100.0	INLET TEMPERATURE (DEG F)			
VARY	25	117	250.0	ENVIRONMENTAL TEMPERATURE (DEG F)			
VARY	25	118	1280.0	HEAT BLOCK TEMPERATURE (DEG F)			
ID**	27	-0	FLOWMETER TO MEASURE THE FLOW OUT OF THE AIR STERILIZER.				
KBAS	27	-0	29 10	25 3 14 5 6 7 8 9			203
NSTR	27	-00		METER TOTAL FLOW			
ID**	28	-0	FLOW JUNCTION PROVIDES MEANS OF INTRODUCING AIR FROM WATER LOOP				
KBAS	28	-0	6	23 3 142526171819 232 3 14 5 6 7 8 9			24
NSTR	28	-02		SECONDARY PRESSURE# MIX PRESSURE			
ID**	41	1	FLUSH WATER TANK WITH INTERNAL HEAT EXCHANGER OUTLET FLOW				
ID**	41	2	DETERMINED BY GPOLY LOGIC STEADY STATE AND TRANSIENT HEAT BL				
KBAS	41	-0	30	45 4 141516 11819 -124 0 1			43
NSTR	41		12012	OUTFLO=GPOLY.INTERNAL HX			
VARY	41	51	100.	TANK TEMP			DEG-F
VARY	41	54	75.	AMBIENT GAS TEMPERATURE			F
VARY	41	60	80.	STRUCTURE TEMP.			F
VARY	41	61	2.77778	CONDUCTANCE BETWEEN TNK WALL/STRUCTURE			B/H-F

VARY	41	67	.578	EFFECTIVE UA BETWEEN HEAT EXCHANGER TO FLUID	
VARY	41	68	40.0	MAXIMUM CAPACITY OF TANK (LB)	
VARY	41	69	30.	TOTAL FLUID WEIGHT IN TNK	LB
VARY	41	70	100.	FLUID TEMP IN TNK	F
VARY	41	71	.48077	FLUID VOLUME IN TNK	FT**3
VARY	41	72	15.	FLUID PRESS IN TNK	PSI
VARY	41	85	30.	WEIGHT OF CONDENSABLE LIQUID	LB
VARY	41	88	1.	SPECIFIC HEAT OF FLUIDS IN TNK	B/LB-F
VARY	41	92		THERMAL CAPACITANCE OF TNK SELL	BTU/F
VARY	41	93	100.	CONDUCTANCE INTERVAL FLUID/TNK	B/HR-F
VARY	41	94	5.0	MAX ALLOWABLE TEMP CHANGE	F
VARY	41	95		EXTERNAL HT LOAD TO FLUID	BTU/HR
VARY	41	96		EXTERNAL HT LOAD TO SHELL	BTU/HR
ID**	43	-0		PIPING TEE PROVIDING FOR MEANS OF STORING THE ECS CONDENSATE	
KBAS	43	-0	7	41 4 141516 11819 -51 4 141516 11819	44
NSTR	43	-0			
ID**	44	-0		PUMP USED FOR CIRCULATING AND PUMPING THE FLUSH WATER.	
KBAS	44	-0	22	43 4 141516 11819	45
NSTR	44	-00002		ADD HEAT SPECIFIED NO OTHER CALCULATIONS REQRO	
ID**	45	-0		VALVE DIRECTING THE FLOW TO EITHER THE TANK OR WASTE MANG SYS	
KBAS	45	-0	10 5	44 4 141516 11819 4 141516 11819	46
NSTR	45	-00		USE UNIVERSAL SPLIT RATIO ALL CONSTITUENTS	
ID**	46	-0		PIPING TEE DIRECTING FLOW TO URINE SYS AND/OR SOLID PROC SYS	
KBAS	46	-0	10 5	-45 4 141516 11819 4 141516 11819	47
NSTR	46	-00		USE UNIVERSAL SPLIT RATIO ALL CONSTITUENTS	
ID**	47	-0		VALVE SIMULATED BY DUMMY COMPONENT TO THE COMMODE	
KBAS	47	-0	49	46 4 141516 11819	48
NSTR	47	-00		VALVE CONTROLLING FLOW TO COMMODE PASS DATA	
ID**	48	-0		PIPING TEE DIRECTING FLOW TO THE DEFECTION AND/OR TRASH UNIT	
KBAS	48	-0	10 5	-46 4 141516 11819 4 141516 11819	49
NSTR	48	-00		USE UNIVERSAL SPLIT RATIO ALL CONSTITUENTS	
ID**	49	-0		VALVE SIMULATED BY DUMMY COMPONENT TO TRASH PROCESSING UNIT.	
KBAS	49	-0	49	48 4 141516 11819	50
NSTR	49	-00		VALVE CONTROLLING FLO TO SHREDDER PASS DATA	
ID**	50	-0		VALVE SIMULATED BY DUMMY COMPONENT, TO THE DEFECTION UNIT.	
KBAS	50	-0	49	-48 4 141516 11819	21
NSTR	50	-00		VALVE CONTROLLING FLO TO CRAPPER PASS DATA	
ID**	51	-0		CONDENSATE INPUT FROM ECS PRESPARATION/RESPIRATION	
KBAS	51	-0	49	4 141516 11819	
NSTR	51	-00		NONE REQRO	
VARY	51	2	70.04	ECS CONDENSATE INPUT TEMP(DEG F))	
VARY	51	17	0.834	ECS CONDENSATE INPUT FLOW (LB/HR)	
ID**	61	-0		FLOW METERING OF GASSES OUTFLOWIN FROM CONENSING TNK	
KBAS	61	-0	29 17	-237 3 14	62
NSTR	61	-01		METER ALL CONSTITUENT FLOWS	
ID**	62	-0		FLOW CONTROL VALVE SIMULATED BY LTOMIX	
KBAS	62	-0	10	237 0 1 -65 0 1	63
NSTR	62	-03		USE PRESS FROM PRINC SOURCE	
ID**	63	-0		WATER PUMP USED IN PUMPING FLOW FROM EITHER/TO WATER ACCU.	
KBAS	63	-0	22	62 0 1	64
NSTR	63	-00002		ADD HT SPECIFIED NO OTHER CALCS REQRO	
ID**	64	1		TEE WHICH PROVIDES A MEANS OF FLOWING WATER FROM THE ACCUMULATOR	
ID**	64	2		TANK OR FROM THE CONDENSING TANK	
KBAS	64	-0	10	63 0 1	65
NSTR	64	-00		PRIMARY SOURCE PRESSURE USED	
VARY	64	65	1.0	PUMP DIRECTLY TO ROTABLE WATER UNIT	
ID**	65	-0		WATER SIDE OF WATER ACCUMULATOR CYCLIC OPERATION	
KBAS	65	-0	30	64 0 1	66
NSTR	65	12010		OUTFLO=GPOLY,NO INTERNAL HX	

VARY	65	51	85.0	TANK TEMP	DEG-F
VARY	65	54	75.	AMBIENT GAS TEMPERATURE	DEG-F
VARY	65	55	1.	THERMAL CONDUCTANCE BETWEEN SURFACE AND AMB	B/HR-F
VARY	65	57	80.	AMB WALL TEMP	DEG-F
VARY	65	58	.22	RADIATION , FA, FACTOR	FT**2
VARY	65	60	80.	STURCTURE TEMP	DEG-F
VARY	65	61	1.	CONDUCTANCE (KA/X) TANK TO STRUCTURE	B/HR-F
VARY	65	68	10.	CAPACITY OF TANK	LBS.
VARY	65	69	10.	INITIAL WEIGHT OF TNK	LBS.
VARY	65	70	85.	INITIAL TEMPERAT OF TNK	F-DEG
VARY	65	71	.16	TANK VOLUME	FT**3
VARY	65	72	15.	TANK PRESSURE	PSI
VARY	65	75	10.	WEIGHT OF LIQUID IN TNK	LB
VARY	65	88	1.	CP OF FLUID IN TANK	B/LB-F
VARY	65	89	62.4	DENSITY OF FLUID IN TNK	LB/FTCU
VARY	65	90	3.6	VISCOSITY OF WATER	
VARY	65	91	18.	MOLECUCULAR WEIGHT OF FLUID IN TANK	LB/LB-MOLE
VARY	65	92	.5	THERMAL CAPACITANCE OF TNK SHELL	B//OI
VARY	65	94	5.0	MAX ALLOWABLE TEMP CHANGE	F-DEG
ID**	66	-0		FLOW METER FOR MEASUREMENT OF WATER FLOW.	
KBAS	66	-0	29 4	-64 0 1	9
NSTR	66	-00		METER TOTAL FLOW ONLY	
ID**	80	-0		TEE PROVIDING FLOW TO 2 OTHER SPLITS NO. 81 AND 82 SIMULTD BY SPLIT	
KBAS	80	-0	10	66 0 1 0 1	81
NSTR	80	-00		UNIVERSAL SPLIT RATIO FOR AL CONSTITUENT FLOS	
ID**	81	-0		TEE PROVIDING FLOW TO TNKS 1 AND 2 SIMULATED BY SPLIT	
KBAS	81	-0	10	80 0 1 0 1	82
NSTR	81	-00		UNIVERSAL SPLIT RATIO FOR AL CONSTITUENT FLOS	
ID**	82	-0		TEE PROVIDING FLOW TO TNKS 3 AND 4 SIMULATED BY SPLIT	
KBAS	82	-0	10	-80 0 1 0 1	83
NSTR	82	-00		UNIVERSAL SPLIT RATIO FOR AL CONSTITUENT FLOS	
ID**	83	-0		POTABLE STORAGE TNK 4 (TANKG)	
KBAS	83	-0	30	82 0 1 115 0 1	84
NSTR	83	12012		OUTFLO=GPOLY,INTERNAL HX	
VARY	83	51	150.	TANK TEMP	DEG-F
VARY	83	54	75.	AMBIENT GAS TEMP	F
VARY	83	57	75.	AMBIENT WALL TEMP	F
VARY	83	60	80.	STRUCTURE TEMP	F
VARY	83	61	3.15369	CONDUCTANCE TNK WALL/STRUCTURE	B/HR-F
VARY	83	64	0.	CONDUCTANCE TNK WALL/INSULATION SURFACE	BTU/HR-F
VARY	83	67	6.656254	THERMAL CONDUCTANCE OF HEATING TUBES	
VARY	83	68	41.5	MAXIMUM FLUID WEIGHT IN TANK (LB)	
VARY	83	69	20.0	INITIAL WEIGHT OF WATER IN TANK (LB)	
VARY	83	70	150.	TEMP OF FLUID	F
VARY	83	71	.609	FLUID VOLUME IN TNK	FT**3
VARY	83	72	15.	PRESS	PSI
VARY	83	88	1.0	SPECIFIC HEAT OF FLUID IN TNK	BTU/LB-F
VARY	83	92	1.0	THERMAL CAPACITANCE OF TANK (BTU/DEG F)	
VARY	83	93	100.	THERMAL CONDUCTANCE BETWEEN FLUID AND WALL (BTU/H-F)	
VARY	83	94	5.0	MAX ALLOWABLE TEMP. CHANGE (DEG :)	
ID**	84	-0		POTABLE STORAGE TNK 3 (TANKG)	
KBAS	84	-0	30	-82 0 1 114 0 1	85
NSTR	84	12012		OUTFLO=GPOLY,INTERNAL HX	
VARY	84	51	150.	TANK TEMP	DEG-F
VARY	84	54	75.	AMBIENT GAS TEMP	F
VARY	84	57	75.	AMBIENT WALL TEMP	F
VARY	84	60	80.	STRUCTURE TEMP	F
VARY	84	61	3.15369	CONDUCTANCE TNK WALL/STRUCTURE	B/HR-F
VARY	84	64	0.	CONDUCTANCE TNK WALL/INSULATION SURFACE	BTU/HR-F

VARY	84	67	6.656254	THERMAL CONDUCTANCE OF HEATING TUBES	
VARY	84	68	41.5	MAXIMUM FLUID WEIGHT IN TANK (LB)	
VARY	84	69	20.0	INITIAL WEIGHT OF WATER IN TANK (LB)	
VARY	84	70	150.	TEMP OF FLUID	F
VARY	84	71	.609	FLUID VOLUME IN TNK	FT**3
VARY	84	72	15.	PRESS	PSI
VARY	84	88	1.0	SPECIFIC HEAT OF FLUID IN TNK	BTU/LB-F
VARY	84	92	1.0	THERMAL CAPACITANCE OF TANK (BTU/DEG F)	
VARY	84	93	100.	THERMAL CONDUCTANCE BETWEEN FLUID AND WALL (BTU/H-F)	
VARY	84	94	5.0	MAX ALLOWABLE TEMP. CHANGE (DEG :)	
ID**	85	-0	JUNCTION TO MERGE FLO FROM TNKS 3 AND 4 (LIQMIX)		
KBAS	85	-0	7	83 0 1 -84 0 1	87
NSTR	85	-00		USE MIN PRESSURE AS MIXED FLO PRESS	
ID**	86	-0	POTABLE STORAGE TNK 2 (TANKG)		
KBAS	86	-0	30	-81 0 1 113 0 1	88
NSTR	86	12012		OUTFLO=GPOLY,INTERNAL HX	
VARY	86	51	150.	TANK TEMP	DEG-F
VARY	86	54	75.	AMBIENT GAS TEMP	F
VARY	86	57	75.	AMBIENT WALL TEMP	F
VARY	86	60	80.	STRUCTURE TEMP	F
VARY	86	61	3.15369	CONDUCTANCE TNK WALL/STRUCTURE	B/HR-F
VARY	86	64	0.	CONDUCTANCE TNK WALL/INSULATION SURFACE	BTU/HR-F
VARY	86	67	6.656254	THERMAL CONDUCTANCE OF HEATING TUBES	
VARY	86	68	41.5	MAXIMUM FLUID WEIGHT IN TANK (LB)	
VARY	86	69	20.0	INITIAL WEIGHT OF WATER IN TANK (LB)	
VARY	86	70	150.	TEMP OF FLUID	F
VARY	86	71	.609	FLUID VOLUME IN TNK	FT**3
VARY	86	72	15.	PRESS	PSI
VARY	86	88	1.0	SPECIFIC HEAT OF FLUID IN TNK	BTU/LB-F
VARY	86	92	1.0	THERMAL CAPACITANCE OF TANK (BTU/DEG F)	
VARY	86	93	100.	THERMAL CONDUCTANCE BETWEEN FLUID AND WALL (BTU/H-F)	
VARY	86	94	5.0	MAX ALLOWABLE TEMP. CHANGE (DEG :)	
ID**	87	-0	POTABLE STORAGE TNK 1 (TANKG)		
KBAS	87	-0	30	81 0 1 112 0 1	86
NSTR	87	12012		OUTFLO=GPOLY,INTERNAL HX	
VARY	87	51	150.	TANK TEMP	DEG-F
VARY	87	54	75.	AMBIENT GAS TEMP	F
VARY	87	57	75.	AMBIENT WALL TEMP	F
VARY	87	60	80.	STRUCTURE TEMP	F
VARY	87	61	3.15369	CONDUCTANCE TNK WALL/STRUCTURE	B/HR-F
VARY	87	64	0.	CONDUCTANCE TNK WALL/INSULATION SURFACE	BTU/HR-F
VARY	87	67	6.656254	THERMAL CONDUCTANCE OF HEATING TUBES	
VARY	87	68	41.5	MAXIMUM FLUID WEIGHT IN TANK (LB)	
VARY	87	69	20.0	INITIAL WEIGHT OF WATER IN TANK (LB)	
VARY	87	70	150.	TEMP OF FLUID	F
VARY	87	71	.609	FLUID VOLUME IN TNK	FT**3
VARY	87	72	15.	PRESS	PSI
VARY	87	88	1.0	SPECIFIC HEAT OF FLUID IN TNK	BTU/LB-F
VARY	87	92	1.0	THERMAL CAPACITANCE OF TANK (BTU/DEG F)	
VARY	87	93	100.	THERMAL CONDUCTANCE BETWEEN FLUID AND WALL (BTU/H-F)	
VARY	87	94	5.0	MAX ALLOWABLE TEMP. CHANGE (DEG :)	
ID**	88	-0	JUNCTION TO MERGE FLO FROM TNKS 1 AND 2 (LIQMIX)		
KBAS	88	-0	7	86 0 1 -87 0 1	90
NSTR	88	-00		USE MIN PRESSURE AS MIXED FLO PRESS	
ID**	89	-0	JUNCTION PROVIDES MEANS OF UTILIZING EMERGENCY POTABLE WATER SUPPL		
KBAS	89	-0	7	85 0 1 -90 0 1	91
NSTR	89	-00		USE MIN PRESSURE AS MIXED FLO PRESS	
ID**	90	-0	EMERGENCY POTABLE STORAGE TNK (TANKG)		
KBAS	90	-0	30	0 1 116 0 1	89

NSTR	90	12012	OUTFLO=GPOLY, INTERNAL HX		
VARY	90	51 150.	TANK TEMP		DEG-F
VARY	90	54 75.	AMBIENT GAS TEMP		F
VARY	90	57 75.	AMBIENT WALL TEMP		F
VARY	90	60 80.	STRUCTURE TEMP		F
VARY	90	61 3.15369	CONDUCTANCE TNK WALL/STRUCTURE		B/HR-F
VARY	90	64 0.	CONDUCTANCE TNK WALL/INSULATION SURFACE		BTU/HR-F
VARY	90	67 6.656254	THERMAL CONDUCTANCE OF HEATING TUBES		
VARY	90	69 41.0	INITIAL WEIGHT OF WATER IN TANK (LB)		
VARY	90	70 150.	TEMP OF FLUID		F
VARY	90	71 .609	FLUID VOLUME IN TNK		FT**3
VARY	90	72 15.	PRESS		PSI
VARY	90	88 1.0	SPECIFIC HEAT OF FLUID IN TNK		BTU/LB-F
VARY	90	92 1.0	THERMAL CAPACITANCE OF TANK (BTU/DEG F)		
VARY	90	93 100.	THERMAL CONDUCTANCE BETWEEN FLUID AND WALL (BTU/H-F)		
VARY	90	94 5.0	MAX ALLOWABLE TEMP. CHANGE (DEG :)		
ID**	91	-0 JUNCTION PROVIDES OF MERGING ALL POTABLE WATER SUPPLY (LIOMIX)			
KBAS	91	-0 7	89 0 1	-88 0 1	99
NSTR	91	-00	USE MIN PRESSURE AS MIXED FLO PRESS		
ID**	92	-0 TEE PROVIDING FLOW TO WATER COOLER/HEATER SIMULATED BY SPLIT			
KBAS	92	-0 10	91 0 1	0 1	93
NSTR	92	-00	UNIVERSAL SPLIT RATIO FOR AL CONSTITUENT FLOS		
ID**	93	-0 WATER COOLER SIMULATED BY ANYHX			
KBAS	93	-0 4	92 0 1	-118 0 1	95
NSTR	93	200020000 0000001 COUNTERFLOW, LIQ-LIQ --SS CALCULATION			
VARY	93	66 12.0	OVERALL UA BTU/HR-F		
ID**	94	-0 WATER HEATER SIMULATED BY ANYHX			
KBAS	94	-0 4	-92 0 1	-3 0 2	96
NSTR	94	200020000 0000001 COUNTERFLOW, LIQ-LIQ --SS CALCULATION			
VARY	94	66 12.0	OVERALL UA BTU/HR-F		
ID**	95	-0 FLOWMETER SIMULATED BY FLOMET			
KBAS	95	-0 29 4	93 0 1		94
NSTR	95	-00	METER TOTAL FLO ONLY		
ID**	96	-0 FLOWMETER SIMULATED BY FLOMET			
KBAS	96	-0 29 4	94 0 1		5
NSTR	96	-00	METER TOTAL FLO ONLY		
ID**	99	-0 WATER DELIVERY PUMP			
KBAS	99	-0 22	91 0 1		92
NSTR	99	-00002	ADD HEAT ONLY		
ID**	100	-0 LOW TEMPERATURE RADIOISOTOPE HEATER			
KBAS	100	-0 37.14 9	137 0		102
NSTR	100	-00	USE CURRENT SIMULATION TIME		
KARY	100	16	3	NUMBER OF NODES	
KARY	100	17	9	NUMBER OF HEAT TRANSFER TERMS	
KARY	100	19 0000001000		STORAGE TERM NO 1 (BTU/F)	
KARY	100	20 0000002000		STORAGE TERM NO 2 (BTU/F)	
KARY	100	21 0000003000		STORAGE TERM NO 3 (BTU/F)	
KARY	100	22 3000001082		FLUID CONVECTION SOURCE TO NODE TERM (B/HR-F)	
KARY	100	23 5000002001		CONDUCTANCE NODE 1 TO NODE 2 (B/HR-F)	
KARY	100	24 5000003001		CONDUCTANCE NODE 1 TO NODE 3 (B/HR-F)	
KARY	100	25 5000003002		CONDUCTANCE NODE 2 TO NODE 3 (B/HR-F)	
KARY	100	26 6000003081		CONDUCTANCE NODE 3 TO SURROUNDINGS (B/HR-F)	
KARY	100	27 1000002080		HEAT GENERATION TERM BTU/HR	
VARY	100	65 2.5			
VARY	100	66 0.		PERIOD (NOT RORD)	
VARY	100	67 0.		INITIAL DISPLACEMENT OF INDEPENDENT VARBL NOT RORD	
VARY	100	68		CALCULATION TIME STEP. (SECONDS)	
VARY	100	69 185.0		TEMPERATURE OF NODE 1 (DEG F)	
VARY	100	70 640.0		TEMPERATURE OF NODE 2 (DEG F)	



VARY 100	71	170.27	TEMPERATURE OF NODE 3	DEG-F
VARY 100	72	2.	CAPACITANCE NODE 1 (BTU/F)	
VARY 100	73	20.	CAPACITANCE NODE 2 (BTU/F)	
VARY 100	74	240.	CAPACITANCE NODE 3 (BTU/F)	
VARY 100	75	603.0	FLUID CONVECTION WCP SOURCE TO NODE 1 (B/HR-F)	
VARY 100	76	8.61	CONDUCTION NODE 1 TO NODE 2 (B/HR-F)	
VARY 100	77	8.57	CONDUCTION NODE 1 TO NODE 3 (B/HR-F)	
VARY 100	78	2.45	CONDUCTION NODE 2 TO NODE 3 (B/HR-F)	
VARY 100	79	18.8	CONDUCTION NODE 3 TO SOURCE, SURROUNDINGS (B/HR-F)	
VARY 100	80	5120.	HEAT GENERATED (BTU/HR)	
VARY 100	81	80.	EFFECTIVE TEMPERATURE OF THE SURROUNDINGS	
VARY 100	82	179.0	TEMPERATURE OF INLET STREAM (DEG F)	
ID** 101	-0	EXPANSION TANK	SIMULATED BY TANKS	
KBAS 101	-0	30		
NSTR 101	12010		OUTFLO=GPOLY, NO INTERNAL HX	
VARY 101	51	100.	TANK TEMP	DEG-F
VARY 101	54	75.	AMBIENT GAS TEMPERATURE	DEG-F
VARY 101	55	.01	THERMAL CONDUCTANCE BTWN SURFACE OF INSULATION/GAS	
VARY 101	57	70.	AMBIENT WALL TEMPERATURE	DEG-F
VARY 101	58	1.2	THERMAL RADIATION FA FACTOR	FT**2
VARY 101	60	140.	STRUCTURE TEMPERATURE	DEG-F
VARY 101	61	.005	CONDUCTANCE KA/X TO THE STRUCTURE	B/HR-FT-F
VARY 101	64	0.	NO INSULATION	
VARY 101	68	3.5	FLUID CAPACITY OF WATER ACCUMULATOR	LBS
VARY 101	69	2.0	TOTAL FLUID WEIGHT IN TNK	LBS.
VARY 101	70	100.	FLUID TEMP. IN TNK	F
VARY 101	71	.0320512	FLUID VOLUME IN TNK	FT**3
VARY 101	72	15.0	FLUID PRESS IN TNK	PSI
VARY 101	75	2.0	WEIGHT OF CONDENSABLE LIQUID	LBS
VARY 101	88	1.0	SPECIFIC HEAT OF FLUIDS IN TNK	B/LB-F
VARY 101	92		THERMAL CAPACITANCE OF TNK SHELL	BTU/F
VARY 101	93		CONDUCTANCE BETWEEN INTERNAL FLUID TNK	BTU/HR-F
VARY 101	94	5.0	MAX ALLOWABLE TEMP CHANGE	F
VARY 101	95		EXTERNAL HT LOAD TO FLUID	BTU/HR
VARY 101	96	0.	EXTERNAL HT TO SHELL	BTU/HR
ID** 102	-0	LIQMIX FROM EXPANSION TANK		
KBAS 102	-0	7	101 0 1	-100 0 1 103
NSTR 102	-01			MIX PRESS DEPENDENT ON PRINCIPAL SOURCE
ID** 103	-0	SPLIT TO PUMPS		
KBAS 103	-0	10	102 0 1	0 1 104
NSTR 103	-00			USE UNIVERSAL SPLIT RATIO FOR ALL FLOWS
ID** 104	-0	COOLANT PUMP NO. 1		
KBAS 104	-0	22	103 0 1	
NSTR 104	-00002			ADD HEAT SPECIFIED NO OTHER CALCULATIONS RORD
VARY 104	85	10.	HEAT ADDED TO FLUID STREAM	WATTS
ID** 105	-0	COOLANT PUMP NO. 2		
KBAS 105	-0	22	-103 0 1	
NSTR 105	-00002			ADD HEAT SPECIFIED NO OTHER CALCULATIONS RORD
ID** 106	-0	LIQMIX FROM PUMPS		
KBAS 106	-0	7	104 0 1	-105 0 1 107
NSTR 106				MAX PRESS OF TWO SOURCES
ID** 107	-0	SPLIT AROUND EVAPORATOR		
KBAS 107	-0	10	106 0 1	0 1 108
NSTR 107	-00			USE UNIVERSAL SPLIT RATIO
ID** 108	-0	SPLIT TO EVAPORATOR		
KBAS 108	-0	10	107 0 1	0 1 109
NSTR 108	-00			USE UNIVERSAL SPLIT RATIO
VARY 108	65	0.07064672	SPLIT TO EVAPORATOR	
ID** 109	-0	EVAPORATOR COOLING LIQUID CONTROL VALVE		

KBAS 109	-0 10	-108 0	1	0	1	112
NSTR 109		1	USE SS SOLUTION DURING TRANSIENT, MAX PRESS			
VARY 109	1 360.	HTNG FLUID FLOWRATE				LB/HR
VARY 109	2 157.5	HTNG FLUID TEMP.				F
VARY 109	3 15.	HTNG FLUID PRESS				PSI
VARY 109	4 15.	HTNG FLUID PRESS				PSI
ID** 110	-0 LIQMIX FROM EVAPORATOR					
KBAS 110	-0 7	-109 0	1	-239 0	1	111
NSTR 110	-03	MIX PRESS DEPENDENT ON PRINCIPAL SOURCE				
ID** 111	-0 LIQMIX FROM BYPASS AND EVAPORATOR					
KBAS 111	-0 7	110 0	1	107 0	1	134
NSTR 111		MAX PRESS OF TWO SOURCES				
ID** 112	-0 SPLIT TO POTABLE WATER STORAGE TANK NO.1					
KBAS 112	-0 10	108 0	1	0	1	113
NSTR 112	-00	USE UNIVERSAL SPLIT RATIO				
VARY 112	65 0.1322	SPLIT TO TANK NO. 1				
ID** 113	-0 SPLIT TO POTABLE WATER STORAGE TANK NO.2					
KBAS 113	-0 10	112 0	1	0	1	114
NSTR 113	-00	USE UNIVERSAL SPLIT RATIO				
VARY 113	65 0.15233	SPLIT TO TANK NO. 2				
ID** 114	-0 SPLIT TO POTABLE WATER STORAGE TANK NO.3					
KBAS 114	-0 10	113 0	1	0	1	115
NSTR 114	-00	USE UNIVERSAL SPLIT RATIO				
VARY 114	65 0.1797	SPLIT TO TANK NO. 3				
ID** 115	-0 SPLIT TO POTABLE WATER STORAGE TANK NO.4					
KBAS 115	-0 10	114 0	1	0	1	116
NSTR 115	-00	USE UNIVERSAL SPLIT RATIO				
VARY 115	65 0.21907	SPLIT TO TANK NO. 4				
ID** 116	-0 SPLIT TO POTABLE WATER STORAGE TANK NO.5					
KBAS 116	-0 10	115 0	1	0	1	117
NSTR 116	-00	USE UNIVERSAL SPLIT RATIO				
VARY 116	65 0.2805	SPLIT TO TANK NO. 5				
ID** 117	-0 SPLIT TO WATER HEATER					
KBAS 117	-0 10	116 0	1	0	1	118
NSTR 117	-00	USE UNIVERSAL SPLIT RATIO				
VARY 117	65 0.36842	SPLIT TO WATER HEATER				
ID** 118	-0 BYPASS LIQMIX FROM COOLING LOOP(NORMALLY HEATING LOOP FLOW)					
KBAS 118	-0 7	-117 0	1	2 0	2	120
NSTR 118		MAX PRESS OF TWO SOURCES				
ID** 119	-0 LIQUID SPLIT TO COOLING LOOP(NORMALLY HEATING LOOP)					
KBAS 119	-0 10	-93 0	1			128
NSTR 119	-00	USE UNIVERSAL SPLIT RATIO				
VARY 119	65 0.	ALL FLOW TO HTNG LOOP				
ID** 120	-0 SPLIT TO WATER COOLER(NORMALLY NO HEATING FLOW)					
KBAS 120	-0 10	117 0	1	0	1	121
NSTR 120	-00	USE UNIVERSAL SPLIT RATIO				
VARY 120	65 0.0	0.0 PER CENT FLOW TO WATER COOLER				
ID** 121	-0 SPLIT TO AIR HEATER AND WARM FLUSH WATER TANK					
KBAS 121	-0 10	120 0	1	0	1	122
NSTR 121	-00	USE UNIVERSAL SPLIT RATIO				
VARY 121	65 0.58333333	SPLIT TO AIR HEATER				
ID** 122	-0 LIQUID CONTROL VALVE TO AIR HEATER					
KBAS 122	-0 10	-121	1	0	1	124
NSTR 122	-00	USE UNIVERSAL SPLIT RATIO				
VARY 122	65 1.	ALL FLOW BYPASSES AIR HEATER.				
ID** 123	-0 LIQMIX AT AIR HEATER					
KBAS 123	-0 7	-204 0	1	122 0	1	126
NSTR 123	-03	MIX PRESS DEPENDENT ON PRINCIPAL SOURCE				
ID** 124	-0 LIQUID CONTROL VALVE TO WARM FLUSH WATER TANK					

KBAS 124	-0 10	121 0 1	0 1	41
NSTR 124	-00	USE UNIVERSAL SPLIT RATIO		
VARY 124	65 0.	ALL HTNG FLUID FLOWS THROUGH FLUSH WATR TNK		
ID** 125	-0 LIQMIX AT MAIN FLUSH WATER TANK			
KBAS 125	-0 7	-41 0 1	124 0 1	123
NSTR 125		MAX PRESS OF TWO SOURCES		
ID** 126	-0 LIQMIX FROM AIR HEATER AND FLUSH WATER TANK			
KBAS 126	-0 7	125 0 1	-123 0 1	127
NSTR 126		MAX PRESS OF TWO SOURCES		
ID** 127	-0 LIQMIX FROM WATER COOLER(NORMALLY NO FLOW)			
KBAS 127	-0 7	126 0 1	5 0 1	119
NSTR 127		MAX PRESS OF TWO SOURCES		
ID** 128	-0 LIQMIX FROM WATER HEATER			
KBAS 128	-0 7	127 0 1	-119 0 1	129
NSTR 128		MAX PRESS OF TWO SOURCES		
ID** 129	-0 LIQMIX FROM WATER NO.5			
KBAS 129	-0 7	128 0 1	90 0 1	130
NSTR 129		MAX PRESS OF TWO SOURCES		
ID** 130	-0 LIQMIX FROM WATER NO.4			
KBAS 130	-0 7	129 0 1	83 0 1	131
NSTR 130		MAX PRESS OF TWO SOURCES		
ID** 131	-0 LIQMIX FROM WATER NO.3			
KBAS 131	-0 7	130 0 1	84 0 1	132
NSTR 131		MAX PRESS OF TWO SOURCES		
ID** 132	-0 LIQMIX FROM WATER NO.2			
KBAS 132	-0 7	131 0 1	86 0 1	133
NSTR 132		MAX PRESS OF TWO SOURCES		
ID** 133	-0 LIQMIX FROM WATER NO.1			
KBAS 133	-0 7	132 0 1	87 0 1	110
NSTR 133		MAX PRESS OF TWO SOURCES		
ID** 134	-0 LIQMIX TO HEAT EXCHANGER			
KBAS 134	-0 7	111 0 1	-133 0 1	136
NSTR 134		MAX PRESS OF TWO SOURCES		
ID** 135	-0 WATER SUPPLY			
KBAS 135	-0 49	0 1		
NSTR 135	-00	DUMMY COMPONENT PROVIDE AIR TMROUGH HX MIN RORI		
VARY 135	1 60.	FLOWRATE OF CLNT TO RITE TEMP CNTRL LB/HR		
VARY 135	2 75.	TEMP OF CLNT. DEG-F		
VARY 135	3 14.7	PRESS PSI		
VARY 135	4 14.7	PRESS PSI		
ID** 136	-0 SPLIT FOR BYPASSING COOLANT IN TEMP CONTROL OF RADIOISOTOPE			
KBAS 136	-0 10	135 0 1	0 1	137
NSTR 136		MAX PRESS OF TWO SOURCES		
ID** 137	-0 HEAT EXCHANGER			
KBAS 137	-0 4	134 0 1	136 0 1	139
NSTR 137	-0200020000	CRUSSFLO,GAS-LIQD,UA INPUT NO SIZING.		
VARY 137	1 426.	HTNG FLU LOW TEMP RITE=3.8 GPM LB/HR		
VARY 137	1 603.0	LOW TEMPERATURE LOOP FLOW RATE (LB/HR)		
VARY 137	2 179.0	LOW TEMPERATURE LOOP TEMPERATURE(DEG F)		
VARY 137	3 15.0	PRESS PSI		
VARY 137	4 15.0	PRESS PSI		
VARY 137	66 50.0	EFFECTIVE UA		
ID** 139	-0 FLOW JUNCTION PERMITS FLOW RITE TEMP CONTROL HX TO MIX			
KBAS 139	-0 7	136 0 1	137 0 1	100
NSTR 139	-01	PRI PRESS DETERMINES MIX PRESS.		
ID** 180	-0 HEAT BLOCK ASSY. SIMULATED BY SUBROUTINE THERML			
ID** 180	1 HEAT BLOCK SIMULATED BY SUBROUTINE THERML			
KBAS 180	-0 37 46 36			181
KARY 180	16	9	NUMBER OF NODES	

KARY 180	17	36	NUMBER OF HEAT TRANSFER TERMS
KARY 180	19	0000001000	NODE 1 HEAT STORAGE TERM FOR RADIOISOTOPE+CORE
KARY 180	20	0000002000	NODE 2 HEAT STORAGE TERM BLOCK TO HEAT PIPE
KARY 180	21	0000003000	NODE 3 HEAT STORAGE TERM BLOCK TO AIR STERILIZER
KARY 180	22	0000004000	NODE 4 HEAT STORAGE TERM BLOCK TO PYROLYSIS 1
KARY 180	23	0000005000	NODE 5 HEAT STORAGE TERM BLOCK TO PYROLYSIS 3
KARY 180	24	0000006000	NODE 6 HEAT STORAGE TERM PYROLYSIS UNIT NO. 2
KARY 180	25	0000007000	NODE 7 HEAT STORAGE TERM BLOCK TO INCINERATOR
KARY 180	26	0000008000	NODE 8 HEAT STORAGE TERM INNER WALL
KARY 180	27	0000009000	NODE 9 HEAT STORAGE TERM INNER WALL
KARY 180	28	1000001000	HEAT GENERATION TERM ISOTOPI
KARY 180	29	1000002000	HEAT GENERATION TERM HEAT PIPE
KARY 180	30	1000003000	HEAT GENERATION TERM AIR STERILIZER
KARY 180	31	1000004000	HEAT GENERATION TERM PU 1
KARY 180	32	1000005000	HEAT GENERATION TERM PU 2
KARY 180	33	1000006000	HEAT GENERATION TERM PU 3
KARY 180	34	1000007000	HEAT GENERATION TERM INCINERATOR
KARY 180	35	5000001002	CONDUCTION NODE 1 TO NODE 2
KARY 180	36	5000001003	CONDUCTION NODE 1 TO NODE 3
KARY 180	37	5000001004	CONDUCTION NODE 1 TO NODE 4
KARY 180	38	5000001005	CONDUCTION NODE 1 TO NODE 5
KARY 180	39	5000001006	CONDUCTION NODE 1 TO NODE 6
KARY 180	40	5000001007	CONDUCTION NODE 1 TO NODE 7
KARY 180	41	5000002003	CONDUCTION NODE 2 TO NODE 3
KARY 180	42	5000002007	CONDUCTION NODE 2 TO NODE 7
KARY 180	43	5000003004	CONDUCTION NODE 3 TO NODE 4
KARY 180	44	5000003007	CONDUCTION NODE 3 TO NODE 7
KARY 180	45	5000003008	CONDUCTION NODE 3 TO NODE 8
KARY 180	46	5000004005	CONDUCTION NODE 4 TO NODE 5
KARY 180	47	5000004008	CONDUCTION NODE 4 TO NODE 8
KARY 180	48	5000005006	CONDUCTION NODE 5 TO NODE 6
KARY 180	49	5000005008	CONDUCTION NODE 5 TO NODE 8
KARY 180	50	5000006007	CONDUCTION NODE 6 TO NODE 7
KARY 180	51	5000006008	CONDUCTION NODE 6 TO NODE 8
KARY 180	52	5000007008	CONDUCTION NODE 7 TO NODE 8
KARY 180	53	5000008009	CONDUCTION NODE 8 TO NODE 9 TERM IN WALL TO O WAL
KARY 180	54	6000009114	CONDUCTION NODE 9 TO AMBIENT TEMPERATURE
VARY 180	65	10.	SOLUTION CONVERGENCE TOLERANCE DEG-F
VARY 180	69	1260.	TEMP NODE 1 HEAT BLOCK RADIOISOTOPE DEG-F
VARY 180	70	1200.	TEMP NODE 2 HEAT BLOCK TO HT PIPE DEG-F
VARY 180	71	1200.	TEMP NODE 3 HEAT BLOCK AIR STERILIZER DEG-F
VARY 180	72	1200.	TEMP NODE 5 HEAT BLOCK PYROLYSIS UNIT1DEG-F
VARY 180	73	1200.	TEMP NODE 5 HEAT BLOCK PYROLYSIS UNIT2DEG-F
VARY 180	74	1200.	TEMP NODE 6 HEAT BLOCK PYROLYSIS UNIT3DEG-F
VARY 180	75	1200.	TEMP NODE 7 HEAT BLOCK INCINERATOR DEG-F
VARY 180	76	810.	TEMP NODE 8 HEAT BLOCK INNER SHELL -F
VARY 180	77	290.0	OUTER WALL TEMPERATURE (DEG F)
VARY 180	78	7.0	HEAT STORAGE TERM NODE 1 RADIOISOTOPE REG BTU/F
VARY 180	79	5.0	HEAT STORAGE TERM NODE 2 HP REGION BTU/F
VARY 180	80	3.0	HEAT STORAGE TERM NODE 3 AS REGION BTU/F
VARY 180	81	3.0	HEAT STORAGE TERM NODE 4 PU1 REGION BTU/F
VARY 180	82	3.0	HEAT STORAGE TERM NODE 5 PU2 REGION BTU/F
VARY 180	83	3.0	HEAT STORAGE TERM NODE 6 PU3 REGION BTU/F
VARY 180	84	4.0	HEAT STORAGE TERM NODE 7 INCIN REGION BTU/F
VARY 180	85	4.1	HEAT STORAGE TERM NODE 8 INNER WALL BTU/F
VARY 180	86	4.2	HEAT STORAGE TERM NODE 9 OUTSIDE WALL BTU/F
VARY 180	87	1574.	HEAT GENERATED BY RADIOISOTOPE BTU/HR
VARY 180	88	100.	HEAT LOST TO HP BTU/HR
VARY 180	89	5.	HEAT LOST TO AS BTU/HR

VARY 180	90	68.	HEAT LOST TO PU1	BTU/HR
VARY 180	91	68.	HEAT LOST TO PU2	BTU/HR
VARY 180	92	68.	HEAT LOST TO PU3	BTU/HR
VARY 180	93	50.	HEAT LOST TO INCIN	BTU/HR
VARY 180	94	1.667	CONDUCTION,A/L, TERM NOD 1/NOD 2TO	HEAT PIPE
VARY 180	95	4.5	CONDUCTION,A/L, TERM NOD 1/NOD 3TO	AIR STERLIZEER
VARY 180	96	5.0	CONDUCTION,A/L, TERM NOD 1/NOD 4TO	PU 1
VARY 180	97	5.0	CONDUCTION,A/L, TERM NOD 1/NOD 5TO	PU 2
VARY 180	98	5.0	CONDUCTION,A/L, TERM NOD 1/NOD 6TO	PU 3
VARY 180	99	5.0	CONDUCTION,A/L, TERM NOD 1/NOD 7TO	INCINERATOR
VARY 180	100	.27	CONDUCTION,A/L, TERM NOD2/NOD3	HP/AS (FT)
VARY 180	101	.25	CONDUCTION,A/L, TERM NOD2/NOD7	HP/INCIN (FT)
VARY 180	102	.25	CONDUCTION,A/L, TERM NOD3/NOD4	AS/PU1 (FT)
VARY 180	103	.25	CONDUCTION,A/L, TERM NOD3/NOD7	AS/INCIN (FT)
VARY 180	104	0.64	CONDUCTION,A/L, TERM NOD 3/NOD 8TO	AS TO I WALL
VARY 180	105	.25	CONDUCTION,A/L, TERM NOD4/NOD5	PU1/PU2 (FT)
VARY 180	106	0.59	CONDUCTION,A/L, TERM NOD 4/NOD 8TO	PU1 TO I WALL
VARY 180	107	.25	CONDUCTION,A/L, TERM NOD5/NOD6	PU2/PU3 (FT)
VARY 180	108	0.59	CONDUCTION,A/L, TERM NOD 5/NOD 8TO	PU2 TO I WALL
VARY 180	109	.25	CONDUCTION,A/L, TERM NOD6/NOD7	PU3/INCIN (FT)
VARY 180	110	0.59	CONDUCTION,A/L, TERM NOD 6/NOD 8TO	PU3 TO I WALL
VARY 180	111	0.64	CONDUCTION,A/L, TERM NOD 7/NOD 8TO	INCIN TO I WALL
VARY 180	112	2.3	CONDUCTION,A/L, TERM NOD 8/NOD 9TO	I WALL TO O WALL
VARY 180	113	5.7	CONDUCTION,A/L, TERM NOD 9/NOD10TO	O WALL TO TS
VARY 180	114	80.0	EFFECTIVE TEMP OF THE SURROUNDINGS.	DEG-F
ID** 181	-0		HEAT PIPE NEAR-ISOTHERMA HEAT REJECTION DEVICE	
KBAS 181	-0	57		125
KARY 181	16	1	TABLE DEFINES SODIUM BOILING TEMP VS PRESSURE	
VARY 181	65	1280.	HEAT PIPE CONTROL TEMPERATURE	F
VARY 181	66	1280.	TEMP OF SOURCE FROM WHICH HEAT IS REJECTED F	
VARY 181	67	38.116	CONDUCTANCE KA/L SOURCE TO HEAT PIPE	A/HR-F
VARY 181	68	0.0416	HEAT PIPE DIAMETER	FT
VARY 181	69	.541667	LENGTH OF CONDENSING REGION OF HEAT PIPE	FT
VARY 181	70	4.4E-6	MASS OF INERT GAS IN THE HEAT PIPE	LBS
VARY 181	71	38.69	GAS CONSTANT OF INERT GAS USED	FT/ R
VARY 181	72	80.	HEAT PIPE TEMP AT COLD END OF HEAT PIPE	F
VARY 181	73	2.225E-4	VOLUME OF INERT GAS RESERVOIR HEAT PIPE	
VARY 181	76	8.5	HEAT TRANSFER COEFFICIENT SURFACE/H.P.	V/HR-FTSQ-F
VARY 181	77	75.	AMBIENT TEMP	F
VARY 181	79	1.638	NOT USED	
VARY 181	80	8.5		
VARY 181	81	15.8	THERMAL CONDUCTIVITY OF HEAT PIPE MATERIAL	B/HR-FT-F
VARY 181	82	.002916	HEAT PIPE THICKNESS	FT.
ID** 201	-0		DUMMY COMPONENT PROVIDING THE FECES INPUT INTO SYS	
KBAS 201	-0	49	4 141516 11819	215
NSTR 201	-00		NONE RORD	
ID** 202	-0		FECES BLENDER SIMULATED BY DUMMY COMPONENT	
KBAS 202	-0	49	215 4 141516 11819	205
NSTR 202	-00		NONE RORD	
ID** 203	-0		DUMMY COMPONENT PROVIDING THE AIR FLOW TO DEFECATION UNIT	
KBAS 203	-0	49	2	204
NSTR 203	-00		NONE RORD	
ID** 204	-0		AIR HEATER SIMULATED BY SUBROUTINE ANYHX.	
KBAS 204	-0	4	203 2 -122 0 1	201
NSTR 204	2	1	AIR HEATER STEADY CALCULATION	
VARY 204	66	12.0	OVERALL UA BTU/HR-F	
ID** 205	-0		A JUNCTION PROVIDING MEANS OF INTRODUCING URINE TO EVAPORATOR	
KBAS 205	-0	7	202 4 141516 11819 23 4 141516 11819	208
NSTR 205	-01		USE PRIMARY FLO PRESS AS MIXED FLO PRESS	

ID** 206	-0	JUNCTION PROVIDE MEANS OF DUMPING WASTE FROM SHREDDER TO EVAP	
KBAS 206	-0 7	205 4 141516 11819 -207 4 141516 11819 218	
NSTR 206	-01	USE PRIMARY FLO PRESS AS MIXED FLO PRESS	
ID** 207	-0	TRASH SHREDDER SIMULATED BY VERSATILE SUBROUTINE ALTCOM	
KBAS 207	-0 7	208 4 141516 11819 -49 4 141516 11819 206	
NSTR 207	-00	NONE RQRD	
ID** 208	-0	DUMMY COMPONENT DEFINING THE TRASH INPUT	
KBAS 208	-0 49	4 141516 11819	207
NSTR 208	-00	NONE RQRD	
KBAS 209	-0 7	206 4 141516 11819 -218 4 141516 11819 210	
NSTR 209	-01	OUTFLO-DEFINED BY GPOLY SS+TRANS HT BAL	
ID** 210	-0	EVAPORATOR SIMULATED BY LIQ-GAS HX WITH EVAPORATION OF WATR	
KBAS 210	59	209 4 1516 11819 -204 3 14 5 6 7 8 9 221	
NSTR 210		11 USE SS SOLUTION DURING TRANSIENT	
KARY 210	17	239	COMPONEN PROVIDING HTNG FLUID TO EVAPORATOR
KARY 210	18	2	TABLE RELATING NUCLEATE BOILING Q/A VS DELTMP
VARY 210	51 110.		
VARY 210	52	EFFECTIVE THERMAL CONDUCTANCE WALL/SURR.	BTU/HR-F
VARY 210	53 0.	TOTAL HEAT LOSS TO SURROUNDINGS	BTU/HR
VARY 210	54 75.	AMBIENT GAS TEMPERATURE	DEG-F
VARY 210	55 1.	CONDUCTANCE, INSULATION SURFACE/AMB. GAS	BTU/HR-F
VARY 210	56	CONVECTIVE HEAT LOSS TO AMBIENT GAS	BTU/HR
VARY 210	57 75.	AMBIENT WALL TEMP	DEG-F
VARY 210	58 .2	RADIATION FA FACTOR SURFACE TO AMB WALL	FTSQ
VARY 210	59	RADIATIVE HEAT LOSS TO AMBIENT WALL	BTU/HR
VARY 210	60 85.	STRUCTURE TEMPERATURE	DEG-F
VARY 210	61 2.5	CONDUCTANCE KA/X EVAP WALL TO STRUCTURE	BTU/HR-F
VARY 210	62	CONDUCTANCE HEAT LOSS TO STRUCTURE	BTU/HR
VARY 210	63	INSULATION SURFACE TEMPERATURE	DEG-F
VARY 210	64 .1	CONDUCTANCE KA/X EVAP WALL TO INSUL SURF.	BTU/HR-F
VARY 210	65 92.5	EFFECTIVE CONDUCTANCE UA WALL/HTNG FLUID	B/HR-DEL-F
VARY 210	66 0.	ELECTRICAL HEAT INPUT TO THE EVAP WALL	BTU/HR
VARY 210	67 1.418	HEIGHT OF THE EVAPORATOR	FT
VARY 210	68 1.73	DIAMETER OF THE EVAPORATOR	FT
VARY 210	69 0.	EFFECTIVE HEIGHT OF THE LIQUID SLURRY IN EVAP (FT)	
VARY 210	70 .15	MINIMUM LIQUID LEVEL ALLOWABLE NORM OPRTN FT	
VARY 210	71 .342	MAXIMUM LOD. LEVL ALLOWABLE NORMAL OPERATION (FT)	
VARY 210	72 0.	HEAT LOST BY VAPORIZING LOD TO AIR FLOWNG	BTU/HR
VARY 210	73 0.	EFFECTIVE GAS SPACE IN THE EVAPORATOR	FT**3
VARY 210	74 0.	HEAT FLUX FROM EVAPORATOR WALL TO LIQUID	BTU/HR
VARY 210	75 0.	HEAT FLUX, Q/A, TERM BETWEEN WALL/LIQUID	B/HR-FT2
VARY 210	76 .2	GAS SPACE IN EVAP NOT SUBJECT TO LOD FILL	FT**3
VARY 210	77 0.	VAPORIZATION RATE OF WATER	LB/HR
VARY 210	78 .0471489	BACTERIAL DECOMPOSITION FACTOR FOR UREA	1/HR
VARY 210	79 0.	DECOMPOSITION RATE OF UREA IN THE EVAP	LB/HR
VARY 210	80 0.	LIEQUID LOST IN DECOMPOSITION REACTION	LB/HR
VARY 210	81 1.17	RATIO OF LIQUID TO SOLID IN SOLIDS RESERVOIR	
VARY 210	82 2.28	FILTRATION FACTOR OF LOD/SOLID FILTER	LB/HR
VARY 210	83 0.	FILTRATION RATE OF LIQUID SOLID COMPONENT	LB/HR
VARY 210	84 .1	FRACTION VAPORIZED VS. MAX THAT CAN TO SAT AIR (DEC)	
VARY 210	85 .51	SOLID RESERVOIR CAPACITY	LB.
VARY 210	86 0.	LIQUID LOSS DUE TO FILTRATION RATE	LB/HR
VARY 210	87 0.	SOLID SP. FLO NO. 2 REMOVAL RATE	LB/HR
VARY 210	88 0.	SOLID SP. FLO NO. 3 REMOVAL RATE	LB/HR
VARY 210	89 0.	SOLID SP. FLO NO. 4 REMOVAL RATE	LB/HR
VARY 210	90 0.	SOLID SP. FLO NO. 5 REMOVAL RATE	LB/HR
VARY 210	91 0.	SOLID SP. FLO NO. 6 REMOVAL RATE	LB/HR
VARY 210	92	NH3 GENERATION RATE IN UREA DECOMPOSITION	LB/HR
VARY 210	93	CO2 GENERATION RATE IN UREA DECOMPOSITION	LB/HR

VARY 210 94 1.	SS CONVERGENCE CRITERIA FOR TEMP PREDICT	PER-CENT
VARY 210 95 200.	POWER REQUIRED TO DRIVE THE EVAP IMPELLER	WATTS
VARY 210 96	HTNG FLUID INLET TEMPERATURE (DEG-F)	
VARY 210 97	HTNG FLUID OUTLET TEMPERATURE (DEG-F)	
VARY 210 98 .1064	PRESSURE DROP COEFFICIENT FOR FLOW EVAP/CNDNSR.	
VARY 210 99 .000142	EFFECTIVE ORIFICE AREA BETWEEN EVAP/DDWNSRM (FISQ)	
VARY 210 100 1.4	DOWNSTREAM PRESS (CONDENSER)	(PSI)
VARY 210 101 26.15439	TOTAL WEIGHT OF LIQUID SLURRY IN EVAPORATOR	LBS
VARY 210 102 105.	TEMPERATURE OF LIQUID/SOLID SLURRY	F-DEG
VARY 210 103 .4	PRESS.	PSI
VARY 210 104 .4	PRESS.	PSI
VARY 210 105	NOT USED	
VARY 210 106	NOT USED	
VARY 210 107	NOT USED	
VARY 210 108	NOT USED	
VARY 210 109	NOT USED	
VARY 210 110	NOT USED	
VARY 210 111	NOT USED	
VARY 210 112	NOT USED	
VARY 210 113	NOT USED	
VARY 210 114 0.		
VARY 210 115 .173217	AMOUNT OF UREA IN THE EVAPORATOR	LBS
VARY 210 116 .9811741	AMOUNT OF OTHER SOLIDS IN THE EVAPORATOR	LBS
VARY 210 117 25.0	AMOUNT OF LOD IN EFAP	LBS
VARY 210 118 0.	SPECIAL FLOW NO 5	LB
VARY 210 119 0.	SPECIAL FLOW NO 6	LB
VARY 210 120 0.	TOTAL WEIGHT OF GAS IN EVAP SYS.	LB
VARY 210 121 102.	TEMP. OF GAS IN EVAP SYS.	F-DEG
VARY 210 122 .4	PRESSURE GAS IN EVAP SYS	PSI
VARY 210 123 .4	PRESS.	PSI
VARY 210 124 0.	WEIGHT OF NON-CONDENSABLES IN EVAP	LB
VARY 210 125 0.	WEIGHT OF VAPOR IN EVAP.	LB
VARY 210 126 0.	WEIGHT OF ENTRAINED LIQUID (ALWAYS ZERO)	LB
VARY 210 127 .22	SPECIFIC HEAT OF NON CONDENSABLE GAS	BTU/LB-F
VARY 210 128 35.	MOLECULAR WEIGHT OF NON-CONDENSABLE GAS	LB/LB-MOLE
VARY 210 129 0.	WEIGHT OF O2 IN EVAP	LB.
VARY 210 130 0.	WEIGHT OF N2 IN EVAP	LB.
VARY 210 131 0.	WEIGHT OF CO2 IN EVAP	LB.
VARY 210 132 0.	WEIGHT OF NH3 IN EVAP (TRACE CONTAMINANT)	LB.
VARY 210 133 0.	WEIGHT OF SPECIAL FLOW NO 1 IN EVAP	LB.
VARY 210 134 0.	WEIGHT OF SPECIAL FLOW NO 2 IN EVAP	LB.
VARY 210 135 0.	WEIGHT OF SPECIAL FLOW NO 3 IN EVAP	LB.
VARY 210 136 0.	WEIGHT OF SPECIAL FLOW NO 4 IN EVAP	LB.
VARY 210 137 0.	WEIGHT OF SPECIAL FLOW NO 5 IN EVAP	LB.
VARY 210 138 0.	WEIGHT OF SPECIAL FLOW NO 6 IN EVAP	LB.
VARY 210 139 0.	TOTAL WEIGHT OF LIQUID/SOLID IN SOLID RES	LB.
VARY 210 140	TEMPERATURE OF LIQUID/SOLID IN SOLID RES	DEG-F
VARY 210 141	PRESSURE OF LIQUID/SOLID IN SOLID RES	PSI
VARY 210 142	PRESSURE OF LIQUID/SOLID IN SOLID RES	PSI
VARY 210 143	NOT USED RC(5)	
VARY 210 144	NOT USED RC(6)	
VARY 210 145	NOT USED RC(7)	
VARY 210 146	NOT USED RC(8)	
VARY 210 147 0.	NOT USED RC(9)	
VARY 210 148 0.	NOT USED RC(10)	
VARY 210 149 0.	NOT USED RC(11)	
VARY 210 150 0.	NOT USED RC(12)	
VARY 210 151 0.	NOT USED RC(13)	
VARY 210 152 0.	SOLID/LIQUID COMPONENT NO 1 IN SOLID RES	LBS

VARY 210 153 0.	SOLID/LIQUID COMPONENT NO 2 IN SOLID RES	LBS
VARY 210 154 0.	SOLID/LIQUID COMPONENT NO 3 IN SOLID RES	LBS
VARY 210 155 0.	SOLID/LIQUID COMPONENT NO 4 IN SOLID RES	LBS
VARY 210 156 0.	SOLID/LIQUID COMPONENT NO 5 IN SOLID RES	LBS
VARY 210 157 0.	SOLID/LIQUID COMPONENT NO 6 IN SOLID RES	LBS
VARY 210 158 26.4	THERML CAPACITANCE OF EVAPORATOR LESS LQD	BTU/LB-F
VARY 210 159 26.146	THERMAL CAPACITANCE OF LIQUID IN EVAP	BTU/LB-F
ID** 213 1	INCINERATOR TO ACCOMPLISH THERMAL DECOMPOSITION AND VAPORIZTN	
ID** 213 2	OF 95 PER CENT OF SOLIDS TO MINIMIZE OZ RERRMNTS FOR COMBUSTN	
KBAS 213 -0 58	210 4 11516	214
NSTR 213 -0	NONE RORD	
VARY 213 54 200.0	AMBIENT GAS TEMPERATURE DEG F	
VARY 213 55 .16	THERMAL LONDUCTANCE TO GAS (BTU/HR- DEG F)	
VARY 213 69 5650.0	HEAT OF REACTION BTU/LB OXYGEN	
VARY 213 71 1080.00	TOTAL CYCLE TIME, SECONDS	
VARY 213 72 300.00	TIME TO CONSTANT RATE PERIOD,SECONDS	
VARY 213 73 600.00	TIME TO END OF DRYING PERIOD,SECONDS	
VARY 213 74 180.0	OXYDATION TIME PERIOD,SECONDS	
VARY 213 75 60.0	GAS VENTING TIME PERIOD,SECONDS	
VARY 213 77 2.00	NUMBER OF OXYDATION-VENT CYCLES, N.O.	
VARY 213 79 0.31	INITIAL MASS OF SOLIDS IN INCINFRATOR(LB)	
VARY 213 84 0.44		
VARY 213 88 15.0	ENVIRONMENTAL PRESSURE,PSI	
VARY 213 89 1280.0	HEAT BLOCK TEMPERATURE,DEG F	
VARY 213 90 0.0000167		
VARY 213 91 2.2223	EVAPORATOR SURFACE AREA,FT2	
VARY 213 92 1000.0	HEAT OF VAPORIZATION,BTU/LB	
VARY 213 93 2.68	THERMAL CONDUCTANCE FROM HEAT SOURCE TO SHUTTLE	
VARY 213 94 0.4	EFFECTIVE THERMAL CAPACITANCE OF SHUTTLE BTU/LB-F	
ID** 214 -0	ASH COLLECTOR SIMULATED BY FLOMET TO METER THE INCIN END PROD	
KBAS 214 29 17	213 3 142526171819	61
NSTR 214 -01	METER ALL CONSTITUENTS IN INCINERATOR	
ID** 215 -0	JUNCTION PROVIDES MEANS OF ADDING FLUSH WATER TO DEFCTN UNIT	
KBAS 215 -0 7	201 4 141516 11819 -50 4 141516 11819	202
NSTR 215 -01	USE PRIMARY PRESS AS MIXED FLO PRESS	
ID** 218 -0	WASH WATER TNK	
KBAS 218 -0 30	219 4 141516 11819	209
NSTR 218 -011000	TANKG NO TRANS HT BAL INITIAL VOLUME	
VARY 218 51 100.	TANK TEMP	DEG-F
VARY 218 54 75.	AMBIENT GAS TEMP	F
VARY 218 55 2.5	CONDUCTANCE INSUL SURFACE/AMBIENT GAS	B/HR-F
VARY 218 57 75.	AMBIENT WALL TEMP	F
VARY 218 58	THERMAL RADIATION FA FACTOR	FT**2
VARY 218 60 75.	STRUCTURE TEMP	F
NSTR 218 12010	OUTFLO=GPOLY,NO INTERNAL HX	
VARY 218 61 1.5	EFFECTIVE CONDUCT BTWN TNK WALL/SURROUNDNG	B/HR-F
VARY 218 64 0.	CONDUCTANCE TNK WALL/INSULATION SURFACE	BTU/HR-F
VARY 218 68 41.75	CAPACITY OF WASH WATER TANK	LBS
VARY 218 69 38.	WEIGHT FLUID IN TNK	LB
VARY 218 70 150.	TEMP OF FLUID	F
VARY 218 71 .609	FLUID VOLUME IN TNK	FT**3
VARY 218 72 15.	PRESS	PSI
VARY 218 75 38.	WEIGHT OF CONDENSABLE LIQUID	LB
VARY 218 88 1.0	SPECIFIC HEAT OF FLUID IN TNK	BTU/LB-F
VARY 218 92 .88	THERMAL CAPACITANCE OF TNK SHELL	BTU/F
VARY 218 93 100.	CONDUCTANCE LQD TO TNK WALL	
VARY 218 85 38.0	WEIGHT OF WATER IN THE WASH TANK (LBS)	
VARY 218 94 5.0	MAX ALLOWABLE TEMP CHANGE	F
VARY 218 95	EXTERNAL HT LOAD TO FLUID	BTU/HR



VARY 218	96	EXTERNAL HT LOAD TO SHELL	BTU/HR
ID** 219	-0	DUMMY COMPONENT PROVIDES WASH WATER FLOW TO WASH WATER TANK	
KBAS 219	-0 49	4 141516 11819	
NSTR 219	-0	NON RORD	
VARY 219	2 80.0	WASH WATER TEMPERATURE (DEG F)	
VARY 219	17 1.	WASH WATER INPUT	
ID** 220	-0	DUMMY COMPONENT PROVIDING O2 FLOW FOR COMBUSTION IN PYROLYSIS	
KBAS 220	-0 30	2	
NSTR 220	-01211	DUMMY COMPONENT NO NSTRS REQRO.	
VARY 220	2 70.	O2 TEMP	
VARY 220	3 900.	O2 PRESS	
VARY 220	4 900.	O2 PRESS	
VARY 220	69 5.06462	TOTAL FLUID WEIGHT IN TANK	LBS
VARY 220	70 70	O2 TEMP IN TANK	DEG-F
VARY 220	71 1.	FLUID VOLUME IN TANK	
VARY 220	72 900.	O2 PRESSURE IN TANK	PSI
VARY 220	73 5.06462	WEIGHT OF NON-CONDENSABLES IN TANK	LBS
VARY 220	76 .22	NON-CONDENSABLE CP IN TANK	
VARY 220	77 32.	NON-CONDENSABLE MOLECULAR WEIGHT IN TANK	LBS/LB-
VARY 220	78 5.06462	WEIGHT OF O2 IN TANK	
VARY 220	88 .22	SPECIFIC HEAT OF FLUID IN TANK	BTU/LB-F
VARY 220	89 5.06462	DENSITY OF FLUID IN TANK	LB/FT**3
VARY 220	90 .08	VISCOSITY OF FLUID IN TANK	
VARY 220	91 32.	MOLECULAR WEIGHT OF FLUID IN TANK	LB/LB-MOLE
ID** 221	-0	TEE PROVIDES MEANS FOR INJECTING O2 INTO THE STREAM	
KBAS 221	6	220 2 14 210 3 14 5 6 7 8 9	232
NSTR 221	-02	TEE WITH SECONDARY PRESSURE USED AS MIXED PRE	
ID** 223		REGENARATIVE HX COMPONENT OF THE PYROLYSIS UNIT. COMPUTES INLET FL	
KBAS 223	-0 4	232.3 14 5 6 7 8 9 -224 3 14 5 6 7 8 9	224
NSTR 223	-0200010000	COUNTERFLOW, NO SIZING, INPUT UA, GAS-GAS HX	
VARY 223	66 21.6729	OVERALL UA	BTU/HR-F
ID** 224		PYROLYSIS UNIT SIMULATED BY CATARN	
KBAS 224	13 4	223 3 14 5 6 7 8 9	226
NSTR 224	-0	NONE REQRO	
KARY 224	16	2 ONE TRACE CONTAMINANT REACTING	
VARY 224	1 2.6067	TOTAL FLOW	
VARY 224	2 1200.	TEMPERATURE OF EXITING GAS STREAM	DEG-F
VARY 224	3 1.09	PRESS	PSIA
VARY 224	4 1.09	PRESS	PSIA
VARY 224	5 .008	TOTAL NON-CONDENSABLE FLOW	LB/HR
VARY 224	6 2.475	CONDENSABLE VAPOR FLD	LB/HR
VARY 224	8 .24	NON-CONDENSABLE SPECIFIC HEAT	B/LB-F
VARY 224	9 28.	NON-CONDENSABLE MOLECULAR WEIGHT	LB/MOL
VARY 224	11 .008	TOTAL N2 FLOW THROUGH PYROLYSIS UNIT	LB/HR
VARY 224	52	EFFECTIVE SUMMED CONDUCTANCE	B/HR-F
VARY 224	54 75.	AMBIENT GAS GEMP.	F
VARY 224	55 0.	AMBIENT GAS/INSULATION UA	B/HR-F
VARY 224	57 75.	AMBIENT WALL TEMP	F
VARY 224	58 0.	RADIATION FACTOR	FT**2
VARY 224	59 0.	HT. LOSS BY RADIATION	B/HR
VARY 224	60 1250.	STRUCTURE TEMP.	F
VARY 224	61 100.0	PYRO TO STRUCTURE CONDUCTANCE KA/X	B/HR-F
VARY 224	64 0.	INSULATION SURFACE/PYRO HT. TRANSFER COND.	B/HR-F
VARY 224	65 .9	REMOVAL EFFICIENCY FOR 1ST CONSTITUENT	
VARY 224	66 10.	HEAT OF COMBUSTION FOR 1ST CONSTITUENT	B/LB
VARY 224	67 0.	RATIO OF CO2 PRODUCED	BL/LB
VARY 224	68 1.5	RATIO OF H2O PRODUCED TO THE 1ST CONSTIT.	BL/LB
VARY 224	70 0.	HTR. POWER REQRO.	WATTS
VARY 224	71 1.	UNIT LUMPED THERMAL CAPACITANCE	B/F

VARY 224	72	1200.	COMPONENT INITIAL TEMP.	F
VARY 224	73	.99996	REMOVAL EFFICIENCY OF HYDROGEN	
VARY 224	74	40000.	HEAT OF COMBUSTION OF PYROLYSIS REACTION	BTU/LB
VARY 224	75	0.	RATIO OF CO2 PRODUCED	LB/LB
VARY 224	76	9.	RATIO OF H2O PRODUCED	LB/LB
ID** 226			REGENERATIVE HX COMP OF THE PYROLYSIS UNIT COMPUTES OUTLET FLOW	
NSTR 226	-0200010000		COUNTERFLOW, NO SIZING, INPUT UA, GAS-GAS HX	
KBAS 226	4	224 3 14 5 6 7 8 9 -232 3 14 5 6 7 8 9		233
VARY 226	66	21.6729	OVERALL UA	BTU/HR-F
ID** 232	-0		VALVE TO DIRECT AIR FLOW TO STERILIZER DURING DEFECATION MOVE.	
KBAS 232	10	221 3 14 5 6 7 8 9 3 14 5 6 7 8 9		223
NSTR 232	-00		UNIVERSAL SPLIT RATIO	
ID** 233	-0		SIMULATION OF HT LOSS THROUGH PIPING SIMULATED BY PIPE	
KBAS 233	16	226 3 14 5 6 7 8 9		237
NSTR 233	-01		USE STD-FWD DIFF SOLUTION	
VARY 233	54	75.	AMBIENT GAS TEMPERATURE	F
VARY 233	55	0.		
VARY 233	57	75.	AMBIENT WALL TEMP	F
VARY 233	58	0.		
VARY 233	60	95.	STRUCTURE TEMP	F
VARY 233	61	2.85	CONDUCTANCE KA/X OUTR SURFAC/STRUCTURE	B/HR-F
VARY 233	64	5.6	CONDUCTANCE (UA) OF INSULATION	B/HR-F
VARY 233	66	5.	LENGTH OF DUCT	FT
VARY 233	67	1.	INSIDE DIAM OF DUCT	IN
VARY 233	68	2.5	LOCATION OF BRACKET ALONG LENGTH OF DUCT	FT
VARY 233	69	1.3	THERML CONDCTNCE GAS/DUCT WALL	B/HR-F
VARY 233	71	.1	DUCT WALL THICKNESS	IN
VARY 233	72	488.	DUCT DENSITY	LB/FT**3
VARY 233	73	.11	DUCT SPECIFIC HEAT	
VARY 233	74	0.	MULTIPLYING FACTOR LUMPING THERML CAPACITANCE	
ID** 237	1		CONDENSER SIMULATED BY G189 COMPONENT SUBROUTINE CONDENSER RECEIVES	
ID** 237	2		GASEOUS FLOW AND CONDENSES OUT THE VAPOR AND VENTS THE NON-CONDENS-	
ID** 237	3		ABLE GASSES THE LIQUID FLOW OUT IS DETERMINED BY GPOLY IN TRANSIENT	
ID** 237	4		IN STEADY STATE LIQUID FLOW OUT IS SET EQUAL TO CONDENSATION RATE	
KBAS 237	60	233 3 14 5 6 7 8 9 3 14 5 6 7 8 9		213
NSTR 237	1		11 PRESSURE USED AS CRITERIA USE SS DURING TRNS	
KARY 237	16	238	COMPONENT NUMBER DEFINING THE COOLNT SUPPLY	
KARY 237	17	5	CONDENSER DIFFUSION COEFFICIENT VS SAT PRESSURES	
VARY 237	3	.63	PRESSURE	PSI
VARY 237	4	.63	PRESSURE	PSI
VARY 237	22	.63	PRESSURE	PSI
VARY 237	23	.63	PRESSURE	PSI
VARY 237	65	0.		
VARY 237	66		EFFECTIVE ORIFICE AREA OF VENT VALVE	FT**2
VARY 237	67	0.01	DOWNSTREAM PRESS TO WHICH GASSES VENT	PSI
VARY 237	68		DEW POINT TEMPERATURE OF THE GASSES	DEG-F
VARY 237	69		CONDENSER CONDENSATION RATE	LB/HR
VARY 237	70		HEAT TRANSFERRED TO WALL FROM GAS	B/HR
VARY 237	71	.29173	RADIUS OF EVAPORATOR	FT
VARY 237	72	1.592	HEIGHT OF EVAPORATOR	FT
VARY 237	73		HEIGHT OF LIQUID LEVEL IN THE CONDENSER	FT
VARY 237	74	8.8	MASS OF LIQUID IN CONDENSER	LBS
VARY 237	75	117.	EFFECTV CONDCTNCE (UA) CLNT FLUID/WALL	B/HR-F
VARY 237	76	0.02	CONVERGENCE TOLERANCE (TEMPERATURE)	
VARY 237	77	2.0	CONDENSER LUMPED THERMAL CAPACITANCE	B/F
VARY 237	78		CONDENSATE LUMPED THERMAL CAPACITANCE	B/F
VARY 237	79	.6	FRACTION OF VENT GAS THAT IS VAPOR	(N.D.)
VARY 237	80	.0746	CONDUCTANCE KA/L BETWEEN TWO WALL NODES	B/HR-F
VARY 237	81	.1	CONDUCTANCE CONDENSER/SURROUNDINGS	B/HR-F

VARY 237 82 70.	EFFECTIVE TEMP. OF SURROUNDINGS	F
VARY 237 83 1.	CONVECTIVE HEAT TRNSFR COEFF GAS/WALL	B/HR-F-F
VARY 237 84	CONDUCTANCE,KA/L, TERM WALL/FLUID	B/HR-F
VARY 237 85 480.	HEAT TRANSFER COEFFICIENT CONDENSATION/WALL	
VARY 237 86 0.	CLNT TEMP AT NODE ,TC1	F
VARY 237 87 0.	CLNT TEMP AT NODE 2 , TC2	F
VARY 237 88 0.	CNDNSR WALL NODE 1 TEMP, TW1	F
VARY 237 89	CONDENSER WALL NODE TEMP 2 TW2	TW2
VARY 237 90		
VARY 237 91	TOTAL WEIGHT OF GAS IN THE CONDENSER	LBS
VARY 237 92	TEMP OF GAS IN THE CNDWSR	F
VARY 237 93	PRESSURE	PSI
VARY 237 94	PRESSURE	PSI
VARY 237 95	NON-CONDENSABLE TOTAL	LBS
VARY 237 96	VAPOR IN CNDNSR	LBS
VARY 237 97 0.	VALUE IS ZERO SUPER SATURATION NOT ACCEPTD	
VARY 237 98	SP HEAT OF N-C GASSES IN CNDNSR	B/LB-F
VARY 237 99	MOLCLR WEIGHT OF N-C GASSES IN CNDNSR	LB/LB-MOLE
VARY 237 100	WEIGHT OF O2 IN THE CNDNSR O2	LB.
VARY 237 101	WEIGHT OF N2 IN THE CNDNSR N2	LB.
VARY 237 102	WEIGHT OF CO2 IN THE CNDNSR CO2	LB.
VARY 237 103	WEIGHT OF NH3 IN THE CNDNSR NH3	LB.
VARY 237 104	SPECIAL FLO NO 1 WEIGHT H2	LB.
VARY 237 105	SPECIAL FLO NO 2 WEIGHT	
VARY 237 106	SPECIAL FLO NO 3 WEIGHT	
VARY 237 107	SPECIAL FLO NO 4 WEIGHT	
VARY 237 108	SPECIAL FLO NO 5 WEIGHT	
VARY 237 109	SPECIAL FLO NO 6 WEIGHT	
ID** 238 1	ALTERNATE COMPONENT PROVIDING THE COOLNT FLOW TO THE CONDENSER	
KBAS 238 -0 49	-9 0 1	
VARY 238 1 266.	COOLANT FLO THROUGH CONDENSER	LB/HR
VARY 238 2 35.	COOLNT TEMP.	F
VARY 238 3 15.	PRESS.	PSI
VARY 238 4 15.	PRESS.	PSI
ID** 239 1	THE DUMMY COMPONENT PROVIDING THE HEATING FLUID FLOW TO THE EVAPORAT	
ID** 239 2	IT RECEIVES FLOW FROM COMP. 109 A VALVE	
KBAS 239 -0 49	109 0 1	
VARY 239 1 360.	HTNG FLUID FLOWRATE	LB/HR
VARY 239 2 150.	HTNG FLUID TEMP	F
VARY 239 3 15.	HTNG FLUID PRESS	PSI
VARY 239 4 15.	HTNG FLUID PRESS	PSI
TABL 1 10 2	7 LOG LIN	
TITL 1	20BOILING TEMPERATURE FOR SODIUM AS A FUNCTION OF PRESSURE	
VALU 1 100 I	2.11 21.1 106.7 175.25 636.0 2116.0	
VALU 1 110 0	800.0 1000. 1200. 1300.0 1400.0 1600.0	
VALU 1 120 I	5731.2	
VALU 1 130 0	1800.0	
TABL 2 1 3	9 5 LIN LIN LIN	
TITL 2 2	HEAT FLUX COEFFICIENT Q/A FUNCTION OF DTMP AND PRESS FOR EVAP	
VALU 2 21 31	.541 1.07 2.225 9.39 14.696	
VALU 2 22 30 1.5	1.5 4.30 8.00 17. 35.0	
VALU 2 23 30 3.0	4.2 9.0 23.0 71.0 135.0	
VALU 2 24 30 4.0	7.20 14.0 38.0 120.0 220.	
VALU 2 25 30 6.0	20.0 40.0 93.0 310. 600.	
VALU 2 26 30 8.0	58.0 120. 240.0 800. 1060.	
VALU 2 27 30 10.	165.0 310.0 620. 2100.0 4400.0	
VALU 2 28 30 12.	480.0 900.0 1060.0 60000. 88000.	
VALU 2 29 30 20.	7400. 12000. 30000. 60000. 88000.	
VALU 2 30 30 38.	42000. 102600. 155800. 300000. 372400.	

TABL	4	10	2	5	LIN	LIN		
TITL	4	20CONDUCTANCE OF NICKEL 600 ALLOY USED FOR HEAT BLOCK BTU/HR-FT-F						
VALU	4	400	1	212.	572.	932.	1292.	1652.
VALU	4	410	0	8.	9.825	11.65	14.475	15.3
TABL	5	10	2	5	LIN	LIN		
TITL	5	20HEAT TRANSFER COEFFICIENT CONDENSATION OF VAPOR ON CONDENSER WALL						
VALU	5	500	1	0.000	3.00	5.0	10.0	15.0
VALU	5	510	0	0.00	15.0	50.	400.	1000.
TABL	6	10	2	3	LIN	STP		
TITL	6	20 FLUSH WATER FLOW FOR URINAL(ONE CYCLE/HOUR) CYCLE=3600.0						
VALU	6	100	21	0.0	300.0	419.0		
VALU	6	110	20	0.0	20.0	0.0		
TABL	7	1	2	3	LIN	STP		
TITL	7	2 FLUSH WATER FLOW FOR BLENDER(ONE CYCLE/6 HOURS) CYCLE=21600.0						
VALU	7	10	21	0.0	1000.0	1058.		
VALU	7	11	20	0.0	75.0	0.0		
TABL	8	10	2	3	LIN	STP		
TITL	8	20FLUSH WATER FLOW FOR SHREDDER (ONE CYCLE/6 HRS) CYCLE=21600.0						
VALU	8	10021	0.0		7200.0	7258.0		
VALU	8	11020	0.0		105.0	0.0		
TABL	9	10	2	3	LIN	STP		
TITL	9	20 AIR FLOW FOR URINAL(ONE CYCLE/HOUR) CYCLE=3600.0						
VALU	9	100	21	0.0	300.0	558.0		
VALU	9	110	20	0.0	56.0	0.0		
TABL	10	10	2	3	LIN	STP		
TITL	10	20 URINE FLOW INPUT (ONE CYCLE/HOUR) CYCLE=3600.0						
VALU	10	100	21	0.0	300.0	358.0		
VALU	10	110	20	0.0	35.0	0.0		
TABL	11	10	2	4	LIN	LIN		
TITL	11	20COOLANT FLOW--WATER COOLER --IF NEG. BYPASS TO HEATER						
VALU	11	10021	0.0		100.0	200.0	400.0	
VALU	11	11020	0.0		0.0	0.0	0.0	
TABL	12	10	2	4	LIN	LIN		
TITL	12	20LOW TEMP HEAT TO HEATER-- IF NEGATIVE BYPASS TO COOLER						
VALU	12	10021	0.0		100.0	200.0	400.0	
VALU	12	11020	0.0		0.0	0.0	0.0	
TABL	13	10	2	3	LIN	STP		
TITL	13	20TRASH INPUT --COMPONENT 208 (LB/HR )						
VALU	13	10021	0.0		7200.0	7258.0		
VALU	13	11020	0.0		18.0	0.0		
TABL	14	1	2	3	LIN	STP		
TITL	14	2 SOLIDS INPUT --COMPONENT 201 (LB/HR)						
VALU	14	10	21	0.0	1000.0	1058.0		
VALU	14	11	20	0.0	18.0	0.0		
TABL	15	1	2	3	LIN	STP		
TITL	15	2 AIR FLOW FOR BLENDER --COMPONENT 203(LB/HR)						
VALU	15	10	21	0.0	1000.0	1300.0		
VALU	15	11	20	0.0	132.00	0.0		
TABL	16	10	2	2	LIN	LIN		
TITL	16	20WATER COOLER (COMPONENT 93) DEMAND RATE (LB/HR)						
VALU	16	10021	0.0		200.0			
VALU	16	11020	1.123		1.123			
TABL	17	100	2	2	LIN	LIN		
TITL	17	20WATER HEATER DEMAND RATE (COMPONENT 94) LB/HR						
VALU	17	10021	0.0		200.0			
VALU	17	11020	1.123		1.123			
PLOT00								RITE SYSTEM
PLOT21	210	25	0.0	4.0				EVAPORATION RATE(LB/HR)
PLOT22			100.0	130.0				TEMPERATURE (DEG-F)

PLOT2	210	2			VAPOR TEMP. (DEG-F)
PLOT2	210	51			WALL TEMP. (DEG-F)
PLOT22			1100.0	1300.0	HEAT BLOCK TEMPERATURES (DEG-F)
PLOT2	180	69			ISOTOPE TEMPERATURE
PLOT2	180	71			NODE WITH AIR STER.
PLOT25					AIR STERILIZER TEMPERATURES
PLOT2	25	80			AIR OUTLET TEMP.
PLOT2	25	116			AIR INLET TEMP.
PLOT2	25	118			HEAT BLOCK TEMP.
PLOT2	25	71			OUTER CYLINDER TEMP.
PLOT2	25	74			STERILE CHAMBER TEMP
PLOT21	213	2	700.0	1500.0	INCINERATOR TEMPERATURE
ENDC					
ENDR					

## Appendix C

### OUTPUT DATA

Appendix C describes the output formats chosen to define the output of the RITE simulation. The first output format consists of the card images of all cards being input for the current simulation case. These data include the control cards (TAPE and ENDC), the case data cards (BASIC and CASE card, namelist \$CASE1 data, and namelist \$PROP1 data), and modifications to the component K and V array data. Since these cards are listed in Appendix B they are not repeated here.

The second section of output is shown in Figure C-1 reflects the Input Editor's interpretation of the data entered on the BASIC card. The output includes the case title, case number, and the number of components in the simulation.

The third section of output shown in Figure C-2 contains the Fortran system output format for all the variables defined under the namelist \$CASE1. If the variables were not input they contain a default value of zero. The fourth and fifth sections shown in Figure C-3 contain a similar output format of the variables contained in the fluid property namelist \$PROP1.

The sixth page of output shows the beginning of the printout of the edited data after the Input editor has sorted and merged the input data from tape and cards and arranged them in groups of cards by numerical order for the components and tables and by card type within the component or table groups. The heading "BASIC CASE DATA" is printed at the beginning and it is followed by the case title as input on the CASE card.

BASIC CASE NO. 1

THE EC/LS MODEL HAS ~~239~~ COMPONENTS. THE LAST OF WHICH TO BE SOLVED IS NO. 139

THE MODEL IS TRANSIENT, WITHOUT PRESSURE DROP DATA ALLOCATION.

Figure C-1. INPUT EDITOR INTERPRETATION OF BASIC CARD

# SCASE1

DTIME = 0.0E+02,

KCHOUT = 0,

KPRNT = 6,

KPRUN = 0,

KPTINV = 1, 0, 0, 0,

KPUNCH = 0,

KRUN = 1,

MAXCI = 0,

MAXLP = 0,

MAXSLP = 6,

MAXSSI = 3,

MINSSI = 2,

NPASPD = 0,

PGMIN = 0.0,

PLMIN = 0.0,

START = 0.0,

TIME MX = 0.3E+03,

TMAX = 0.2E+04,

TMIN = 1.45E+03,

WTHAX = 0.1E+05,

IUT15 = 0,

IPONLY = 0,

YPLOTS = 0.0, 0.0, 0.0, 0.0, 0.0,

YPLOTE = 0.0, 0.0, 0.0, 0.0, 0.0,

IFPLOT = 0,

ICLK = 0,

ICPLMT = 0,

IPPLMT = 0,

IPGLMT = 0,

SEND

Figure C-2. NAMELIST  
\$CASE1 DATA



**SPR021**

RCNR

-c-b

The next section of output, Figure C-4, shows the initial solution path and the component numbers not in the solution path but used for data storage.

The G189A program then automatically prints out an expanded and labelled form of the K and V array data for all components in the solution path and others which are not in the solution path but are given subroutine numbers to produce a printout. This printout, a partial output is shown in Figure C-5, is given at the beginning of the run, at the end of steady state calculations and at the end of transient calculations.

The RITE simulation is set up to print out the V-array data for each component in the solution path at every system pass. Samples of this output are shown in Figures C-6 to C-8. The definition of the V-array variable is given in the G-189 manual and in the documented input cards described in Appendix B.

An option is available in the G189A program to produce SD-4060 graphical output. The output plots are of the form of a specified V array location (component number, reference location) versus mission times. Typical output plots are shown in Section 5.

The G189A program contains many alternate output options which are available to the programmer. These options are described in detail in the G189 Manual.

~~100, 102, 103, 104, 105, 106, 107, 108, 109, 112, 113, 114, 115, 116, 117, 118, 120, 121, 122, 124,~~  
~~41, 43, 44, 45, 46, 47, 48, 49, 50, 51, 52, 53, 54, 55, 56, 57, 58, 59, 60, 61, 62, 63,~~  
~~202, 205, 208, 207, 206, 208, 209, 210, 221, 232, 223, 224, 226, 233, 237, 213, 214, 61, 62, 63,~~  
~~64, 65, 66, 9, 2, 3, 80, 81, 82, 83, 84, 85, 86, 88, 90, 89, 91, 99, 92,~~  
~~93, 95, 94, 96, 5, 6, 10, 6, 100, 101, 125, 123, 126, 127, 119, 128, 129, 130, 134, 132,~~  
~~133, 110, 111, 134, 136, 137, 139,~~

~~ALTERNATE COMPONENTS - (COMPONENTS WHICH MAY BE INCLUDED IN SOLUTION PATH AS A RESULT OF AN INSTRUCTION CHANGE~~  
~~OR COMPONENTS USED FOR DATA STORAGE)~~  
~~- (FLAG BY INPUTTING A SUBROUTINE NO. IN KK(1,1), THIS ALSO FORCES PRINTOUT OF DATA) -~~

~~1, 51, 101, 135, 210, 230, 238, 239,~~

~~COMPONENT ROWS NOT SPECIFIED IN THE SOLUTION PATH OR AS AN ALTERNATE PATH OR AS DATA STORAGE~~

~~4, 7, 11, 12, 13, 14, 15, 16, 17, 18, 19, 20, 26, 29, 30, 31, 32, 33, 34, 35,~~  
~~36, 37, 38, 39, 40, 42, 52, 53, 54, 55, 56, 57, 58, 59, 60, 67, 68, 69, 70, 71,~~  
~~72, 73, 74, 75, 76, 77, 78, 79, 97, 98, 138, 140, 141, 142, 143, 144, 145, 146, 147, 148,~~  
~~149, 150, 151, 152, 153, 154, 155, 156, 157, 158, 159, 160, 161, 162, 163, 164, 165, 166, 167, 168,~~  
~~169, 170, 171, 172, 173, 174, 175, 176, 177, 178, 179, 182, 183, 184, 185, 186, 187, 188, 189, 190,~~  
~~191, 192, 193, 194, 195, 196, 197, 198, 199, 200, 211, 212, 216, 217, 222, 225, 227, 228, 229, 230,~~  
~~231, 234, 235, 236,~~

Figure C-4. INITIAL SOLUTION PATH

[illegible]

51-COMP SOURCE TEMP	0.	0.	0.	0.	0.	0.	0.	0.	0.
52-EFF SUMMED COND	0.	0.	0.	0.	0.	0.	0.	0.	0.
53-COMP TOTAL LOSS	0.	0.	0.	0.	0.	0.	0.	0.	0.
54-AM IEIT GAS TEMP	0.	0.	0.	0.	0.	0.	0.	0.	0.
55-AM IEIT HA	0.	0.	0.	0.	0.	0.	0.	0.	0.
56-AM CONV Q LOSS	0.	0.	0.	0.	0.	0.	0.	0.	0.
57-AM WALL TEMP	0.	0.	0.	0.	0.	0.	0.	0.	0.
58-AM SCRIPT(F)=A	0.	0.	0.	0.	0.	0.	0.	0.	0.
59-AM RAD Q LOSS	0.	0.	0.	0.	0.	0.	0.	0.	0.
60-STRUCTURE TEMP	0.	0.	0.	0.	0.	0.	0.	0.	0.
61-STRUCTURE KAX	0.	0.	0.	0.	0.	0.	0.	0.	0.
62-STRUCTURE LOSS	0.	0.	0.	0.	0.	0.	0.	0.	0.
63-TEMPERATURE TEMP	0.	0.	0.	0.	0.	0.	0.	0.	0.
64-TEMPERATURE KAX	0.	0.	0.	0.	0.	0.	0.	0.	0.

SUBROUTINE DEPENDENT V ARRAY DATA - - -

VR( 1, 65)= 0.	VR( 1, 66)= 0.	VR( 2, 65)= 0.	VR( 5, 65)= 0.	VR( 8, 65)= 0.
VR( 3, 66)= 0.	VR( 0, 67)= 0.	VR( 8, 68)= 0.	VR( 9, 65)= .92400	VR( 9, 65)= 0.

COMPONENT NO. = SUBR. TYPE =	1 ALTCOM	2 SPLIT	3 L10MIX	5 SPLIT	6 L10MIX	8 FLOMET	9 SPLIT	10 L10MIX
1-SUBR NO./EXV/EXX	42000000	10000000	7000000	10000000	7000000	29004000	10000000	7000000
2-PRI-SUB/FLO CODE	0	900	200	9400	500	1000	100	600
3-PRI-SPEL TYP 1-3	20000	20000	20000	20000	20000	20000	20000	20000
4-PRI-SPEL TYP 4-7	-0	-0	-0	-0	-0	-0	-0	-0
5-SEC-SUB/FLO CODE	-0	0	12000	0	11900	-0	0	-23300
6-SEC-SPEL TYP 1-3	-0	20000	20000	20000	20000	-0	20000	20000
7-SEC-SPEL TYP 4-7	-0	-0	-0	-0	-0	-0	-0	-0
8-NEXT COMP/CAVIN	-0	300000	8000000	600000	1000000	18000000	200000	800000
9-COMP NSTR 1-9	-0	0	100000000	0	110000000	0	0	300000000
10-COMP NSTR 10-18	-0	-0	-0	-0	-0	-0	-0	-0
11-NOFL/HLFL/TPASS	0	0	0	0	0	0	0	0
12-PRI-VISC/DENSITY	0	0	0	0	0	0	0	0
13-PRI-CP/DP/OP/UP	0	0	0	0	0	0	0	0
14-SEC-VISC/DENSITY	0	0	0	0	0	0	0	0
15-SEC-CP/DP/OP/UP	0	0	0	0	0	0	0	0

Figure C-5 (continued)



COMP NO 211 EVAP SUB NO 30 PRI SJR 209 SEC SCH 204 COMP PASS NO 1 TIME \* 60.0 SEC

FAILURE FLAG -- COMP = 0 LOOP = 0

A(1) = 0.	0.	0.	0.	0.	0.	0.	0.	0.	0.	0.	0.	0.	0.
0.	0.	0.	0.	0.	0.	0.	0.	0.	0.	0.	0.	0.	0.
B(1) = 0.	0.001	0.001	0.001	0.001	0.001	0.001	0.001	0.001	0.001	0.001	0.001	0.001	0.001
0.	0.	0.	0.	0.	0.	0.	0.	0.	0.	0.	0.	0.	0.
R( 1) = 0.	105.40	1.1174	1.1174	0.	0.	0.	0.	0.	0.	0.	0.	0.	0.
R( 11) = 0.	0.	0.	0.	0.	0.	0.	0.	0.	0.	0.	0.	0.	0.
R( 21) = 105.40	1.1174	1.1174	1.05994E-02	2.3514	0.	0.	0.	0.	0.	0.	0.	0.	0.
R( 31) = 5.97916E-03	4.62026E-03	0.	0.	0.	0.	0.	0.	0.	0.	0.	0.	0.	0.
R( 41) = 0.	0.	0.	0.	0.	0.	0.	0.	0.	0.	0.	0.	0.	0.
R( 51) = 110.63	2.5924	7.7534	75.000	1.0000	3.0757	75.000	20000	.64995	85.000	0.	0.	0.	0.
R( 61) = 2.3000	75.627	7.7534	.10000	22.500	0.	1.4180	1.7300	.17659	.15000	0.	0.	0.	0.
R( 71) = .34200	0.	3.4166	243.6	315.94	.20000	2.3514	4.71489E-02	8.12653E-03	2.44289E-03	0.	0.	0.	0.
R( 81) = 1.1700	2.2800	1.1700	.10000	.51000	1.1774	1.51001E-02	8.55335E-02	.11774	0.	0.	0.	0.	0.
R( 91) = 0.	4.62026E-03	5.12966E-03	1.7000	7473.8	180.87	122.80	.10640	1.42000E-04	.63000	0.	0.	0.	0.
R(101) = 26.154	105.40	1.1174	.40000	0.	0.	0.	0.	0.	0.	0.	0.	0.	0.
R(111) = 0.	0.	0.	0.	.17322	.95117	25.000	0.	0.	1.04011E-02	0.	0.	0.	0.
R(121) = 105.40	1.1174	.40000	9.20324E-05	1.03001E-02	0.	.22000	35.000	0.	0.	0.	0.	0.	0.
R(131) = 2.62117E-03	2.02545E-05	0.	0.	0.	0.	0.	0.	0.	105.40	0.	0.	0.	0.
R(141) = 1.1174	0.	0.	0.	0.	0.	0.	0.	0.	0.	0.	0.	0.	0.
R(151) = 0.	0.	0.	0.	0.	0.	0.	0.	0.	0.	0.	0.	0.	0.

Figure C-7. V-ARRAY OUTPUT FOR THE EVAPORATOR

COMP NO 237 CHANGE= SUBR NO 140 PRI SRT. 233 SEC SRT. 0 COMP PASS NO 2 TIME = 60.0 SEC  
FAILURE FLAGS -- COMP= 0 LOOP= 0

A(1)= 2.3591 136.23 1.0000 1.0000 5.1246E-03 2.3539 0. 24126 29.109 1.28690E-03  
3.79948E-03 9.41246E-05 0. 1.11793E-03 5.95710E-06 7.77115E-06 0. 0. 0. 0.  
GPA= .43956 WTHA= 18.015 RHQA= 5.10083E-03 VISCA= 0. XKA= 0.  
B(1)= 0. 0. 0. 0. 0. 0. 0. 0. 0. 0.  
0. 0. 0. 0. 0. 0. 0. 0. 0. 0.  
GPR= 0. WTHB= 0. RHQR= 0. VISCB= 0. XKB= 0.  
R( 1)= 0. 56.634 5.63000 5.63000 0. 0. 0. 0. 0. 0.  
R( 11)= 0. 0. 0. 0. 0. 0. 0. 0. 0. 0.  
R( 21)= 92.923 5.63000 5.63000 5.19465E-03 1.83729E-02 0. 24126 29.109 1.28690E-03 3.79948E-03  
R( 31)= 9.45246E-05 0. 1.11793E-03 5.95710E-06 7.77115E-06 0. 0. 0. 0. 0.  
R( 41)= 0. 0. 0. 0. 0. 0. 0. 0. 0. 0.  
R( 51)= 0. 0. 0. 0. 0. 0. 0. 0. 0. 0.  
R( 61)= 0. 0. 0. 0. 2479.3 3.95954E-06 1.00000E-02 81.650 2.3227 0.  
R( 71)= .29173 1.5920 5.5745 3.9000 117.10 2.00000E-02 2.0000 -0. 60000 7.46000E-02  
R( 81)= .10000 70.000 1.0000 -0. 430.00 48.202 48.273 78.069 48.502 -.67146  
R( 91)= 5.95916E-04 92.923 5.63000 5.63000 1.27973E-04 4.64943E-04 0. 24126 29.109 3.17035E-05  
R(101)= 9.36020E-05 2.32866E-16 0. 2.75900E-10 1.46775E-07 1.91446E-07 0. 0. 0. 0.  
R(111)= 0. 0. 0. 0. 0. 0. 0. 0. 0. 0.

C-10

Figure C-8. V-ARRAY OUTPUT FOR THE CONDENSER



## Appendix D

### GPOLY SUBROUTINES

The G189A program contains two subroutines GPOLY1 and GPOLY2, to be used to add user coded Fortran logic which is peculiar to the RITE simulation. The GPOLY1 subroutine is called before component solution and the GPOLY2 subroutine is called after component solution. The subroutines are used to modify data computed by the component subroutines, to insert new logic, and to exercise control over the order and method of solution used for the system simulation.

The logic contained in the RITE simulation GPOLY subroutines is described in detail in Section 3. A complete listing of the RITE GPOLY1 and GPOLY2 subroutines follows.

```

SUBROUTINE GPOLY1
COMMON /COMP/ DS(15),N,NA1,NB1,NO,NCAB,NCFL,NEXT,NEXV,NK
1 NKEX,NKS,NKT,NLFL,NPT,NPASS,NPF,NPF7(6),NQ,NS,NSF,NSPT(6)
2 NSTR(18),NSJRR,JV,NVT,Y(12)
COMMON /RARRAY/ IMAXR,R(1)
COMMON /ECLST1/ KCHOUT,KPRNT,KPTINV(4),KWIT,KWIT1,KWIT2,
1 KWIT3,KWIT4,NUFF
COMMON /KANDV/ K
COMMON /MISC/ DTIME,GRAB,KFLSYS,KOUTPT,KPDROP,KSYPAS,KTE
1 LPSUM(5),MAXCI,MAXLP,MAXSLP,MAXSSI,NCOMPS,NEWDT,NLAST,NF
2 MINSSI,PGMIN,PLMIN,START,STEADY,TIME,TIMEX,TMAX,TMIN,WT
COMMON /CASE/ NCASE,NRSCS
COMMON /PROPTY/ CP,CP(99),CPCONL,CPCONV,CPCO2,CPDIL,CPO
1 GAGAS,RHOG,RHO(99),V1800,V180(99),VISGAS,WTMO,WTM(99),W
2 WTDIL,WTMTG,XK0,XK(99),XKGAS,XKLIQ,VISLIQ
COMMON /SOURCE/ A(19),B(19),CPA,CPR,IA1,IB1,NA,NB,NPFS,N
1 NSF,NSFST(6),RHOA,RHOB,VISGA,VISCO,WTMA,WTMB,XKA,XKB
COMMON /POW/ POWER
COMMON /VLOC/ IP,IS,ICTIG,IV,IVT,IEV,INEXK
LOGICAL POWER
DIMENSION V(1),K(1)
EQUIVALENCE (V,K)
LOGICAL STEADY

```

C  
C  
C  
C  
C  
C

THIS LOGIC ALLOWS US TO GET THE CORRECT INLET FLOW  
TEMPERATURE TO COMPONENT 10

```

IF(I.NE.10) GO TO 10
R(1)=VV(9,20)
R(2)=VV(237,87)
10 CONTINUE

```

C  
C  
C  
C  
C  
C  
C  
C  
C  
C

\*\*\*\*\* END OF COMPONENT 10 GPOLY LOG

GPOLY1 LOGIC FOR LOW TEMP. HEATING LOOP

INTERFACE OF THERMAL C100 TO C139

```

IF(I.NE.100) GO TO 100
R(82)=A(2)
R(75)=CPA*A(1)
DO 1100 J=1,19
1100 R(J)=A(J)
100 CONTINUE

```

C  
C  
C  
C  
C  
C  
C  
C  
C

EVAPORATOR COOLING LIQUID CONTROL VALVE (COMPONENT 109)

```

IF(I.NE.109) GO TO 109
IF(STEADY.OR.(NSTR(16).GT.0)) GO TO 109

```

```

TEMP=VV(210,2)
IF(TEMPS.GT.105)R(45)=1.0
IF(R(65).GT.0.0.AND.TEMPS.LT.104.7)R(45)=0.0
IF(TEMPS.LT.104.7)R(65)=0.0
09 CONTINUE

```

# GPOLY1 LOGIC FOR AIR LOOP

## AIR HEATER CONTROL LOGIC COMPONENT 122

```

IF(W.NE.122)GO TO 122
AIRTEP=VV(204,2)
R(65)=0.0
IF(AIRTEP.GT.110.0)R(65)=1.0
122 CONTINUE

```

## WARM FLUSH WATER TANK CONTROL LOGIC

```

IF(W.NE.124)GO TO 124
IF(STEADY) GO TO 124
TTANK=VV(41,70)
IF(TTANK.LT.94.0)R(65)=0.0
IF(TTANK.GT.96.0)R(65)=1.0
IF(R(65).GT.0.0.AND.TTANK.LT.94.0)R(65)=0.0
124 CONTINUE

```

## LTHL HX CONTROLLER (COMPONENT 136)

```

IF(W.NE.136)GO TO 136
RITET=VV(100,2)
R(65)=0.0
IF(RITET.GT.190.0)R(45)=1.0
136 CONTINUE

```

# GPOLY1 LOGIC FOR FLUSH WATER LOOP

COMPONENT 41 WARM FLUSH WATER TANK

IF(1.NE.41) GO TO 41

4J=VALUE(6,CYCLE(3600,0,0,0),0,0)

4S=VALUE(8,CYCLE(21000,0,0,0),0,0)

4F=VALUE(7,CYCLE(21000,0,0,0),0,0)

R(17)=WU+WF+WS

R(1)=R(17)

4S1=VV(51,17)

A(1)=W51

A(2)=70.0

A(17)=A(1)

SPLIT45=0.0

IF(R(17).NE.0.0)SPLIT45=R(17)/(R(17)+4S1)

CALL SV(SPLIT45,45,65)

SPLIT46=0.0

IF(R(17).NE.0.0)SPLIT46=(WS+WF)/R(17)

CALL SV(SPLIT46,46,65)

SPLIT48=0.0

IF(4S+WF.GT.0.0)SPLIT48=WF/(WS+WF)

CALL SV(SPLIT48,48,65)

41 CONTINUE

## GPOLY1 LOGIC FOR AIR LOOP

IF(1.NE.22) GO TO 22

THIS MOD COMPENSATES FOR FLO CODE 4

THIS MOD COMPENSATES FOR FLO CODE 4

R(3)=15.

R(4)=15.

R(8)=1.0

R(9)=1.0E+10

22 CONTINUE

IF(1.NE.25) GO TO 25

THE FOLLOWING CARDS INTERFACE AIR STERILIZER  
TO THE SYSTEM

FLOW CONVECTION TERMS

PC 1025-J=1,19

1025

R(J)=A(J)

R(116)=A(2)

R(106)=A(1)\*CPA

R(104)=A(1)\*CPA

R(102)=A(1)\*CPA

R(99)=A(1)\*CPA

R(96)=A(1)\*CPA

R(93)=A(1)\*CPA

HEAT BLOCK TEMPERATURE AT NODE 6 HING TO AIR STERILIZER

R(118)=VV(184,76)

25 CONTINUE

GPOLV1 LOGIC FOR SOLID WASTE LOOP

IF (N.NE. 210) GO TO 210

R(3)=14.7

R(4)=14.7

NSTR(4)=0

IF (VALUE(15,TIME,0.0),GT.0.0) NSTR(4)=1

CALL SV(0.0,232,65)

IF (NSTR(4).GT.0) CALL SV(1.0,232,65)

R(100)=VV(237,22)

210 CONTINUE

GPOLV1 LOGIC FOR INCINERATOR

IF (N.NE. 213) GO TO 213

R(51)=R(2)

IF (TIME.LT.10.OR.TIME.GT.60.0) GO TO 213

R(76)=1.0

R(78)=0.2

R(79)=1.31

213 CONTINUE

IF (N.NE. 221) GO TO 221

A(1)=R(13)\*.7651

A(5)=A(1)

A(10)=A(1)

A(2)=70.

W02TNK=VV(220,69)

A(8)=.22

A(9)=32.

W02TNK\*W02TNK=A(1)\*DTIME

A(3)=W02TNK\*1545./32\*530./144.

A(4)=A(3)

CALL SV(W02TNK,220,69)

CALL SV(W02TNK,220,73)

CALL SV(W02TNK,220,78)

CALL SV(W02TNK,220,89)

CALL SV(A(3),220,3)

CALL SV(A(3),220,4)

CALL SV(A(3),220,72)

221 CONTINUE

BYPASS VAPOR LOOP IF FLOW=0.0

IF (N.NE. 232) GO TO 232

IF (VALUE(15,TIME,0.0),GT.0.0) NEXT=213

232 CONTINUE

IF(4 .NE. 224) GO TO 224  
 SET HEAT BLOCK (STRUCTURE) TEMP  
 R(60)=1200.0  
 PYROLYZE NH3 = 1/2 N2 + 3/2 H2  
 A(14)=A(13)\*1.5\*(WTM(14)/WTMT0)  
 A(11)=A(13)\*0.5\*(WTM(11)/WTMT0)  
 A(13)=0.

SIMULATE ROSCH REACTION ~~2H2+CO2=C+2H2O~~

CRBJER=A(12)\*12./44.\*.984191  
 A(14)=A(14)-.98491\*A(12)\*4./44.  
 A(12)=.015809\*A(12)

224 CONTINUE

IF(5 .NE. 8) GO TO 8  
 THIS IOD BYPASSES HEAT BLOCK DURING STEADY STATE  
 IF(STEADY)-NEXT=181  
 8 CONTINUE

GPOLY1 LOGIC FOR COOLING LOOP

COMPONENT 2 FLOW SPLIT TO WATER HEATER  
 IF(4.NE.2)GO TO 2  
 IF(ICOOL.GE.0.1)GO TO 2000  
 R(65)=0.0  
 IF(ICOOL.GT.0.1) GO TO 2000  
 R(65)=1.0  
 CALL SV(1.0,5,65)  
 GO TO 2

2000 CALL SV(0.0,5,65)

2 CONTINUE

GPOLY1 LOGIC FOR RITE LOOP

IF(7 .NE. 180) GO TO 180  
 SET HEAT BLOCK INTERFACES  
 HEAT LOSS TO HEAT PIPE  
 R(88)=VV(181,75)  
 HEAT LOSS TO AIR STERILIZER  
 R(89)=VV(25,83)\*(VV(25,116)-VV(25,80))-VV(25,108)\*(VV(25,61)-VV(25,117))-(VV(25,107)\*(VV(25,70)-VV(25,117)))  
 HEAT LOSS TO PYROLYSIS UNIT NO1  
 R(90)=0.0  
 HEAT LOSS TO PYROLYSIS UNIT NO2  
 R(91)=R(90)  
 HEAT LOSS TO PYROLYSIS UNIT NO3  
 R(92)=R(90)

# HEAT LOSS TO INCINERATOR

R(93)=VV(213,66)

180 CONTINUE

IF(I.NE.181) GO TO 181

SET INTERFACE TEMP HEAT BLOCK TO HEAT PIPE

R(66)=VV(180,70)

181 CONTINUE

CONDENSER- OUTPUT=INPUT

IF(I.NE.237) GO TO 237

R(1)=R(69)

237 CONTINUE

GPOLY1 LOGIC FOR POTABLE WATER STORAGE LOOP

THE POTABLE WATER TANKS WILL BE FILLED IN THIS ORDER 83,84,87,86

COMPONENT SPLIT 80 == SPLIT TO TWO TANK GROUPS

IF(I.NE.80) GO TO 80

R(65)=1.0

IF(VV(83,69).GT.41.5-A(1)\*DTIME/3600.0.AND.VV(84,69).GT.41.5-A(1)\*DTIME/3600.0)R(65)=0.0

80 CONTINUE

COMPONENT SPLIT 81=SPLIT TO TANKS 86 AND 87

IF(I.NE.81) GO TO 81

R(65)=0.0

IF(VV(86,69).LT.41.5-A(1)\*DTIME/3600.0)R(65)=1.0

81 CONTINUE

COMPONENT 82 SPLIT TO TANKS 83 AND 84

IF(I.NE.82) GO TO 82

R(65)=0.0

IF(VV(83,69).GT.41.5-A(1)\*DTIME/3600.0)R(65)=1.0

82 CONTINUE

COMPONENT 83 TANK NO. 1

IF(I.NE.83) GO TO 83

R(1)=0.0

WTOT=VALUE(16,TIME,0.0)+VALUE(17,TIME,0.0)

IF(R(69).GT.WTOT\*DTIME/3600.0) R(1)=WTOT

83 CONTINUE

COMPONENT 84 TANK NO: 2

IF(I.NE.84)GO TO 84  
R(1)=0.0  
WTOT=VALUE(16,TIME,0.0)+VALUE(17,TIME,0.0)  
IF((VV(83,1).LE.0.0).AND.(WTOT.GT.0.0).AND.(R(69).GT.WTOT\*DT  
13600.0)) R(1)=WTOT  
84 CONTINUE

COMPONENT 87 TANK NO: 3

IF(I.NE.87)GO TO 87  
R(1)=0.0  
WTOT=VALUE(16,TIME,0.0)+VALUE(17,TIME,0.0)  
IF((VV(85,1).LE.0.0).AND.(WTOT.GT.0.0).AND.(R(69).GT.WTOT\*DT  
13600.0)) R(1)=WTOT  
87 CONTINUE

COMPONENT 86 TANK NO: 4

IF(I.NE.86)GO TO 86  
R(1)=0.0  
WTOT=VALUE(16,TIME,0.0)+VALUE(17,TIME,0.0)  
IF((WTOT.GT.0.0).AND.(VV(85,1).LE.0.0).AND.(VV(87,1).LE.0.0).  
1(R(69).GT.WTOT\*DTIME/3600.0)) R(1)=WTOT  
86 CONTINUE

COMPONENT 90 EMERGENCY TANK

IF(I.NE.90)GO TO 90  
R(1)=0.0  
WTOT=VALUE(16,TIME,0.0)+VALUE(17,TIME,0.0)  
-90 CONTINUE  
IF(WTOT.GT.0.0.AND.VV(85,1).LE.0.0.AND.VV(88,1).LE.0.0)R(1)=WTOT

COMPONENT 92 SPLIT TO WATER HEATER AND COOLER

IF(I.NE.92)GO TO 92  
W93=VALUE(16,TIME,0.0)  
W94=VALUE(17,TIME,0.0)  
R(69)=(W94/(W93+W94))  
92 CONTINUE  
THIS CARD MODIFIES THE KL8) DATA TO ALLOW VAPOR LOOP  
IF(I.NE.137)GO TO 137  
137 CONTINUE

RETURN  
END



```

SUBROUTINE GPOLY2
COMMON /COMP/ DS(15), NA1, NB1, NC, NCAB, NCFL, NEX1, NEXV, NK,
1 AKEX, NKS, NKT, NLFL, NP, NPASS, NPF, NPFT(6), NO, NS, NSF, NSFT(6),
2 NSTR(18), NSJDR, NV, NVT, V(12)
COMMON /RARRAY/ IAXP, R(1)
COMMON /KANDV/ K
COMMON /MISC/ DTIME, GRAV, KFLSYS, KOUTPT, KPDROP, KSYNAP, KTRANS,
1 LPSUM(5), MAXC1, MAXLH, MAXSLP, MAXSSI, NCOMPS, NEWDT, NLAST, NPASPD,
2 MIASSI, PG, IJ, PLMIN, START, STEADY, TIME, TIMEMX, THAX, TMIN, WTMAX
COMMON /CASE/ NCASE, IRSCS
COMMON /PROPTY/ CPA, CPT(99), CPGONL, CPGONV, CPCO2, CPDIL, CPOXY, CPTG,
1 GAGAS, RHOO, RHO(99), VISCO, VISC(99), VISGAS, WTM0, WTM(99), WTMCON,
2 WTDIL, WTNTO, XK0, XK(99), XKGAS, XKLIO, VISLIO
COMMON /SOURCE/ AT(9), BT(9), CPA, CPG, IA1, IB1, NA, NB, NPPS, NPFST(6),
1 NSFS, NSFST(6), RHOA, RHOB, VISCA, VISCB, WTHA, WTNB, XKA, XKB
COMMON /POW/ POWER
LOGICAL POWER
DIMENSION V(1), K(1)
EQUIVALENCE (V, K)
LOGICAL STEADY

```

\*\*\*\*\* THIS PRINTS V-ARRAY DATA FOR EACH COMPONENT

GPOLY2 LOGIC LOW TEMPERATURE HEATING LOOP

~~IF(1, NE, 100) GO TO 100~~

INTERFACE OF THERMAL C100 TO C139

R(2)=R(69)

100 CONTINUE

COOLANT PUMP NO.1 COMPONENT 104

~~IF(1, NE, 104) GO TO 104~~

R(70)=0.0151\*A(1)

104 CONTINUE

THIS CHANGE COMPENSATES FOR HEAT LOSS IN PIPE

~~IF(1, NE, 108) GO TO 108~~

~~R(21)=R(21)-4.0~~

108 CONTINUE

GPOLY 2 LOGIC FOR FLUSH WATER LOOP

FLUSH WATER PUMP=COMPONENT 44

OSCILLATION OF PUMP POWER

```

IF(J.NE.44) GO TO 44
R(70)=0.0
IF(R(1).GT.0.0)R(70)=0.70527+0.0473*R(1)
44 CONTINUE

```

SPOLY2 LOGIC FOR AIR LOOP

COMPONENT 21 URINE AIR SUPPLY

```

IF(J.NE.21) GO TO 21

```

```

STANDARD AIR INPUT=14.7 PSI 70 DEG F
WATER VAPOR 1.05 PERCENT
OXYGEN 26.0 PERCENT
NITROGEN 72.0 PERCENT
CARB-DIOXIDE 10.65 PERCENT
WAIR=VALUE(9,CYCLE(3600,0.0,0),0.0)
R(6)=0.0105*WAIR
R(10)=0.263*WAIR
R(11)=0.72 *WAIR
R(12)=0.0065*WAIR

```

```

R(2)=25.6
INPUT= HUMIDIFICATION
HUR=VALUE(10,CYCLE(3600,0.0,0),0.0)

```

```

URINE CONSTITUANTS=94.0 PERCENT WATER
URINE CONSTITUANTS=3.0
2.0 PERCENT UREA

```

```

R(3)=15.0
R(4)=15.0
R(17)=0.042*HUR
R(18)=0.028*HUR
R(19)=0.03*HUR

```

```

21 CONTINUE

```

AIR LOOP FAN = COMPONENT 24 -FAN ELECTRICAL POWER

```

IF(J.NE.24)GO TO 24

```

```

R(72)=0.0
IF(A(1).GT.0.0)R(72)=117.0+0.91*A(1)
24 CONTINUE

```

```

IF(J.NE. 25) GO TO 25
INTERFACE COMPONENT 25 TO 27

```

R(2)=R(80)

25 CONTINUE

# BPOLY2 LOGIC FOR SOLID WASTE LOOP

## COMPONENT 203 BLENDER AIR SUPPLY

IF(I.NE.203) GO TO 203

WAI RB=VALUE(15,CYCLE(21000,0,0,0),0.0)

R(6)= 0.0105\*WAI RB

R(10)= 0.263 \*WAI RB

R(11)= 0.72 \*WAI RB

R(2)=80.0

R(3)=14.7

R(4)=14.7

IF(WAI RB.GT.0.1)CALL SV(0,23,204,72)

R(12)= 0.0055\*WAI RB

203 CONTINUE

## COMPONENT 201--SOLIDS INPUT

IF(I.NE.201) GO TO 201

R(2)=98.6

R(3)=14.7

R(4)=14.7

R(16)=VALUE(14,CYCLE(21000,0,0,0),0.0)

201 CONTINUE

## BLENDER--COMPONENT 202 POWER CALCULATION

R(65) IS SET EQUAL TO BLENDER ELECTRICAL POWER

IF(I.NE.202)GO TO 202

R(65)=0.0

IF(A(1) .GT.0.0)R(65)=372.0

202 CONTINUE

## COMPONENT 208 TRASH SUPPLY

IF(I.NE.208)GO TO 208

R(16)=VALUE(13,CYCLE(21000,0,0,0),0.0)

R(2)=80.0

R(3)=14.7

R(4)=14.7

208 CONTINUE

# SHREDDER --COMPONENT 207 ELECTRICAL POWER

IF (N.NE.207) GO TO 207

R(65)=0.0

IF (A(1).GT.0.0.OR.R(1).GT.0.0) R(65)=977.0

207 CONTINUE

IF (N.NE.210) GO TO 210

210 CONTINUE

IF (N.NE.224) GO TO 224

PREDICT AMOUNT NOX GENERATE IN THE PYROLYSIS UNI

SPECIAL FLOW 2 (A(15))=NO2

SPECIAL FLOW 3 (A(16))=NO

IF (A(11).LT.0.06+A(10)) GO TO 2241

A(15)=.00294 \*A(10)\*46./32.

IF (R(11).LT.0.06\*R(10)) GO TO 2241

R(15)=.0032\*R(10)\*46./32.

R(16)=.0032\*R(10)\*30./16.

R(11)=R(11)+R(15)\*14./46.+R(16)\*14./30.

R(10)=R(10)+R(15)\*32./46.+R(16)\*16./30.

2241

CONTINUE

224

CONTINUE

RETURN

END